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LITHUANIAN RESEARCH CENTRE FOR AGRICULTURE AND FORESTRY

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**ENVIRONMENTALLY INDUCED OXIDATIVE STRESS IN GREEN PEA
AND ITS MANAGEMENT MEASURES**

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Agricultural Sciences, Agronomy (A 001)

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ŽIRNIUOSE IR JO VALDYMO PRIEMONĖS**

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ABBREVIATIONS

$^1\text{O}_2$ – singlet oxygen
3Chl – triplet state of Chl
ABTS – 2,2 -azino-bis (3-ethylbenzothiazoline-6-sulphonic acid)
AsA – ascorbate
APX – ascorbate peroxidase
BCF – bioconcentration factor
CAT – catalase
DHAR – dehydroascorbate reductase
DPPH – 2-diphenyl-1-picrylhydrazyl
ETC – electron transport chain
FAD – flavin adenine dinucleotide
FRAP – Fe^{2+} reducing antioxidant power assay
GPOX – glutathione peroxidase
GPX – guaiacol peroxidase
GR – glutathione reductase
GSSG – glutathione disulfide
GST – glutathione S-transferases
 H_2O_2 – hydrogen peroxide
LHCs – light-harvesting complexes
MDA – malondialdehyde
MDHAR – monodehydroascorbate reductase
NADH – nicotinamide adenine dinucleotide
NPs – nanoparticles
 $\text{O}_2^{\bullet -}$ – superoxide radicals
 OH^{\bullet} – hydroxyl radicals
PS II – photosystem II
ROS – reactive oxygen species
SM – substrate moisture
SOD – superoxide dismutase
TBA – thiobarbituric acid
TF – translocation factor
Ti – tolerance index
TPC – total phenolic compound

INTRODUCTION

Relevance of the topic

Agricultural crops are increasingly adversely affected by heavy rains or droughts, temperature extremes, strong winds, and the spread of new pathogens due to a changing climate (Shahzad et al., 2021; Rivero et al., 2022). Drought stress is the most common environmental factor limiting crop productivity (Basu et al., 2016; Liliane et al., 2020). Soil drought characteristics can vary from water deficit and depletion to long periods of water deficit or when the soil water content is below total capacity (Seleiman et al., 2021). Drought stress reduces photosynthesis by reducing leaf area and photosynthetic rate. In addition, during drought, the plant closes its stomata to reduce water loss through transpiration, reducing CO₂ and nutrient uptake (Khan et al., 2018; Gambetta et al., 2020; Kapoor et al., 2020; Ozturk et al., 2021). These induced damages are due to the plant's redox balance disruption, resulting in the formation of reactive oxygen species (ROS) (Yang et al., 2021). ROS causes oxidative stress in plant cells. Plants have adapted to survive under stressful conditions by developing an antioxidant system (Basu et al., 2016; Kapoor et al., 2020). Balancing between oxidative and reductive reactions is essential in maintaining redox homeostasis in plants, leading to plants forming an enzymatic and non-enzymatic antioxidant system. External measures, such as the application of various nanoparticles (NPs), can be used to reduce the harmful effects of adverse environmental factors on plants. (Kandhol et al., 2022; Ghani et al., 2022; Ahmad et al., 2022). The application of NPs has been discussed in recent decades to improve the resistance of various crops to abiotic stress. However, to date, there are still considerable gaps in knowledge on the benefits of NPs for plant growth and yields. Besides, there is still no consensus on the effects of NPs on the environment, exclusively on plant oxidative stress biomarkers and antioxidant systems.

Nanoparticles are materials of up to 100 nm, characterized by unique shapes, surface charge, and a large surface area (Modena et al., 2019; Raval et al., 2019). Therefore, in studying the effects of NPs on plants, it is essential to study the properties of NPs suspensions, as well as different methods of application pathways (Tarafdar et al., 2012; Mittal et al., 2020). Furthermore, it is crucial to choose the optimal concentration, which can be different for each type of plant and cause the opposite effect as expected.

The green pea (*Pisum sativum* L.) is one of the most popular plants of the legume family, which is exceptionally sensitive to drought (Nadeem et al., 2019). Molybdenum and boron are essential for pea-specific nodule bacteria and their differentiation into a nitrogen-fixing form. However, molybdenum and boron have not yet been studied on the nanoscale. In addition, the

use of silicon and copper NPs, due to their exceptional properties in regulating oxidative stress, makes this work not only scientific but also of practical importance.

Research hypothesis. Likely, the enzymatic and non-enzymatic antioxidant system induced by the application of nanoparticles neutralizes the oxidative burst in drought-affected pea plants.

The subject of research. The green pea (*Pisum sativum* L.) plant oxidative and antioxidant systems.

The aim. To select the most appropriate concentration and method of application of SiO₂, MoO₃, B₂O₃, CuO nanoparticles for green peas (*Pisum sativum* L.) by enhancing their resistance to drought and the combined effects of heavy metal copper and drought.

Research tasks:

1. To evaluate the effects of SiO₂, CuO, MoO₃, and B₂O₃ nanoparticles on morphological parameters, oxidative stress, and antioxidant system in green peas.
2. To investigate the effects of agrometeorological drought and SiO₂, CuO, MoO₃, and B₂O₃ nanoparticles on oxidative stress and antioxidant system in green peas.
3. To investigate the combined effect of environmental factors (agrometeorological drought, heavy metals, and nanoparticles) on oxidative stress and antioxidant system in green peas.
4. To evaluate the application pathway of various NPs in strengthening the activity of the green pea antioxidant system and resistance to adverse environmental factors.

Defended statements:

1. Using optimal concentrations of SiO₂, MoO₃, B₂O₃, and CuO nanoparticle suspensions for green peas grown in a substrate of normal moisture stimulate the activity of the antioxidant and oxidative system, which positively affects morphological parameters and increases plant productivity.
2. The resistance of peas to agrometeorological drought is increased by suspensions of SiO₂, MoO₃, B₂O₃, and CuO nanoparticles. It enriches the mineral nutrition of plants and activates enzymatic and non-enzymatic antioxidants, which reduce the content of oxidative biomarkers and positively affect the morphological parameters and yield of green peas.
3. Suspensions of SiO₂, MoO₃, and B₂O₃ nanoparticles increase the resistance of peas to the combined effects of agrometeorological drought and heavy metal copper by strengthening the antioxidant system, weakening the impact of oxidative stress, and

increasing the tolerance index of plants to copper. Conversely, the effects of CuO nanoparticles act synergistically with excess copper by increasing oxidative biomarkers and reducing productivity in green peas.

4. The effect of nanoparticle SiO₂, MoO₃, B₂O₃, and CuO suspensions on the antioxidant system, morphological parameters, and accumulation of macro- and microelements in peas depends on their stability, surface charge, and mode of exposure.

Novelty. The study provided new knowledge about the effect of SiO₂, CuO, MoO₃, and B₂O₃ nanoparticles on oxidative stress biomarkers, antioxidant system, and macro- and microelement changes in pea plants when the plants were grown in normal moisture substrate, moisture deficit conditions and was exposed to combined drought and heavy metal copper stress. It has been found that nanoparticle suspensions have positive and negative effects, which depend on the properties of the applied particles, their concentration, and the application method. The investigated zeta potential of SiO₂, CuO, MoO₃, and B₂O₃ nanoparticles showed that all suspensions were stable and anionic, and nano-sized particles were found in all aqueous suspensions. The most optimal concentrations of nanoparticles that had a positive effect on oxidative stress biomarkers, antioxidant system, macro- and microelement changes, and productivity of peas were selected: 50 ppm - SiO₂, CuO, MoO₃, 12.5 ppm - B₂O₃. It was also found that spraying with CuO and B₂O₃ nanoparticles affects pea productivity more effectively than watering, and watering plants with MoO₃ nanoparticle suspension was more effective than spraying.

Practical importance. Research on the effects of metal-based NPs on pea enriches the knowledge about the benefits or risks of NPs to plants, helps to include a wider variety of NPs, and expand the use of nanotechnology in crop production while contributing to the improvement of general agricultural practices. All this has great value for the development of agronomic science. In addition, the thesis has a social benefit, as the development of such research would help meet the needs for plant-based food in society and improve the quality of plant-based products.

Approval of dissertation. The results of the research were published in 2 scientific articles published in international journals with the Clarivate Analytics Web of Science citation index of the Scientific Information Institute database (Horticulturae (Q1, IF – 2.331), Journal of Soil Science and Plant Nutrition (Q1, IF – 3.872)) journals indexed in Clarivate Analytics Web of Science database and one scientific article in a peer-reviewed journal (Sodininkystė ir daržininkystė). The main research results were presented at five international conferences: International Conference on Agriculture and Horticulture (ICAH2020, oral presentation, online,

2020); 11th Scandinavian Plant Physiology Society Ph.D. Student conference (SPPS, poster presentation, online, 2020); 17th International Conference of young scientists on energy and natural sciences issues (CYSENI, oral presentations, online, 2021, 2022); 1st International Electronic Conference on Horticulturae (IECHo, poster presentation, online, 2022); 31st International Horticultural Congress (IHC, poster presentation, France, 2022)

Volume and structure of the dissertation. The dissertation consists of an introduction, literature analysis, description of research methods, analysis of results, conclusions, list of used literature, and list of publications published together with co-authors and appendices. Fifteen figures and 22 tables illustrate the literature analysis, description of the research methodology, and results. There are 295 sources in the bibliography, and the volume of the work is 155 pages.

1. LITERATURE REVIEW

1.1 The response of plants to environmental change

Plants are highly complex multicellular organisms with many organs and tissues above or below ground. To meet environmental challenges (abiotic and/or biotic stress), plants must adapt their development and metabolism to ensure survival. The concept of stress consists of four stages: alarm, resistance, exhaustion, and regeneration (Selye et al., 1936; Lichtenthaler, 1998). Plants are at a particular optimal physiological stage, depending on the environmental conditions of growth, light, water, and availability of minerals before stress. Stressors or combined stresses lead to the first three phases of the stress response and then to the regeneration phase if the damage is not too severe after removing the stressor. At the beginning of stress, plants react with a decrease in photosynthesis, metabolite transport, and ion uptake and transfer. Due to this decrease in metabolic activity, plants deviate from the usual physiological standard. Acute damage and senescence occur quickly in plants with only weak stress resistance mechanisms or none - in such plants, the resistance minimum is low (Selye et al., 1936; Lichtenthaler, 1998). However, most plants with stress-coping mechanisms, such as acclimatization of metabolic flows, repair processes, and long-term metabolic and morphological adaptations, activate theirs in the alarm phase. This general adaptation of the plant sets a new physiological standard, which is the optimal physiology stage under the influence of a stressor and corresponds to the maximum plant resistance. In the case of long-term stress or a dose of stress that overloads the plant's stress-coping mechanism, the exhaustion stage (termination phase) occurs. This causes severe damage to the plant and, ultimately, cell death. However, when stressors are removed right before senescence processes become dominant, plants regenerate and move to new physiological standards (regeneration phase). The time of exhaustion and the stage at which the stressors are removed from the plant determine to which new physiological standard the plants will grow within the minimum and maximum resistance limits (Lichtenthaler, 1998).

However, the approach to stress has become broader, and it is not limited to general adaptation syndrome and only adverse effects on live organisms nowadays. Scientists have noticed that a weak stimulus, causing a lower impact of stressors, can increase plant productivity and positively affect metabolic processes. Hormesis is a phenomenon where low doses of environmental factors can stimulate plant mechanisms to cope with stress, increasing the growth and production of secondary metabolites, and wound healing capacity to ultimately maintain homeostasis (Poschenrieder et al., 2013; Duarte-Sierra et al., 2020). Modern scientists distinguish two possible effects of stressors on plants: oxidative distress as harmful and

oxidative eustress as positive (Sies, 2019). Excessive oxidative stress is detrimental to plant biomolecules, but maintenance of physiological oxidative levels, oxidative eustress, is essential for plants growing under optimal conditions due to redox signaling (Seis et al., 2017).

Acclimating plants to stress conditions successfully requires an efficient, timely, and coordinated response that spans most, if not all, plant parts and tissues (Zandalinas, 2020). Plants have developed several systemic signaling pathways that allow the transmission of different stress signals from a particular part of a plant. It should be noted that different plant cells and tissues can communicate with each other through simplastic, vascular, and apoplastic connections. That allows short, long-distance intercellular communication (systemic signaling) (Sevilem et al., 2013). The systemic response to the biotic stimulus is called systemic acquired resistance and is usually caused by the interaction of certain cells with pathogens (bacteria, fungi, viruses). A systemic response to abiotic stimulus (heat, cold, high light, UV, salinity, water deficiency, osmotic stress) is called systemic acquired acclimatization. A systemic response to mechanical stress that can be induced by biotic (e.g., insect feeding) or abiotic stress (wind, hail, heavy rain) is called a systemic wound reaction (Conrath, 2006, Perez and Brown, 2014). These systemic reactions warn distant and non-stressed plant tissues of a biotic or abiotic threat and activate resistance or acclimatization pathways in these tissues.

Among the main systemic signals in plants are electrical, calcium, reactive oxygen species (ROS) and hydraulic waves, and plant hormones. Systemic signals spread from the affected tissue to the whole plant in a few minutes (Gilroy et al., 2016; Vega-Muñoz et al., 2020; Fichman and Mittler, 2020; Zandalinas, 2020). Due to the rapid and systematic transmission of calcium, electricity, and ROS waves, the plant adapts to light, heat, and other stressors (Gilroy et al., 2016). Most research has focused on how these systemic signals respond to a single stress stimulus affecting a specific plant leaf or root. Still, plants are often exposed to more than one environmental stress simultaneously, known as a "stress combination" (Lamers et al., 2020). Acclimation of plants to a combination of stress (such as a combination of drought and heat) includes responses to each stress that affects the plant simultaneously (Zandalinas, 2020).

1.2 Oxidative stress in plants

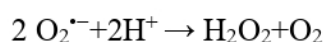
Oxygen (O₂), which has appeared in the Earth's atmosphere mainly as a product of photosynthesis, is a double-edged sword for aerobic organisms. It allows efficient energy production by enzymatic combustion of organic compounds but also damages the formation of reactive oxygen species (ROS) that react with all biomolecules (Bartosz, 1997). Over millions of years of evolution, aerobic organisms have adapted to the threat posed by O₂ by developing antioxidant systems that reduce the harmful effects of ROS and even by using ROS as a defense

and possibly as secondary signaling (Mansoor et al., 2022). As a result, each aerobic cell has a dynamic balance between ROS-generating reactions, thus promoting non-specific oxidations and antioxidants (Figure 1). The shift in this balance in favor of oxidative reactions is called oxidative stress (OS) (Inzé and Montagu, 1995; Das and Roychoudhury, 2014; Noctor and Foyer, 2016).

OS results in redox imbalance due to the formation of ROS (Demidchik, 2015). The use of O₂ in many essential metabolic processes of living systems has had an evolutionary cost because its metabolism leads to the production of ROS. Cells generate both non-radical and free-radical ROS. Free radicals include hydroxyl radicals (OH•), superoxide radicals (O₂^{•-}), organic hydroperoxides (ROO•), organic hydroperoxyl radicals (RO₂•), and alkoxy radicals (RO•). Non-radicals include singlet oxygen (¹O₂) and hydrogen peroxide (H₂O₂). The most prevalent cellular ROS in plants is H₂O₂, OH•, O₂^{•-}, and ¹O₂ (Zhou et al., 2014; Inzé and Montagu, 1995; Das and Roychoudhury, 2014; Noctor and Foyer, 2016).

The superoxide anion radical (O₂^{•-}) is the primary ROS generated in cells, leading to a cascade of reactions and the generation of secondary ROS, either directly or via enzymatic and metal-catalyzed processes, depending on the cellular compartment (Huang et al., 2019; Gill and Tuteja, 2010). The O₂^{•-} is generated during PSI's acyclic electron transport chain (ETC) (Foyer, 2018). Typically, cytochrome c oxidase interacts with O₂ to form H₂O, but when O₂ reacts with different ETC components, O₂^{•-} can be created. O₂^{•-} is moderately reactive, does not cause much damage by itself, and has a short half-life of 2–4 μs (Gill and Tuteja, 2010). However, it is rapidly converted to H₂O₂ by superoxide dismutase (SOD). Thus, preventing its accumulation in cells and protecting proteins from damage, especially those containing Fe-S clusters (Imlay, 2006; Gill and Tuteja, 2010). The most important reaction of O₂^{•-} is dismutation. Dismutation can be non-enzymatic or catalyzed by SOD when it reacts with another O₂^{•-} molecule. It results in the reduction of one of its molecules to H₂O₂ and the oxidation of the other to oxygen (Das and Roychoudhury, 2014; Sharma et al., 2012; Dumanović et al., 2021).

Hydrogen peroxide (H₂O₂) is a moderately reactive ROS usually generated by the neutralization of O₂^{•-} via univalent reduction or protonation. However, it can be formed non-enzymatically due to low pH or a reaction catalyzed by the enzyme SOD (Das and Roychoudhury, 2014).



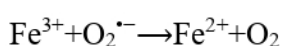
H₂O₂ plays a dual role: at low concentrations, it is involved in signal transduction; at high concentrations, it is toxic to the cell. It acts as a signal for regulating essential physiological processes such as senescence, photorespiration and photosynthesis, stomatal movement, cell cycle, and growth and development at low intracellular concentrations. Under normal

conditions, the amount of H₂O₂ in plant leaves ranges from about 1 μmol per gram of fresh tissue weight (10 μmol/L H₂O₂ in peroxisomes) (Raja et al., 2022). The elimination half-life of H₂O₂ is 1 ms, which allows it to cross membranes through aquaporins (Bienert et al., 2006) and cover the cell over long distances (Foyer et al., 1997) and cause oxidative damage (Vilchis-Landeros et al., 2020). High intracellular concentrations (10–100 μmol L⁻¹ in peroxisomes) of H₂O₂ oxidize the residues of cysteine (-SH) and methionine (-SCH₃). Further, oxidize the thiol groups of the Cu/Zn-SOD and Fe-SOD enzymes. Excess production of H₂O₂ in plant cells during oxidative stress is caused by factors such as chilling, drought, UV radiation, wounding, intense light, and pathogens (Sharma et al., 2012).

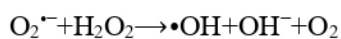
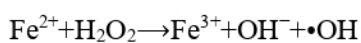
Limited internal carbon dioxide (CO₂) concentration and fixation due to stomatal closure, oxygenation via ribulose 1,5-bisphosphate is favored, resulting in enhanced photorespiration. As a result, 70% more H₂O₂ is formed in the case of water shortage than when plants grow in favorable conditions (Das and Roychoudhury, 2014). The main sources of H₂O₂ production in plant cells are the ETC in the chloroplast, endoplasmic reticulum (ER), cell membrane, mitochondria, photorespiration, and β-oxidation of fatty acids. Additional hydrogen peroxide is produced by photooxidation by NADPH oxidase and xanthine oxidase. (Anjum et al., 2016).

O₂^{•-} and H₂O₂ are moderately reactive, but the full damage caused by them is realized only after transformation into more reactive species. An indirect effect of H₂O₂ is expressed by crossing the cellular membranes via peroxiporins and by Fenton (Fenton, 1984) and Haber-Weiss reaction (Koppenol, 2001) with the iron (Fe²⁺) or copper (Cu⁺) ions, resulting in the formation of more potent toxic species such as OH• (Dumanović et al., 2021).

The Haber-Weiss reaction:



The Fenton reactions:



The hydroxyl radical (OH•) is the most reactive and toxic ROS due to its short half-life, positive redox potential (close to +2 V), and high affinity for biomolecules, non-selectively oxidize sugars, amino acids, DNA, lipids, proteins, and metals, leading to damage or genetic instability (Bano et al., 2021; Dumanović et al., 2021; Sharma et al., 2012).

Lipid peroxidation, which leads to the destruction of proteins and membrane damage, is caused by OH• (Ayal et al., 2014; Repetto et al., 2012). Plants have no enzymatic system to eliminate this radical, but mechanisms to maintain iron homeostasis have evolved in cells. Plants' propensity to oxidative stress is also determined by the absorption of metals from the soil (Rai et al., 2021). Special efforts are made to prevent the reaction of transition metals and H₂O₂

by sequestering them. In this way, plants are provided with ferritin and metallothionein, which help to accumulate iron, zinc, and copper (Ravet and Pilon, 2013).

Singlet oxygen ($^1\text{O}_2$) reacts with the antenna system's triplet state of chlorophyll (Chl) (Krieger-Liszkay, 2005; Ma et al., 2009; Dogra and Kim, 2019). The lack of intracellular CO_2 stimulates the formation of $^1\text{O}_2$ during stressors. $^1\text{O}_2$ is particularly dangerous for photosystems I and II (PSI, PSII) (Krieger-Liszkay, 2005). Its half-life is shorter, about a few ns, but it can diffuse 100 nanometers in the cell and damage pigments, nucleic acids, proteins, and lipids (Dogra and Kim, 2019). Plants have been able to adapt and efficiently remove $^1\text{O}_2$ using tocopherol, plastoquinone, and β -carotene. Genes responsible for protection against photooxidative stress can be regulated by $^1\text{O}_2$ (Foyer, 2018; Dmitrieva et al., 2020).

Plants are particularly affected by $^1\text{O}_2$ because they are rich in Chl, which acts as a photosensitizer, resulting in the continuous formation of $^1\text{O}_2$ in the leaves. Chl absorbs light in light-harvesting complexes (LHCs), photosystem II (PS II) reaction centers, and internal antennas (Krieger-Liszkay, 2005; Triantaphylidès and Havaux, 2009; Dogra and Kim, 2019). Its excited state is long-lived enough that the excitation energy can be converted into an electrochemical potential using the charge separation method during photosynthesis (Dmitrieva et al., 2020). However, when energy is not efficiently used, the excited triplet state of Chl (3Chl) near molecular oxygen can provide sufficient energy to cause the formation of $^1\text{O}_2$ (Triantaphylidès and Havaux, 2009). $^1\text{O}_2$ is one of the most critical ROS because it can reduce the activity of PS II by degrading the D1 protein, which causes pigment "bleaching" (Tripathy and Oelmüller, 2012). Plants use two strategies to protect themselves from photoinhibition. For prevention, it is the dissipation of excessive 3Chl excitation energy in PS II antennas in the form of heat (non-photochemical quenching). Another method is based on the ability of PS II to transfer electrons to nearby acceptors such as carotenoids and α -tocopherol. Due to this, excess energy is released in the form of heat and returns to its ground state (Derks et al., 2015; Roach and Krieger-Liszkay, 2014). This transfer of Chl and carotenoids is efficient, 5% 3Chl remains, and this incomplete quenching is evidence that antenna pigments are considered a potential source of $^1\text{O}_2$ in chloroplasts, which tend to damage the D1 protein (Roach and Krieger-Liszkay, 2014).

1.2.1 Sources of reactive oxygen species

Redox balance is a crucial factor that ensures the smooth functioning of proteins and enzymes involved in essential metabolic pathways. Plant cell organelles chloroplasts, mitochondria, peroxisomes, plasma membranes, endoplasmic reticulum (ER), and the cell wall

produce ROS during normal and stress conditions (Sharma et al., 2012; Zhou and Shao, 2014; Noctor and Foyer, 2016; Bano et al., 2021).

The chloroplast consists of an orderly arranged system of thylakoid membranes, where photosynthetic mechanisms responsible for light capture and efficient light collection are located. PSI and PSII photosystems are the core of the light collection system in thylakoids and the main ROS formation sources. The formation of $O_2^{\cdot-}$ in PS via the Mehler reaction ($2O_2 + 2F_{d_{red}} \rightarrow 2O_2^{\cdot-} + 2F_{d_{ox}}$) is determined by abiotic stressors such as water stress, CO_2 limitation, and excess light (Cruz de Carvalho, 2008; Foyer, 2018; Das and Roychoudhury, 2014). Cu/Zn SOD enzymes in PSI convert $O_2^{\cdot-}$ to H_2O_2 . Fe-S clusters can also be accomplices of electron leakage from the PSI ETC (Imlay, 2006; Gill and Tuteja, 2010). Through the QA and QB electron acceptors, electron penetration into PS II occurs and is responsible for the formation of $O_2^{\cdot-}$ (Pospíšil, 2016). The superoxide radical converts to a more toxic ROS such as $OH\cdot$ via the H_2O_2 intermediate via a Fenton reaction at the Fe-S centers (Imlay, 2006; Gill and Tuteja, 2010). PSII also produces 1O_2 , and this occurs in two ways: by the formation of triplet Chl ($Chl \rightarrow ^3Chl$), which reacts with dioxygen to release singlet oxygen ($^3Chl + ^3O_2 \rightarrow Chl + ^1O_2$), when environmental stress disturbs the balance between energy utilization and LHC) PSII generates 1O_2 when the ETC is too reduced (Dogra and Kim, 2019). Accumulation of 1O_2 in chloroplasts leads to peroxidation of membrane lipids. Polyunsaturated fatty acids (PUFA) and membrane proteins are particularly affected due to damage to the PSII P680 reaction center (Foyer, 2018; Li and Kim, 2021). The primary source of ROS production in plants is the chloroplast. Controlling and scavenging ROS in the chloroplast is necessary to ensure the continued survival of stressed plants.

To a lesser extent, harmful ROS such as H_2O_2 and $O_2^{\cdot-}$ are also produced in mitochondria (Huang et al., 2016; Cruz de Carvalho, 2008). Mitochondria of plants participate in photorespiration and have O_2 and carbohydrate circulation. The mitochondrial ETC (mtETC) is the site of ROS production, as it generates electrons with sufficient energy. Complex I, III, and IV are the main components of mtETC responsible for ROS production (Møller, 2001). Complex I directly reduce O_2 to $O_2^{\cdot-}$ in its flavoprotein domain using nicotinamide adenine dinucleotide dehydrogenase (NADH). When the bound of NAD^+ substrates is lacking, reverse electron flow from complex III to complex I occur, and ROS production is enhanced. By the principle of ATP hydrolysis, this reverse flow of electrons is controlled (Vanlerberghe, 2013). An electron is transferred to cytochrome c1 when ubiquinone is completely reduced. This results in an unstable ubisemiquinone half-radical that promotes electron leakage to O_2 , thereby generating $O_2^{\cdot-}$ in complex III. (Rhoads et al., 2006). Although $O_2^{\cdot-}$ is the predominant ROS in mitochondria, it is converted to H_2O_2 by ascorbate peroxidase and Mn-SOD (Sharma et al.,

2012). About 1-5% of the total O₂ consumed by mitochondria is directed to the production of H₂O₂ (Møller, 2001). Mitochondrial ROS production is activated under abiotic stress conditions (Vanlerberghe, 2013). Stress affects the close connection between ETC and ATP synthesis, resulting in a significant reduction of electron carriers (ubiquinone pool), which influences the formation of ROS (Huang et al., 2016; Nadarajah, 2020). During drought, to compensate for the lower rate of chloroplast ATP synthesis, mitochondrial ATP synthesis increases due to increased respiration and, thus, the production of ROS (Pastore et al., 2007; Cruz de Carvalho, 2008; Miller et al., 2010). Two enzymes neutralize ROS in mitochondria, alternative oxidase and mitochondrial Mn-SOD (Huang et al., 2016; Møller, 2001; Vanlerberghe, 2013).

Peroxisomes are metabolically plastic because the amount of enzyme in them can change depending on environmental conditions, plant species, and cell and tissue type (Corpas et al., 2019). Peroxisomes are 0.1–1 µm spherical microbodies bound by a single membrane and are the main sites of intracellular H₂O₂ production due to integrated oxidative metabolism (Corpas et al., 2019; Smirnoff and Arnaud). Xanthine is converted to uric acid by a reaction catalyzed by xanthine oxidoreductase. This hydroxylase enzyme, which contains flavin adenine dinucleotide (FAD), molybdenum, iron, and sulfur, exists in two interconvertible forms. Oxygen-dependent xanthine oxidase and NAD-dependent xanthine dehydrogenase may be present. As a result, either superoxide radicals or NADH can be formed during this reaction. Allantoin and H₂O₂ are then produced by urate oxidase (Corpas et al., 2008; Corpas et al., 2019; Kostić et al., 2015). Another pathway for the formation of H₂O₂ occurs when the stomata close during stress. As a result, increased photorespiration initiates glycolate production. Glycolate oxidase in peroxisomes oxidizes glycolate, releasing H₂O₂ (Del Rio et al., 1996; Rojas et al., 2012). In addition, fatty acid β-oxidation and flavin oxidase pathways influence unbalanced ROS production in peroxisomes.

The apoplast is the intercellular space filled with water and gas, formed between the cell membranes, the intermycelial and interfibrillar space of the cell walls, and the xylem, which extends to the rhizoplane and cuticle (Farvardin et al., 2020). In the apoplast, CO₂ is converted into a soluble form, which then enters the cytosol and participates in photosynthesis (Tränkner et al., 2018). It becomes an important site for H₂O₂ production when a stress signal with abscisic acid occurs. Enzymes such as NADPH oxidases, peroxidases, oxalate oxidases, and polyamine oxidases (Xue et al., 2009; Kärkönen and Kuchitsu, 2015) produce H₂O₂ and O₂^{•-} in the apoplast.

The endoplasmic reticulum can be a source of O₂^{•-} production. The reaction depends on NAD(P)H electron transfer involving CytP₄₅₀. After the reaction of the substrate RH with Cyt P450, the compound is reduced with the flavoprotein, resulting in the formation of a radical

intermediate Cyt P₄₅₀ R⁻. Cyt P₄₅₀ R⁻ reacts with triplet oxygen (³O₂) to form the oxygenated complex Cyt P₄₅₀-ROO⁻. This oxygenated complex can be reduced with cytochrome b or complexes to release O₂^{-•} (Sharma et al., 2012).

The plasma membrane surrounds the entire plant cell. It is responsible for transmitting information about the changing environment necessary for the cell's survival (Schapire et al., 2008). At the plasma membrane, the transfer of electrons from cytosolic NADPH to O₂ is dependent on NADPH oxidase, resulting in the formation of O₂^{-•} (Das and Roychoudhury, 2014). Then, either spontaneously or catalyzed by SOD, O₂^{-•} is converted to H₂O₂ (Schopfer and Liskay, 2006). NADPH oxidase is essential in defense of plants against abiotic stresses. Hydroperoxidation of PUFA with the participation of lipoxygenase is induced in the cell wall under stress conditions, and polyamines or diamines are used by diamine oxidases, resulting in the formation of ROS in the cell wall (Hasanuzzaman et al., 2020).

1.3 Plant antioxidant system

Balancing between oxidative and reductive reactions is vital in maintaining redox homeostasis in plants. This led to the formation of an antioxidant system in plants. Antioxidants are molecules that inhibit or slow down the reactions of free radicals, thereby preventing cell damage. Antioxidants in plants are divided into two groups: enzymatic and non-enzymatic. Enzymatic antioxidants (Table 1.3.1) include superoxide dismutase (SOD), ascorbate peroxidase (APX), guaiacol peroxidase (GPX), catalase (CAT), glutathione reductase (GR), glutathione peroxidase (GPOX), glutathione S-transferases (GST), dehydroascorbate reductase (DHAR), and monodehydroascorbate reductase (MDHAR).

SOD is essential in oxidative stress, as it catalyzes the dismutation of O₂^{-•}, thereby reducing the risk of OH[•] formation through metal-catalyzed reactions. SOD is found in the plant in three forms: Fe SOD, Mn-SOD, and Cu/Zn-SOD (Suman et al., 2021). APX is one of the most critical antioxidants in plants, with an essential role in scavenging H₂O₂ (Amir et al., 2019). Its principal function is to neutralize ROS generated during stress. It removes H₂O₂ by using ascorbate (AsA) as an electron donor in the ascorbate-glutathione (AsA/GSH) cycle. Four isoforms of the APX family are distinguished according to localization: tAPX, gmAPX, sAPX, and cAPX (Habi, 2014). The highest amounts of CAT are found in peroxisomes, with smaller quantities of CAT found in mitochondria (Mhamdi et al., 2010). Three CAT genes exist: class I CAT in photosynthetic tissues, class II CAT in vascular tissue, and class III in seeds and young seedlings (Willekens et al., 1995). CAT binds H₂O₂ produced in peroxisomes during photorespiration and β-oxidation of fatty acids. Four heme subunits with Fe²⁺ ions are included in the composition of CAT. They catalyze the dissociation of two H₂O₂ molecules into water

and oxygen. The GPX enzyme contains a heme group; its main function is to remove excess H_2O_2 during normal metabolism and stress. It is critical for lignin biosynthesis and is also important for degrading indole acetic acid (IAA) using H_2O_2 , thus protecting the plant from the effects of stress. GPX electron donors are guaiacol and pyragol. GPX is active in the cytosol, vacuoles, and cell wall. GPX activity depends on plant species and stress conditions (Mika and Lüthje, 2003; Das and Roychoudhury, 2014). GR is known as flavoprotein oxidoreductase. The highest amounts of this enzyme are found in chloroplasts, less in mitochondria and cytosol (Das and Roychoudhury, 2014). GR belongs to the AsA-GSH cycle and, by maintaining a reduced GSH state, plays an essential role in the defense system against ROS. GR catalyzes the reduction of GSH, a molecule involved in much metabolic regulation and antioxidant processes in plants, where GR catalyzes the NADPH-dependent disulfide bond reaction of GSSG and is, therefore important in supporting GSH formation. GR and GSH are crucial in determining plant tolerance to various stresses.

GPOX are thiol-based enzymes that protect plant cells from oxidative stress damage by scavenging ROS by catalyzing the reduction of H_2O_2 and other organic hydroperoxides (Eshdat et al., 1997). According to their localization and composition, eight different genes were found encoding GPOX activity in plants (Bela et al., 2015). Glutathione-dependent peroxidase activity can be associated with glutathione transferase isozymes. They are essential in detoxifying lipid hydroperoxides and other reactive molecules under stress conditions (Hasanuzzaman et al., 2019). GST is one of the most versatile enzymes that catalyze reactions involving the conjugation of glutathione (GSH; γ -Glu-Cys-Gly) with electrophilic compounds (Moons, 2005). One of the functions of GST is to scavenge ROS. GSTs can also function as anthocyanin transporters into the vacuole, phenylpropanoid and auxin carriers, and enzymes in tyrosine catabolism. Plant GSTs have been divided into seven classes: GSTU (Tau), GSTF (Phi), GSTL (Lambda), GSTT (Theta), GSTZ (Zeta), DHAR (dehydroascorbate reductase) (Estévez et al., 2020). MDAR reduces MDHA to AsA using electron donors such as NADH/NADPH. Considering that MDHA is an unstable radical, MDAR must quickly convert it to AsA because if it is not there, it is spontaneously converted to AsA and dehydroascorbate (DHA) (Eltayeb et al., 2007). In the DHAR-catalyzed reaction, DHA is reduced to AsA using GSH as the reducing agent (Eltayeb et al., 2006). Therefore, DHAR and MDAR are critical components in maintaining reduced AsA levels and are extremely important for oxidative stress resistance (Yin et al., 2010).

Table 1.3.1. Enzymatic antioxidants functions and sites in plants

Enzymatic antioxidant	Function (as catalysator or directly involved)	Location in the cell
Superoxide dismutase, SOD (EC 1.15.1.1)	$O_2^- + O_2^- + 2H^+ \rightarrow 2H_2O_2 + O_2$	Fe-SOD in chloroplast; Mn-SOD in mitochondria and peroxisomes. Cu/Zn-SOD in chloroplast and cytosol.
Ascorbate peroxidase, APX (EC 1.11.1.11)	$H_2O_2 + AsA \rightarrow 2H_2O + DHA$	tAPX – thylakoid, gmAPX - glyoxisome membrane, sAPX – chloroplast, cAPX – cytosol.
Catalase, CAT (EC 1.11.1.6)	$2H_2O_2 \rightarrow 2H_2O + O_2$	Peroxisomes and mitochondria.
Guaiacol peroxidase, GPX (EC 1.11.1.7)	$H_2O_2 + GSH \rightarrow H_2O + GSSG$	Cytosol, vacuoles, and cell wall
Glutathione reductase, GR (EC 1.6.4.2)	$GSSG + NADP(H) \rightarrow GSH + NAD(P)^+$	Chloroplast, mitochondria, and cytosol
Glutathione peroxidase, GPOX (EC 1.11.1.9)	$2 GSH + H_2O_2 \rightarrow GSSG + 2 H_2O$	GPX1, GPX7 - chloroplast GPX2 - chloroplast, mitochondria, and cytosol GPX3 - mitochondria, Golgi apparatus GPX4 - mitochondria and cytosol GPX5 - plasma membrane and endoplasmic reticulum GPX6 – chloroplast, plasma membrane, apoplast GPX8 – nucleus and cytosol
Glutathione S-transferases, GST (EC 2.5.1.18)	$RX + glutathione = HX + R-S-glutathione$	Tau - cytosol and nucleus Phi - cytosol and chloroplast Theta - peroxisome and nucleus Zeta - cytosol DHAR - cytosol, chloroplast, and peroxisome Lambda - cytosol, chloroplast, and peroxisome
Monodehydroascorbate reductase, MDHAR (EC 1.6.5.4)	$H^+ + 2MDHA + NADH \rightarrow 2AsA + NAD^+$	Chloroplast, peroxisome, mitochondria, and cytosol

The main non-enzymatic antioxidants in plants are ascorbic acid, phenols, tocopherols, flavonoids, carotenoids, and glutathione (Gill and Tuteja, 2010; Hasanuzzaman et al., 2019). Ascorbic Acid (AsA) is one of the most powerful and widely studied antioxidants. It can donate

electrons for various enzymatic and non-enzymatic reactions (Akram et al., 2017). The catalysis of L-galactan- γ -lactone dehydrogenase produces the majority of AsA in plant cells in plant mitochondria via the Smirnoff-Wheeler pathway (Linster et al., 2007). The majority of AsA is concentrated in the apoplast. AsA can be oxidized in two steps: oxidation to MDHA, which can be converted to AsA and DHA. AsA reacts with H_2O_2 , $OH\cdot$, $O_2^{\cdot-}$ and regenerates α -tocopherol from the tocopheroxyl radical, thereby protecting membranes from oxidative damage (Das and Roychoudhury, 2014). AsA in the chloroplast is also a cofactor for violaxanthin deep oxidase, where it is involved in the production of xanthophylls, which are directly engaged in quenching the excessive excitation energy of PS II (Davey et al., 2000). It also protects the activity of metal-binding enzymes.

There are four tocopherol isomers: α -, β -, γ -, δ -. α -tocopherol has the highest antioxidant capacity (Sadiq et al., 2019). It belongs to lipophilic antioxidants, which are effective scavengers of ROS and lipid radicals, making them essential defenders and essential components of biological membranes. α -tocopherol is synthesized from γ -tocopherol by γ -tocopherol-methyltransferase. Tocopherols protect lipids and other chloroplast membrane components by reacting with 1O_2 and quenching its excess energy, thereby protecting PSII structurally and functionally. Tocopherol is also an effective free radical scavenger. It reacts with lipid radicals $RO\cdot$, $ROO\cdot$, and $RO\cdot$ at the membrane-water interface, where α -tocopherol reduces them and is itself converted to $TOH\cdot$. $TOH\cdot$, which is then reduced by interaction with GSH and AsA (Munné-Bosch and Alegre, 2002). Tocopherols are synthesized only in photosynthetic organisms.

Carotenoids also belong to a family of lipophilic antioxidants, such as lycopene, β -carotene, xanthophyll, lutein, and zeaxanthin, that can be found in most plant tissues. In addition, they belong to a group of antenna molecules that absorb light between 450 and 570 nm and transfer the energy to the ChL molecule. Carotenoids demonstrate their antioxidant activity by protecting photosynthetic mechanisms in four ways (Ramel et al., 2012; Das and Roychoudhury, 2014): 1) reacting with LPO products; 2) remove 1O_2 and generate heat as a byproduct, 3) react with $3Chl^*$ and excited Chl (Chl^*) to prevent 1O_2 formation, and 4) dissipates excess excitation energy through the xanthophyll cycle.

Phenols are a group of secondary metabolites such as flavonoids, tannins, hydroxycinnamate esters, and lignin (Roaa, 2020; Grace et al., 2000). The antioxidant properties of phenols are due to their constituent aromatic ring with $-OH$ or $-OCH_3$ substituents, which are adapted to scavenge free radicals (Pietta, 2000). They can donate an electron or a hydrogen atom to neutralize ROS. Phenols can modify the kinetics of peroxidation by changing the lipid

package, which determines the reduced fluidity of the membrane (Arora et al., 2000). These changes can limit the diffusion of free radicals and inhibit the peroxidation reaction.

Glutathione (GSH) is a thiol tripeptide (γ -glutamyl-cysteinyl-glycine) abundantly found in chloroplasts, ER, cytosol, vacuoles, peroxisomes, mitochondria, and even the apoplast (Das and Roychoudhury, 2014). It is important for cell differentiation, cell growth and division, cell senescence, and expression of stress-responsive genes. This versatility of GSH is due to its high reducing potential. GSH scavenges ROS by reducing them when present and generating GSSG as a byproduct (Garcia-Caparrós et al., 2021). GSH also aids in the formation of phytochelatin via phytochelatin synthase, which helps to chelate heavy metal ions, thereby eliminating another potential source of ROS generation in plants (Roychoudhury et al., 2012). Therefore, a delicate balance between GSH and GSSG is necessary to maintain the cell's redox state.

GSH and most enzymatic antioxidants are involved in the ascorbate-glutathione cycle (AsA-GSH) (Figure 1.3.1), which is essential in maintaining the redox state in plants (Roychoudhury et al., 2012; Garcia-Caparrós et al., 2021). The cycle begins when H_2O_2 is detoxified in the AsA-GSH cycle by APX, which in turn oxidizes two AsA molecules (Dumanović et al., 2021). This results in the formation of the short-lived radical monodehydroascorbate, which can spontaneously convert to AsA and dehydroascorbate (DHA) and/or be reduced to AsA by NAD(P)H catalysis (Gallie, 2013). Unlike AsA, DHA has no antioxidant properties and is converted back to AsA. This occurs using DHA reductases and the addition of two electrons from GSH. GSH-dependent DHAR activity is expressed in chloroplasts, mitochondria, and peroxisomes (Nakano and Asada, 1981). GSH is regenerated from the oxidized state by catalyzing the reaction of glutathione disulfide (GSSG) with GR and using electrons from NAD(P)H, thus closing the regeneration cycle of AsA and GSH (Roychoudhury et al., 2012; Dumanović et al., 2021).

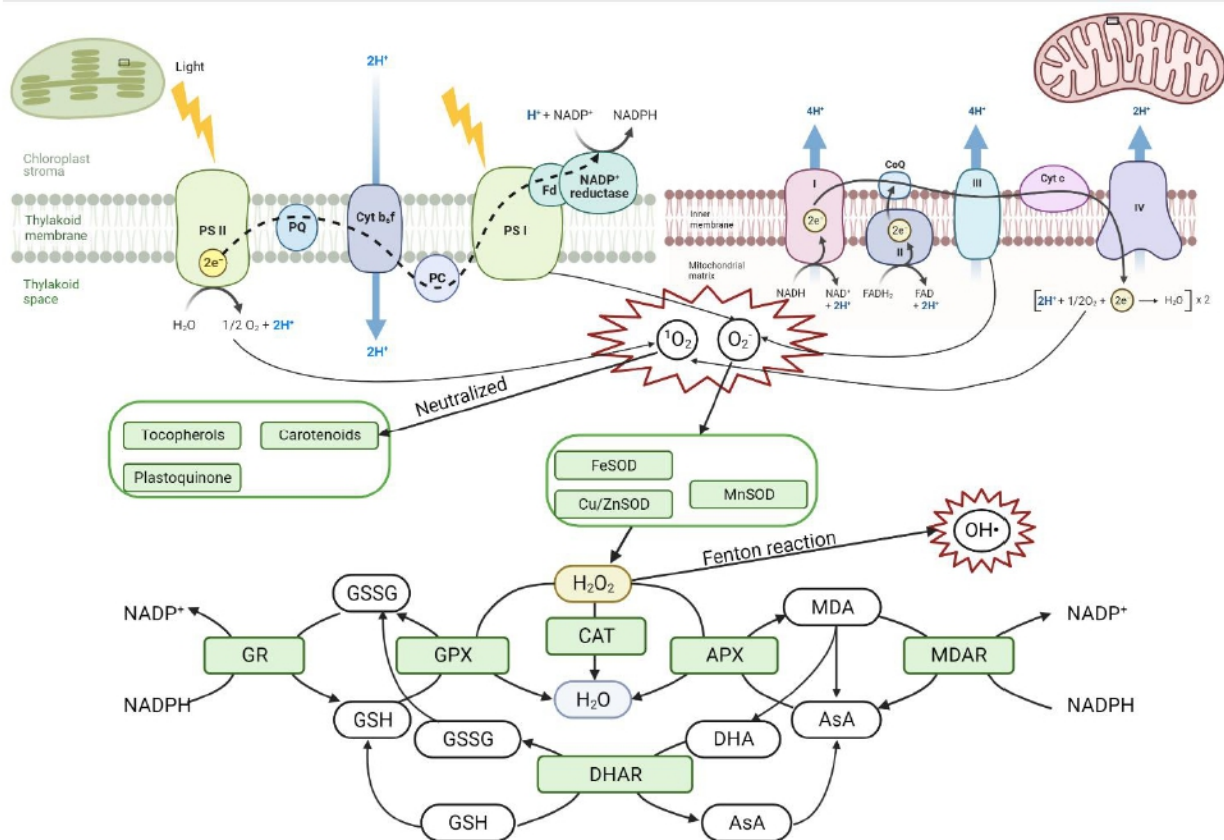


Figure 1.3.1. Main sources of ROS production and antioxidant defense mechanisms in the plant cell

1.4 Specific effects of drought and heavy metals stresses on plants

The impact of abiotic stress on plants in both natural and agricultural environments is a crucial topic due to the potential effects of climate change on precipitation patterns and temperature extremes, salinization of agricultural lands due to irrigation and fertilization, and maintaining increased agrarian productivity in marginal lands (Mitter, 2006; Pinheiro and Chaves, 2011; Corwin, 2021). In field conditions, plants can experience several different abiotic stresses at the same time, for example: decreased water availability during drought or increased excessively during flood, extreme temperatures, including heat waves and freezing, reduced availability of nutrients from the soil (or, conversely, the accumulation of toxic ions during salt stress), heavy metals, excess light (especially when photosynthesis is limited) or increased soil hardness limiting root growth (Seleiman et al., 2021). Different stresses cause different responses in plants, which requires plants to use other acclimation processes. In addition, when several strains affect plants simultaneously, their adverse effects overlap, and the plant becomes more stressed (Rivero et al., 2022).

1.4.1 Effects of drought (water deficit) on plants

Drought is one of the significant constraints limiting crop production worldwide. Crop growth models predict that this issue will be more severe in the future. The conceptual definition

of drought is related to physical processes such as meteorological drought resulting from a lack of precipitation, agricultural drought resulting from a lack of soil moisture, hydrological drought resulting from water shortages in lakes and streams, and water shortages resulting from water management (Mukherjee et al., 2010; Zargar et al., 2011). Drought impairs normal growth, disturbs water relations, and reduces plant water use efficiency (Farooq et al., 2012). The diversity of plant species growing in climate regions with arid conditions suggests that some plants are drought-tolerant. Drought resistance is a broader term applied to plant species with adaptive traits that allow them to escape, avoid, or tolerate drought stress (Basu et al., 2016; Bowles et al., 2021; Santos et al., 2022). "Drought escape" is the ability of a plant species to complete its life cycle before the onset of drought (Carraro et al., 2022, Abid et al., 2018). Plants survive drought stress through two distinct mechanisms: rapid phenological development and developmental plasticity, which can alter their vegetative and reproductive growth according to water availability (Basu et al., 2016; Shavrukov et al., 2017). Plants using rapid phenological development grow faster, form a minimal number of seeds against water deficits, and are thought to have no particular physiological, morphological, or biochemical adaptations. Plants with developmental plasticity mechanisms inhibit growth during drought stress, forming few flowers and seeds, but when more moisture appears, they grow indefinitely, forming many seeds (Carraro et al., 2022, Abid et al., 2018; Osakabe et al., 2014). "Drought avoidance" is when plants retain a relatively higher amount of water in their tissues despite reduced substrate moisture. This is achieved through various adaptations: reducing water loss (reducing transpiration, transpiration area, radiation absorption) and optimizing water uptake (increasing hydraulic conductivity and rooting) (Chaves and Oliveira, 2004; Chaves et al., 2002; Santos et al., 2022). "Drought tolerance" is when plants can withstand a decrease in the amount of water in the tissues by maintaining cell turgor, osmotic regulation, cell elasticity, and increasing protoplasmic resistance (Ashraf et al., 2011; Zivcak et al., 2016).

A sudden plant reaction is a stomatal closure during drought stress. However, stomatal closure reduces water loss through reduced transpiration and CO₂ and nutrient uptake, thus altering metabolic pathways such as photosynthesis (Basu et al., 2016; Chaves and Oliveira, 2004). Drought stress reduces photosynthesis by reducing leaf area and photosynthetic rate. During drought, a lack of intracellular CO₂ leads to the accumulation of photosynthetic electron transport components, leading to the generation of ROS (Gupta et al., 2020). However, plants have developed adaptive responses to mitigate drought-induced damage to photosynthesis, such as thermal dissipation of light energy, dissociation of light-harvesting complexes from photosynthetic reaction centers, xanthophyll, and water-water cycling. Disturbances in plant metabolism are caused by changes in photosynthetic carbon metabolism during drought stress.

Biochemical photosynthetic efficiency depends on ribulose-1,5-bisphosphate (RuBP) regeneration and ribose-1,5-bisphosphate carboxylase/oxygenase (RuBisCO) activity (Portis, 1992) in drought-stressed plants.

Plants growing in dry soils may have xeromorphic properties that reduce transpiration (Chaves et al., 2002). Transpiration reduction under drought conditions could be achieved by shedding leaves and reducing leaf number, size, and branching (Basu et al., 2016). Sclerophylly is another adaptation to combat drought stress, where plants develop stiff leaves that do not suffer permanent damage from wilting and can fully function when conditions return to normal (Basu et al., 2016; Chaves et al., 2002). During drought stress, reduced stomatal conductance is associated with reduced expression of aquaporin genes and reduced chloroplast surface area. The interaction of the leaf developmental stage and light availability with drought is also known to alter the differentiation of mesophyll and chloroplasts, influencing photosynthetic capacity (Chaves et al., 2002; Basu et al., 2016; Gupta et al., 2020). Another important factor is stomatal density and size. Scientists have found that under mild drought stress, the density of stomata increases and decreases under severe drought. All these plant adaptations positively affect water use efficiency (WUE) (Bertolino et al., 2019).

Crops important for agriculture first sense the lack of water through the root system (Ghosh and Xu, 2014, Kim et al., 2020). Drought stress does not affect primary root growth, but lateral root growth is significantly reduced, inhibiting the activation of lateral root meristems (Ghosh and Xu, 2014). Lateral root meristem activity is regulated through the ABA signaling cascade. Auxin signaling is also attenuated. As a result, the plant may begin to form small roots that increase water uptake by providing a more absorbent surface area (Kim et al., 2020).

Osmotic regulation (OA) is the process by which the water potential decreases and solutes accumulate in dividing cells, thereby helping to maintain turgor (Zivcak et al., 2016; Blum, 2017). Cell division and growth in plants depend on water availability and turgor. During drought stress, cell growth is inhibited by OA, although turgor in plant leaves and stems shows little or no reduction. OA is associated with stomatal conductance, photosynthesis, leaf water volume, and growth maintenance under drought conditions. In plants, the content of water decreases, and the salt concentration and mechanical resistance increase when growing in drought conditions (Bertolino et al., 2019). In addition, inorganic cations, organic acids, free amino acids, and carbohydrates accumulate. The rate at which cell volume will decrease depends on the elasticity of the cells or tissues due to reduced water uptake. Water moves out of the plant cells, and the water potential of the cells decreases. The osmotic potential decreases because the solutes in the cells are concentrated. If the cells accumulate solutes during dehydration and the osmotic potential decreases, it is like the decrease in water potential, and

turgor is maintained (Chen and Jiang, 2010). In practice, turgor maintenance is more often mentioned when the accumulation of soluble substances does not correspond to the decrease in water potential and turgor decreases. The maintenance of turgor depends on the elasticity of tissues and the speed of stress development, which leads to OA. It is noteworthy that when plant water deficit develops slowly, the degree of OA is higher (Blum, 2017).

1.4.2 Effects of heavy metals on plants

The group of metals and metalloids with an atomic density greater than 4 g/cm³ or at least five times that of water is collectively called "heavy metals". However, the chemical properties of heavy metals have a more significant influence in terms of toxicity (Raskin et al., 1994). Heavy metals are considered nickel (Ni), cadmium (Cd), lead (Pb), cobalt (Co), silver (Ag), chromium (Cr), copper (Cu), zinc (Zn), iron (Fe), and arsenic (As) (Nagajyoti et al., 2010; Ravet et al., 2013; Viehweger, 2014; Morkunas, 2018). The excess of these elements in nature can be formed naturally (rock outcroppings or geological parent material) or due to anthropogenic factors (agricultural, industrial, atmospheric, and domestic waste) (Nagajyoti et al., 2010; Viehweger, 2014; Morkunas, 2018). Metals can interact with macromolecules and affect the structure and activity of proteins, so their balance is particularly important in plants (Ravet et al., 2013). Biological activity is essential for these transition metals Mo, Mn, Fe, Cu, and Ni. These metal ions bind to the active centers of certain proteins and participate in many oxidation-reduction reactions. Plants have evolved different mechanisms to maintain the balance of metal ions. As the first defense stage, plants exposed to heavy metals try to reduce or prevent their entry into the root cells. They work to confine metal ions in the apoplast by binding them to the cell wall or cell exudates or by inhibiting long-distance transport (Rsdkin et al., 1994; Morkunas, 2018). If the cell fails to protect itself in this way, it tries to transport, chelation, trafficking, and sequester the metal into the vacuole. When this occurs, plants activate oxidative stress defense mechanisms and synthesize stress-related proteins and signaling molecules such as heat shock proteins, reactive oxygen species, and hormones (Emamverdian et al., 2015).

Increased concentration of metals in the soil causes toxic effects on the environment. In Europe, the average concentration of Cu in the soil is about 11.4 - 17 mg Cu kg⁻¹. Using Cu preparations in agriculture as a fungicide can increase its amount in the soil hundreds of times (Printz et al., 2016). In small amounts, Cu is beneficial to both plants and humans, but an increase in Cu concentration in the soil can cause harmful changes in plants (Yruela, 2005). The bioavailability of Cu depends on total soil Cu content, soil pH, and cation exchange capacity. Meanwhile, the activity of free Cu²⁺ in the soil solution increases as the soil pH decreases, ultimately reducing Cu adsorption (Adrees et al., 2015). Plant roots also play an important role

in Cu bioavailability, but this depends on the plant species and the availability of Cu in the soil (Yruela, 2009). Similarly, Cu bioavailability is also affected by physical, chemical and biological processes occurring at the soil-root interface in the rhizosphere, such as changes in pH or dissolved organic matter.

Excess Cu inhibits plant growth and disrupts photosynthetic electron transport. The most common effect of higher Cu concentration in the growing medium is a reduction in the amount of photosynthetic pigments. A decrease in the amount of chlorophyll and damage to the structural photosynthesis apparatus leads to a decrease in photosynthesis. Cu has an inhibitory effect on both photosystems, but photosystem II is the most sensitive to excess Cu (Adrees et al., 2015). At high concentrations of Cu, oxidative stress is induced in the plant, and ROS production is generated, which leads to reduced plant biomass and yield (Fernandes and Henriques, 1991). Scientists have published a study on the effects of the heavy metal Cu on peas (Hattab et al., 2009). They found that the excess amount of Cu negatively affected the morphological parameters of peas (length of roots and shoots, leaf area), the content of chlorophylls and carotenoids, and significantly reduced photosynthetic rate (Hattab et al., 2009).

The effect of trace metals depends not only on the chemical properties, concentration, and density of the metal but also on the type of plant (Viehweger, 2014.). The toxic effect of Cu on biomass depends on plant species, for example, wheat (An, 2006) and sorghum (*Sorghum bicolor* L.) (Kasim, 2006) were more sensitive to Cu stress compared to maize plants and showed a decreasing trend with increasing Cu excess. Oilseed rape growth was more inhibited than mustard under Cu stress (Feigl et al., 2013). For soybeans, a significant decrease in biomass was caused by a concentration of >100 ppm Cu, while for chickpeas, a concentration of >60 ppm Cu was sufficient (Adhikari et al., 2012). Plants that accumulate metals or metalloids in their tissues hundreds or thousands of times higher than usual for most plants are hyperaccumulators. Plants can be metal scavengers, metal indicators, and metal hyperaccumulators according to their sensitivity to heavy metals (Pajuelo et al., 2011). Plants of the *Brassicaceae* family are characterized by hyperaccumulation of almost all heavy metals (Pandey and Bajpai, 2019). Due to Rhizobium bacteria, the Legumes family is characterized by hyperaccumulation of As, Cd, Cu, Zn, and Pb heavy metals in roots (Pajuelo et al., 2011). Such plants are adapted to survive in heavy metal environments. They have well-developed metal detoxification mechanisms involving mycorrhiza, cell exudates, plasma membrane, heat shock proteins, phytochelatin (PCs), metallothioneins (MTs), organic acids and amino acids (Singh et al., 2016). Extracellular plants play the role of mycorrhizae and extracellular exudates in the plasma membrane, falling by absorbing heavy metals or causing metal ion resorption. On the

other hand, intracellular heat shock proteins, MTs, organic acids, amino acids, and PCs also play essential roles in the tolerance of different heavy metals (Kumar et al., 2016).

1.5 Agronomic measures for stress mitigation in crops

Agronomic measures used to mitigate the effects of drought and heavy metals on crops range from cultivar selection and sowing time to tillage, cropping systems, fallowing, mulching, use of soil inoculants, irrigation, and nutrient management (Nguyen et al., 2018).

Selecting suitable crops and cultivars is perhaps the most important factor in soil suitability (Nguyen et al., 2018). Not all plants are suitable for growing in sand or sandy clay because water and elements do not stagnate. Besides, not all plants can grow in clay, where in wet conditions, water stagnates, and in drought conditions, it dries out, hardens, reduces the mobility of trace elements, and makes it difficult for roots to grow. The selection of crops and varieties for drought and heavy metal resistance should be based on the tolerance level (Reinhardt et al., 2021). It is crucial to consider the vegetation period - when the crop or variety matures under favorable conditions for survival. For example, under drought conditions, it is better to choose early-maturing crop varieties to grow before the drought stress peak (Rosenow et al., 1983; Shavrukov et al., 2017). It is also recommended to choose varieties with short stems and a small leaf surface area to help reduce transpiration in the event of stress. It would be helpful to select varieties with deep and wide root systems to increase water availability in soil (Wasson et al., 2012). If the soil is contaminated with heavy metals, it would be better to choose varieties that hyper-accumulate those metals (Varsha et al., 2010). As a new tool, geovisualization application was developed to help farmers choose the suitable variety for cultivation (Peter et al., 2020). Maps can show the most convenient places to grow certain varieties or, according to your location and parameters such as rainfall, seasonality, and geographical limit, suggest which variety would be the most suitable to grow in such conditions.

Another critical factor is the choice of sowing time. If the climate tends to be dry, the plants should be sown as early as possible so that they have time to develop deeper roots that will increase the availability of water, as well as the plants, have time to flower and produce crops before the drought sets in (Nguyen et al., 2018). In addition, it is essential to consider the temperature of the soil for sowing. For example, when sowing wheat, gram, peas, and turnips, the soil temperature may be lower (Singh and Dhaliwal, 1972) so they can be sown in early spring, while maize is better suited to warmer soil, so it is suitable for later sowing (Bollero et al., 1996). When planted at the right time, plants will grow healthier and stronger, making them more resistant to, or even avoiding, stressful conditions.

Tillage practices also affect soil hydraulic properties, such as hydraulic conductivity, which determines soil moisture retention (Azooz et al., 1996). Reducing tillage reduces the soil's total porosity and macropore volume but increases water accumulation in delicate pores. Smaller diameter pores (<7.5 μm) are classified as efficient pores that retain water in the soil and make it available to plants. However, pores with a larger diameter (>150 μm) are efficient pores where gravity allows water to drain freely and not stagnate in water (Bhattacharyya et al., 2006). According to this, a tillage method should be chosen that leads to the formation of smaller pores.

Covering the soil surface with an organic or inorganic protective layer (mulching) reduces water evaporation by up to 28% and retains 8-22% more moisture in the soil, thus supporting crop growth in unfavorable conditions associated with water scarcity (Rana et al., 2016; Nguyen et al., 2018; Sekara et al., 2019). However, a large amount of mulch is required to cover cultivated areas, which complicates the use of this tool.

Watering plants is inefficient because about 50% of the water used during irrigation is wasted (Nguyen et al., 2018). Optimizing water use is especially important when growing plants in dry environments (Hsiao et al., 2007). The most effective way is to irrigate plants at night, thereby reducing the amount of water that evaporates. Another way to save water is to modernize irrigation systems to a precise type and irrigate crops only during critical stages of growth.

Crop rotation is recognized to reduce the impact of climate change on soils and yields (Bowles et al., 2020). Crop rotation improves yields, restores soil health, and breaks weed and pathogen cycles. In addition, crop rotation increases crop resistance to drought and other stresses through improved soil properties, increased soil water capture and storage, and an abundance of beneficial soil microbes. It is important to note that leguminous plants are useful in crop rotation (Fischer et al., 2002). This is determined by the Rhizobium bacteria in their roots, which fix atmospheric nitrogen and convert it into a form available to plants, thus enriching the soil.

Crop treatment with arbuscular mycorrhizal fungi and rhizobacteria helps plants to resist various stresses. Arbuscular mycorrhizal fungi integrate diverse mechanisms of cumulative stress response tolerance in plants by regulating the activity of antioxidant enzymes, osmolytes, and phytohormone synthesis (Hashem et al., 2018). Plant growth-promoting rhizobacteria contribute to plant stress tolerance mechanisms include phosphorus solubilization, siderophore production, nitrogen fixation, organic acid, and 1-aminocyclopropane-1-carboxylic acid deaminase, chitinase, and glucanase production (Al-Garni, 2006; Egamberdieva et al., 2017; Saghafi et al., 2018).

Plant growth regulators or phytohormones such as auxin, gibberellin, abscisic acid, jasmonic acid, and salicylic acid are said to be strongly associated with plant resistance to drought, heavy metals, heat waves, flooding stresses. Plants treated with phytohormones

increase water potential and chlorophyll content, stimulate antioxidant activity, regulate oxidative stress, and reduce transpiration, thereby increasing stress tolerance. (Egamberdieva et al., 2017; Nguyen et al., 2018).

Targeted nutrient management at macro and micro levels can improve plant resistance to stress and maintain or even increase crop yields. The most critical macronutrients are phosphorus (P), calcium (Ca), and potassium (K). At the same time, the essential micronutrients are selenium (Se), silicon (Si), copper (Cu), zinc (Zn), iron (Fe), molybdenum (Mo), and boron (B). Application of P and K to drought-stressed plants increased root growth, stomatal conductance, photosynthesis, membrane stability, and leaf water potential. Additional K and Ca use in plants improves water uptake, which helps regulate stomatal permeability, osmosis, and high tissue water potential and increases resistance to temperature stress (Waraich et al., 2012). Micronutrients such as B, Mn, and Se activate plants' physiological, biochemical, and metabolic processes, thereby increasing plant stress resistance. Using Se and salicylic acid increases the activity of enzymatic antioxidants and reduces the effect of ROS on membranes. Se increases the water uptake capacity of roots under drought conditions, and the addition of Si to drought-stressed plants increased relative water content by increasing proline and glycine betaine content. Applying Si alone or in combination with K increased dry matter yield in chickpea plants exposed to drought (Ahlawat et al., 2007). The exogenous application of Si has been reported to reduce drought stress in wheat, rice, pea, and chickpea. To reduce the effects of drought, the external use of N, K, Mg, B, Zn, and Si elements can reduce the production of ROS and increase the activity of enzymatic antioxidants SOD, CAT, and PX in plant cells (Waraich et al., 2011). This results in reduced photooxidation, maintenance of chloroplast membrane integrity, and increased or maintained photosynthetic rates in crop plants, improving water use efficiency. Nutrients P, K, Mg, and Zn can improve root growth, it increases water availability, which helps regulate stomata, increases the rate of photosynthesis, and increases water use efficiency (Fageria, 2001). Applications of K, Ca, and chloride (Cl) help regulate osmosis in plant cells during stress. Using Zn and Fe increases the relative amount of water and positively affects the accumulation of proteins and trace elements in plants. B improves nodule number and mass in soybean and pea grown under drought conditions when applied through the leaves. Besides, it is important to mention another form of nutrient management related to using nano-sized elements in agronomy. The application of nanoparticles in agronomy is a relatively new field, so the following chapter is devoted to an overview of their effects and applications.

1.6 Characterization and application of nanoparticles in agriculture

Nanotechnology is an advanced field of science, the use of which prevails from standard household chemicals and cosmetics to precision agriculture, the development of medicines and medical devices, and their use in space technology for effective shielding and energy storage. According to the International Organization for Standardization, nanotechnology is the knowledge and control of 1-100 nm nanoscale objects and organisms (ISO/TC 229). NPs stand out from other bulk materials because they are composed of three layers (Khan et al., 2019; Ijaz et al., 2020). A surface layer is functionalized with metal ions, polymers, and surfactants. The shell layer is usually chemically different from the core material (Kang et al., 2018). The core is essentially the central part of the NP and usually refers to the NP itself. Nanoparticles can be divided into six groups: carbon-based NPs, metal and metal oxide NPs, ceramics NPs, semiconductor NPs, polymeric NPs, and lipid-based NPs.

The properties of nanomaterials differ not only from their bulk counterparts but also from different nanoforms of the same chemical material (Schwim et al., 2014). Therefore, different synthesis methods can create other morphological properties, sizes, shapes, and surface areas of NPs from the same chemical substance. These parameters can affect the reactivity and photocatalytic activity of the particles, which can lead to different cell penetration abilities, effects, and toxicity/benefits levels (Ridolfo et al., 2020).

Nanotechnology has been specifically applied in agricultural applications as nano - pesticides, -herbicides, -insecticides, -fungicides, -fertilizers. Specific nanoparticles (NPs) or their derivatives can be applied to increase plant productivity and protect plants from weeds, insects, and pathogens without harming soil and water (Prasad et al. 2017). The properties of NPs differ from other micro-macro elements due to size manipulation (Guo et al., 2013). NPs can be synthesized in two ways. Bottom-Up method, during which NPs are formed from relatively more straightforward materials using template support synthesis, plasma/flame laser pyrolysis, spraying synthesis, atomic or molecular condensation, and spinning (Khan et al., 2019; Wang and Xia, 2004). This method includes the biological synthesis of NPs by bacteria, yeast, fungi, plants, and algae, otherwise known as green synthesis (Ijaz et al., 2020). Top-down synthesis methods are characterized by destructiveness, starting from a larger molecule, which is decomposed into smaller units and converted into NPs. This method synthesizes NPs using mechanical melting, sputtering, laser ablation, chemical etching, and electro-explosion (Khan et al., 2019; Ijaz et al., 2020). Studying the effects of various nano-derivatives on nature is essential to avoid pollution and their possible toxic effects (Prasad et al., 2017; Pestovsky and Martínez-Antonio, 2017; Singh et al., 2021).

1.6.1 Routes of entry and movement of nanoparticles in plants

The chemical and morphological properties of NPs affect their absorption and accumulation in plant tissues. NPs can enter the plant through the roots (irrigation application) and the leaves (foliar application). In the soil, NPs can interact by directly diffusing through the root surface or be transported to the plant with the help of specific protein carriers or through microorganisms and specific surface compounds that can facilitate or hinder their absorption. Notably, the effect of NPs depends on soil type, granulometric composition, and pH (Simonin et al., 2015; Ben-Moshe et al., 2013). For NPs to reach the xylem, they must overcome the ectoderm, cortex, endoderm, and Casparian strip (Schwab et al., 2016). NPs in plants often accumulate near the root tips, especially the root cap and root hairs (Geisler-Lee et al., 2012; Sun et al., 2020). Accumulation of NPs on the root surface (mucilage) may coincide with reduced root–shoot translocation, which is often associated with the positive surface charge of NPs (Sun et al., 2020). Positively charged NPs were more likely to accumulate on the negatively charged mucilage than negatively charged NPs. Negatively charged NPs exhibit higher translocation rates due to less association with the negatively charged mucilage (Spielman-Sun et al., 2019).

The interaction of pathogenic organisms and symbiotic bacteria with roots begins in the region of cell wall elongation, where they strongly interact with plant mucus and exudates. Microorganisms living with plants create a symbiotic microenvironment, biomineralized products, and soil aggregates. The symbiotic microenvironment can affect the mobility and uptake of metal and their oxide NPs in plants (Schwab et al., 2016; Tian et al., 2019). Rhizobia, symbionts of legumes (Fabaceae) are N-fixing bacteria (*Rhizobium leguminosarum*). They create holes in the cell walls of the root hair to induce the formation of bacteria-containing nodules. These nodules help legumes survive the effects of heavy metal accumulation. For soybean and chickpea symbiotic bacteria *Bradyrhizobium japonicum* iron (Fe₃O₄) and molybdenum (Mo) NPs suspensions have a positive impact associated with stimulation of plant nodule formation, increase in their size and weight (Ghalamboran, 2011; Taran et al., 2014).

Widespread mycorrhizal fungi can form symbiotic relationships with higher plants and increase the effective surface area of plant roots up to 10 times. It also improves plant water and nutrient uptake, which may affect NPs uptake (Tian et al., 2019). In addition, mycorrhizae can protect roots from heavy metal stress due to increased glutathione levels in the hyphae. Fungi form symbiotic spores through mantles around the root tip and elongation zone, penetrating the apoplast or through arbuscular punctate penetration into cell walls (Ban et al., 2021). Overall, scientific studies with a negative effect on symbiotic bacteria have been found, but it should be

noted that the impact mostly depends on the concentration and particle sizes, so more extensive studies on this topic are needed.

After NPs enter the plant through the roots in the ways mentioned above, they can move further through the ectoderm, cortex, and endoderm to the Casparian strip. The Casparian strip is a layer of interstitial cell walls between endodermal cells sealed by lipophilic hydrocarbons such as lignin and suberin. It forms a mostly water-impermeable barrier between the cortex and the blood vessels in the apoplast. The Casparian strip is the last barrier for apoplastically traveling NPs before entering the vasculature (Lv et al., 2019; Schwab et al., 2015). Such travel around cells can take place through intracellular spaces and longitudinal channels in cell walls. NPs with properties unsuitable for translocation through the Casparian strip accumulate near it (Schwab et al., 2015). However, most NPs in plants travel symplastically directly through cells. In symplastic transport, NPs must cross the cell membrane via fluid-phase endocytic or non-endocytic pathways and move from cell to cell via plasmodesmata, sieve plates or primary pit fields (Su et al., 2019).

The rate of transpiration in plants correlates with the absorption rate of NPs. This means that if plants evaporate less water or slowly, NPs accumulation is small or occurs more slowly (Dietz and Herth, 2011). Transpiration is proportional to the amount of water entering the plant, which is often determined by the plant's leaf area. Thus, some NPs accumulate in the roots, while others move through the plant's xylem along with water and mineral substances to the plant's leaves (Su et al., 2019). NPs in leaves can both positively and negatively affect stomatal rotation, photosynthetic rate, and chlorophyll formation, induce oxidative stress and hormesis or activate antioxidant activity (Dietz and Herth, 2011).

Using the foliar spray method, NPs can form a protective surface on the leaf or enter the plant through the waxy layer and cuticle (Schwab et al., 2015; Su et al., 2019; Lv et al., 2019; Spielman-Sun et al., 2019). Wax crystals of various sizes and shapes are formed on the surface of the cuticle of plants, which increases the roughness and resistance of the leaf surface. The molecular permeability of waxes and cuticles increases as the lipophilicity and diffusion coefficient of the substance increase. The permeability can be increased by using surfactants that dissolve epicuticular waxes. Lipophilic molecules accumulate in cuticles above anticlinal cell walls and then move along pectin through the apoplast (Schwab et al., 2015).

Considering the resistance of the cuticle, it is likely that NPs do not cross it directly but enter the plant through the thinned and more permeable areas of the cuticle. Such areas on leaves can include trichomes and hydathodes. Hydathodes are small openings at the top of the leaf that allow the cuticle to absorb water and other substances (Banerjeet al., 2019; Jauneau et al., 2020). Trichomes, unicellular and multicellular formations in the epidermis, are found on most plant leaf surfaces (Levin, 1973; Hülskamp, 2004). Their main function is protective, but they also have secretory properties that make them permeable to polar substances. Stomata are most abundant on the abaxial side of the leaf. The permeability of the stomata to water and carbon dioxide is regulated by guard cells. Guard cells use osmotic pressure to open the stomata during the day and close them at night. Their size can vary from 18 to more than 300 μm^2 depending on the plant type and growing conditions, and they are more permeable to polar substances (Eckerson, 1908). For these reasons, these are potentially permeable sites for NPs deposition and uptake after foliar application. Also, wounds found in plants, a pathway for many plant pathogens (Savatin et al., 2014), can facilitate the penetration of NPs directly into the plant cytoplasm. It has been described that NPs ranging in size from 4-10 nm can cross the epidermis by disrupting the wax layer, and fluorescently labeled 50 nm NPs can accumulate in the epidermis. Also, the charge of the particles is significant for the effect, as the negatively charged plant cell walls act as an ion exchange surface, which favors the penetration of cationic NPs rather than anionic NPs (Nadmint et al., 2013).

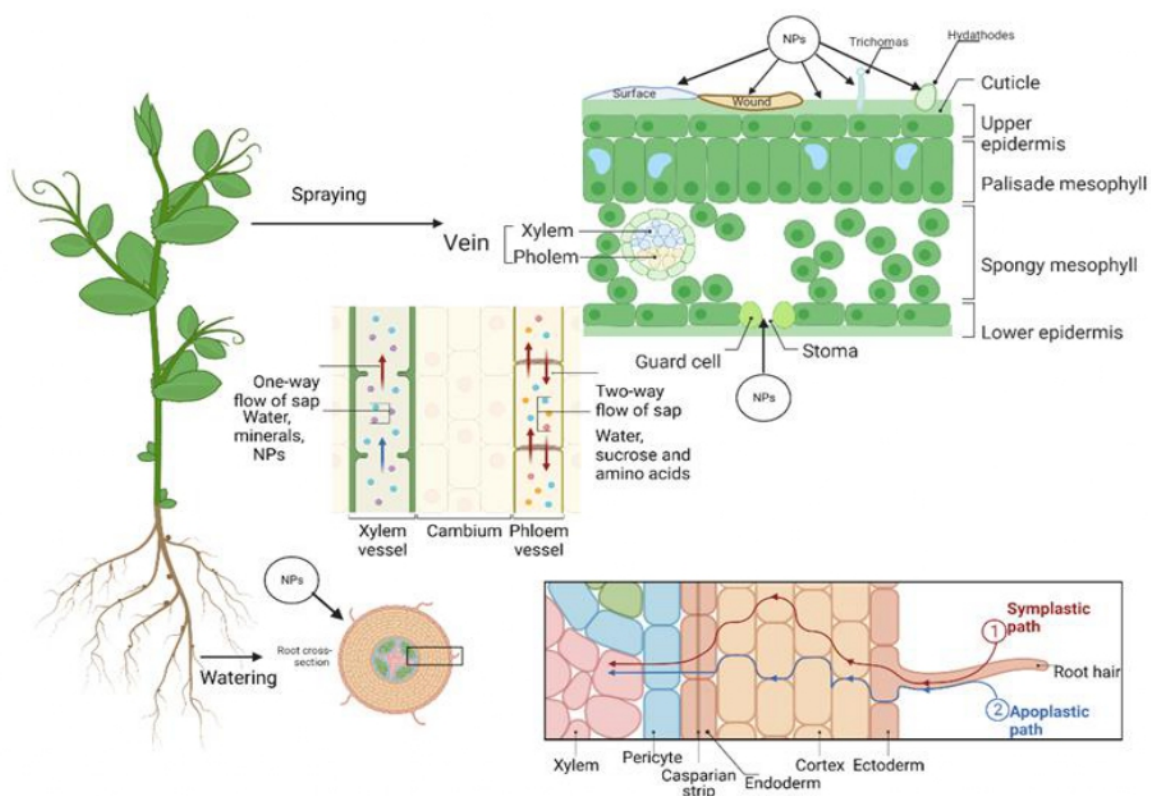


Figure 1.6.1. Routes of entry and movement of nanoparticles in plants

1.6.2 Application of nanoparticles in stress mitigation

The application of nanoparticles in agronomy to reduce the effects of stresses needs to be carefully studied. NPs can have both positive and negative impacts, unlike bulk elements. As mentioned earlier, the elements P, Ca, K, Se, Si, Cu, Zn, Fe, Mo, and B are essential for plant resistance to stressors. However, NPs synthesized from these elements may have different properties.

K NPs on pea plants have not yet been studied, especially the combination of their effects with stresses. However, a positive impact of K_2O NPs on plants growing in sandy soil was observed, as MDA and ion leakage decreased, Chl content increased significantly in leaves, and the use of these NPs can promote K^+ uptake, modulate Na^+ levels and reduce cell wall damage (Lo'ay et al., 2021). It was also reported that applying K NPs-based mulch increased the average grain yield and yield quality of maize grown under normal substrate moisture (Kandil et al., 2020).

One of the NPs with a strong effect is calcium phosphate (CaP) NPs. They were tested using plants exposed to drought, salinity, and heavy metals (Nasrallah et al., 2022). CaP NPs enhance cellular defense mechanisms, especially enzymatic antioxidants CAT, POX, and SOD, increase the total amount of phenolic compounds and increase the accumulation of soluble sugars, resulting in a reduction of oxidative stress biomarkers. Another critical compound containing the crucial elements Fe and P is $FePO_4$ NPs. After treating cucumbers and corn grown in hydroponics with $FePO_4$ NPs, an increase in the amount of P and Fe, as well as an effect on the absorption of manganese and zinc, as well as lengthening of the roots, which led to a lower shoot-to-root ratio, were found (Sega et al., 2020). However, these NPs have not yet been studied concerning stress.

Exogenous treatment of plants with CaO NPs significantly reduces Cd and As-induced growth inhibition in barley. CaO NPs stimulate the absorption of nutrients and increase the antioxidant defense capacity by stimulating the activity of enzymatic antioxidants by stimulating the expression of the genes encoding antioxidant enzymes (Nazir et al., 2022^a; Nazir et al., 2022^b).

The positive effects of silica (SiO_2) NPs have been found against the impact of heavy metals such as mercury (Li et al., 2020), cadmium (Ali et al., 2019), and chromium (Tripathi et al., 2015) on plants. It reduced the accumulation of these metals in plants, increased antioxidant activity, and reduced oxidative stress. SiO_2 NPs showed a positive effect against strains related to water deficit (Behboudi et al., 2018; Zahedi et al., 2020; Sutulienė et al., 2021) and salinity (Kalthé et al., 2018). NPs maintained or increased plant biomass, productivity, Chls, carotenoids, and polyphenols and reduced H_2O_2 and MDA levels in plants. It also had a positive effect on the activity of plant enzymatic antioxidants such as CAT, SOD, and APX and positively affected the

water content of plants. It was also found that SiO₂ effectively protects plants from cold stress by maintaining photosynthetic activity and increasing the chlorophylls and carotenoids content (Elsheery et al., 2020). Depending on the discussed publications, the sizes of the SiO₂ particles usually vary from 5-30 nm and effective concentrations from 25-1000 ppm.

Zinc (ZnO) NPs can increase proline content and activate SOD, CAT, and APX enzyme activities, promote photosynthetic pigment synthesis, correct osmoregulation, and reduce MDA and Na levels in plants that have been affected by water deficit stress (Abdel Latef et al., 2017; Faizan et al., 2021; Semida et al., 2021; Sun et al., 2021). ZnO NPs can also protect the plant from the effects of Pb, Cd, and Hg heavy metals (Venkatachalam et al., 2017; Hussain et al., 2018; Raghieb et al., 2020). Using ZnO NPs in plants exposed to heavy metals can significantly increase plant growth rate, biomass, photosynthetic pigment, and protein content, inducing the overexpression of antioxidant enzymes and initiating desirable genetic changes. In addition, ZnO also has positive effects against cold stress (Elsheery et al., 2020; Song et al., 2021), as it can restore Chl accumulation and significantly reduce chilling-induced oxidative stress by reducing H₂O₂, MDA, proline and increasing the activities of key antioxidant enzymes SOD, CAT, and POD. It should be noted that ZnO NPs can be toxic to plants. This can depend on the concentration and particle size used, the type of plant, and even the route of exposure (García-Gómez et al., 2017).

Iron (Fe) NPs are commonly used as Fe₂O₃ and FeO, and they positively affect plant resistance to drought, salinity, and heavy metals such as cadmium (Rizwan et al., 2019; Adreaset al., 2020; Manzoor et al., 2021). The use of Fe NPs in plants can increase the content of Chl and affect the accumulation of important macroelements such as N, P, K. In addition, the increasing concentration of Fe NPs decreases the concentration of electrolyte leakage, MDA and H₂O₂ in plant leaves, but increases the content of phenolic compounds and activates SOD and POD. Also, the accumulation of Fe is significantly increased by using NPs instead of the bulk structure so that it can be effectively applied in biofortification.

Copper (CuO) NPs are often toxic to plants growing under normal conditions (Tang et al., 2016; Dai et al., 2018). CuO NPs can significantly reduce the activity of complexes I and III in the mitochondrial electron transport chain, block electron transfer from NADH to ubiquinone and from ubisemiquinone to ubiquinol, and cause oxidative stress (Dai et al., 2018). However, when CuO NPs of appropriate size and concentration are applied to plants stressed by salinity and drought, can eliminate adverse effects. For example, in wheat affected by salinity stress, using CuO-NPs in different ways can compensate for the harmful effects of salinity on DNA damage and cytosine methylation (Hosseinpour et al., 2021). Using Cu NPs maintained leaf water status as well as Chl and carotenoid content under drought conditions and increased the

activity of ROS scavenging enzymes and antioxidants in maize (Van Nguyen et al., 2022). To the lack of publications, Cu NPs on plants under stress have not yet been widely studied. There is a lack of data on their effects on heavy metal stress and on plants with symbiotic relationships with bacteria, which may lead to different results.

The use of selenium (Se) NPs on drought-affected plants positively affects photosynthesis, increases metabolite accumulation, activates antioxidants, and maintains the osmotic state of cells, thus increasing growth and plant biomass (Zahedi et al., 2021). Treatment with Se NPs reduced the adverse effects of cold on sugarcane, maintained the photochemical efficiency of PSII, and increased the content of Chl and carotenoids. In addition, foliar application with Se/SiO₂ increases drought resistance in strawberries by increasing the activity of antioxidant enzymes CAT, APX, GPX, and SOD and reducing MDA and H₂O₂ levels (Zahedi et al., 2020). Treatment with Se NPs reduced the harmful effects of cold on sugarcane, maintained the photochemical efficiency of PSII, and increased the content of Chl and carotenoids (Elsheery et al., 2020). Applying chitosan-coated (Cs) Se NPs by foliar spray increased plant growth and yield by increasing the expression of photosynthetic pigments in leaves and increasing overall photosynthetic capacity in salinity-stressed plants (Sheikhalipour et al., 2021). Application of Cs–Se NPs also increased enzymatic antioxidant activity and decreased H₂O₂ content in leaves, thereby reducing oxidative damage under salinity stress conditions. Cs-Se NPs increased the expression of phenolics and flavonoids, ascorbate, and anthocyanins.

Plants need molybdenum (Mo) in small quantities, but it is crucial for symbiotic bacteria, which can improve plant's properties by acting synergistically with the plant. For example, green synthesized Mo NPs and *Bacillus* sp. were used to reduce As's harmful effects on wheat (Ahmed et al., 2022). Mo NPs induced the phyto-beneficial traits of strain ZH16, such as IAA synthesis, phosphate solubilization, and ACC deaminase activity under As stress. The combined use of bacterial strain ZH16 and Mo NPs reduced As accumulation in wheat, maintained their growth parameters, and improved nutrient accumulation. In addition, Mo NPs positively affected chickpeas and their symbiotic bacteria by promoting nodulation and increasing their number (Taran et al., 2014). Other studies have shown that Mo NPs can regulate nitrate accumulation and nitrate reductase activity in spinach (Abbasifar et al., 2020), and affect the antioxidant system in rice (Sharma et al., 2021). Still, there are no studies on their effects on water deficit stress.

Boron (B) NPs have only been studied in plants grown under normal conditions. Plants treated with B NPs showed higher Chl content, higher MDA and H₂O₂ concentrations, and more active SOD and CAT enzymes (Daglioglu et al., 20.) Besides, B NPs can also influence shoot height, flower number, and B, N, and K accumulation in soybeans (Dimkpa et al. 2017).

2. RESEARCH OBJECT, CONDITIONS, AND METHODS

Object of research

The research object is the green pea (*Pisum sativum* L.) plants. Peas were chosen as model plants because of their susceptibility to drought and their significant role in crop rotation. In addition, due to their ability to fix atmospheric nitrogen in a symbiotic association with *Rhizobium* bacteria and fulfill the nitrogen demand of the subsequent crops, they can be used for both human food and animal feed. Green pea (*Pisum sativum* L.) cultivar ‘Respect’ (Maribo Seed International ApS, Denmark) was used in all experiments. It is a medium-early semi-leafless pea variety.

Preparation of nanoparticles suspension

Silica (SiO₂) NPs (particle size: 20-30nm; purity: 99%), copper oxide (CuO) NPs (particle size: 25-55 nm; purity: 99.95%), molybdenum trioxide (MoO₃) NPs (particle size: 35-45 nm; purity: 95%), boron trioxide (B₂O₃) NPs (particle size: 100nm; purity: 99.9%) were used in the experiments (US Research Nanomaterials, Inc, Houston, TX USA). The NPs with concentrations of 12.5 ppm, 25 ppm, 50 ppm, and 100 ppm were suspended in deionized water and ultrasonically dispersed for 60 min. The NPs size and suspension stability were measured using Delsa™ Nano Submicron Particle Size (Beckman Coulter Instruments. Corporation, Fullerton, California) and Zeta Potential device (Dispersion Technology Inc., Bedford Hills, New York).

The obtained data showed that all prepared NPs suspensions were strongly anionic and stable, also monodispersed according to the polydispersity index (PDI) (Table 1). The pH value of all suspensions was equalized to 7.

Table 2.2.1. Properties of SiO₂, B₂O₃, CuO, MoO₃ NPs suspension in DI water: zeta potential, results represent the mean ± standard error, polydispersity index, and percentage of nanoparticles between 1 – 100 nm

	SiO ₂ NPs	B ₂ O ₃ NPs	CuO NPs	MoO ₃ NPs
Zeta potential (ζ; mV)	-20.64 ± 0,333	-28.54±0.223	-26.68±0.631	-24.92±0.314
The polydispersity index (PI)	0.34	0.237	0.245	0.218
NPs size 1-100 nm in suspension, %	70%	43%	54%	68%
pH value	6.6	8.4	8.6	7.8

Experimental design

A visually summarized methodology is presented in Figure 2.3. Before sowing, green pea seeds were sterilized in 5% sodium hypochlorite solution for 15 min to ensure surface sterility (Lehotai et al., 2011) and rinsed gently with deionized water several times. Then, the seeds were soaked in water for 24 h.

The following experiments were performed:

The initial (primary) experiment was performed to determine the appropriate concentrations of SiO₂, B₂O₃, CuO, and MoO₃ NPs for pea plants. The research was carried out in a walk-in controlled environment growth chamber (4 × 6 m, h = 3.2 m) at the Institute of Horticulture, Lithuanian Research Centre for Agriculture and Forestry. The three seeds were sown in each 500 mL plastic pot which was treated as a single biological replication, and 16 pots were used for each treatment. Plants were seeded and grown in peat substrate, pH 6 (Profi 1, JSC Durpeta, Lithuania) for 30 days. The substrate's average amounts of nutrients (mg L⁻¹) were as follows: N, 110; P₂O₅, 50; K₂O, 160. The average amounts of microelements (mg L⁻¹) Fe (4,5), Mn (0,5), Cu (0,1), B (0,02), Mo (0,03), and Zn (0,04), were also present. Electrical conductivity (EC) varied between 2.0 and 2.5 mS cm⁻¹ (± 0.03 mS cm⁻¹). The growth conditions were as follows: 16 h photoperiod, 20 ± 2/16 ± 2 °C day/night ambient air temperature, 60 % relative air humidity, ~220 μmol m⁻² s⁻¹ photon flux density of photosynthetically active radiation (PAR) provided by high-pressure sodium (HPS) lamp (SON-T Agro, 400 W; Philips, Somerset, NJ, USA). At the 14 BBCH growth stage, when plants unfolded their 3-4 true leaf or had 3-4 tendrils developed (14 days after sowing) (Meier, 2018), peas were watered with 100 mL of SiO₂, CuO, Mo, or B NPs suspension each containing 0.0125; 0.025; 0.05; 0.1 ppm concentrations. After that, all plants were watered daily with deionized water, and no additional fertilizer or nutrient solution was added. Randomization and regular spatial rearrangement of 10 pots per treatment were applied to minimize the potential effect of the growth chamber on plant response. Others were left unrotated to avoid border effects. At the end of the experiment, when plants reached the 31 BBCH growth stage – the beginning of stem elongation (Meier, 2018) the peas were harvested for biometric, non-destructive measurements of leaf chlorophyll and nitrogen balance (NBI) indexes and biochemical analyses. After a thorough examination and analysis of the affected plants, three concentrations of each NPs were selected for use in further experiments. These experiments aimed to select the most effective concentrations of different NPs against the effects of drought on pea plants.

The following experiments were performed over a two-year spring-summer period (2019-2020) in two greenhouses (3 × 6 m, h = 2 m) at the Lithuanian Research Centre for Agriculture

and Forestry, Institute of Horticulture, Babtai, Lithuania (55°05'08.4"N 23°48'03.5"E, at an altitude of 51 m; moderate climate zone of the northern hemisphere). For experiments, pea seeds were prepared for sowing as described above. Ten seeds were sown in 10-L volume plastic pots (7 pots per treatment, arranged randomized), and filled with ~ 8 kg of soil mixture (volume of 7:1 soil to perlite ratio, respectively). The soil was heavy loam particle size distribution, pH 7.4 ± 0.1 ; concentration of humus – $3.6 \pm 0.1\%$; P_2O_5 - 243 ± 8 mg kg^{-1} ; K_2O - 348 ± 37 mg kg^{-1} ; NH_4 – 4 ± 0.6 mg kg^{-1} ; NO_3 – 22 ± 0.9 mg kg^{-1} ; SiO_2 – 39 ± 0.8 mg kg^{-1} . Pea seedlings were thinned to 7 plants per pot five days after sowing. After 16 days of cultivation, the peas were fertilized with 7 g pot^{-1} ammonium nitrate. The peas were sprayed with fungicides because the green pea cultivar ‘Respect’ is more susceptible to powdery mildew, even when grown in a greenhouse. Pots were irrigated with water by graduated cylinder daily to 80% of substrate moisture (SM) using a substrate moisture sensor (Delta-T devices, HH2 moisture meter, Cambridge, United Kingdom) for 35 days. Plants were grown under a natural day-length photoperiod. The average day/night temperature was $22.2/14.4$ °C; relative air humidity - $58/77 \pm 5\%$ before exposure; during the ten days of drought treatment, the average day/night temperature was $25.4/16.6$ °C, and the relative air humidity was $53/75 \pm 5\%$, data were measured throughout the experiment (Termio+ data logger, Poland) at the first year.

When the peas reached the 40 BBCH growth stage (Meier, 2018), they were watered (100 ± 1 mL per pot) or foliar sprayed until full wetting (ca. 14 ± 0.5 mL $plant^{-1}$) with solutions containing different concentrations of SiO_2 , MoO_3 , CuO and B_2O_3 NPs: 0 (watered or sprayed with water, NPs-untreated), 12.5 ppm, 25 ppm, and 50 ppm. After the application of NPs, the watering of one part of the pea plants was stopped, and drought stress was initiated (30% SM). In contrast, another part of the pea and control plants was irrigated with water to maintain regular soil moisture (80% SM) throughout the experiment. These regimes were applied for ten days until harvest. After each treatment, plants were harvested after reaching the BBCH 50 growth stage (Meier, 2018) to assess their morphophysiological responses.

In subsequent experiments, we sought to determine whether NPs suspensions could protect plants from the complex effects of agrometeorological drought and copper (Cu) as heavy metal.

The conditions of the following experiment: the average day/night temperature was $24.2/14.4$ °C; relative air humidity - $54/75 \pm 5\%$ before exposure; during the ten days of drought treatment, the average day/night temperature was $26.2/17.0$ °C, and the relative air humidity was $50/73 \pm 5\%$. Soil preparation, pea sowing, and treatment with NPs were kept the same as described above. In addition, aqueous solution of 160 mg kg^{-1} $CuSO_4$ was added to half of the pots to cause Cu as heavy metal stress. When the peas reached the 40 BBCH growth stage

(Meier, 2018), they were watered (100±1 mL per pot) or foliar sprayed until full wetting (ca. 14±0.5 mL per plant) with solutions containing one concentration of each SiO₂ (50 ppm), CuO (50 ppm), MoO₃ (50 ppm), B₂O₃ (12.5 ppm) NPs selected from previous experiments. After the application of NPs, the watering of one part of the pea plants was stopped, and drought stress was initiated (30% SM). In contrast, another part of the pea and control plants was irrigated with water to maintain regular soil moisture (80% SM) throughout the experiment. These regimes were applied for ten days until harvest. After each treatment, plants were harvested after reaching the BBCH 50 growth stage (Meier, 2018) to assess their morphophysiological responses.

Morphological and non-destructive measurements

Ten plants per treatment were randomly selected (n = 10) for biometric measurements. First, the shoots were separated from the roots, and then shoot height, root length, fresh weight (FW), and dry weight (DW) were determined. FW and DW were measured with an electronic scale (Mettler Toledo AG64, Columbus, OH, USA). DW was determined following forced-air convection drying at 105°C to a constant dry weight (VENTICELL 222, MBT, Czech Republic). After shoot FW determination, ten matured plants per treatment were floated on deionized water for 24h, and turgid weights (TW) were measured. Relative water content (RWC) was calculated using the following equation (Baris and Weatherley, 1962):

$$\text{RWC, \%} = \frac{(\text{FW} - \text{DW})}{(\text{TW} - \text{DW})} \times 100 \quad (1)$$

The leaf area was measured using an automatic leaf area meter (AT Delta-T Devices, Wallingford, UK) expressed as cm² g⁻¹. For the calculation of specific leaf area (SLA), the total plant leaf area (n=10) was divided by the shoot DW. The root/shoot ratio was determined as the ratio of root DW to aboveground DW.

Pods were collected from each pea plant, and the average number of heads/pods m⁻² was counted (A, average number of pods m⁻²). Then, the number of grains in the pods in each variant was calculated, and the average (B, average number of grains per pod) was derived. The weight of 100 grains (C, weight of 100 grains of peas) was calculated. The pea yield was calculated according to the following formula (Koiter and Bill Ashton, 2021):

$$\text{The pea yield (t ha}^{-1}\text{)} = (A \times B \times C) / 10,000 . \quad (2)$$

Antioxidant properties and total phenolic compounds

Non-enzymatic antioxidant activity

Antioxidant properties of pea leaves were evaluated as the DPPH (2-diphenyl-1-picrylhydrazyl), ABTS (2,2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid)) diammonium salt, radical scavenging activities, and Fe²⁺ reducing antioxidant power assay (FRAP). Also, the total contents of phenolic compounds were determined. Extracts were prepared by grinding 0,3 g of plant leaves with liquid nitrogen and diluting them with 5 mL of 80% methanol. After 24 h, the samples were centrifuged for 10 min at 3000 rpm (Hermle Z300K, Germany), extracts were filtered through cellulose filters, and the supernatant was used for further analyses. All biochemical analysis was performed in 3 biological replications. Each of the three biological replicates consisted of at least three conjugated plants and was repeated in three analytical replicates.

The total content of phenolic compounds was determined as gallic acid equivalents. 250 µL aliquot of the sample extract was mixed with 250 µL of 10% (w/v) Folin-Ciocalteu reagent, 500 µL of 1 M Na₂CO₃ solution, and 2 mL of distilled water (Ainsworth and Gillespie, 2007). After 20 min in the dark incubation, the absorbance was measured at 765 nm (M501, Spectronic Camspec Ltd., UK). The total phenolic compounds quantity mg g⁻¹ was calculated from the calibration curve of the gallic acid (0.01 – 0.1 mg mL⁻¹, R²=0,99).

The ABTS (2,2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid)) radical cation was obtained by incubating the 7 mM ABTS stock solution (100 mL) with 2.45 mM potassium persulfate (K₂S₂O₈; final concentration) and allowing the mixture to stand in the dark at room temperature for 12–16 h before use (Re et al., 1999). Thereafter, 50 µL of the prepared sample was mixed with 2 mL of ABTS solution (ABTS stock solution was diluted 1:7), and the absorbance was measured after 11 min (plateau phase) at 734 nm (M501, Spectronic Camspec Ltd., UK). The ABTS scavenging activity of pea leaves extracts was calculated as the difference between the initial absorbance and after reacting for 10 min. A calibration curve was determined using Trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid; 97% purity; Sigma-Aldrich, USA) as an external standard with a range of concentrations from 0.1 to 0.8 mM (R² = 0.99). It was expressed as ABTS µmol scavenged per 1 g of fresh weight (µmol g⁻¹ FW).

For DPPH (2-diphenyl-1-picrylhydrazyl) assay, a stable 126,8 µM DPPH (100% purity; Sigma-Aldrich, USA) solution was prepared in methanol (Sharma and Bhat, 2009). Subsequently, 1 mL of the DPPH solution was transferred to a test tube and mixed with 100 µL of the diluted pea extract with 400 µL methanol. The absorbance was scanned at 515 nm (M501, Spectronic Camspec Ltd., UK) while reacting for 16 min. The free radical scavenging capacity was expressed as µmol of DPPH radicals scavenged per 1 g of fresh weight (µmol g⁻¹ FW). A calibration curve was determined using Trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid; 97% purity; Sigma-Aldrich, USA) as an external standard with a range of concentrations from 0.1 to 0.6 mM (R² = 0.99).

The FRAP method is based on reducing ferric ions (Fe^{3+}) to ferrous ions (Fe^{2+}). The fresh working solution was prepared by mixing 300 mM, pH 3.6 acetate buffer, 10 mM TPTZ (2,4,6-tripyridyl-s-triazine) solution in 40 mM HCl, and 20 mM $\text{FeCl}_3 \times 6\text{H}_2\text{O}$ at 10:1:1 (v/v/v) (Benzie and Strain, 1996). 20 μL of the sample was mixed with 3 mL of working solution and incubated in the dark for 30 min. Readings of the colored product (ferrous tripyridyl-triazine complex) were then taken at 593 nm. A calibration curve was determined using $\text{Fe}_2(\text{SO}_4)_3$ (Iron (III) sulfate; 97% purity; Sigma-Aldrich, USA) as an external standard with a range of concentrations from 0.005 to 0.5 mM ($R^2 = 0.99$). The antioxidant power is expressed as Fe^{2+} antioxidant capacity ($\text{Fe}^{2+} \mu\text{mol g}^{-1} \text{FW}$).

Enzymatic antioxidant activity

The extracts used to determine the activity of superoxide dismutase (SOD), catalase (CAT), ascorbate peroxidase (APX), glutathione reductase (GR), and guaiacol peroxidase (GPX) in pea leaves were prepared by grinding 0.5 g of fresh sample with liquid nitrogen and diluting within 5 mL extraction buffer (100 mM potassium-phosphate buffer, pH 7.8, containing 0.1 mM EDTA). After centrifugation for 10 min at 3000 rpm (Hermle Z300K, Germany), the supernatant was collected and used for the assays of enzymatic activities. All steps in the preparation of the enzyme extract were carried out at 4 °C.

For soluble protein determination, the dye-binding method and bovine serum albumin as standard were used to determine soluble proteins. 30 μL of enzyme extract was mixed with 1.5 mL of Bradford reagent diluted by 1:5 with DI water. Absorbance was read after 2 min through a spectrophotometer (M501, Spectronic Camspec Ltd., UK) at 595 nm (Bradford, 1976).

Total SOD activity was estimated by the inhibition of the photochemical reduction of nitroblue tetrazolium (NBT) by the enzyme (Dhindsa et al., 1981). 3 mL of reaction mixture consisted of 13 mM methionine, 75 μM NBT, 100 mM potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 50 μL enzyme extract, and 13 μM riboflavin. The tubes were under 150 $\mu\text{mol m}^{-2} \text{s}^{-1}$ for 1 min to initiate the reaction and then covered. The absorbance was recorded after 30 min by spectrophotometer (M501, Spectronic Camspec Ltd., UK) at 560 nm, and one unit of enzyme activity was taken as that amount of enzyme, which reduced the absorbance reading to 50% in comparison with tubes lacking enzyme expressed as unit $\text{mg}^{-1} \text{protein min}^{-1}$.

CAT activity was measured as the disappearance of H_2O_2 (Aebi, 1984). 100 μL enzyme extract was added in 1.275 mL of 0.1 M phosphate buffer (pH 7.8, containing 0.1 mM EDTA). The reaction started by adding 125 μL of 30 mM H_2O_2 . The decrease in absorbance measured by spectrophotometer (M501, Spectronic Camspec Ltd., UK) at 240 nm was observed for 1 min,

and enzyme activity was computed by calculating the amount of H₂O₂ decomposed ($\mu\text{mol H}_2\text{O}_2 \text{ mg}^{-1} \text{ protein min}^{-1}$).

APX activity was assayed by recording the decrease in optical density due to ascorbic acid at 290 nm (Nakano and Asada, 1981). The 1 mL assay mixture contained 0.1 M potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 0.5 mM ascorbic acid, 0.1 mL enzyme extract, and 0.1 mL of 30 mM H₂O₂ was added to initiate the reaction. The decrease in absorbance was measured spectrophotometrically (M501, Spectronic Camspec Ltd., UK) for 1 min. The extinction coefficient of $2.8 \text{ mM}^{-1} \text{ cm}^{-1}$ for reduced ascorbate was used to calculate the enzyme activity that was expressed as $\mu\text{mol AsA mg}^{-1} \text{ protein min}^{-1}$.

Measuring GR activity based on the decrease in the absorbance of oxidized glutathione (GSSG), at 340 nm (Sofa et al., 2005). The reaction mixture contained 0.1 M potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 1 mM GSSG, 100 μL enzyme extract and 75 μL 0.1 mM NADPH added last to initiate the reaction. The decrease in absorbance measured by spectrophotometer (M501, Spectronic Camspec Ltd., UK) was recorded every 5 min until 20 min. An absorption coefficient of $6.22 \text{ mM}^{-1} \text{ cm}^{-1}$ was used for calculations. GR activity was defined as $\mu\text{mol NADPH mg}^{-1} \text{ protein min}^{-1}$.

GPX activity measurements are based on the increase in the absorbance of oxidized guaiacol at 470 nm (Kvaratskhelia et al., 1997). The reaction mixture contained 0.1 M potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 31 mM guaiacol, 100 μL enzyme extract, and 75 μL 3.6 mM H₂O₂ added last to initiate the reaction. The increase in absorbance measured by spectrophotometer (M501, Spectronic Camspec Ltd., UK) was recorded for 2 min. Therefore, GPX activity was expressed as $\mu\text{mol H}_2\text{O}_2 \text{ mg}^{-1} \text{ protein min}^{-1}$.

Oxidative stress biomarkers

The extracts used to determine the concentration of lipid peroxidation and hydrogen peroxide (H₂O₂) in pea leaves were prepared by grinding 0.1 g of fresh sample with liquid nitrogen and diluting with 4 mL of 0.1% TCA (trichloroacetic acid). After centrifugation for 10 min at 3000 rpm (Hermle Z300K, Germany), the supernatant was used for further analyses.

For H₂O₂ measurements in plant leaves, 500 μL of the supernatant was added to 1 mL of 1 M potassium iodide (KI). The absorbance of the mixture was scanned at 390 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., UK). A calibration curve was determined using H₂O₂ (30% hydrogen peroxide) as an external standard with a range of concentrations from 0,6 – 24,3 mM ($R^2 = 0.99$). The content of H₂O₂ is expressed in fresh weight ($\mu\text{mol g}^{-1} \text{ FW}$) (Velikova et al., 2000).

The thiobarbituric acid (TBARS) test determines malondialdehyde (MDA) content in pea leaves samples as the end product of lipid peroxidation. 500 μL of the supernatant was added to 1 mL 0.5% (w/v) thiobarbituric acid (TBA) in 20% trichloroacetic acid (TCA). The mixture was incubated in boiling water for 30 min. The reaction stopped after the samples had cooled. The samples were centrifuged at $10\,000\times g$ for 5 min, and the absorbance of the supernatant was measured at 532 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., UK). The value for non-specific absorbance at 600 nm was subtracted (Heath and Packer, 1968). The amount of MDA–TBA complex (red pigment) in leaves was calculated and expressed as $\text{nmol g}^{-1}\text{ FW}$:

$$C_{\text{MDA}} = (A_{532} - A_{600}) / E_{\text{MDA}} \quad (3)$$

C_{MDA} – concentration of MDA, μM

A_{532} , A_{600} – Absorbance at wavelengths

E_{MDA} – MDA extinction coefficient $155\text{ mM}^{-1}\text{ cm}^{-1}$.

Determination of macro- and microelements

The macro- and microelements quantity in pea leaves, stems, and roots were determined using the microwave digestion technique combined with inductively coupled plasma optical emission spectrometry. Complete digestion of dry plant material (0.3 g) was achieved with 8 mL 65% HNO_3 using a microwave digestion system Multiwave GO (Anton Paar GmbH, Graz, Austria). The digestion program was as follows: (1) 170 $^{\circ}\text{C}$ reached within 3 min, digested for 10 min; (2) 180 $^{\circ}\text{C}$ reached within 10 min, digested for 10 min. Full-digested samples were diluted to 50 mL with deionized water. The elemental profile was analyzed by an ICP–OES spectrometer (Spectro Genesis, SPECTRO Analytical Instruments, Kleve, Germany). The operating conditions employed for ICP–OES determination were 1300 W RF power, 12 L min^{-1} plasma flow, 1 L min^{-1} auxiliary flow, 0.8 L min^{-1} nebulizer flow, and 1 mL min^{-1} sample uptake rate. The analytical wavelengths chosen were P I 213.618 nm, K I 766.491 nm, S I 182.034 nm, Ca II 445.478 nm, Mg II 279.079 nm, Fe II 259.941 nm, Zn I 213.856 nm, Mn II 259.373 nm, Cu I 324.754 nm. The calibration standards were prepared by diluting a stock multi-elemental standard solution (1000 mg L^{-1}) in 6.5% (v/v) nitric acid and by diluting stock phosphorus and standard sulfur solutions (1000 mg L^{-1}) in deionized water. The calibration curves for all the studied elements were in the range of 0.01–400 mg L^{-1} . The contents of macro and microelements in the dry weight of pea are presented (Viršilė et al., 2020).

Bio-concentration factor, translocation factor, and tolerance index calculation

Bioaccumulation is the ability of plants to absorb elements and retain them. The efficiency of this process depends on environmental conditions, and the plant type most influences the

elements' retention. Due to the complexity of the methods, the bioconcentration factor (BCF) of specific elements was calculated to assess the environmental risk that may arise from the substances under investigation. Depending on the component, a high BCF value means a low element solubility in water and a high relative octanol-water partition coefficient besides a high soil adsorption coefficient.

Therefore, BCF was calculated as an index of the pea plants' ability to accumulate the heavy metal copper Cu. It was calculated as the ratio of the Cu concentration (mg kg^{-1}) in the soil and the Cu concentration (mg kg^{-1}) in the pea plant (Zacchini et al., 2009):

$$\text{BCF} = \frac{C_{\text{shoot}}}{C_{\text{soil}}} \quad (4)$$

The translocation factor (TF) was estimated to determine the relative translocation of specific elements from roots to shoots of pea plants (Zacchini et al., 2009). TF was calculated using the following formula:

$$\text{Tf} = \frac{C_{\text{shoot}}}{C_{\text{root}}} \times 100 \quad (5)$$

C_{shoot} and C_{root} correspond to element concentrations (mg kg^{-1}) in the pea shoot and roots for which TF was calculated.

The tolerance index (Ti) was calculated to evaluate the plant's ability to grow under the influence of the heavy metal Cu (Zacchini et al., 2009):

$$\text{Ti} = \frac{\text{Dry weight of pea plants grown under copper stress}}{\text{Dry weight of pea plants grown under control conditions}} \times 100 \quad (6)$$

Statistical Analysis

All the values were expressed as mean \pm standard deviation. Data were analyzed using the Analysis of Variance (ANOVA) test followed by Tukey HSD at $p \leq 0.05$ to identify significant differences. Differences were analyzed between variants when peas were grown under normal conditions and separately between peas grown under drought conditions. The micro- and macro-elemental analysis results were also compared between control - peas grown in a drought not treated with NPs (SM 30%) and treated with NPs and between control peas grown under normal conditions (SM 80%) not treated with NPs and treated with NPs.

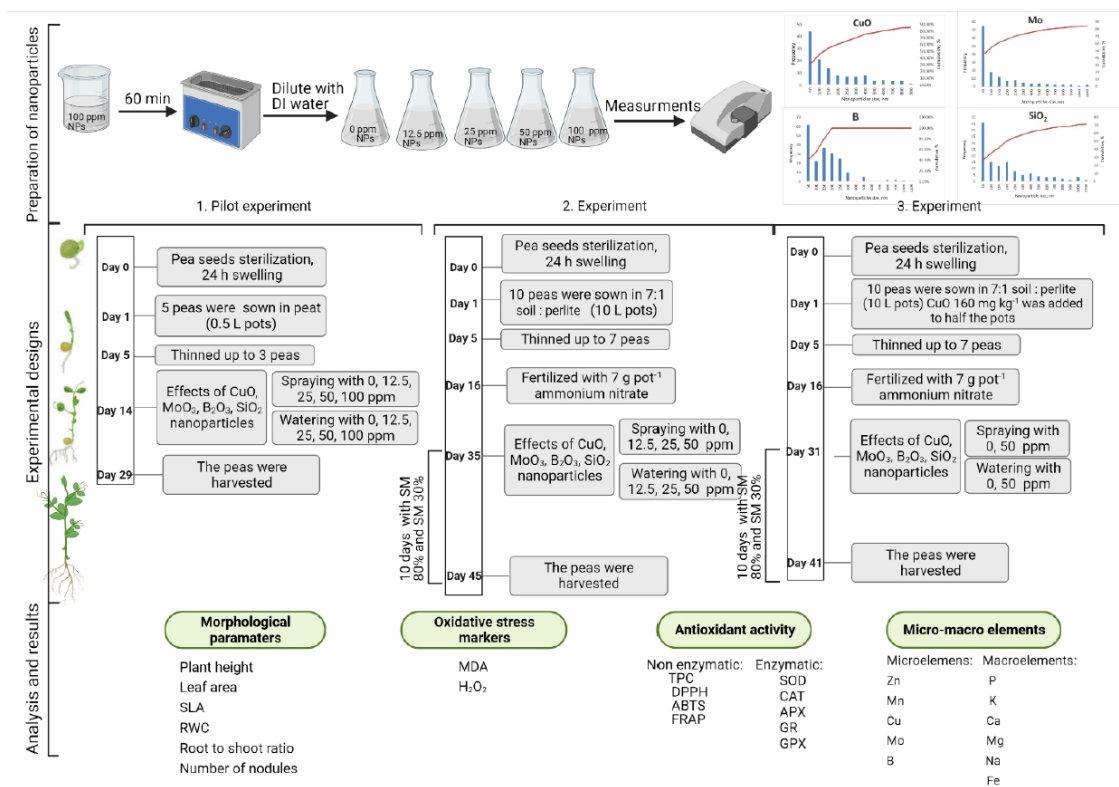


Figure 2.3.1. Methodology: preparation of nanoparticle suspensions, experiments design, analysis, and results

3. RESULTS

This section analyzes the effect of SiO₂, MoO₃, B₂O₃ and CuO nanoparticles (ND) on pea seedlings grown in optimal conditions (3.1), fully developed pea plants grown under agrometeorological drought (3.2) and complex copper and agrometeorological drought (3.3).

3.1 Effect of nanoparticles on the morphological parameters and antioxidant system in pea seedlings

The first experiment was conducted to determine how pea seedlings react to the effect of nanoparticles (NPs) of different concentrations when growing them under optimal conditions. This section analyses the effects of silica (SiO₂), copper oxide (CuO), molybdenum trioxide (MoO₃), and boron trioxide (B₂O₃) NPs on pea morphological parameters and antioxidant activity when the plants were grown in a closed climate-controlled chamber.

The results showed that using SiO₂ NPs increased peas' height by 28-45%, depending on the increasing concentration (Table 3.1.1). When peas were watered with the suspension of concentrations of 12.5, 25, 50, and 100 ppm, the average fresh weight increased by 81, 104, 116, and 109%, respectively. The dry mass increased from 62 to 135% when the concentration of SiO₂ NPs in the solution increased. Root elongation was increased by 76% after watering plants with the suspension of 100 ppm concentration, and with other solutions, it got longer, up to 106%. Root fresh and dry weight increased with higher concentrations of SiO₂ NPs. The study confirmed the positive effect of SiO₂ NPs on pea height, fresh and dry biomass, and root length at all concentrations in the solution (Table 3.1.1).

MoO₃ NPs and SiO₂ NPs had a more positive effect on peas with increasing concentration in the solution (Table 3.1.1). The higher concentration of MoO₃ NPs in the solution promoted an increase in the height of peas from 23 to 48%. In addition, an increase in shoot fresh up to 61% and dry up to 84% biomass were determined. As a result, root length was longer by 56 - 85%, and their biomass accumulated more intensively, both fresh 58 - 121% and dry 30 - 71% according to used concentrations of 12.5 – 100 ppm.

The lowest CuO NPs concentration in the solution had the most positive effect on the morphological parameters of the pea (Table 3.1.1). Pea height increased by 41, 32, and 16%, shoot fresh biomass increased by 87, 78, and 41%, and shoot dry biomass increased by 47, 50, and 35% using 12.5, 25, and 50 ppm suspensions, compared to NPs untreated plants. Root length was affected by all concentrations of CuO NPs, lengthening them by 85 - 108%. Increasing concentrations of CuO NPs in the solution caused a positive effect in fresh root biomass from 108% to 28%, but it should be emphasized that the increase in biomass slowed down with increasing concentration.

The effect of B₂O₃ NPs was closer to the development of CuO NPs - the decreasing concentration of B₂O₃ NPs had a more positive impact on the morphological parameters of peas (Table 3.1.1). Watering B₂O₃ NPs suspension at 12.5, 25, and 50 ppm of concentrations increased pea's height by 52, 21, and 2%, fresh biomass by 95, 60, and 14%, and dry biomass by 116, 35, and 52%, respectively. Root length was increased from 92 to 11%, and fresh biomass was accumulated more by 91 – 21% using the B₂O₃ NPs solutions.

Table 3.1.1. Impact of SiO₂, CuO, MoO₃, B₂O₃ NPs (12,5; 25; 50; 100 ppm) on P. sativum L. morphological parameters: plants shoot height, fresh and dry weight (FW, DW), root length, FW, and DW. 0 – control plants watered with deionized water. Mean values in columns with different letters are significantly different from the control at p < 0.05 (n=10) by Tukey (HSD) test

Nanoparticle concentration, ppm	Shoot			Root			
	Height, cm	FW, g	DW, g	Length, cm	FW, g	DW, g	
H ₂ O 0	20.0 b	1.3 b	0.240 b	4.5 b	0.365 b	0.044 b	
SiO ₂	12.5	25.7 a	2.4 a	0.388 a	9.1 a	0.624 a	0.073 a
	25	28.2 a	2.7 a	0.465 a	9.2 a	0.782 a	0.082 a
	50	28.7 a	2.9 a	0.501 a	9.2 a	1.049 a	0.089 a
	100	28.8 a	2.8 a	0.564 a	7.8 a	1.166 a	0.105 a
CuO	12.5	28.1 a	2.5 a	0.353 a	9.1 a	0.763 a	0.069 a
	25	26.3 a	2.4 a	0.361 a	8.3 a	0.668 a	0.058 b
	50	23.3 a	1.9 a	0.323 a	8.4 a	0.501 a	0.055 b
	100	20.3 b	1.2 b	0.217 c	9.3 a	0.469 a	0.037 c
MoO ₃	12.5	24.6 a	1.7 b	0.259 b	6.9 a	0.627 a	0.063 b
	25	24.7 a	1.5 b	0.304 a	7.4 a	0.577 a	0.058 b
	50	27.7 a	2.2 a	0.380 a	7.3 a	0.667 a	0.065 a
	100	29.6 a	2.5 a	0.388 a	8.3 a	0.809 a	0.076 a
B ₂ O ₃	12.5	30.3 a	2.6 a	0.518 a	8.6 a	0.696 a	0.061 a
	25	24.3 a	2.1 a	0.323 a	7.3 a	0.598 a	0.053 b
	50	20.4 b	1.5 b	0.366 a	4.9 b	0.444 a	0.044 b
	100	19.3 c	1.4 b	0.270 b	5.5 a	0.341 b	0.043 b

FW – fresh weight; DW – dry weight

All applied NPs suspensions influenced the antioxidant response in pea leaves (Figure 1). Increasing concentration of SiO₂ NPs increased total phenolic compounds (TPC) content by 10 (12.5 ppm), 22 (25 ppm), 30 (50 ppm), and 32% (100 ppm) in pea leaves (Figure 1A). The same effect was observed in the 2-diphenyl-1-picrylhydrazyl (DPPH) and 2,20-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) free radical scavenging activities (Figure 1B, C). The activity of DPPH radical scavenging increased up to 35%, corresponding to the increasing concentration of SiO₂ NPs in the solution. ABTS increased by 12%, 24%, and 25% after applying the solution containing 25 ppm, 50 ppm, and 100 ppm SiO₂ NPs, respectively.

MoO₃ NPs effectively induce higher TPC content by 29, 31, 33, and 24% after exposure to suspensions containing 12.5, 25, 50, and 100 ppm NPs (Figure 3.1.1A). The DPPH and ABTS free radicals scavenging activities showed the same tendency. The DPPH free radical scavenging activity increased intensively at 12.5 ppm - 28%, 25 ppm - 29%, and 50 ppm - 34%, but after watering with 100 ppm, the activity increased less significantly - 18% (Figure 3.1.1B). The ABTS free radical scavenging activity was also affected by increasing MoO₃ NPs concentration by 24, 31, 36, and 23%, respectively (Figure 3.1.1C). In general, a MoO₃ concentration of 100 ppm was observed to have a weaker effect on TPC accumulation and antioxidant capacity in pea leaves than lower concentrations.

TPC content in pea leaves increased by 25, 19, 18, and 15% after exposure to 12.5, 25, 50, and 100 ppm CuO NPs (Figure 3.1.1A). Also, stimulation of DPPH free radical scavenging activity by 31, 28, 23, and 21%, and ABTS free radical scavenging activity by 21, 18, 17, and 15% after exposure to CuO NPs at 12.5, 25, 50, 100 ppm were determined (Figure 3.1.1B, C). The results showed that the effect on antioxidant capacity decreased with increasing concentration of CuO NPs in the suspension.

B₂O₃ NPs concentrations of 12.5, 25, 50, and 100 ppm strongly stimulated TPC accumulation in peas, increasing it by 23, 24, 28, and 22%, respectively (Figure 3.1.1A). On the other hand, the DPPH free radical scavenging activity was weakly affected (Figure 3.1.1B). At the same time, B₂O₃ NPs concentrations of 12.5, 25, 50, and 100 ppm increased the activity of ABTS free radical scavenging by 12, 20, 30, and 22%, respectively (Figure 3.1.1C).

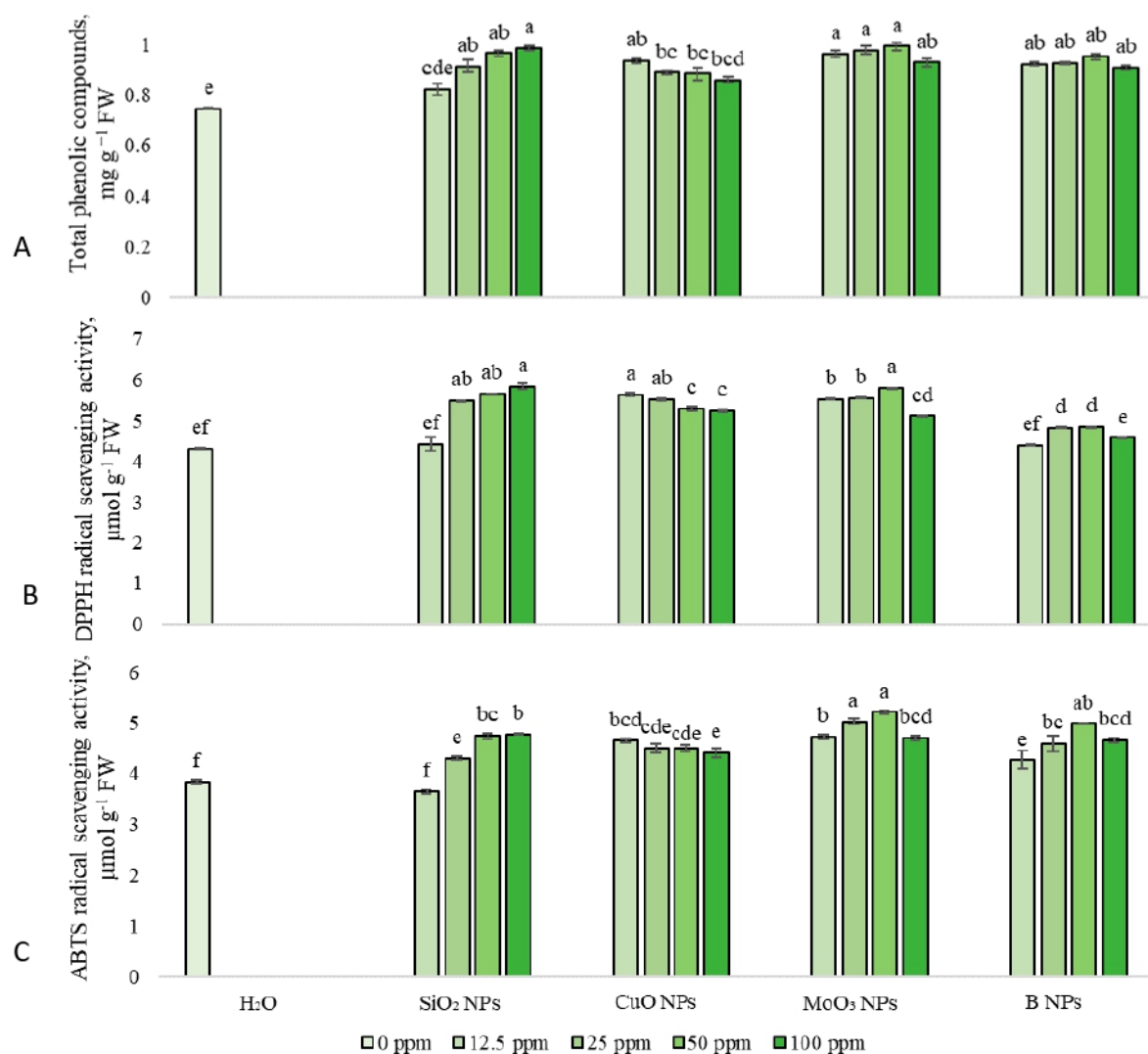


Figure 3.1.1. Influence of SiO₂, CuO, MoO₃, and B₂O₃ (B in the figure) NPs (0; 12.5; 25; 50, 100 ppm) on total phenolic compounds (A), DPPH (B) free radicals scavenging activity, and ABTS (C) free radicals scavenging activity in *P. sativum* L. H₂O – control plants watered with deionized water. Values are mean ± SE of nine replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

SiO₂ NPs were effective in reducing hydrogen peroxide (H₂O₂) and malondialdehyde (MDA) concentrations in pea leaves (Figure 3.1.2A, B). The concentration of H₂O₂ decreased up to 10% in pea leaves after treatment with 25, 50, and 100 ppm SiO₂ solutions. A slight increase in H₂O₂ (6%) was determined after irrigation with SiO₂ NPs at 12.5 ppm. MDA, which indicates lipid peroxidation, was effectively attenuated by 17, 30, 43, and 35% after irrigating peas with SiO₂ NPs solutions containing 12.5, 25, 50, and 100 ppm concentrations.

MoO₃ exposure with concentrations of 50 and 100 ppm decreased H₂O₂ content in pea leaves by 6 and 14%. Furthermore, lipid peroxidation weakened with increasing concentrations of MoO₃ NPs. This was shown by the decrease in MDA concentration by 11, and 14% depending on higher concentration.

12.5 ppm CuO NPs suspension substantially reduced H₂O₂ (17%) and MDA (30%) content in pea leaves. A slightly weaker effect was detected after the application of 25 ppm suspension. However, a significant increase in H₂O₂ content (26%) was found after irrigation with CuO NPs solution at 100 ppm.

B₂O₃ NPs at concentrations of 12.5 and 25 effectively reduced the amount of H₂O₂ by 14 and 15%. However, H₂O₂ content did not differ from the control plants after exposure to a concentration of 100 ppm B₂O₃ NPs. The amount of MDA in pea leaves significantly decreased with concentrations of 12.5 and 25 ppm B₂O₃ NPs. A significant increase in MDA content was found after applying suspension with 100 ppm B₂O₃ NPs concentration.

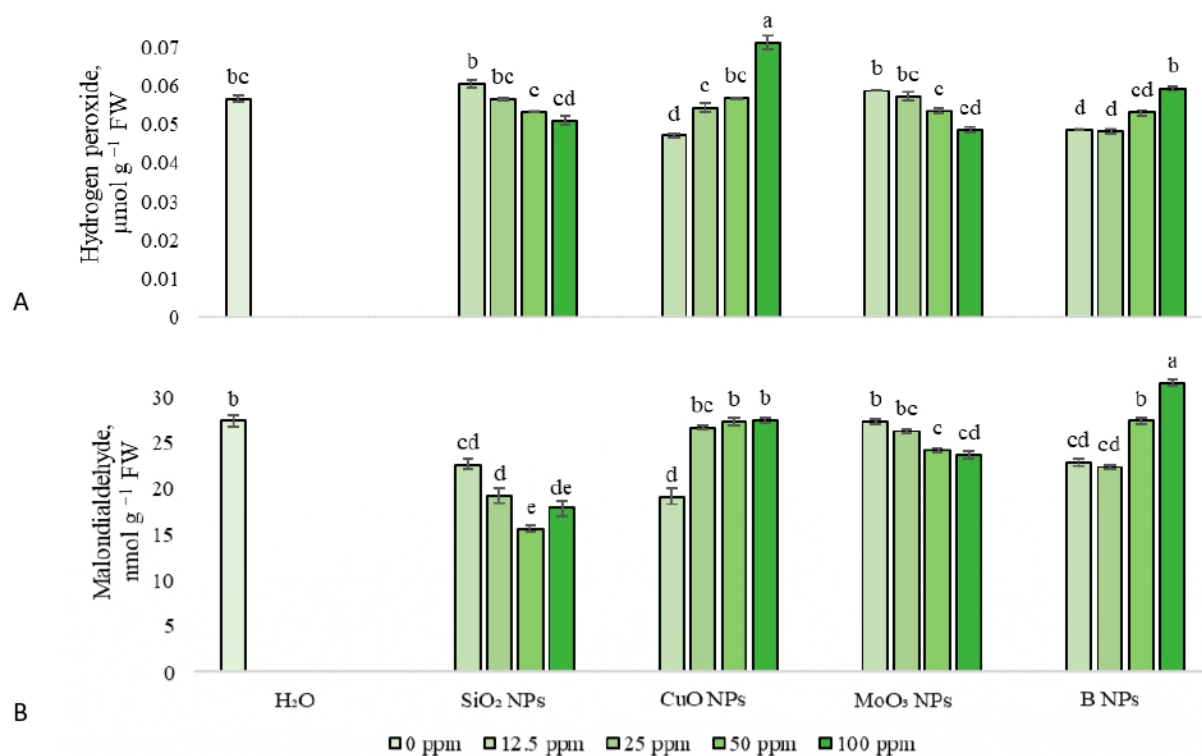


Figure 3.1.2. Influence of SiO₂, CuO, MoO₃, and B₂O₃ (B in the figure) NPs (0; 12.5; 25; 50, 100 ppm) on hydrogen peroxide (A) and malondialdehyde (B) content in *P. sativum* L. H₂O – control plants watered with deionized water. Values are mean ± SE of nine replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Summary

- *As the concentration of SiO₂ NPs increased, the morphological parameters of peas, the activity of antioxidant radicals, and TPC accumulation increased, and the concentration of H₂O₂ and MDA in peas decreased. However, no significant difference was found between 50 and 100 ppm concentrations, and it was chosen to use 12.5, 25, and 50 ppm concentrations for further studies.*

- *MoO₃ NPs showed a strong positive effect on morphological parameters with increasing concentration. However, the antioxidant response was considered: a concentration of 100 ppm was rejected due to a reduced antioxidant response.*

- *CuO NPs showed that using a concentration of 100 ppm is useless or even harmful to peas, so it was chosen to use concentrations of 12.5, 25, and 50 ppm for further experiments.*

- *B₂O₃ NPs concentration of 100 ppm showed an increase in H₂O₂ and MDA content in pea leaves and a decrease in pea morphological parameters, so it was rejected and not used in further studies.*

3.2 Effects of nanoparticles on drought-stressed pea plants

This section discusses the effects of NPs on drought-stressed peas grown at 30% substrate moisture (SM 30%) compared to plants grown at regular substrate moisture (SM 80%). The influence of NPs on morphological parameters, antioxidant system activity, and macro- and microelement accumulation in peas is discussed in detail. The different effects of NPs on peas were highlighted by applying two methods of exposure: foliar spraying and watering through the roots. This section discusses the results obtained between plants affected by drought and plants affected by drought and NPs. The subsections are divided according to the NPs, at the first paragraph were discussed the watered NPs effects on plants, and the second one is the foliar NPs application through the leaves.

3.2.1 Effects of silica nanoparticles on peas under different substrate moisture

Impact on morphological parameters

Irrigation of peas with 12.5, 25, and 50 ppm SiO₂ NPs suspensions significantly affected pea height (Table 3.2.1.1, 80% SM) when substrate moisture was sufficient. Also, plants irrigation with 50 ppm suspension stimulated leaf area development effectively. In addition, spraying with 12.5, 25, and 50 ppm SiO₂ NPs suspension increased leaf area by up to 15%. Relative water content (RWC) increased by 6% when plants were irrigated with a concentration of 50 ppm suspension. The ratio of root to shoot increased by 7, 34, and 68% when watering with suspensions of 12.5, 25, and 50 ppm, respectively. However, the influence of SiO₂ NPs decreased by 70, 23, and 4% after spraying with solutions containing 12.5, 25, and 50 ppm NPs.

No differences were found in specific leaf area (SLA) after watering peas with SiO₂ NPs, while spraying with 12.5 ppm reduced SLA by 13%. A positive influence was found on nodulation. The number of nodules increased to 5 times when peas were watered with SiO₂ NPs solutions. After spraying with SiO₂ NPs solutions, the number of nodules in pea roots increased slightly to 1.4 times. A 41% increase in pea yield was found after 50 ppm SiO₂ NPs suspension was applied to plants.

SiO₂ NPs showed a significant positive effect on pea plants as they grow in the substrate with insufficient moisture content (Table 3.2.1.1, 30% SM). Pea height increased by 23, 20, and 18% after irrigation and by 8, 11, and 24% after spraying with solutions containing 12.5, 25, and 50 ppm SiO₂ NPs. Leaf area increased by 11 and 13% after peas irrigation with 25 and 50 ppm and by 12, 18, and 10% after foliar application with 12.5, 25, and 50 ppm SiO₂ NPs solutions. RWC increased by 7% and 10% after irrigation of peas with 12.5 and 50 ppm SiO₂ NPs suspension when plants were grown under 30% substrate moisture. There was a substantial increase in root-to-shoot ratio (up to 30%) after irrigating peas with SiO₂ NPs solutions or spraying with 50 ppm concentration. However, the root-to-shoot ratio decreased after peas were sprayed with 12.5 and 25 ppm SiO₂ NPs solutions. The SiO₂ NPs did not affect SLA of peas grown under drought conditions, except for plants sprayed with 25 ppm solution. The application of SiO₂ NPs induced the formation of nodules in pea roots. This is shown by the 30% increase in nodules number after watering with 50 ppm suspension and 33%, 50%, and 180% increase after spraying with 12.5, 25, and 50 ppm solutions, respectively. A significant increase in the pea yield (up to 43%) of drought-affected peas was found after the plants were irrigated with 12.5, 25, and 50 ppm SiO₂ NPs suspensions. A positive effect of 40, 35, and 45% on yield was found after peas were sprayed with the suspensions containing mentioned concentrations.

SiO₂ NPs slightly increased H₂O₂ concentration in pea leaves, 12.5 ppm - 25%, 25 ppm - 20%, 50 ppm - 12% when watered and 12.5 ppm - 28% when peas were sprayed and grown under 80% substrate moisture (Figure 3.2.1.1A). Knowing that plants need peroxide in small amounts, it can be noted that such an increase is optimal for plants and does not have a harmful effect. The concentration of malondialdehyde (MDA) in the plant indicates lipid peroxidation. MDA concentration in pea leaves was affected by SiO₂ NPs (Figure 3.2.1.1B, 80% SM). After watering with suspensions containing 12.5 and 25 ppm of SiO₂ NPs, the MDA content in the plant increased significantly. Pea exposure by watering or spraying with 50 ppm of SiO₂ NPs suspension significantly reduced the MDA content in pea leaves.

Table 3.2.1.1. Impact of drought stress and SiO₂ NPs (12.5; 25; 50 ppm) on P. sativum L. height, leaf area, specific leaf area (SLA), relative water content (RWC), root-to-shoot ratio, and the number of nodules.

0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%.

Mean values within columns followed by different letters differ significantly at $p < 0.05$ ($n=10$)

according to Tukey (HSD) test

	SiO ₂ NPs, ppm	Plants height, cm	Leaf area, cm ²	SLA, m ² kg ⁻¹	RWC, %	Root/shoot ratio	Nr. of nodules	Yield, t ha ⁻¹	
SM 80%	Watered	0	28.4 d	36.1 c	5.3 ab	82.5 b	7.8 c	1.7 c	3.9 b
		12.5	30.1 bc	36.3 c	4.7 ab	83.6 ab	8.3 b	5.0 ab	3.8 b
		25	32.5 b	37.6 b	5.1 ab	85.7 ab	10.4 ab	6.7 ab	3.7 b
	Sprayed	50	34.4 a	47.2 a	4.8 ab	87.6 a	13.0 a	9.3 a	5.5 a
		12.5	29.4 cd	38.8 b	4.6 c	84.6 ab	13.2 a	3.3 bc	3.4 bc
		25	29.3 cd	39.3 ab	5.6 a	83.8 ab	9.5 ab	4.0 b	4.0 b
SM 30%	Watered	50	29.2 c	41.5 ab	5.1 ab	84.8 ab	8.1 b	3.3 bc	5.4 a
		0	26.0 e	33.1 d	5.0 b	53.0 d	9.2 b	2.0 b	2.5 c
		12.5	32.0 d	33.2 d	3.8 bc	56.9 ab	11.7 a	2.0 b	3.3 ab
	Sprayed	25	31.2 a	36.6 bc	4.0 bc	56.6 ab	10.5 a	2.0 b	3.2 ab
		50	30.7 bc	37.3 b	4.8 b	53.9 d	11.9 a	2.7 b	4.5 a
		12.5	28.1 cd	37.0 b	5.1 ab	54.7 c	8.9 bc	2.7 b	3.4 ab
Sprayed	25	28.9 ab	39.0 ab	6.2 a	55.4 bc	8.5 c	3.0 b	3.2 ab	
	50	30.7 cd	43.1 a	4.9 b	58.9 a	9.2 b	2.0 b	4.6 a	

Effects on oxidative stress biomarkers

Focusing on the effects of drought on the concentration of H₂O₂ in the pea, it has increased to a concentration that is dangerous to the plant, causing an imbalance in the cells. In pea leaves affected by drought, the concentration of H₂O₂ increased 2.5 times compared to plants grown in normal substrate moisture (Figure 3.2.1.1A, 30% SM). Watering with 12.5, 25, and 50 ppm SiO₂ NPs solutions positively affected plants by reducing H₂O₂ concentration by 15, 25, and 27%, while spraying with 25 and 50 ppm SiO₂ NPs solutions reduced H₂O₂ by 13 and 21%, respectively. An increase in MDA concentration of over 26% was found when plants were exposed to drought compared with peas grown under 80% SM (Figure 3.2.1.1B). In the results, when peas were watered or sprayed with SiO₂ NPs solutions, the amount of MDA in plants decreased significantly compared to untreated NPs.

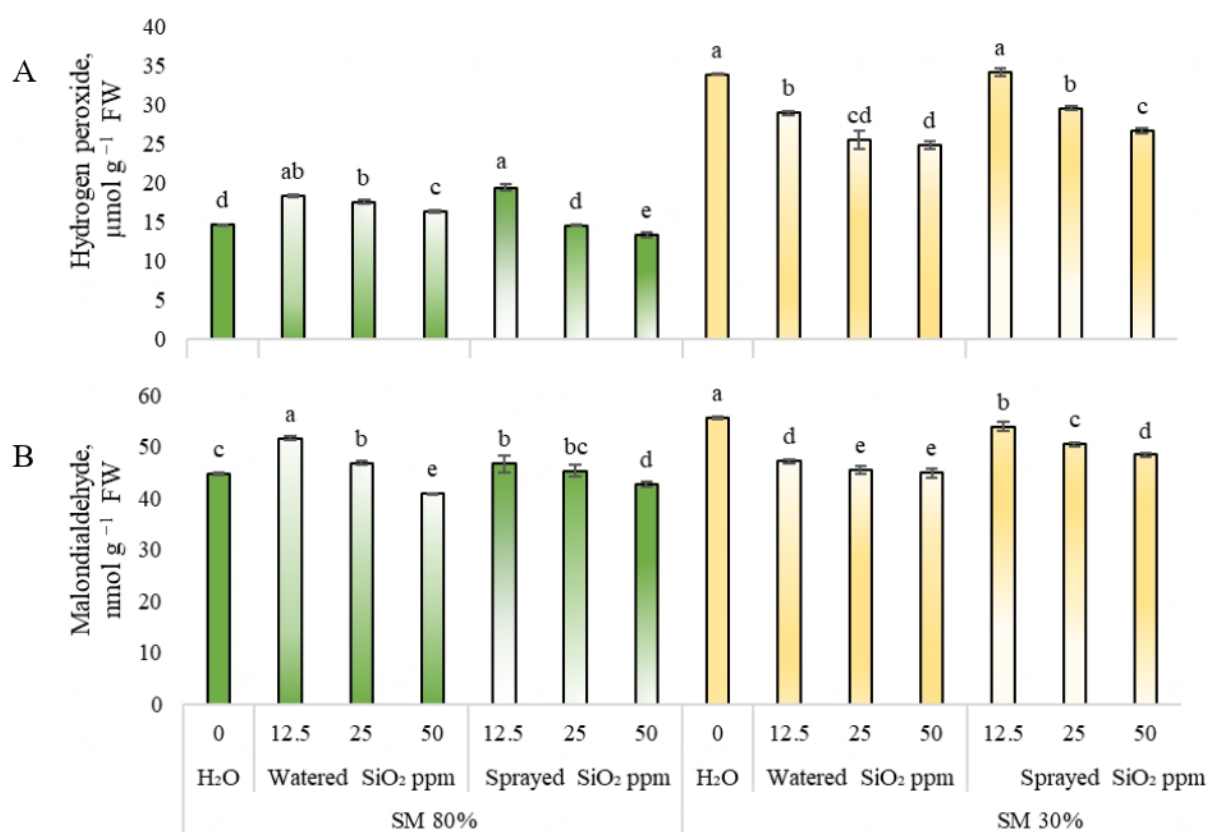


Figure 3.2.1.1. Influence of drought stress and SiO₂ NPs (0; 12,5; 25; 50 ppm) on hydrogen peroxide (A) and malondialdehyde (B) content in *P. sativum* L. H₂O – control plants sprayed or watered, with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of nine replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Effects on non-enzymatic antioxidants

The inhibitory effect of SiO₂ NPs on the antioxidant system of peas grown in 80% substrate moisture was determined (Figure 3.2.1.2, 80% SM). The TPC content decreased by 17% when the plants were watered with the suspension at 12.5 ppm concentration, and after spraying with suspensions containing 12.5, 25, and 50 ppm of SiO₂ NPs decreased it by 21, 23, and 12%, respectively. (Figure 3.2.1.2A). A statistically reliable effect in FRAP antioxidant power was caused by irrigation and foliar application of SiO₂ NPs suspensions (Figure 3.2.1.2D). The impact on the DPPH free radical scavenging activity was multifaceted. The activity of the DPPH radical scavenging decreased by 13% after treatment through roots with 25 ppm suspension, but after foliar application of 12.5 ppm suspension, it increased by 25% (Figure 3.2.1.2B). ABTS free radical scavenging activity increased significantly after spraying peas with the suspension of 50 ppm (Figure 3.2.2C).

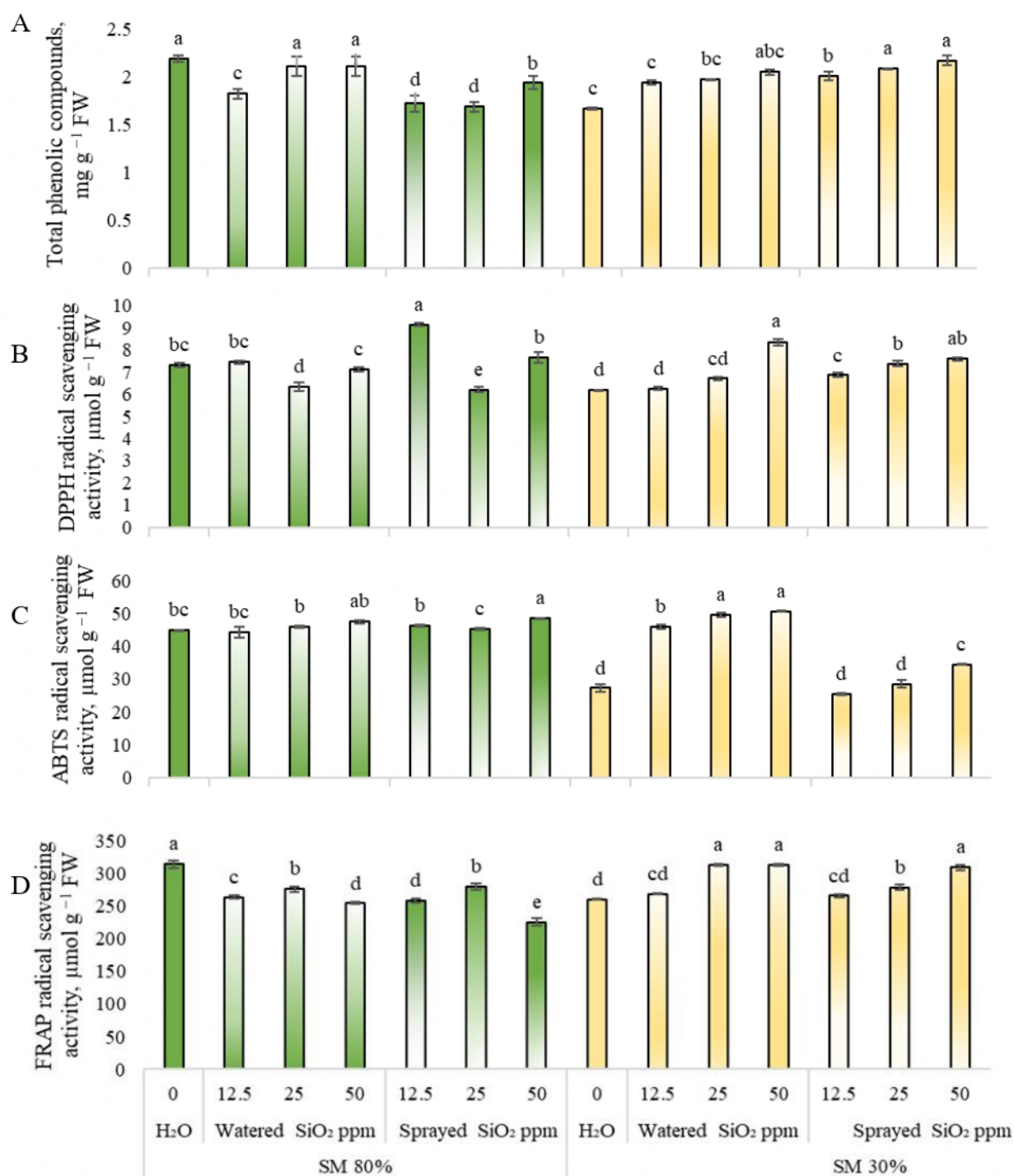


Figure 3.2.1.2. Influence of drought stress and SiO₂ NPs (0; 12,5; 25; 50 ppm) on A – total phenolic compounds (TPC), B – DPPH free radical scavenging activity, C – ABTS free radical scavenging activity D – FRAP antioxidant power in *P. sativum* L. H₂O – control plants, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Opposite results were found when studying the effects of SiO₂ NPs on pea plants grown in the substrate with reduced moisture content (Figure 3.2.1.2, 30% SM). Significant increases of 16, 18, and 23% in TPC content were found in pea leaves after irrigation with increasing

concentrations of SiO₂ NPs, and 20, 25, and 30% increases in TPC after spraying with 12.5, 25, and 50 ppm suspensions (Figure 3.2.1.2A). The results showed that irrigation with 50 ppm suspension of SiO₂ NPs DPPH free radical scavenging activity increased by 35%; after spraying with 12.5, 25, and 50 ppm suspensions - by 11, 19, and 23%, respectively (Figure 3.2.1.2B). FRAP antioxidant power increased to 20% after irrigation with 25 and 50 ppm SiO₂ NPs suspensions and up to 19% after spraying drought-affected peas (Figure 3.2.1.2D). SiO₂ NPs and drought strongly affected ABTS free radical scavenging activity (Figure 3.2.1.2C). The ABTS free radical scavenging activity increased by 69, 82, and 86% after using SiO₂ NPs through roots at 12.5, 25, and 50 ppm concentrations. However, the effect was different during spraying because only 50 ppm concentration of suspension increased it up to 26%.

Effects on enzymatic antioxidants

SiO₂ NPs stimulated the CAT activity after spraying peas grown under sufficient substrate moisture with 12.5 ppm by 93% and 25 ppm by 89% (Figure 3.2.1.3B, 80% SM). An inhibitory effect was found for the APX and GR enzymes (Figure 3.2.1.3A, D 80% SM). APX activity was inhibited after watering or spraying 50 ppm suspension by 32% and 38%. GR activity inhibition by 13% at a concentration of 50 ppm SiO₂ NPs applied through the roots and after spraying with 50 ppm by 19% were found. The activities of GPX and SOD were strongly affected by SiO₂ NPs (Figure 3.2.1.3E, C 80% SM). GPX activity increased 9, 8, and 5 times after irrigating peas with suspensions of 12.5, 25, and 50 ppm SiO₂ NPs, and after spraying, all used concentrations increased GPX activity by 9 times. SOD activity increased after 12.5, 25, and 50 ppm SiO₂ NPS application: by watering 49, 44, and 47%, and by spraying 42, 47, and 48%.

CAT activity in pea leaves after exposure to drought and 12.5, 25, and 50 ppm SiO₂ NPs increased by 36, 84, and 153% after watering and by 13, 41, and 119% after spraying (Figure 3.2.1.3B 30% SM). Increasing SiO₂ NPs concentration increased SOD activity after watering up to 37% and after spraying up to 23% (Figure 3.2.1.3C 30% SM) in drought-affected pea plants. GR activity was affected by 25 and 50 ppm SiO₂ NPs increasing the activity by 21 and 100% after watering. Furthermore, an even more substantial effect of up to 128% on GR activity was found after spraying with SiO₂ NPs drought-affected peas (Figure 3.2.1.3D 30% SM). Notably, decreases in APX and GPX activities were found (Figure 3.2.1.3A, E 30% SM). APX activity decreased by 37 and 16% after watering with 12.5 and 25 ppm SiO₂ NPs, with an even more substantial negative effect of up to 37% after spraying. GPX activity was inhibited after watering drought-affected peas with 12.5 ppm SiO₂ NPs by 19% while spraying inhibited the activity by up to 30%. A slight GPX activity increase was found only after irrigating peas with a concentration of 50 ppm SiO₂ NPs.

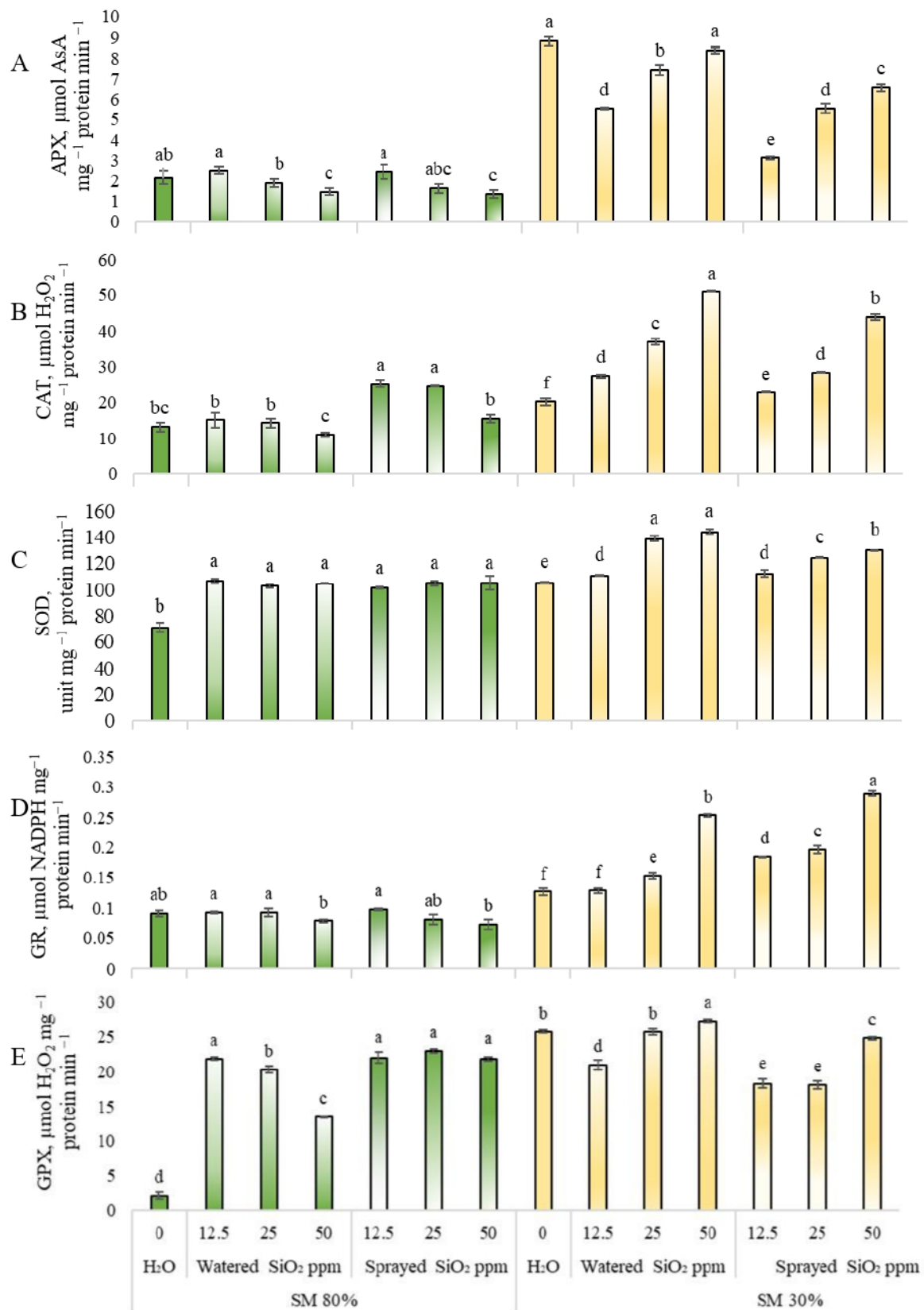


Figure 3.2.1.3. Response of A – ascorbate peroxidase (APX), B – catalase (CAT), C – superoxide dismutase (SOD), D – glutathione reductase (GR), and E – guaiacol peroxidase (GPX) activity to drought stress and SiO₂ NPs (0; 12.5; 25; 50 ppm) in *P. sativum* L. H₂O – control plants sprayed or watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean \pm SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Changes in macro- and microelement composition

The effects of SiO₂ NPs suspension on macro- and microelements in pea's leave, stems, and roots are presented in Table 3.2.1.2. The results showed that drought reduces P, Mg, Na, Mn, Mo, and B but increases the accumulation of K, Ca, Fe, and Zn in pea plant leaves. Spraying at a concentration of 50 ppm when plants were grown with 30% SM increased the accumulation of K, Ca, Mg, Fe, Mn, and Mo in the pea leaves. SiO₂ NPs solution of 50 ppm had an effect against drought by increasing the amount of all micro- and macroelements in the plant stems except for Na, Fe, and B elements. In addition, the SiO₂ suspension strongly affected the accumulation of all elements in the root, significantly increasing the amount of P, K, Mg, Fe, Zn, Mn, Cu, and Mo.

Table 3.2.1.2. Effects of SiO₂ NPs suspension on macro- and microelements of peas leaves, stem, and root. Mean values in bold indicate a statistically significant difference at $p < 0.05$ according to Tukey (HSD) test ($n=9$)

Treatment SiO ₂ NPs, ppm		Macroelements, mg g ⁻¹ DW							Microelements, µg g ⁻¹ DW				
Leaves		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B	
SM 80%	Watered	0	3.79	18.76	45.73	9.20	2.64	19.86	25.71	42.54	10.32	2.99	33.20
		12.5	3.60	21.27	48.76	9.99	3.41	23.50	33.10	48.30	11.03	2.40	16.51
		25	3.00	14.53	35.74	6.95	1.44	19.78	23.29	30.89	7.78	1.58	15.48
	Sprayed	50	4.91	23.63	46.18	9.32	1.98	20.76	34.70	40.59	9.65	1.60	15.82
		12.5	3.26	19.07	40.76	8.99	3.30	13.55	30.12	43.44	10.23	2.51	17.51
		25	3.41	17.64	43.27	8.55	1.71	15.94	29.23	44.82	9.78	2.46	16.70
SM 30%	Watered	50	4.41	23.63	57.59	11.98	4.15	25.15	42.63	53.93	11.50	2.19	16.31
		0	3.21	20.56	46.19	8.81	2.07	20.03	37.24	37.65	10.02	0.70	11.66
		12.5	3.75	23.65	48.93	9.57	1.65	21.86	37.12	44.00	8.82	1.97	4.20
	Sprayed	25	3.53	22.58	44.91	8.71	1.59	15.38	35.05	38.94	7.59	3.11	9.83
		50	3.01	21.22	42.15	8.16	1.27	18.99	39.01	41.42	7.51	2.82	11.32
		12.5	2.13	24.26	40.72	8.90	5.65	14.47	29.72	27.48	12.08	3.06	3.70
SM 80%	Watered	25	3.08	19.10	43.75	8.26	1.76	14.71	31.78	38.78	7.31	0.38	9.77
		50	4.11	24.69	47.76	9.82	1.82	22.45	35.19	47.62	8.28	1.52	11.75
		0	2.00	17.77	37.81	8.71	7.73	15.49	19.81	26.09	10.08	3.66	1.05
	Sprayed	12.5	1.62	19.78	41.52	8.34	7.82	14.24	26.24	25.42	12.89	2.74	9.28
		25	1.52	15.76	29.50	6.26	5.02	14.60	18.09	18.80	8.46	3.08	4.26
		50	2.26	20.86	42.90	8.31	6.84	17.10	24.35	27.08	12.05	2.63	4.61
SM 30%	Watered	12.5	1.65	17.83	42.06	8.39	7.48	16.60	25.91	29.11	13.03	4.56	8.08
		25	1.49	16.23	36.17	7.36	5.53	12.82	20.52	24.56	9.72	3.38	7.50
		50	2.11	20.71	47.87	9.89	8.73	16.39	25.88	30.40	16.20	5.85	9.87
	Watered	0	1.43	20.13	38.10	7.98	6.83	15.46	22.31	24.19	8.92	1.17	10.57
		12.5	1.89	24.25	43.95	8.53	6.22	19.28	27.43	30.08	12.17	2.55	11.11
		25	1.87	24.75	37.25	7.39	5.33	12.55	20.98	24.52	10.88	1.38	7.94
50	1.74	21.73	37.61	7.06	4.02	13.37	30.57	26.51	9.69	1.33	6.88		

	Sprayed	12.5	2.13	24.26	40.72	8.90	5.65	14.47	29.72	27.48	12.08	3.06	3.70
		25	1.54	20.16	36.91	7.49	5.58	13.67	24.81	23.53	10.96	2.39	9.83
		50	2.18	22.81	42.93	8.52	5.43	13.88	27.44	30.93	13.21	3.86	7.59
	Root		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B
SM 80%	Watered	0	0.81	16.16	10.25	6.91	3.18	258.88	15.88	46.44	18.46	9.40	57.91
		12.5	2.96	37.36	22.85	18.47	8.42	797.28	45.21	149.63	36.13	25.06	27.27
		25	2.72	42.93	23.14	18.55	7.94	758.35	43.47	144.16	37.75	22.71	16.89
	Sprayed	50	3.44	48.48	31.12	21.51	11.42	895.74	56.39	164.75	42.82	27.62	26.67
		12.5	2.78	34.62	24.21	15.82	11.30	667.76	51.21	129.32	38.21	21.71	21.91
		25	5.07	65.36	44.07	31.90	14.62	1591.93	120.9	320.39	65.65	48.16	27.04
		50	3.23	37.08	24.69	19.56	7.49	1138.92	53.11	217.61	42.07	31.86	24.55
SM 30%	Watered	0	1.58	28.33	19.57	15.45	7.33	657.85	34.15	118.04	22.49	16.58	9.84
		12.5	2.22	31.42	20.43	15.60	7.73	774.18	48.50	146.08	31.38	22.69	7.04
		25	2.80	47.33	24.74	18.75	9.86	906.39	46.80	169.12	40.77	24.70	10.07
	Sprayed	50	2.92	42.44	31.10	22.50	12.41	1353.64	68.41	249.53	45.04	36.94	8.78
		12.5	2.48	37.92	23.92	17.01	11.60	960.38	64.23	166.31	38.08	25.69	9.00
		25	3.94	59.53	42.14	31.60	15.60	1321.75	70.93	244.31	61.44	41.37	4.17
		50	3.16	46.26	32.11	22.40	12.75	1398.81	57.86	248.60	47.33	38.15	2.13

Macroelements: P – phosphorus, K – potassium, Ca – calcium, Mg – magnesium, Na – sodium, Fe – iron; Microelements: Zn – zinc, Mn – manganese, Cu – copper, Mo – molybdenum, B – boron. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%.

Table 3.2.1.3. The impact of SiO₂ NPs (12,5; 25; 50 ppm) on *P. sativum* L. grown in the substrate with sufficient (SM 80%) and insufficient (SM 30%) moisture is expressed as a percentage change (%) compared to the control (for SM 80% control means plants grown under SM 80% and NPs untreated; SM 30% control means drought affected but NPs untreated plants) in the heat map. Statistically, significant differences are marked in bold

Treatment SiO ₂ NPs, ppm	SM 80%						SM 30%					
	Watered			Sprayed			Watered			Sprayed		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Plant height	6	15	21	4	3	3	23	20	18	8	11	24
Leaf area	1	4	31	8	9	15	0	11	13	12	18	10
Nodules	200	300	460	100	140	100	0	0	33	33	50	183
RWC	1	4	6	3	2	3	7	7	2	3	4	10
Root/shoot	7	34	68	70	23	4	27	14	30	-3	-8	16
SLA	-11	-3	-9	-13	6	-3	-24	-20	-4	2	23	-25
Yield	-2	-5	40	-14	4	38	37	37	44	40	35	45
ABTS	-1	2	6	3	1	8	69	82	86	-7	5	26
DPPH	2	-13	-3	25	-15	5	1	9	35	11	19	23
TPC	-17	-4	-4	-21	-23	-12	16	18	23	20	25	30
FRAP	-16	-12	-19	-18	-11	-28	3	20	20	2	7	19
HP	26	20	12	78	0	-9	-15	-25	-27	1	-13	-21
MDA	15	5	-9	4	1	-5	-15	-18	-19	-3	-9	-13
GR	2	2	-13	7	-11	-19	2	21	100	46	55	128
GPX	914	841	525	919	965	912	-19	0	6	-29	-30	-4
APX	16	-12	-32	13	-24	-38	-37	-16	-5	-65	-37	-25
SOD	6	15	21	4	3	3	23	20	18	8	11	24
CAT	1	4	31	8	9	15	0	11	13	12	18	10

RWC – relative water content, SLA – specific leaf area, TPC – total phenolic compounds, HP – hydrogen peroxide, MDA – malondialdehyde, GR – glutathione reductase, GPX – guaiacol peroxidase, APX – ascorbate peroxidase, SOD – superoxide dismutase, CAT – catalase. 0 – control plants watered with deionized water, drought stress – 30% substrate moisture (SM 30%).

Summary (Table 3.2.1.3)

- SiO₂ NPs protect peas from drought by strengthening the antioxidant system and activating enzyme-catalyzed reactions.
- SiO₂ suspension reduced the concentration of H₂O₂ and MDA in the pea plant, thus reducing the oxidative stress caused by drought.
- The selected SiO₂ concentration of 50 ppm is the most effective for the pea plant against drought-induced stress.
- No significant difference was found between applying SiO₂ NPs on pea plants by foliar spray or root irrigation.

3.2.2 Effects of copper oxide nanoparticle on peas under different substrate moisture

Impact on pea morphological parameters

CuO NPs had different effects on pea plants grown at 80% SM. The results in Table 3.2.2.1 show that plant height, SLA, and root-to-shoot ratio were not affected by CuO NPs (80% SM). Leaf area decreased by 13% after watering with suspensions of 25 or 50 ppm CuO NPs. Spraying peas with 12.5 and 25 ppm suspensions of CuO NPs slightly decreased the leaf area. RWC was reduced when peas were treated with 50 ppm CuO NPs by watering or spraying. The effect of CuO NPs on pea yield was not significant.

*Table 3.2.2.1. Impact of drought stress and CuO NPs (12.5; 25; 50 ppm) on *P. sativum* L. height, leaf area, specific leaf area (SLA), relative water content (RWC), root to shoot ratio, and the number of nodules. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Mean values within columns followed by different letters differ significantly at $p < 0.05$ ($n=10$) according to Tukey (HSD) test*

	CuO NPs, ppm	Plants height, cm	Leaf area, cm ²	SLA, m ² kg ⁻¹	RWC, %	Root/shoot ratio	Number of nodules	Yield, t ha ⁻¹	
SM 80%	Watered	0	28.4 a	36.1 ab	5.3 a	82.5 b	7.8 a	1.7 c	3.9 a
		12.5	27.8 a	33.4 bc	3.9 ab	83.9 ab	9.7 a	2.0 bc	3.7 a
		25	28.5 a	31.8 c	4.0 ab	84.1 ab	5.5 a	1.0 c	3.8 a
	Sprayed	50	27.3 a	31.5 c	4.4 ab	78.6 cd	7.7 a	5.3 a	3.8 a
		12.5	26.9 a	28.5 c	4.0 ab	80.1 bc	12.6 a	2.7 bc	3.5 ab
		25	27.4 a	29.4 c	5.9 a	80.1 bc	6.2 a	3.7 b	3.5 ab
SM 30%	Watered	50	26.1 ab	38.9 a	5.8 a	79.5 cd	4.9 a	4.7 ab	3.6 a
		0	26.0 ab	33.1 ab	5.0 a	53.0 a	9.2 a	2.0 b	2.5 b
		12.5	27.4 a	33.5 ab	4.4 b	52.6 a	11.2 a	2.3 b	2.6 ab
	Sprayed	25	26.0 ab	28.3 c	4.4 b	49.4 ab	6.9 a	1.7 b	2.5 b
		50	23.7 c	27.2 cd	5.2 ab	45.3 ab	6.9 a	3.7 a	3.3 a
		12.5	26.6 ab	36.9 a	5.7 ab	51.6 a	7.5 a	2.3 b	2.7 ab
Sprayed	25	27.3 a	26.5 d	4.7 ab	50.7 ab	8.3 a	3.0 b	2.9 ab	
	50	21.6 c	32.7 ab	4.9 ab	48.2 ab	6.0 a	3.7 a	3.7 a	

The highest concentration (50 ppm) of CuO NPs reduced pea height after watering or spraying pea plants grown in insufficient substrate moisture (Table 3.2.2.1, 30% SM). Leaf area was reduced by 14 and 18% after the application of suspensions of 25 and 50 ppm and after spraying with 25 ppm suspensions by 20%. SLA decreased by 12% after watering plants with

12.5 and 25 ppm CuO NPs solutions. RWC and root-to-shoot ratio were not affected significantly by CuO NPs treatment. CuO NPs positively affected nodule formation in pea roots when 50 ppm CuO NPs concentration was used. A significant positive effect of 27% on pea yield was found after exposure to drought and irrigation with 50 ppm CuO NPs. The yield of sprayed peas increased up to 47% depending on the concentration of CuO NPs.

Effects on oxidative stress biomarkers

CuO NPs strongly affected both H₂O₂ and MDA accumulation in pea plants grown in the substrate with sufficient moisture (Figure 3.2.2.1A, B). After irrigation with CuO NPs concentrations of 12.5, 25, and 50 ppm, H₂O₂ content increased by 70, 41, and 16%, while after spraying, an increase of 45, 74, and 72% were found in pea leaves (Figure 3.2.2.1A 80% SM). MDA amount in pea leaf cells also significantly increased up to 54% after irrigating or spraying peas with CuO NPs solutions. (Figure 3.2.2.1B 80% SM). From these results, we can conclude that CuO NPs strongly activate lipid peroxidation and can cause oxidative stress in plants.

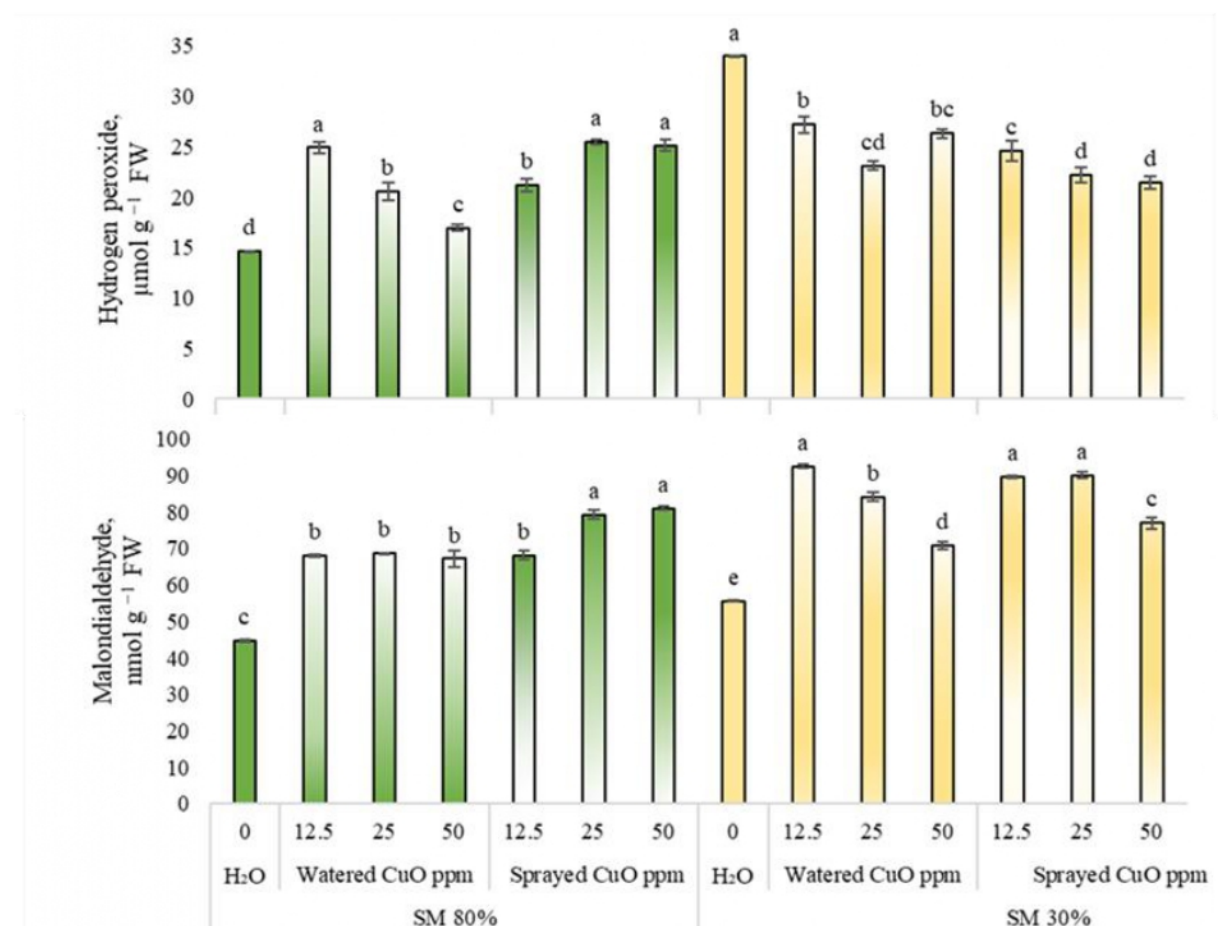


Figure 3.2.2.1. Influence of drought stress and CuO NPs (0; 12.5; 25; 50 ppm) on hydrogen peroxide and malondialdehyde content in *P. sativum* L. H₂O – control plants watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

However, the effect of CuO NPs on the amount of H₂O₂ in the plant changed along with drought exposure (Figure 3.2.2.1A 30% SM). The content of H₂O₂ was reduced remarkably after irrigating or spraying peas with CuO NPs solutions containing different concentrations. The MDA-increasing effect in peas persisted under drought conditions (Figure 3.2.2.1B 30% SM). The MDA content increased to 66 % depending on the CuO NPs concentration and application method.

Effects on non-enzymatic antioxidants

Pea irrigation with 12.5 ppm CuO NPs solution significantly decreased the TPC in pea leaves. In contrast, 25 and 50 ppm exposure significantly increased when plants were grown in the substrate with sufficient moisture (Figure 3.2.2.2A 80% SM). Pea spraying with solutions containing 12.5 and 25 ppm CuO NPs reduced TPC content by 17 and 10%, respectively. A significant increase in ABTS free radical scavenging activity was found after peas were watered with 12.5 ppm and sprayed with 50 ppm CuO NPs solutions (Figure 3.2.2.2C 80% SM). However, inhibition of ABTS free radical scavenging activity was found after watering peas with 25 and 50 ppm or after spraying with 12.5 and 25 ppm CuO NPs compared with control plants.

DPPH free radical scavenging activity decreased by 36 and 13% as peas were irrigated with 12.5 and 50 ppm CuO NPs suspensions, respectively (Figure 3.2.2.2B 80% SM). However, significant stimulation of DPPH free radical scavenging activity was determined after peas were sprayed with concentrations of 25 and 50 ppm CuO NPs. FRAP antioxidant power in pea leaves was strongly affected (Figure 3.2.2.2D 80% SM). It increased 1.7 and 3 times after watering peas with 12.5 and 25 ppm CuO NPs and increased up to 2 times when peas were sprayed with CuO NPs. A decreased value of FRAP was found in peas watered with 50 ppm CuO NPs suspension.

Although more positive indicators were found in pea plants exposed to drought stress and CuO NPs application. For instance, TPC content in pea leaves increased when peas were watered or sprayed with any CuO NPs concentration (Figure 3.2.2.2A 30% SM). In addition, the activity of DPPH free radical scavenging in pea leaves increased with increasing CuO NPs concentration after watering or spraying (Figure 3.2.2.2B 30% SM). ABTS free radical scavenging activity was stimulated up to 70% depending on applied concentration after irrigation and by 50 and 48% after spraying with 25 and 50 ppm CuO NPs solution (Figure 3.2.2.2C 30% SM). A significant reduce in ABTS free radical scavenging activity was found by spraying plants with 12.5 ppm CuO NPs. A positive effect on the FRAP antioxidant power in

pea leaves was observed under all applied CuO NPs concentrations within any application manner (Figure 3.2.2.2D 30% SM).

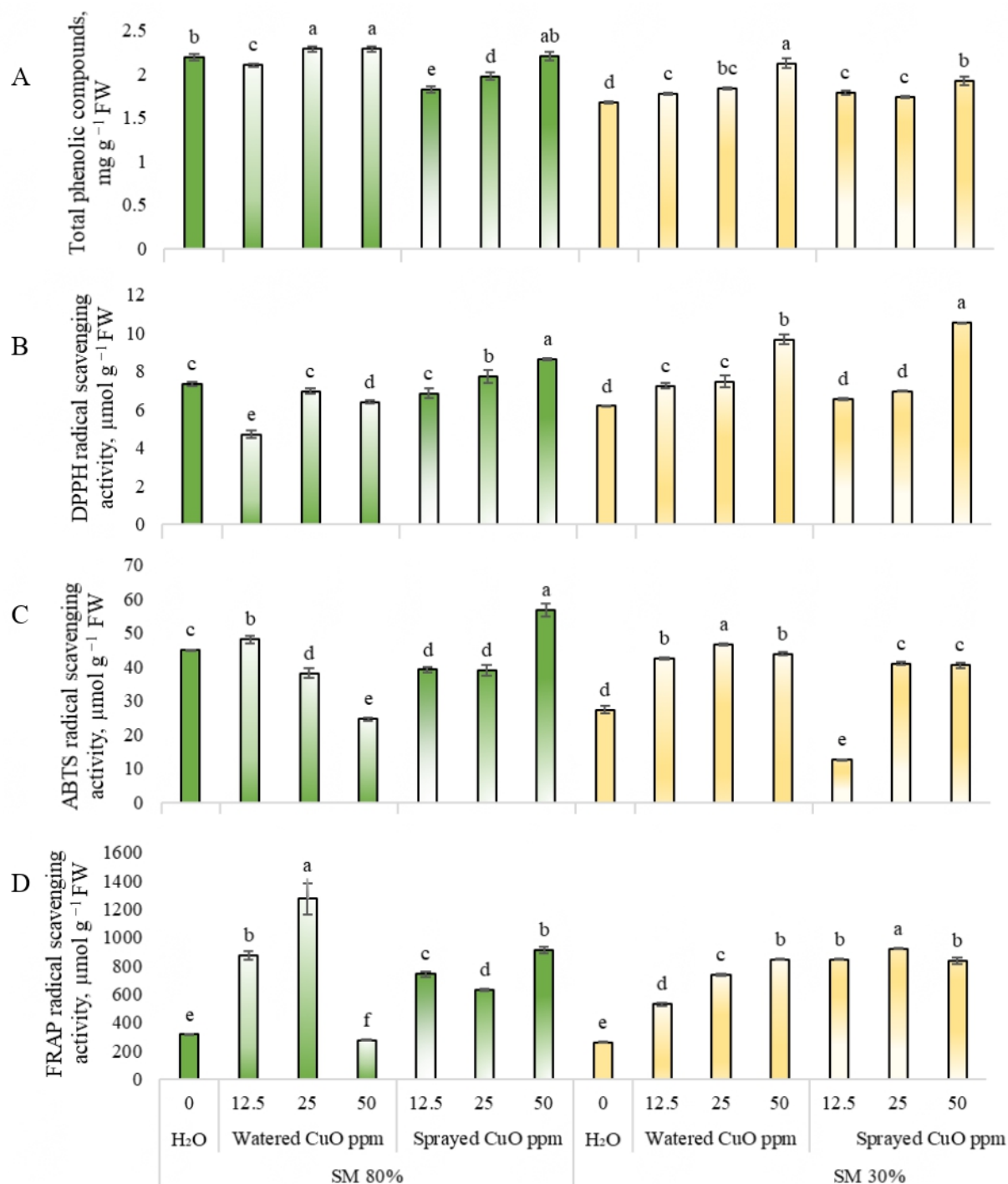


Figure 3.2.2.2. Influence of drought stress and CuO NPs (0; 12,5; 25; 50 ppm) on A – total phenolic compounds, B – DPPH free radical scavenging activity, C – ABTS free radical scavenging activity D – FRAP antioxidant power in *P. sativum* L. H₂O – control plants, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Effects on enzymatic antioxidants

A robust inhibitory effect on CAT and GR activity in pea leaves was found when the plant grew in the substrate with sufficient moisture content (Figure 3.2.2.3A, C 80% SM). CAT activity was reduced by 57, 37, and 63% when peas irrigation with 12.5, 25, and 50 ppm CuO NPs suspensions and by 48, 68, and 65% when sprayed with the same concentrations. Also, GR activity decreased by 37, 27, and 17% when peas were irrigated with 12.5, 25, and 50 ppm CuO NPs, respectively. An inhibitory effect on GR activity was found after spraying plants with concentrations of 25 and 50 ppm CuO NPs. The activity promotion was found for the APX enzyme (Figure 3.2.2.3B 80% SM), as it increased by 3.6, 5.6, and 1.7 times after watering and 4.6, 3.2, and 3 times after spraying with concentrations of 12.5, 25, and 50 ppm CuO NPs respectively. GPX activity (Figure 3.2.2.3E 80% SM) was significantly increased up to 94% after watering plants with any concentration solution, while spraying with CuO NPs had a significant effect only using the solution at 25 ppm concentration. SOD activity (Figure 3.2.2.3D 80% SM) was stimulated by watering peas with concentrations of 25 and 50 ppm and spraying with 12.5 ppm CuO NPs.

The negative effect remained on GR activity in pea leaves (Figure 3.2.2.3C 30% SM) during drought and CuO NPs exposure, reducing it up to 93% depending on applied solution concentration. An inhibitory effect on GPX activity (Figure 3.2.2.3E 30% SM) was also found when peas were sprayed with 12.5, 25, and 50 ppm and watered with 12.5 ppm CuO NPs. A stimulation of GPX activity (Figure 3.2.2.3E 30% SM) was determined only with watering exposure to 25 ppm CuO NPs. CAT activity (Figure 3.2.2.3A 30% SM) was positively influenced up to 1.8 times within any applied CuO NPs concentration. In addition, APX activity (Figure 3.2.2.3B 30% SM) was also significantly stimulated after peas were exposed to CuO NPs. Irrigation of peas with CuO NPs significantly increased SOD activity (Figure 3.2.2.3D 30% SM) up to 33%, depending on the used suspension concentration, and in addition, spraying with 50 ppm induced SOD activity by 53%. However, spraying with 12.5 ppm of CuO NPs had no significant effect but spraying with 25 ppm solution significantly reduced SOD activity.

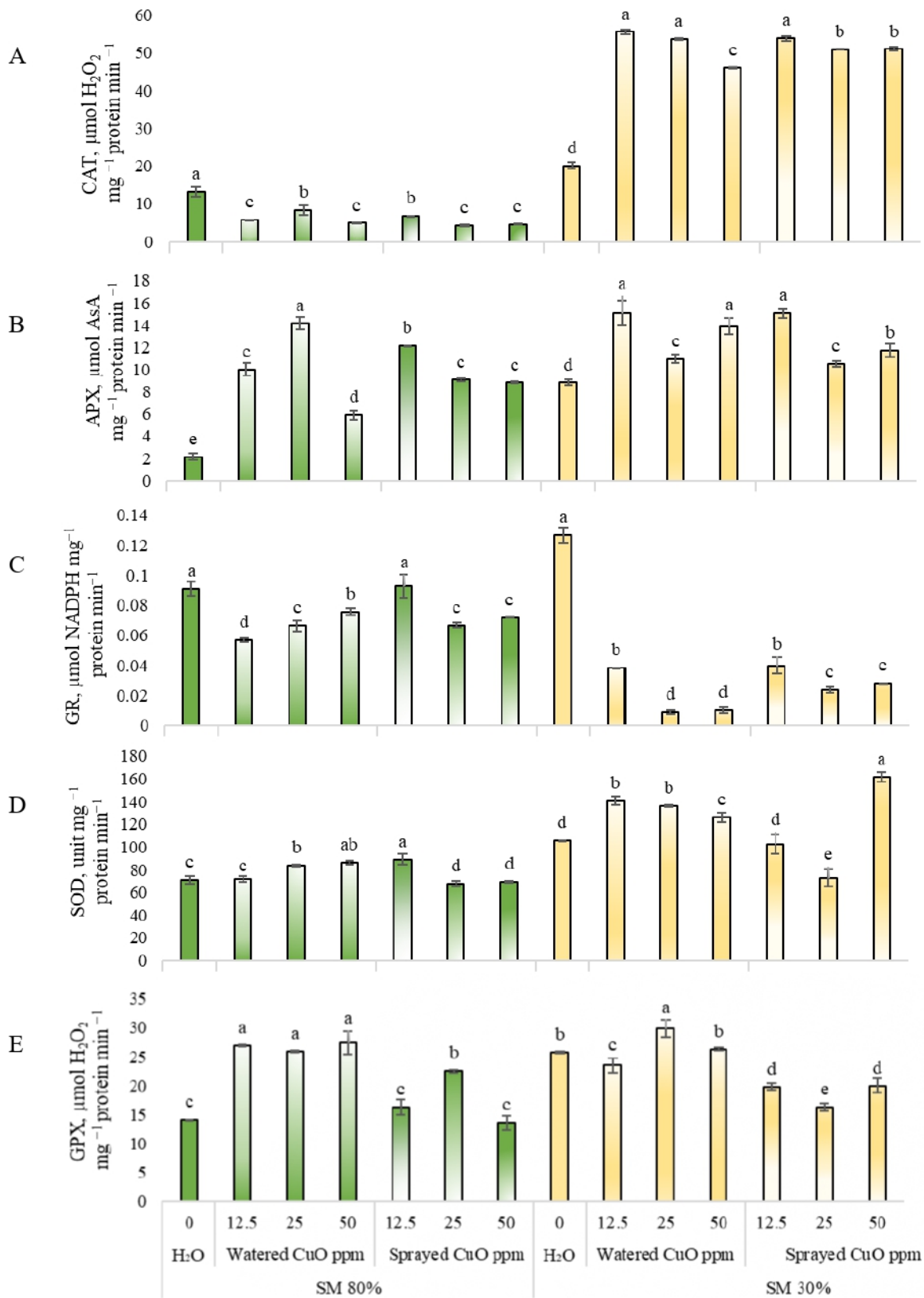


Figure 3.2.2.3. Response of A – ascorbate peroxidase (APX), B – catalase (CAT), C – superoxide dismutase (SOD), D – glutathione reductase (GR), and E – guaiacol peroxidase (GPX) activity to drought stress and CuO NPs (0; 12.5; 25; 50 ppm) in *P. sativum* L. H₂O – control plants sprayed or watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Changes in macro- and microelement composition

Cu content in pea leaves grown at 80% SM increased by 24% after irrigating peas with a concentration of 50 ppm (Table 3.2.2.2). Cu content in pea leaves increased with increasing CuO NPs concentration: 12.5 ppm by 49%, 25 ppm – 112%, and 50 ppm – 233%, the same trend was observed for Cu accumulation in pea stems. An increase in the amount of Cu by 59, 101, and 159% in the roots was found after pea irrigation with 12.5, 25, and 50 ppm CuO NPs suspensions, but an even higher accumulation of 77, 113, and 230% of Cu were found after spraying.

Irrigation with CuO NPs did not cause significant Cu content differences in drought-affected pea plant leaves. However, the increase in Cu content by 14, 35, and 55% in the leaves was found after spraying the drought-affected peas with suspensions at 12.5, 25, and 50 ppm concentrations. In addition, Cu accumulated more in the roots after irrigation compared to the Cu amount in roots not affected by NPs. Spraying with CuO NPs had a positive effect on the accumulation of Cu in all parts of the pea, and it is also worth noting that the amount of Cu increased with the increasing concentration of CuO NPs. Cu content increased in leaves up to 4 times, stems up to 3.5, and roots up to 1.7 times.

The Ca concentration in pea leaves increases during their growth with 80% SM and after spraying or watering at a concentration of 50 ppm. After drought exposure, the Ca content increased when peas were irrigated with 25 ppm and sprayed with 25 and 50 ppm suspensions. The amount of Fe increased when the plants were exposed to CuO NPs suspension with a concentration of 50 ppm. CuO NPs significantly increased the Fe, Zn, and Mn content in roots.

Table 3.2.2.2. Effects of CuO NPs suspension on macro- and microelements of peas leaves, stem, and root. Mean values in bold indicate a statistically significant difference at $p < 0.05$ according to Tukey (HSD) test ($n=9$)

Treatment CuO NPs		Macroelements, mg g ⁻¹ DW							Microelements, µg g ⁻¹ DW				
Leaves		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B	
SM 80%	Watered	0	3.79	18.76	45.73	9.20	2.64	19.86	25.71	42.54	10.32	2.99	3.58
		12.5	3.41	18.29	46.85	9.49	2.21	22.10	33.14	54.10	10.38	2.14	3.35
		25	2.95	16.35	37.41	8.17	2.06	20.53	27.83	49.61	9.14	1.25	4.28
	50	3.98	19.39	48.84	10.20	2.86	31.06	30.50	56.97	12.82	1.54	4.65	
	Sprayed	12.5	3.41	18.29	46.85	9.49	2.21	22.10	33.14	54.10	15.38	2.14	4.19
		25	3.50	18.01	44.24	8.26	2.02	25.91	30.49	49.19	21.88	2.33	2.43
50		4.04	20.45	53.80	10.15	2.11	29.73	37.69	63.38	34.39	2.27	3.87	
SM 30%	Watered	0	3.21	20.56	46.19	8.81	2.07	20.03	37.24	37.65	10.02	0.70	1.30
		12.5	3.31	20.93	48.73	8.51	1.68	27.86	31.14	50.62	9.17	1.50	2.60
		25	3.39	17.85	49.51	9.81	2.55	29.61	33.09	54.21	9.60	1.30	1.96
	50	3.60	19.53	47.44	9.31	1.66	60.76	36.38	64.81	8.36	1.16	2.72	
	Sprayed	12.5	1.86	18.31	44.51	8.02	5.94	18.17	22.23	34.62	15.05	1.61	4.52
		25	3.34	16.45	42.35	8.11	1.59	23.62	30.74	50.56	25.40	1.26	3.98
50		3.14	21.33	52.29	10.12	2.15	24.33	34.36	64.64	51.99	1.54	2.81	
Stem		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B	
SM 80%	Watered	0	2.00	17.77	37.81	8.71	7.73	15.49	19.81	26.09	10.08	3.66	1.05
		12.5	2.18	20.12	47.97	9.24	5.98	19.44	22.37	34.92	11.40	2.54	1.19
		25	1.67	17.29	37.26	7.57	6.41	18.62	21.19	30.99	10.07	3.03	1.11
	50	2.06	17.15	46.37	9.21	8.30	20.21	27.08	33.66	12.05	3.59	0.86	
	Sprayed	12.5	1.77	15.72	46.11	8.55	6.01	17.33	22.28	34.78	15.13	3.86	1.53
		25	2.34	16.58	41.67	7.48	4.93	20.95	26.60	34.47	18.92	3.65	1.33
50		2.10	20.09	51.22	9.58	7.98	25.06	25.33	39.38	30.14	4.53	1.72	
SM 30%	Watered	0	1.43	20.13	38.10	7.98	6.83	15.46	22.31	24.19	8.92	1.17	10.57
		12.5	1.97	21.78	46.83	8.84	5.92	21.48	29.72	33.69	10.19	1.35	5.90
		25	2.19	17.45	48.84	9.50	7.05	29.12	33.34	41.45	12.08	1.38	3.29
50	1.73	20.40	42.92	7.92	7.56	34.94	39.52	36.46	13.81	0.75	4.39		

	Sprayed	12.5	1.86	18.31	44.51	8.02	5.94	18.17	22.23	34.62	15.05	1.61	4.52
		25	1.77	19.38	36.29	7.21	6.45	15.74	27.02	27.93	16.91	1.41	6.40
		50	1.91	22.82	49.94	9.45	8.38	19.89	35.58	38.88	40.97	1.55	5.83
	Root		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B
SM 80%	Watered	0	0.81	16.16	10.25	6.91	3.18	258.88	15.88	46.44	18.46	9.40	2.68
		12.5	2.13	36.49	25.45	15.18	7.28	781.52	40.59	135.20	29.29	12.33	1.89
		25	2.51	34.83	27.74	13.80	10.74	1002.32	51.54	173.10	37.18	10.13	2.51
	50	3.01	42.45	31.05	21.64	9.69	1571.45	84.47	278.69	47.73	12.22	2.41	
	Sprayed	12.5	2.24	39.14	28.93	17.65	7.50	1057.82	48.79	180.70	32.67	7.56	1.05
		25	4.49	69.29	51.35	30.96	12.89	2134.70	104.78	382.71	39.25	8.90	2.01
50		2.63	46.95	33.46	21.66	9.05	1201.86	61.77	205.61	60.94	9.38	1.13	
SM 30%	Watered	0	1.58	28.33	19.57	15.45	7.33	657.85	34.15	118.04	22.49	8.29	6.70
		12.5	2.03	33.83	23.36	14.06	7.53	906.12	47.73	163.88	29.10	6.72	5.19
		25	2.85	37.31	29.91	18.00	9.11	1153.86	60.92	200.61	41.80	8.41	1.34
	50	3.21	39.42	31.21	18.90	10.67	1260.26	110.09	211.62	43.69	8.73	3.01	
	Sprayed	12.5	2.72	40.68	29.84	15.08	9.72	1047.25	65.12	182.99	36.94	6.35	9.59
		25	3.93	67.11	44.51	22.27	15.83	1539.12	79.43	262.40	45.13	9.33	2.40
50		3.09	43.97	39.53	19.78	9.67	1536.57	75.82	278.21	59.78	9.51	8.71	

Macroelements: P – phosphorus, K – potassium, Ca – calcium, Mg – magnesium, Na – sodium, Fe – iron; Microelements: Zn – zinc, Mn – manganese, Cu – copper, Mo – molybdenum, B – boron. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%.

Summary of the chapter (Table 3.2.2.3)

- CuO NPs positively affected the antioxidant capacity and activated APX, SOD, and CAT enzymes but suppressed the activity of GPX and GR enzymes.
- CuO NPs significantly reduced H₂O₂ content but induced an MDA in pea leaves.
- The highest Cu amount was accumulated in the leaves of plants sprayed with 50 ppm NPs suspension as well as in the stem and roots
- CuO NPs suspensions were more effective when sprayed on plants than when watered.

Table 3.2.2.3. The impact of drought stress and CuO NPs (12,5; 25; 50 ppm) on *P. sativum* L. grown in the substrate with sufficient (SM 80%) and insufficient (SM 30%) moisture is expressed as a percentage change (%) compared to the control (for SM 80% control means plants grown under SM 80% and NPs untreated; SM 30% control means drought affected but NPs untreated plants) in the heat map.

Statistically, significant differences are marked in bold

Treatment CuO NPs, ppm	SM 80%						SM 30%					
	Watered			Sprayed			Watered			Sprayed		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Plant height	-2	0	-4	-6	-4	-8	5	0	-9	2	5	-17
Leaf area	-7	-12	-13	-21	-19	8	1	-14	-18	12	-20	-1
Nodules	20	-40	220	-60	120	180	-83	-17	83	17	-100	283
RWC	2	2	-5	-3	-3	-4	-1	-7	-15	-3	-4	-9
Root/shoot	25	-29	-1	62	-20	-36	22	-24	-24	-18	-9	-34
SLA	-26	-24	-17	-24	12	10	-12	-13	4	13	-7	-3
Yield	-4	-2	-1	-9	-11	-7	2	1	27	7	14	47
ABTS	7	-15	-45	-13	-13	26	55	70	60	-54	50	48
DPPH	-36	-5	-13	-7	6	18	17	21	56	6	13	70
TPC	-4	4	4	-17	-10	1	6	10	27	7	4	15
FRAP	177	304	-13	136	100	190	103	182	224	224	253	220
HP	70	41	16	45	74	72	-20	-32	-23	-28	-35	-37
MDA	52	54	50	52	77	81	66	51	27	61	62	38
GR	-37	-27	-17	2	-27	-21	-70	-93	-92	-68	-81	-78
GPX	90	83	94	15	59	-4	-9	16	2	-23	-37	-22
APX	364	556	173	463	320	308	72	25	58	71	19	33
SOD	1	18	21	25	-5	-3	33	29	20	-3	-31	53
CAT	-57	-37	-63	-48	-68	-65	176	166	128	167	153	153
Cu (leaves)	1	-11	24	49	112	233	-8	-4	-17	50	153	419
Cu (stem)	13	0	19	50	88	199	14	35	55	69	89	359
Cu (roots)	59	101	159	77	113	230	29	86	94	64	101	166

RWC – relative water content, SLA – specific leaf area, TPC – total phenolic compounds, HP – hydrogen peroxide, MDA – malondialdehyde, GR – glutathione reductase, GPX – guaiacol peroxidase, APX – ascorbate peroxidase, SOD – superoxide dismutase, CAT – catalase, Cu content in leaves, stem, and root. 0 – control plants watered with deionized water, drought stress – 30% substrate moisture.

3.2.3 Effects of molybdenum trioxide nanoparticle on peas under different substrate moisture

Impact on morphological parameters

Molybdenum trioxide nanoparticle (MoO_3 NPs) suspensions significantly affected the morphological parameters of peas grown under 80% substrate moisture (Table 3.2.3.1). Pea height increased after watering with MoO_3 NPs suspensions of any concentration. The suspension of 50 ppm concentration induced leaf area formation in both watering (25%) and spraying (15%). The RWC in plants was positively affected by MoO_3 NPs, and the most significant increase was found after watering and spraying peas at 50 ppm. The ratio of root to shoot increased by 22% when peas were watered with a 50 ppm concentration of MoO_3 NPs suspension. In addition, 33 and 11% increase was found after spraying with 12.5 and 50 ppm concentration. The SLA was reduced by 28% after peas grown in the substrate with sufficient moisture were watered with 12.5 ppm. SLA reduction of up to 30% was found after spraying with 12.5 and 50 ppm suspensions. MoO_3 NPs concentrations of 25 and 50 ppm strongly influenced the formation of nodules on the roots, as their number was 3 and 5.6 times higher after watering and 1.4 times higher after spraying with 50 ppm compared to the control plants. Yield increased significantly after irrigating peas with 12.5, 25, and 50 ppm MoO_3 solution. A significant increase in yield was observed in plants treated through leaves with 50 ppm MoO_3 and grown under 80% substrate moisture.

MoO_3 NPs effectively reduced the effects of drought on pea morphological parameters (Table 3.2.3.1, 30 % SM). The height of peas increased up to 40% after watering and up to 24% after spraying with MoO_3 NPs solutions. The formation of leaf area increased when drought-affected peas were watered with a suspension of 50 ppm MoO_3 NPs by 30%. Spraying the suspensions of MoO_3 NPs increased leaf area up to 12%. RWC in peas increased by 10% when 50 ppm MoO_3 NPs were applied through roots. The reducing effects of MoO_3 NPs were found on SLA at all concentrations. A decrease of 30% was determined after watering and up to 25% after spraying. Even during drought conditions, a positive effect of MoO_3 NPs on pea root nodule number was recorded. An increase of 1.2 to 5 times was found when peas were watered with 25 and 50 ppm and 1.2 – 2 times when spraying with MoO_3 NPs. The pea yield showed a strong dependence on the increasing concentration of MoO_3 NPs during drought. It increased by 11, 26, and 80% after irrigation with 12.5, 25, and 50 ppm suspensions of MoO_3 NPs, and after spraying increased by 3, 15, and 64%, respectively.

Table 3.2.3.1. Impact of drought stress and MoO₃ NPs (12.5; 25; 50 ppm) on *P. sativum* L. height, leaf area, specific leaf area (SLA), relative water content (RWC), root-to-shoot ratio, and the number of nodules. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Mean values within columns followed by different letters differ significantly at $p < 0.05$ ($n=10$) according to Tukey (HSD) test

	MoO ₃ NPs, ppm	Plants height, cm	Leaf area, cm ²	SLA, m ² kg ⁻¹	RWC, %	Root/shoot ratio	Number of nodules	Yield, t ha ⁻¹	
SM 80%	Watered	0	28.4 d	36.1 bcd	5.3 a	82.5 e	7.8 b	1.7 c	3.9 b
		12.5	30.1 c	35.0 cd	3.8 bc	86.4 cd	7.6 bc	2.3 c	4.1 a
		25	34.4 b	41.3 abc	4.2 abc	87.2 bc	7.9 b	7.0 b	4.1 a
	Sprayed	50	35.5 a	45.2 a	4.0 abc	88.9 a	9.4 ab	11.0 a	4.1 a
		12.5	30.5 c	32.9 d	3.5 c	85.1 d	10.3 a	2.3 c	3.9 b
		25	29.8 c	36.7 bcd	5.0 ab	86.8 c	6.6 c	2.0 c	4.0 ab
SM 30%	Watered	50	33.4 b	41.6 ab	3.7 bc	88.5 ab	8.6 ab	4.0 b	4.2 a
		0	26.0 f	33.1 cd	5.0 a	53.0 b	9.2 ab	2.0 b	2.5 c
		12.5	28.5 d	30.8 d	4.4 ab	60.1 ab	5.1 b	1.0 b	2.8 ab
	Sprayed	25	30.6 c	30.4 d	3.5 b	58.0 ab	9.2 ab	4.3 a	3.2 a
		50	36.3 a	43.1 a	3.6 b	64.4 a	11.3 ab	12.7 a	4.5 a
		12.5	27.0 e	37.1 b	4.4 ab	56.7 b	12.5 a	1.7 b	2.6 b
Sprayed	25	28.6 d	36.3 b	4.1 ab	56.7 b	8.4 ab	4.3 a	2.9 ab	
	50	32.2 b	36.3 b	3.8 b	58.5 ab	10.7 ab	5.7 a	4.2 a	

Effects on oxidative stress biomarkers

H₂O₂ content in normal substrate moisture (SM 80%) grown pea leaves was promoted from 1.2 to 2 times after pea application with 12.5, 25, and 50 ppm MoO₃ NPs suspensions. MoO₃ NPs promoted lipid peroxidation in pea plants grown under SM 80% (Figure 3.2.3.1B 80% SM). This is shown by the MDA concentration increase in peas when they were exposed to a concentration of 12.5 and 50 ppm by watering and by spraying with 12.5 and 25 ppm MoO₃ NPs.

In drought-affected peas, H₂O₂ content decreased by 8, 12, and 20% when plants were irrigated with 12.5, 25, and 50 ppm MoO₃ NPs suspensions, but MDA concentration increased by 9, 11, and 25% (Figure 3.2.3.1A, B 30% SM). Spraying with 12.5, 25, and 50 ppm MoO₃ NPs suspensions similarly reduced H₂O₂ content in pea leaves. The reduced MDA content was determined in peas sprayed with 50 ppm MoO₃ NPs suspension compared to plants grown in drought conditions and untreated with NPs.

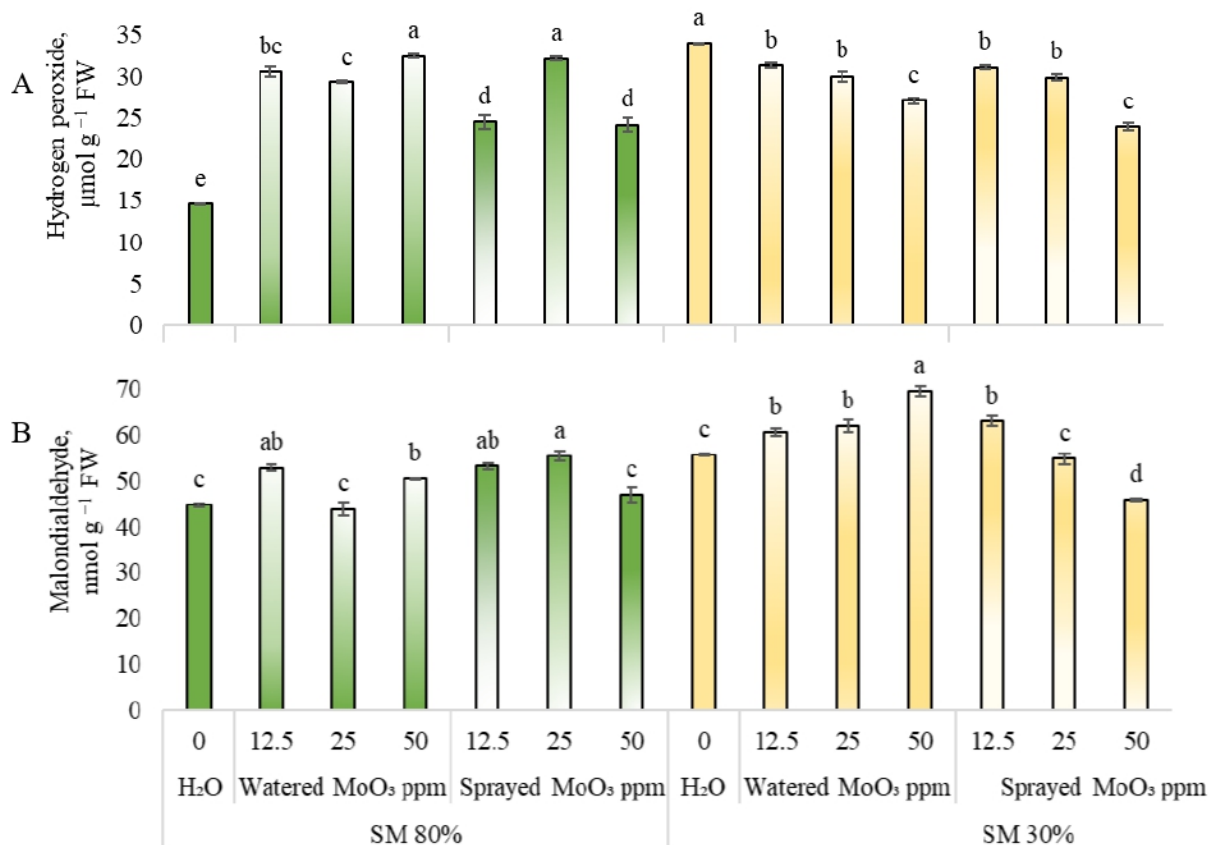


Figure 3.2.3.1. Influence of drought stress and MoO₃ NPs (0; 12,5; 25; 50 ppm) on hydrogen peroxide and malondialdehyde content in *P. sativum* L. H₂O – control plants watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Effects on non-enzymatic antioxidants

MoO₃ NPs have an ambiguous effect on plant non-enzymatic antioxidants in peas growing at 80% SM (Figure 3.2.3.2). The amount of TPC (Figure 3.2.3.2A 80% SM) decreased by 20% after 12.5 ppm MoO₃ NPs exposure through the roots, and after foliar application with 12.5 and 25 ppm, the amount of TPC decreased by 23 and 15%. In pea leaves, FRAP antioxidant power (Figure 3.2.3.2D 80% SM) increased significantly after watering or spraying with MoO₃ NPs solution of any concentration. Irrigation of peas with 25 and 50 ppm MoO₃ NPs solutions influenced the stimulation of DPPH free radical scavenging activity by 27 and 36%, respectively. Spraying with 25 ppm suspension of MoO₃ NPs caused a 7% decrease in DPPH free radical scavenging activity (Figure 3.2.3.2B 80% SM). ABTS free radical scavenging activity (Figure 3.2.3.2C 80% SM) was inhibited by watering peas with 12.5 ppm by 11% and by spraying with 25 ppm - 8%.

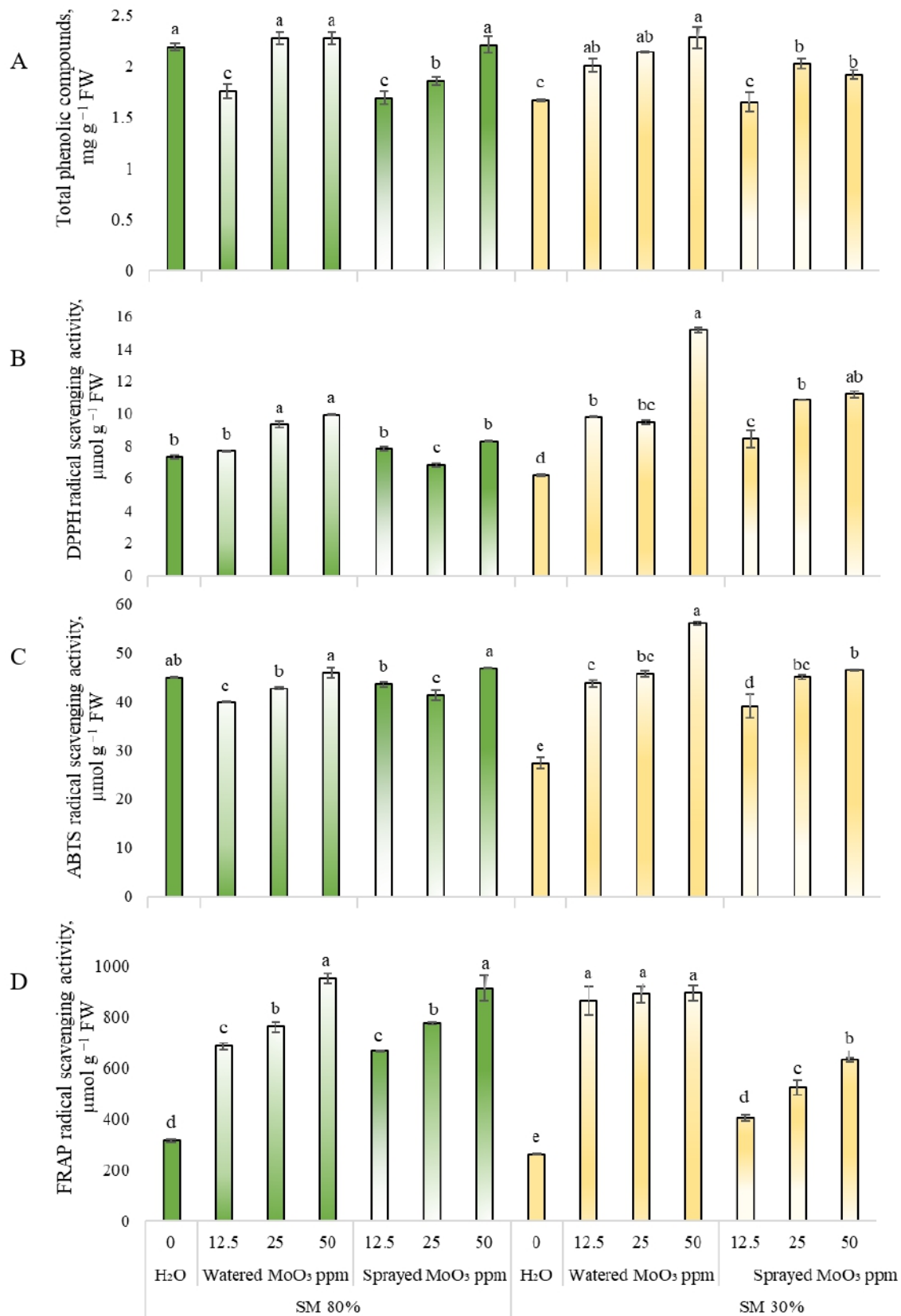


Figure 3.2.3.2. Influence of drought stress and MoO₃ NPs (0; 12,5; 25; 50 ppm) on A – total phenolic compounds, B – DPPH free radical scavenging activity, C – ABTS free radical scavenging activity, D – FRAP antioxidant power in *P. sativum* L. H₂O – control plants, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean \pm SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

The effects of drought (Figure 3.2.3.2, 30% SM) and MoO₃ NPs on the non-enzymatic antioxidants of peas were significant. The amount of TPC (Figure 3.2.3.2A 30%SM) in drought-affected peas was increased up to 37% after exposure with MoO₃ NPs through roots and up to 15% after spraying. The activity of the ABTS free radical scavenging (Figure 3.2.3.2C 30% SM) was particularly enhanced. It increased by 60, 67, and 105% in plants exposed to 12.5, 25, and 50 ppm MoO₃ NPs through roots. Also, foliar exposure to MoO₃ NPs increased ABTS free radical scavenging activity by 43, 65, and 70% at the above concentrations. The results show that exposure through roots to MoO₃ NPs induced DPPH free radical scavenging activity (Figure 3.2.3.2B 30%SM) up to 145% and through leaves up to 81%. The positive influence of MoO₃ NPs of any concentration was significant for FRAP antioxidant power in pea leaves.

Effects on enzymatic antioxidants

A strong effect of MoO₃ NPs on the activity of enzymatic antioxidants was found in plants grown at 80% substrate moisture (Figure 3.2.3.3). CAT activity (Figure 3.2.3.3A 80%SM) increased 41% after spraying with 12.5 ppm MoO₃ NPs suspension. Concentrations of 12.5, 25, and 50 ppm MoO₃ NPs caused significantly higher APX activity (Figure 3.2.3.3B 80% SM) after watering or spraying pea plants. In contrast, the GR activity decreased after watering or spraying with MoO₃ NPs solution of any concentration. MoO₃ NPs strongly induced GPX activity in pea leaves depending on applied concentration and application manner (Figure 3.2.3.3E 80% SM).

Drought exposure and MoO₃ NPs stimulated APX activity in pea leaves (Figure 3.2.3.3B 30% SM). After peas' exposure to 12.5, 25, and 50 ppm MoO₃ NPs, from 2 to 6 times increase in APX activity was found, and after exposure through leaves – 2 – 4 times. Similar results were found for CAT activity in pea leaves during drought stress (Figure 3.2.3.3A 30% SM). An increase in CAT activity of 2 times was found after watering peas with 12.5, 25, and 50 ppm MoO₃ NPs while spraying promoted CAT activity from 1.6 to 1.9 times. Pea plant treatment with 12.5, 25, and 50 ppm MoO₃ NPs through roots inhibited GR activity in leaves by 56, 57, and 31% and foliar exposure by 88, 79, and 71%, respectively. GPX activity (Figure 3.2.3.3E 30% SM) in drought-affected peas stimulated the irrigation with suspensions of 25 and 50 ppm by 43 and 55%, respectively. GPX activity increased by 17 and 56% after plants were sprayed with 12.5 and 50 ppm suspensions of MoO₃ NPs. SOD activity (Figure 3.2.3.3D 30% SM) was inhibited when peas were watered with 25 ppm and increased with the application of 50 ppm MoO₃ NPs by watering or spraying.

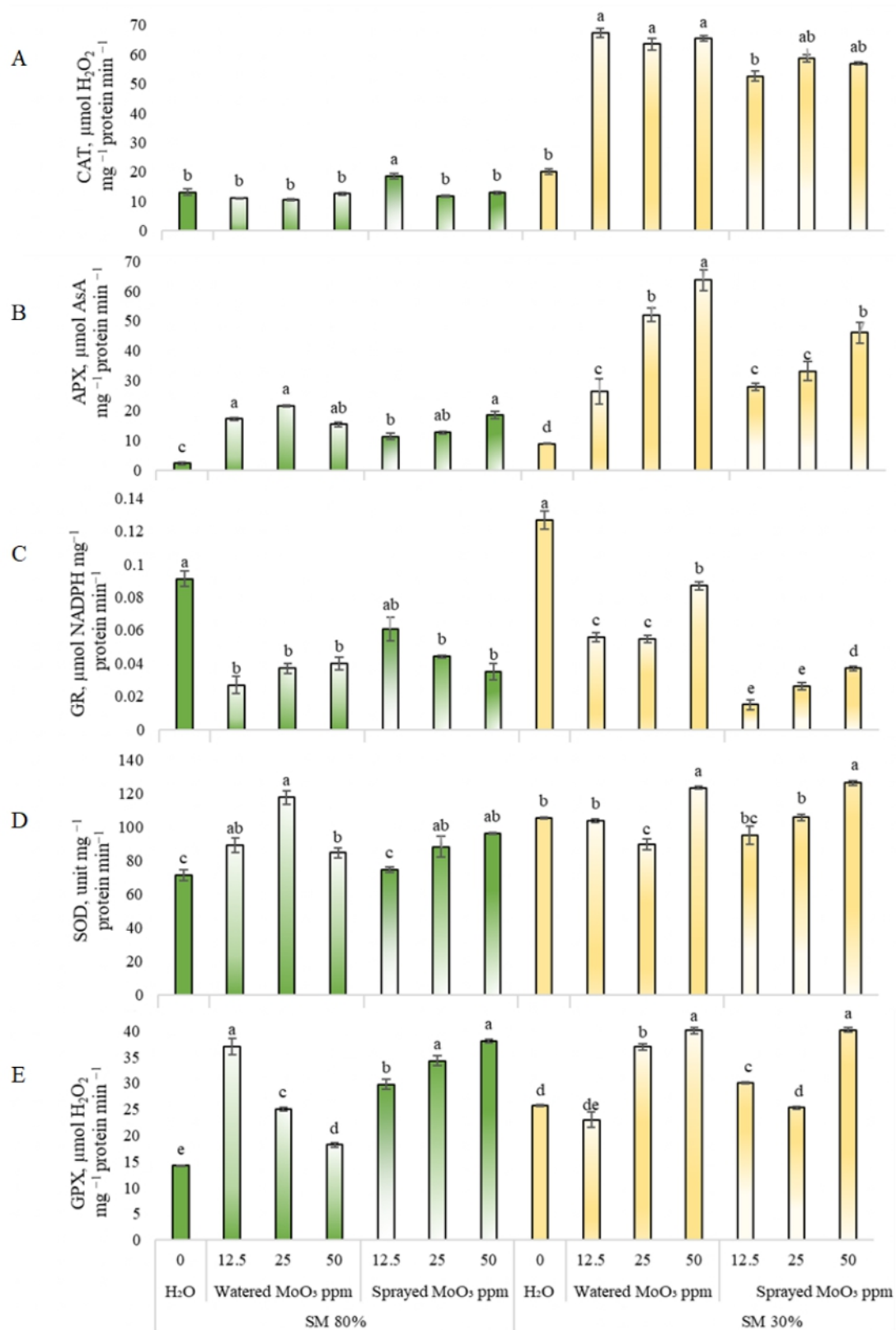


Figure 3.2.3.3. Response of A – ascorbate peroxidase (APX), B – catalase (CAT), C – superoxide dismutase (SOD), D – glutathione reductase (GR), and E – guaiacol peroxidase (GPX) activity to drought stress and MoO₃ NPs (0; 12.5; 25; 50 ppm) in *P. sativum* L. H₂O – control plants sprayed or watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test (p < 0.05)

Changes in macro- and micro-elemental composition

The content of Mo in pea leaves increased with increasing concentration of MoO₃ NPs in suspension (Table 3.2.3.2). For instance, in pea leaves grown under normal conditions (SM 80%), up to 43% increase in Mo amount was found after MoO₃ NPs exposure through roots, and up to 47% after foliar treatment. A similar tendency was observed for the accumulation of Mo content in the stem. Mo content increased up to 34% in pea stems when plants were treated through roots, and an even more substantial increase in Mo content up to 46% was found when plants were sprayed. Mo accumulated 2-3.6 times more in the roots when peas were irrigated with MoO₃ NPs and when they were sprayed 1.3-1.8 times.

Exposure to drought and MoO₃ NPs significantly increased the Mo content in pea leaves and stems. Watering with MoO₃ NPs suspensions was influenced to accumulate 1.6 - 3 times more Mo in leaves, while spraying increased Mo content up to 4.5 times. In addition, a higher Mo accumulation was observed in the stem after spraying the plants with MoO₃ NPs.

Furthermore, MoO₃ NPs had a strong effect in reducing Ca accumulation in pea leaves and stems when they were grown under 80% SM. Accumulation of Mg, Na, and Fe also decreased in stems. However, in the roots, after watering and spraying with MoO₃ NPs, a positive effect on all elements was found.

Table 3.2.3.2. Effects of MoO₃ NPs suspension on macro- and microelements of peas leaves, stem, and root. Mean values in bold indicate a statistically significant difference at $p < 0.05$ according to Tukey (HSD) test ($n=9$)

Treatment MoO ₃ NPs, ppm		Macroelements, mg g ⁻¹ DW						Microelements, µg g ⁻¹ DW					
Leaves		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B	
SM 80%	Watered	0	3.79	18.76	45.73	9.20	2.64	19.86	25.71	42.54	10.32	2.99	3.58
		12.5	3.93	19.33	42.80	8.58	2.26	18.35	24.90	41.32	7.50	3.40	3.36
		25	3.63	17.55	40.75	8.13	2.26	20.82	26.45	46.29	12.77	3.79	4.96
	Sprayed	0	5.48	19.16	37.28	7.67	1.43	17.34	26.56	41.23	9.68	4.27	4.53
		12.5	3.93	19.33	42.80	8.58	2.26	18.35	24.90	41.32	7.50	3.40	3.36
		25	2.66	13.74	32.55	7.62	2.60	15.16	24.01	41.47	9.71	3.76	4.77
SM 30%	Watered	0	3.21	20.56	46.19	8.81	2.07	20.03	37.24	37.65	10.02	0.70	2.92
		12.5	3.31	15.47	35.02	6.58	1.23	14.95	23.90	38.54	8.58	1.83	4.80
		25	3.78	18.30	45.14	8.98	2.16	20.75	30.10	49.45	11.27	2.36	4.42
	Sprayed	0	5.52	21.83	44.86	8.87	1.55	20.00	33.79	48.59	12.45	2.89	4.84
		12.5	2.27	21.20	41.08	8.54	6.33	12.18	15.62	29.62	8.96	0.51	3.20
		25	3.72	18.43	46.85	8.66	1.82	21.55	33.70	53.09	12.75	3.67	4.17
SM 80%	Watered	0	2.00	17.77	37.81	8.71	7.73	15.49	19.81	26.09	10.08	3.66	1.05
		12.5	1.62	15.40	34.36	6.78	4.22	13.28	13.40	24.80	9.26	4.02	2.03
		25	1.53	15.50	36.53	7.30	6.24	11.57	17.93	25.31	8.37	4.91	2.25
	Sprayed	0	2.14	15.46	26.60	5.77	4.74	9.31	12.47	20.17	7.19	4.89	2.79
		12.5	2.61	19.49	43.13	8.73	5.37	13.45	21.52	33.23	10.06	4.73	0.80
		25	1.14	13.24	31.40	6.70	6.22	9.74	13.13	23.71	6.78	5.28	1.61
SM 30%	Watered	0	1.70	14.31	33.33	7.01	5.78	11.22	12.00	21.96	8.83	5.36	2.78
		12.5	1.43	20.13	38.10	7.98	6.83	15.46	22.31	24.19	8.92	1.17	2.11
		25	1.56	15.47	28.56	5.88	3.94	8.34	11.79	19.51	6.15	3.53	2.51
	Sprayed	0	1.83	18.69	38.74	8.32	6.92	13.09	18.92	28.34	8.33	3.58	1.61
		12.5	2.43	20.78	37.50	7.84	5.49	11.96	16.20	26.29	8.27	3.70	2.34
		25											

	Sprayed	12.5	2.27	21.20	41.08	8.54	6.33	12.18	15.62	29.62	8.96	2.51	2.14
		25	1.53	18.87	41.54	7.87	6.29	14.26	17.46	29.51	10.12	2.61	1.70
		50	1.95	18.78	42.06	7.55	4.27	18.25	17.84	30.52	10.02	2.77	1.61
	Root		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B
SM 80%		0	0.81	16.16	10.25	6.91	3.18	258.88	15.88	46.44	18.46	9.40	2.68
	Watered	12.5	2.65	30.54	26.31	19.02	6.09	1542.53	53.70	269.59	16.85	28.18	6.05
		25	1.74	24.06	18.78	13.48	4.74	1085.63	38.42	185.89	16.27	29.43	2.92
		50	2.60	26.79	20.12	13.39	5.66	1098.77	44.99	192.17	19.96	43.19	2.30
	Sprayed	12.5	2.65	35.00	26.38	17.11	6.88	777.16	39.92	134.38	21.16	21.63	5.37
		25	1.71	24.34	21.29	12.18	6.44	756.67	32.37	131.22	16.09	26.04	0.74
50		2.26	27.44	22.20	13.67	5.80	947.33	38.04	162.72	18.57	26.69	0.30	
SM 30%		0	1.58	28.33	19.57	15.45	7.33	657.85	34.15	118.04	14.99	16.58	1.97
	Watered	12.5	1.76	25.67	21.46	13.76	5.91	1123.38	48.53	206.49	18.28	20.12	3.02
		25	1.52	22.28	18.62	12.25	7.10	851.83	39.13	143.28	16.89	21.21	0.91
		50	2.89	38.30	28.11	16.08	10.07	847.52	48.35	152.10	20.37	28.07	4.28
	Sprayed	12.5	2.15	28.40	22.71	11.62	6.85	819.77	35.10	154.88	18.16	22.91	1.54
		25	2.27	35.22	27.72	16.93	7.92	1077.09	53.24	202.91	21.94	25.80	0.61
50		1.67	25.03	20.39	12.45	5.46	1034.68	36.24	183.94	16.29	28.65	1.37	

Macroelements: P – phosphorus, K – potassium, Ca – calcium, Mg – magnesium, Na – sodium, Fe – iron; Microelements: Zn – zinc, Mn – manganese, Cu – copper, Mo – molybdenum, B – boron. 0 – control plants watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%.

Summary of the chapter (Table 3.2.3.3)

- *MoO₃ NPs suspension positively affected plant morphological parameters determined by the effective reduction of oxidative biomarkers, increased total phenolics, and non-enzymatic antioxidant activity under drought conditions.*

- *MoO₃ NPs had an effect in enhancing the activity of CAT, APX, SOD, and GPX but reducing the activity of GR under both drought and normal conditions.*

- *The highest accumulation of Mo was found in pea plants when they were watered with 50 ppm MoO₃ NPs suspension.*

- *Comparing the application methods, MoO₃ NPs through the roots have a more substantial effect on peas.*

Table 3.2.3.3. The impact of drought stress and MoO₃ NPs (12,5; 25; 50 ppm) on *P. sativum* L. grown in the substrate with sufficient (SM 80%) and insufficient (SM 30%) moisture is expressed as a percentage change (%) compared to the control (for SM 80% control means plants grown under SM 80% and NPs untreated; SM 30% control means drought affected but NPs untreated plants) in the heat map.

Statistically, significant differences are marked in bold

Treatment MoO ₃ NPs, ppm	SM 80%						SM 30%					
	Watered			Sprayed			Watered			Sprayed		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Plant height	6	21	25	7	5	17	10	18	40	4	10	24
Leaf area	-3	15	25	-9	2	15	-7	-8	30	12	10	10
Nodules	40	320	560	40	20	140	-50	117	533	-17	117	183
RWC	5	6	8	3	5	7	13	9	21	7	7	10
Root/shoot	-1	2	22	33	-15	11	-45	1	23	36	-8	16
SLA	-28	-22	-24	-34	-6	-30	-12	-30	-29	-12	-18	-25
Yield	6	5	6	1	-12	7	11	26	80	3	15	64
ABTS	-11	-5	2	-3	-8	4	60	67	105	43	65	70
DPPH	5	27	36	6	-7	13	59	53	145	36	76	81
TPC	-20	4	4	-23	-15	1	20	28	37	-1	22	15
FRAP	117	141	202	111	146	190	231	241	242	55	100	142
HP	109	101	122	68	120	65	-8	-12	-20	-9	-12	-30
MDA	18	-2	13	19	24	5	9	11	25	13	-1	-17
GR	-70	-59	-56	-33	-52	-62	-56	-57	-31	-88	-79	-71
GPX	161	76	28	110	142	168	-11	43	55	17	-2	56
APX	692	899	607	423	481	748	198	490	622	216	276	422
SOD	25	65	19	5	24	35	-2	-15	17	-10	0	20
CAT	-16	-19	-4	41	-10	-2	234	215	224	161	191	183
Mo (leaves)	14	27	43	14	26	47	160	234	310	27	421	454
Mo (stem)	10	34	34	29	44	46	201	205	215	114	122	136
Mo (roots)	200	213	360	130	177	184	21	28	69	38	56	73

RWC – relative water content, SLA – specific leaf area, TPC – total phenolic compounds, HP – hydrogen peroxide, MDA – malondialdehyde, GR – glutathione reductase, GPX – guaiacol peroxidase, APX – ascorbate peroxidase, SOD – superoxide dismutase, CAT – catalase, B content in leaves, stem, and root. 0 – control plants watered with deionized water, drought stress – 30% substrate moisture.

3.2.4 Effects of boron nanoparticle on peas under different substrate moisture

Impact on morphological parameters

Pea height increased by 14 and 27% when watered and by 28 and 19% when sprayed with 12.5 and 50 ppm B₂O₃ NPs suspension under sufficient substrate moisture (Table 3.2.4.1, 80% SM). Leaf area also significantly increased when plants were watered with 12.5 and 50 ppm B₂O₃ NPs suspensions while spraying increased it within any applied concentration. Furthermore, a positive effect of B₂O₃ NPs was found in RWC. At the same time, a decrease of 15% in SLA was observed after spraying with 12.5 ppm solution. Root to shoot ratio statistically reliably increased after watering plants at 12.5 ppm by 21%, 25 ppm - by 36%, and 50 ppm - by 68%. This shows that in plants grown at 80% substrate moisture, increasing the concentration of B₂O₃ nanoparticle suspension promotes root growth. In addition, an increase in the root-to-shoot ratio by 68% (12.5 ppm), 18% (25 ppm), and 34% (50 ppm) was observed when plants were sprayed. The results also show that B₂O₃ NPs positively affect the number of nodules on plant roots by increasing their amount up to 5.6 times when plants were watered and up to 3.4 times when plants were sprayed. The results showed that pea irrigation with 50 ppm B₂O₃ NPs had a significant positive effect on yield, while foliar treatment increased pea yield at suspensions containing 12.5 and 25 ppm B₂O₃ NPs.

B₂O₃ NPs strongly affected pea plants grown in drought conditions (Table 3.2.4.1 30% SM). Applied B₂O₃ NPs suspension with different concentrations increased plant height as they were watered or sprayed. Furthermore, watering the plants with 25 and 50 ppm B₂O₃ NPs suspensions increased the leaf area by 30 and 40%, respectively. There was a statistically significant increase in RWC at higher B₂O₃ NPs concentrations. The root-to-shoot ratio increased to 30% after watering drought-affected peas with B₂O₃ NPs solutions, while foliar application increased the ratio to 14%. 50 ppm B NPs concentration influenced the number of root nodules increasing it by 3 times during watering and up to 6 times during spraying. Irrigation with the suspension of 12.5 and 25 ppm B₂O₃ NPs positively affected pea yield. Additionally, spraying drought-stressed peas with 12.5, 25, and 50 ppm B₂O₃ NPs increased yield by 16%.

Table 3.2.4.1. Impact of drought stress and B₂O₃ NPs (12,5; 25; 50 ppm) on *P. sativum* L. height, leaf area, specific leaf area (SLA), relative water content (RWC), root-to-shoot ratio, and the number of nodules. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Mean values within columns followed by different letters differ significantly at $p < 0.05$ ($n=10$) according to Tukey (HSD) test

	B ₂ O ₃ NPs, ppm	Plants height, cm	Leaf area, cm ²	SLA, m ² kg ⁻¹	RWC, %	Root/shoot ratio	Number of nodules	Yield, t ha ⁻¹	
SM 80%	Watered	0	28.4 d	36.1 c	5.3 ab	82.5 d	7.8 c	1.7 d	3.9 b
		12.5	32.4 bc	46.1 a	6.1 a	84.3 c	9.4 b	11.0 a	3.1 c
		25	30.3 cd	39.2 bc	5.1 ab	86.1 ab	10.5 b	5.0 c	4.0 ab
	Sprayed	50	36.1 a	46.8 a	4.7 ab	86.8 ab	13.1 a	9.7 ab	4.7 a
		12.5	36.3 a	49.9 a	4.5 b	87.3 a	13.1 a	7.3 bc	4.4 a
		25	30.9 bcd	44.9 ab	5.0 ab	85.5 bc	9.2 b	5.7 c	4.3 a
SM 30%	Watered	50	33.8 ab	44.6 ab	5.1 ab	85.6 bc	10.4 b	5.0 c	3.9 b
		0	26.0 e	33.1 b	5.0 a	53.0 c	9.2 c	2.0 c	2.5 c
		12.5	28.4 d	27.5 b	4.2 ab	52.6 c	9.6 b	2.3 c	2.9 a
	Sprayed	25	32.7 a	42.9 a	4.8 a	58.3 ab	10.5 a	6.0 b	2.7 ab
		50	30.4 bc	46.4 a	4.2 ab	59.1 a	11.9 a	8.3 b	2.6 bc
		12.5	29.3 cd	31.6 b	4.4 ab	51.9 c	10.2 a	2.0 c	3.0 a
Watered	25	31.2 ab	33.4 b	3.9 ab	55.2 bc	10.5 a	3.3 c	2.9 a	
	50	29.6 cd	31.2 b	3.5 b	57.8 ab	9.6 b	13.7 a	2.9 a	

Effects on oxidative stress biomarkers

The results show that exposure to B₂O₃ NPs through the roots increased the amount of H₂O₂ in plants, regardless of its concentration, when peas were grown under sufficient substrate moisture (Figure 3.2.4.1A 80% SM). When plants were sprayed, a statistically reliable 65% increase in H₂O₂ content was found at 12.5 ppm B₂O₃ NPs. A significant decrease in MDA concentration (Figure 3.2.4.1B 80% SM) was also found in pea leaves as plants were watered or sprayed with a solution containing any concentration of B₂O₃ NPs

Significant inhibition of H₂O₂ and MDA was found as their concentration decreased after plants' exposure to drought and B₂O₃ NPs (Figure 3.2.4.1A, B 30% SM). The amount of H₂O₂ decreased by 18, 24, and 45% after spraying the plants with 12.5, 25, and 50 ppm, and by 22, 37, and 9% after watering. The reduction in the MDA content by 22, 13, and 17% was found after pea irrigation with 12.5, 25, and 50 ppm suspensions of B₂O₃ NPs and after foliar application by 20, 25, and 22%.

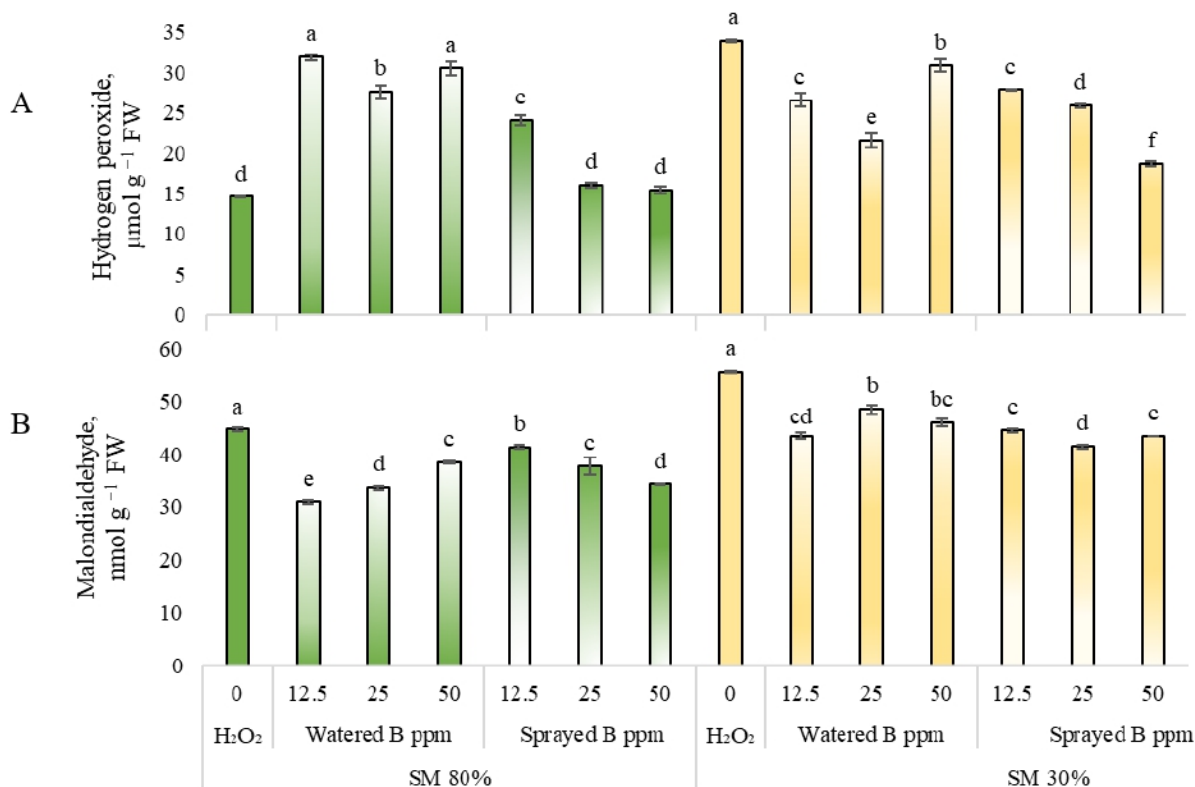


Figure 3.2.4.1. Influence of drought stress and B₂O₃ (B in the figure) NPs (0; 12,5; 25; 50 ppm) on hydrogen peroxide and malondialdehyde content in *P. sativum* L. H₂O – control plants watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Effects on non-enzymatic antioxidants

It was found that at 80% substrate moisture, both watering and spraying with B₂O₃ NPs reduced the TPC in pea leaves up to 30% (Figure 3.2.4.2A 80% SM). B₂O₃ NPs treatment did not affect ABTS free radical scavenging activity (Figure 3.2.4.2C 80% SM). However, it was determined that after spraying peas with 25 and 50 ppm suspensions, the DPPH free radical scavenging activity increased by 25 and 24% (Figure 3.2.4.2B 80% SM). Furthermore, concentrations of B₂O₃ NPs suspensions of 12.5, 25, and 50 ppm increased the FRAP antioxidant power (Figure 3.2.4.2D 80% SM): as plants were watered or sprayed.

The results showed that spraying drought-affected peas with 12.5, 25, and 50 ppm B₂O₃ NPs suspensions increased TPC content to 18% while watering with 12.5 significantly reduced it (Figure 3.2.4.2A 30% SM). ABTS free radical scavenging activity showed its sensitivity to the impact of B₂O₃ NPs (Figure 3.2.4.2C 30% SM); it increased to 73% after watering and 96% after spraying compared to drought-affected plants without NPs exposure. Similar results were found for FRAP antioxidant power in peas (Figure 3.2.4.2D 30% SM). The exposure of drought and B₂O₃ NPs 12.5 and 25 ppm suspension through the roots caused a slight impact (20%) on DPPH free radical scavenging activity (Figure 3.2.4.2B 30% SM). In addition, spraying with

12.5, 25, and 50 ppm B₂O₃ NPs suspensions induced DPPH free radical scavenging activity by 35, 24, and 25%, respectively.

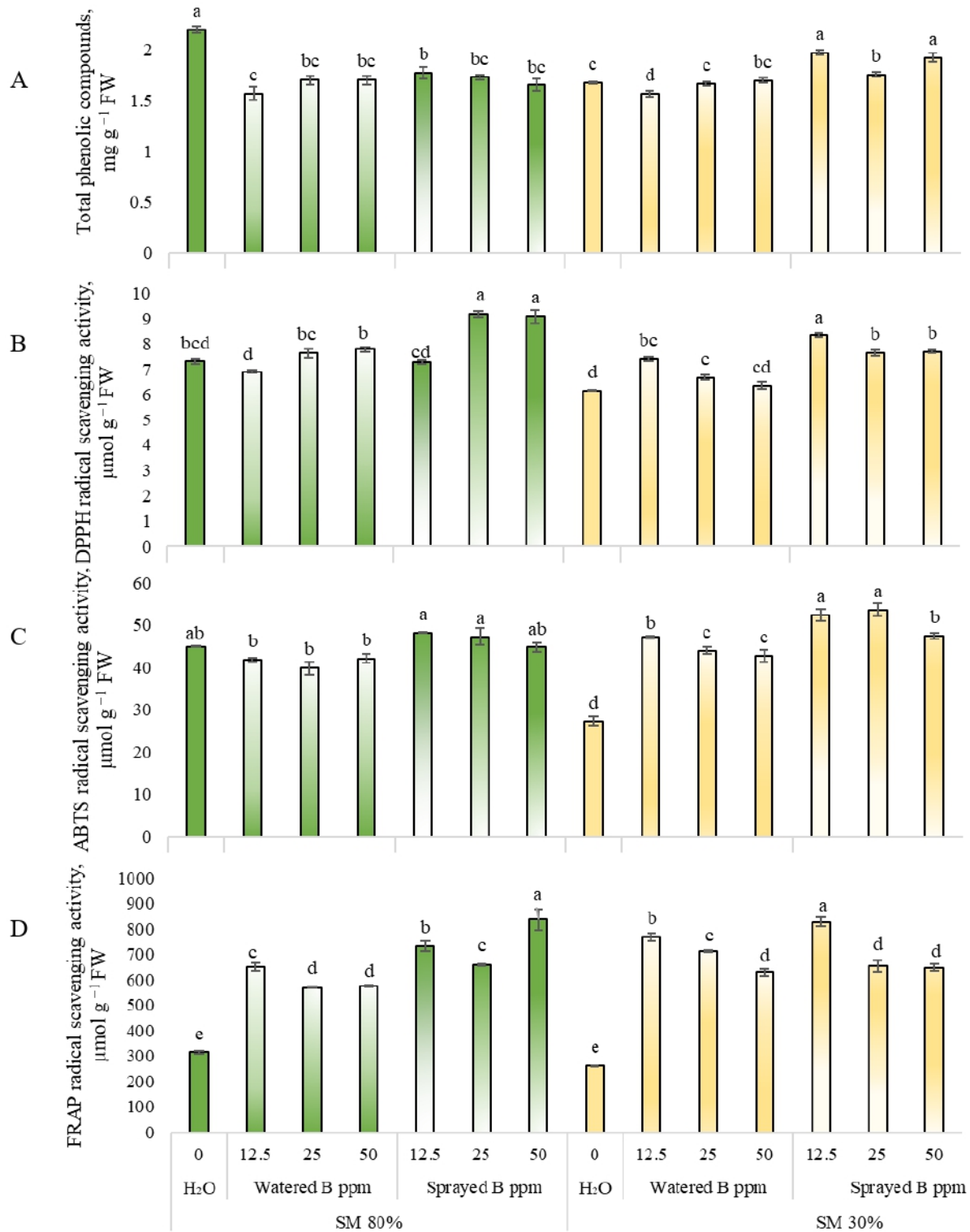


Figure 3.2.4.2. Influence of drought stress and B₂O₃ NPs (B in the figure) (0; 12.5; 25; 50 ppm) on A – total phenolic compounds, B – DPPH free radical scavenging activity, C – ABTS free radical scavenging activity D – FRAP antioxidant power in *P. sativum* L. H₂O – control plants, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean ± SE of three replicates, and different letters differed significantly by Tukey HSD Test (*p* < 0.05)

Effects on enzymatic antioxidants

B₂O₃ NPs induced the activities of CAT, APX, SOD, and GPX in pea leaves when they were grown in 80% SM (Figure 3.2.4.3A, B, D, E). APX activity increased particularly strongly after watering plants with B₂O₃ NPs suspensions, while a slightly weaker effect was caused by spraying. CAT activity increased up to 2 times when plants were watered with suspensions of B₂O₃ NPs. When peas were sprayed, the CAT activity increased by 1.3, 1.8, and 2 times when the concentration was 12.5, 25, and 50 ppm. SOD activity was induced up to 41% by exposure to B₂O₃ NPs through roots, and foliar treatment activated the enzyme up to 46%. GPX activity was distinguished because lower concentrations of 12.5 and 25 ppm had a more substantial positive effect during watering, while higher concentrations of 25 and 50 ppm increased activity more strongly during spraying. B₂O₃ NPs suspension reduced GR activity (Figure 3.2.4.3C 80% SM) when suspensions of concentrations 12.5 and 50 ppm were used for plant watering or spraying.

A substantial decrease in GR activity was caused by drought and B₂O₃ NPs exposure, with a 55% reduction after irrigation and a 45% reduction after spraying (Figure 3.2.4.3C 30% SM). Also, an adverse effect was found on SOD activity (Figure 3.2.4.3D 30% SM) after peas irrigation with 25 and 50 ppm solutions of B₂O₃ NPs, a 36% increase in SOD activity was determined after using 12.5 ppm B₂O₃ NPs suspension. Furthermore, SOD activity was induced up to 51% when drought-affected peas were sprayed with B₂O₃ NPs solution. The strong effect of B₂O₃ NPs on the APX activity (Figure 3.2.4.3B 30% SM) remained in pea leaves as plants were grown in drought conditions. After watering peas with B₂O₃ NPs suspensions, APX activation occurred up to 1.4 times after spraying up to 8 times. Additionally, GPX activity in drought-affected peas (Figure 3.2.4.3E 30% SM) was increased up to 91% after foliar exposure with all B₂O₃ NPs concentrations. CAT activity (Figure 13A 30% SM) was strongly activated by watering or spraying plants with B₂O₃ NPs solutions of any concentration.

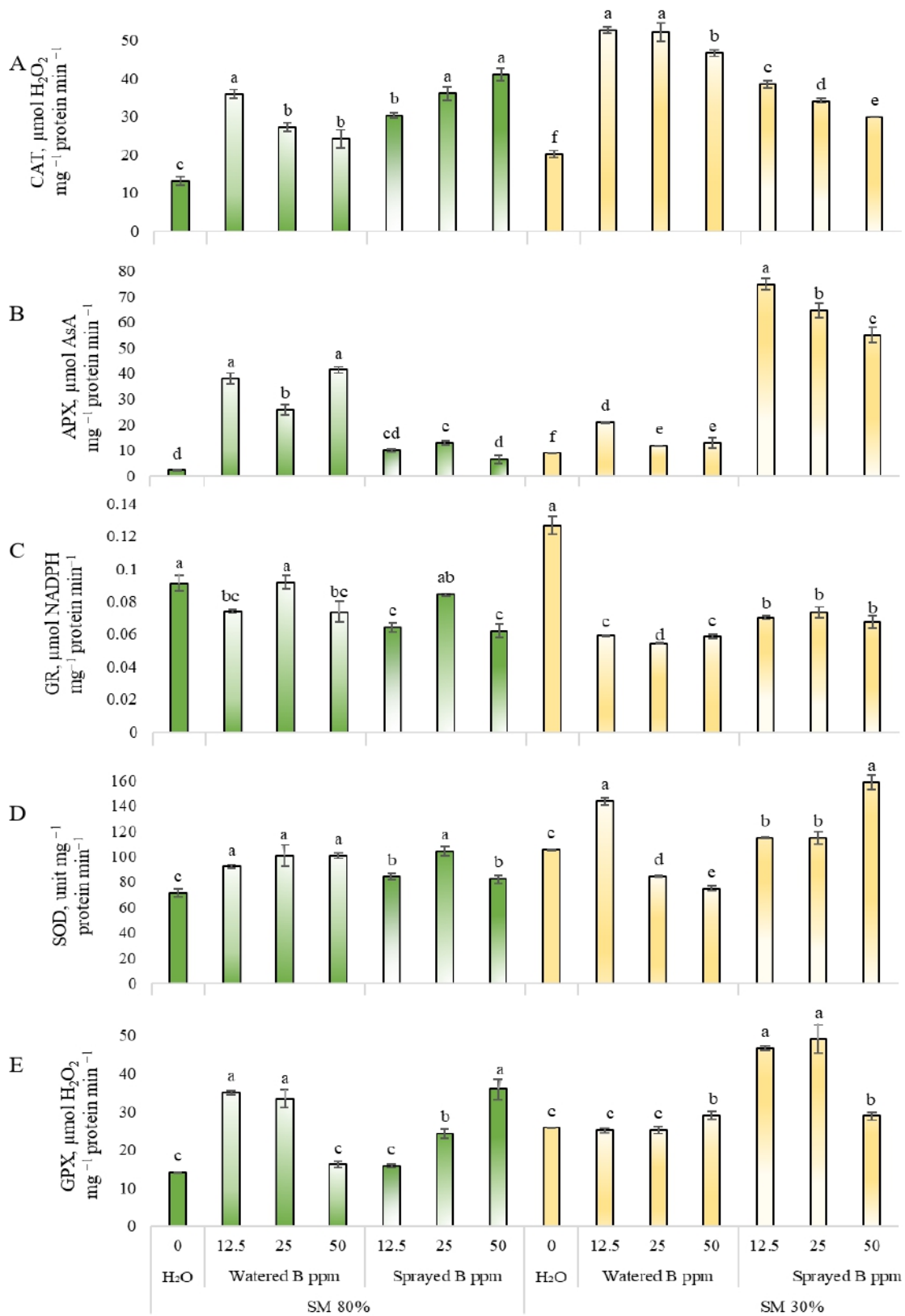


Figure 3.2.4.3. Response of A – ascorbate peroxidase (APX), B – catalase (CAT), C –superoxide dismutase (SOD), D –glutathione reductase (GR), and E –guaiacol peroxidase (GPX) activity to drought stress and B_2O_3 (B in the figure) NPs (0; 12,5; 25; 50 ppm) in *P. sativum* L. H_2O – control plants sprayed or watered with deionized water, substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean \pm SE of three replicates, and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Changes in macro- and micro-elemental composition

B content in peas grown in sufficient substrate moisture (Table 3.2.4.2, 80% SM) leaves increased by 2, 3, and 11 times after watering with B₂O₃ NPs suspensions of 12.5, 25, and 50 ppm after spraying with the same concentrations by 3, 5 and 12 times. B content increased 10 times in the stem after watering with B₂O₃ NPs solutions and up to 17 times after foliar application. Furthermore, a 3, 7, and 20 times increase in B content was found in the roots after watering plants grown during normal substrate moisture conditions with B₂O₃ NPs. Foliar application increased the amount of B in the roots from 1 to 3 times.

When drought-affected peas were sprayed with 25 and 50 ppm B₂O₃ NPs solutions, leaf B content increased 3 and 5 times, respectively. In addition, 3 times increase in B content in the leaves was found when peas were irrigated with suspensions of B₂O₃ NPs. Moreover, substantial increases (up to 18 times) in B content were found in stems after irrigating peas with suspensions containing B₂O₃ NPs. B amount increased 1.5, 3, and 4 times after watering or spraying with the suspensions containing 12.5, 25, and 50 ppm B₂O₃ NPs.

B₂O₃ NPs affected the accumulation of Ca, Mg, Fe, Zn, and Mn in pea leaves. Spraying with B₂O₃ NPs resulted in higher K concentration in the leaves regardless of the exposure. It was found that the application of B₂O₃ NPs decreased the content of Ca in plant stems during both normal substrate conditions and drought. Furthermore, the reduction of Mg content in plant leaves and stems was found after exposure to B₂O₃ NPs solutions. However, pea leaves exposed to water deficit Mo content increased to 7 times. K, Ca, Mg, and Fe increased in roots when peas were grown at 80% SM. Drought and 50 ppm B₂O₃ NPs suspension increased the P, Ca, Mg, Na, and Fe content in roots.

Table 3.2.4.2. Effects of B_2O_3 NPs suspension on macro- and microelements of peas leaves, stem, and root. Mean values in bold indicate a statistically significant difference at $p < 0.05$ according to Tukey (HSD) test ($n=9$)

Treatment B_2O_3 NPs, ppm			Macroelements, $mg\ g^{-1}$ DW						Microelements, $\mu g\ g^{-1}$ DW				
Leaves			P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B
SM 80%	Watered	0	3.79	18.76	45.73	9.20	2.64	19.86	25.71	42.54	10.32	2.99	3.58
		12.5	4.63	22.01	39.80	6.34	2.34	22.78	25.03	40.54	9.43	2.64	6.28
		25	2.87	17.22	32.84	5.69	2.08	16.00	18.50	33.35	5.84	0.27	9.52
	Sprayed	50	4.05	19.65	36.95	6.43	1.51	16.41	22.54	35.32	6.27	1.63	33.20
		12.5	4.63	22.01	39.80	6.34	2.34	22.78	25.03	40.54	9.43	2.64	9.52
		25	3.36	19.22	42.14	6.17	2.33	23.64	23.93	44.82	8.78	1.01	14.07
SM 30%	Watered	50	4.16	19.05	47.94	8.04	3.92	32.44	24.72	50.31	8.36	1.21	35.14
		0	3.21	20.56	46.19	8.81	2.07	20.03	37.24	37.65	10.02	0.70	11.66
		12.5	4.14	21.69	44.88	8.74	2.02	22.28	35.03	48.08	11.52	3.34	30.52
	Sprayed	25	4.07	19.84	47.46	8.23	1.30	22.62	35.06	44.97	13.53	5.30	36.38
		50	5.37	24.67	47.94	8.87	1.61	25.63	43.90	49.17	16.07	4.49	38.23
		12.5	2.22	23.66	36.52	6.09	2.62	18.84	17.67	27.98	13.08	3.47	10.40
SM 80%	Watered	25	4.49	21.91	49.75	8.55	1.51	21.30	39.24	42.55	15.95	5.82	31.71
		50	4.29	21.83	47.76	9.00	1.56	20.81	42.85	49.19	16.90	4.48	53.25
		0	2.00	17.77	37.81	8.71	7.73	15.49	19.81	26.09	10.08	3.66	1.05
	Sprayed	12.5	1.71	15.97	34.96	5.10	4.04	17.45	17.66	26.81	13.08	5.02	3.25
		25	1.44	13.77	28.02	4.32	4.73	10.66	11.81	20.91	9.39	3.55	9.37
		50	2.21	17.38	33.61	5.44	5.96	17.34	16.73	24.64	10.76	3.25	11.71
SM 30%	Watered	12.5	1.87	16.49	29.69	6.02	4.20	10.62	12.38	19.37	7.71	1.23	3.37
		25	1.24	15.42	30.53	6.52	5.61	12.25	12.04	19.65	8.23	1.25	16.86
		50	1.81	14.98	33.27	6.99	5.65	13.58	17.09	23.33	7.07	1.75	18.56
	Sprayed	0	1.43	20.13	38.10	7.98	6.83	15.46	22.31	24.19	8.92	1.17	10.57
		12.5	1.71	20.38	33.69	5.72	5.91	17.62	19.07	29.16	12.00	3.26	12.40
		25	1.50	20.41	30.73	5.14	6.06	11.02	15.44	22.76	11.17	2.98	17.27
Watered	50	2.05	22.79	34.61	5.96	5.48	20.04	16.95	27.47	12.16	2.97	19.39	

	Sprayed	12.5	2.22	23.66	36.52	6.09	7.62	18.84	17.67	27.98	13.08	3.47	9.65
		25	2.17	21.02	38.79	5.88	5.72	17.45	24.28	29.92	13.71	4.21	10.40
		50	1.74	23.02	33.99	5.73	7.13	14.66	16.47	24.82	11.23	2.62	11.11
	Root		P	K	Ca	Mg	Na	Fe	Zn	Mn	Cu	Mo	B
SM 80%	Watered	0	0.81	16.16	10.25	6.91	3.18	258.88	15.88	46.44	18.46	9.40	2.68
		12.5	1.35	21.65	13.90	10.16	3.66	752.51	27.30	133.15	19.50	16.38	11.90
		25	1.40	21.30	13.11	9.61	4.91	532.65	23.54	100.79	17.05	10.79	22.36
	Sprayed	50	1.95	24.23	13.64	9.21	5.57	400.15	33.52	78.58	17.35	6.56	57.91
		12.5	1.73	19.96	12.67	9.44	4.50	548.87	23.91	95.76	16.32	12.93	5.00
		25	1.99	28.53	19.54	14.55	6.28	656.22	29.92	115.93	23.04	15.34	8.09
		50	1.94	25.32	14.36	10.01	5.39	547.88	35.78	97.03	21.05	10.78	11.11
SM 30%	Watered	0	1.58	28.33	19.57	15.45	7.33	657.85	34.15	118.04	22.49	16.58	6.70
		12.5	1.58	29.73	18.37	12.77	7.41	570.32	26.72	102.34	19.85	8.54	9.84
		25	1.88	28.24	19.24	12.89	5.76	693.32	29.81	123.37	22.36	10.63	17.91
	Sprayed	50	2.21	33.13	23.25	14.22	7.96	852.68	98.65	150.67	23.30	12.23	23.68
		12.5	1.91	35.33	19.62	12.79	7.05	642.51	29.47	113.23	23.58	10.32	10.78
		25	2.52	41.54	25.40	16.58	7.84	1035.42	44.91	187.73	29.58	18.86	17.32
		50	2.39	39.01	25.40	17.32	8.59	1044.14	46.56	196.38	29.07	19.95	21.22

Macroelements: P – phosphorus, K – potassium, Ca – calcium, Mg – magnesium, Na – sodium, Fe – iron; Microelements: Zn – zinc, Mn – manganese, Cu – copper, Mo – molybdenum, B – boron. 0 – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%.

Summary of the chapter (Table 3.2.4.3)

- Spraying with B₂O₃ NPs at a 12.5 ppm most effectively stimulated TPC accumulation, antioxidant capacity, and APX, SOD, and GPX enzyme activity in pea leaves exposed to drought.
- B₂O₃ NPs reduced the amount of H₂O₂ and MDA in pea plants grown on a substrate with insufficient moisture.
- B₂O₃ accumulation was most significant in drought-affected peas when irrigated with 25 and 50 ppm B₂O₃ NPs suspensions.
- The most substantial positive effect was found on peas affected by drought after spraying them with 12.5 ppm of B₂O₃ NPs.

Table 3.2.4.3. The impact of drought stress and B₂O₃ NPs (12.5; 25; 50 ppm) on *P. sativum* L. grown in the substrate with sufficient (SM 80%) and insufficient (SM 30%) moisture is expressed as a percentage change (%) compared to the control (for SM 80% control means plants grown under SM 80% and NPs untreated; SM 30% control means drought affected but NPs untreated plants) in the heat map.

Statistically, significant differences are marked in bold

Treatment B ₂ O ₃ NPs, ppm	SM 80%						SM 30%					
	Watered			Sprayed			Watered			Sprayed		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Plant height	14	7	27	28	9	19	9	26	17	13	20	14
Leaf area	28	9	30	38	25	24	-17	30	40	-4	1	-6
Nodules	560	200	480	340	240	200	17	200	317	0	67	583
RWC	2	4	5	6	4	4	-1	10	11	-2	4	9
Root/shoot	21	36	68	69	18	34	4	14	30	11	14	5
SLA	14	-5	-11	-15	-6	-3	-17	-4	-16	-13	-23	-29
Yield	-21	2	20	12	11	-1	16	10	6	19	16	14
ABTS	-7	-11	-6	7	5	0	73	61	56	92	96	74
DPPH	-5	4	7	0	25	24	20	8	3	35	24	25
TPC	-29	-22	-22	-19	-21	-25	-6	-1	1	18	5	15
FRAP	106	81	83	132	109	166	194	174	141	217	151	148
HP	119	89	109	65	10	6	-22	-37	-9	-18	-24	-45
MDA	-31	-25	-14	-7	-15	-23	-22	-13	-17	-20	-25	-22
GR	-19	0	-19	-30	-8	-32	-53	-57	-54	-45	-42	-47
GPX	147	136	14	11	71	153	-3	-2	13	81	91	12
APX	1657	1100	1817	363	498	200	136	33	46	750	634	522
SOD	29	41	41	18	46	15	36	-20	-29	9	9	51
CAT	173	107	84	131	175	214	161	159	132	91	69	48
B (leaves)	18	16	7	8	4	20	-3	36	52	-18	-20	-24
B (stem)	210	794	1017	221	1508	1670	17	63	83	-9	-2	5
B (roots)	344	735	2062	87	202	315	47	167	253	61	158	217

RWC – relative water content, SLA – specific leaf area, TPC – total phenolic compounds, HP – hydrogen peroxide, MDA – malondialdehyde, GR – glutathione reductase, GPX – guaiacol peroxidase, APX – ascorbate peroxidase, SOD – superoxide dismutase, CAT – catalase, B content in leaves, stem, and root. 0 – control plants watered with deionized water, drought stress – 30% substrate moisture.

3.3 Effects of nanoparticles on peas exposed to the combined stress of drought and heavy metal copper

In field conditions, plants can experience several different abiotic stresses simultaneously, for example: decreased water availability during drought or increased excessively during a flood, extreme temperatures, including heavy metals, or increased soil hardness limiting root growth. Different stresses cause different responses in plants, which requires plants to use other acclimation processes. When several stresses affect plants simultaneously, their adverse effects overlap, and the plant suffers stronger strains. This chapter will analyze in detail the impact of NPs on combined drought, and heavy metal copper (Cu) stresses on pea plants.

This section discusses the results obtained between plants affected by drought and heavy metal copper and plants affected by drought, heavy metal copper, and NPs. The subsections are divided according to the NPs. The first paragraph discussed the watered NPs effects on drought, and heavy metal copper exposed plants. The second one is the foliar NPs application on drought, and heavy metal copper exposed plants.

3.3.1 Effects of silica nanoparticles on peas exposed to the combined stress of drought and heavy metal copper

SiO₂ NPs strongly affected peas to protect them from the combined effects of drought and Cu (Table 3.3.1.1). Pea height, root-to-shoot ratio, RWC, and yield increased by 24, 86, 24, and 50%, respectively, when plants were watered with SiO₂ NPs under drought and Cu treatment (WxDxCu). The activity of DPPH free radical scavenging and FRAP were significantly increased by 21 and 10% and the amount of TPC by 21% when plants were affected by WxDxCu. Besides, SiO₂ NPs stimulated GR and CAT activity by 50 and 67%, respectively, when plants were grown under WxDxCu. Irrigation with SiO₂ NPs reduced H₂O₂ and MDA content in leaves by 12 and 14% when plants were treated by WxDxCu. An increase in BCF was found in leaves by 75% and in roots by 94%, as well as in Tf by 1.5 times and Ti by 85%.

Spraying with SiO₂ NPs positively affected peas as they were affected by drought and Cu (SxDxCu). In these plants, an increase in height, RWC, root-to-shoot ratio, and yield was determined by 34, 23, 99, and 44%, respectively. Also, SiO₂ NPs positively affected antioxidant capacity (ABTS, DPPH, FRAP) and increased TPC content in pea leaves during these conditions SxDxCu. However, the spraying of SiO₂ NPs strongly reduced the activities of enzymes such as GPX, APX, and SOD in peas exposed to drought and Cu. In addition, spraying with SiO₂ NPs significantly reduced the content of H₂O₂ and MDA by 57 and 55%. An increase in BCF was found in leaves by 99% and in roots by 81%, as well as in Tf by 200% and Ti by 90% after exposure to SxDxCu.

Table 3.3.1.1. The impact of drought stress, Cu, and SiO₂ NPs (50 ppm) on *P. sativum* L. is presented. H₂O – control plants sprayed or watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean and different letters differed significantly by Tukey HSD Test ($P < 0.05$).

Treatment with SiO ₂ NPs, 50 ppm	80% SM	30 % SM					
	H ₂ O	H ₂ O	Cu	Watered NPs	Sprayed NPs	Cu+Watered NPs	Cu+Sprayed NPs
Morphological parameters							
Plants height	32.9 a	22.2 cd	20.0 d	28.3 ab	30.4 a	24.9 c	26.8 bc
Leaf area	76.9 a	49.9 c	43.9 cd	55.9 b	62.7 ab	44.8 d	50.4 c
No. of nodules	8.3 a	3.7 d	2.0 e	6.0 bc	6.3 b	3.6 d	2.6 de
RWC	80.6 a	41.7 c	29.3 e	47.3 b	45.8 bc	36.3 cd	35.9 cd
Root/shoot	11.3 a	5.4 bc	4.1 c	6.0 bc	8.9 a	6.8 bc	8.3 ab
SLA	9.5 bc	11.0 ab	13.5 a	8.5 b	9.4 bc	8.4 b	9.6 bc
Yield	3.4 a	2.4 b	1.8 c	3.3 a	2.6 b	2.7 b	2.6 b
Non-enzymatic antioxidant activity							
ABTS	113.2 a	103.1 b	96.4 bc	111.4 a	109.1 ab	93.9 c	106.6 a
DPPH	45.1 ab	38.5 c	36.8 cd	45.9 ab	49.2 a	44.5 ab	44.8 ab
TPC	2.7 ab	2.3 bc	1.9 d	2.7 ab	2.9 a	2.3 bc	2.4 b
FRAP	1475.8 a	1158.6 b	939.5 d	1380.3 a	1405.9 a	1027.6 bc	1117.3 bc
Enzymatic antioxidant activity							
Gr	0.053 c	0.074 bc	0.084 b	0.041 c	0.027 c	0.126 a	0.094 b
GPX	11.1 c	15.5 b	44.9 a	8.4 d	8.1d	19.3 b	10.6 cd
APX	25.1 e	51.3 b	74.3 a	61.5a	55.6 ab	48.2 cd	34.5 d
SOD	46.8 c	67.2 b	74.5 a	73.9 a	79.7 a	14.7 de	11.1 e
CAT	16.7 d	31.9 c	33.1 c	37.7 ab	15.2 d	55.8 a	43.2 ab
Oxidative stress biomarkers							
HP	12.3 e	25.5 ab	26.2 a	21.6 bc	19.6 d	23.1 b	20.3 bc
MDA	54.7 f	62.3 de	97.3 a	89.9 b	76.8 d	83.4 bc	74.7 d
Copper accumulation							
BCF (leaves)			0.04 b			0.07 a	0.08 a
BCF (roots)			1.6 b			3.1 a	2.9 a
Tf			2.1 b			5.2 a	6.1 a
Ti			62.7 c			116.2 b	119.3 a

BCF – bioconcentration factor, Tf – translocation factor, Ti tolerance index

Summary of the chapter

- *SiO₂ NPs increased the morphological parameters of peas, stimulated DPPH free radicals scavenging activity and antioxidant power of FRAP, and stimulated the activity of GR and CAT enzymes when plants were exposed to combined stress. Reduced oxidative stress biomarkers (H₂O₂, MDA) in peas exposed to drought, Cu, and watered SiO₂ NPs. Irrigation with SiO₂ NPs increased the TF and Ti.*

- *SiO₂ NPs increased the morphological parameters of peas and induced positive effects on antioxidant capacity and TPC content. Reduced oxidative stress biomarkers in peas exposed to drought, Cu, and sprayed with SiO₂ NPs. Foliar application with SiO₂ NPs increased the TF and Ti.*

- *Pea irrigation with SiO₂ NPs has more statistically reliable effects when peas are affected by drought (Table 3.3.1.2).*

- *More statistically reliable effects were found for peas exposed to drought, Cu, and spraying with SiO₂ NPs (Table 3.3.1.2).*

*Table 3.3.1.2. The impact of drought stress, Cu, and SiO₂ NPs (50 ppm) on P. sativum L. is presented. * Significant difference p<0.05*

Treatment with SiO ₂ NPs, 50 ppm	Watered				Sprayed	
	D	D x Cu	D x NPs	D x Cu x NPs	D x NPs	D x Cu x NPs
Plants height	*	*	*	*	*	*
Leaf area	*	*	*		*	
Nodules	*	*	*	*	*	
RWC	*	*	*		*	*
Root/shoot	*	*				*
SLA		*		*		*
Yield	*	*	*			
ABTS	*	*	*		*	*
DPPH	*	*	*		*	
TPC		*			*	*
FRAP	*	*	*		*	*
Gr		*		*		
GPX	*	*	*	*	*	*
APX	*	*	*	*		*
SOD	*	*	*	*	*	*
CAT	*	*	*	*	*	*
HP	*	*	*	*	*	*
MDA	*	*	*	*		*

3.3.2 Effects of copper oxide nanoparticle on peas exposed to the combined stress of drought and heavy metal copper

A significant increase in the root-to-shoot ratio was found after peas were exposed to drought, Cu, and watered with CuO NPs (Table 3.3.2.1). Besides, a significant decrease was found in leaf area and yield. Furthermore, a reduction in FRAP antioxidant power and TPC content and no statistical significance for DPPH and ABTS free radicals scavenging activity in pea leaves were determined. In these plants, CuO NPs increased the activity of SOD, and no statistical reliability was found for GR, GPX, APX, and CAT enzymes. CuO NPs significantly reduced H₂O₂ content, but no effect on MDA content was detected in peas. Finally, CuO NPs increased BCF by 13% in roots, decreased Tf by 33%, and did not affect Ti.

When pea plants were exposed to drought and Cu, spraying with CuO NPs produced more positive results. An increase in RWC of up to 44% was found, and no differences were determined in all other morphological parameters. In these plants, stimulation of ABTS free radical scavenging activity and increase in TPC content were determined, while no differences were found in enzymatic antioxidants. However, there was a significant decrease in H₂O₂ and MDA in pea leaves. Foliar CuO NPs application did not affect both leaf and root BCF and Ti but significantly reduced Tf.

Table 3.3.2.1. The impact of drought stress, Cu, and CuO NPs (50 ppm) on *P. sativum* L. is presented. H₂O – control plants sprayed or watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean (n=9) and different letters differed significantly by Tukey HSD Test ($P < 0.05$)

Treatment with CuO NPs, 50 ppm	80% SM		30 % SM				
	H ₂ O	H ₂ O	Cu	Watered NPs	Sprayed NPs	Cu+Watered NPs	Cu+Sprayed NPs
Morphological parameters							
Plants height	32.9 a	22.2 cd	20.0 d	25.0 b	24.3 b	19.9 d	21.8 cd
Leaf area	76.9 a	49.9 b	43.9 c	40.6 cd	46.8 bc	24.3 e	40.2 cd
Nr. Of nodules	8.3 a	3.7 b	2.0 cd	2.1 cd	3.1 c	1.7 d	1.9 cd
RWC	80.6 a	41.7 bc	29.3 d	48.1 b	47.2 b	38.2 c	42.2 bc
Root/shoot	11.3 a	5.4 bc	4.1 c	10.2 a	6.8 b	8.1 b	6.1 bc
SLA	9.5 cd	11.0 c	13.5 ab	10.0 c	9.2 cd	13.2 b	14.7 a
Yield	3.4 a	2.4 b	1.8 c	2.4 b	2.7 ab	1.4 de	1.6 cd
Non-enzymatic antioxidant activity							
ABTS	113.2 a	103.1 ab	96.4 bc	108.7 b	111.5 a	90.7 bc	110.2 a
DPPH	45.1 a	38.5 b	36.8 c	36.6 c	35.9 c	35.2 c	35.3 c
TPC	2.7 a	2.3 bc	1.9 c	2.6 ab	2.6 ab	1.2 d	2.4 b
FRAP	1475.8 a	1158.6 b	939.5 de	1082.9 d	1201.8 c	683.4 f	890.0 e

Enzymatic antioxidant activity							
Gr	0.053 c	0.074 b	0.084 ab	0.073 b	0.066 bc	0.094 a	0.083 ab
GPX	11.1 f	15.5 e	44.9 ab	21.4 cd	19.6 cd	47.0 a	37.5 b
APX	25.1 d	51.3 c	74.3 a	63.0 b	63.7 b	70.3 ab	72.3 ab
SOD	46.8 e	67.2 d	74.5 cd	133.3 a	82.6 c	105.8 b	73.8 cd
CAT	16.7 d	31.9 bc	33.1 b	28.0 c	38.5 a	34.0 ab	32.2 bc
Oxidative stress biomarkers							
HP	12.3 d	25.5 ab	26.2 ab	24.8 ab	22.4 bc	19.7 c	20.1 c
MDA	54.7 e	62.3 d	97.3 a	84.5 b	81.9 bc	97.1 a	73.3 c
Copper accumulation							
BCF (leaves)			0.04 ab			0.05 ab	0.06 a
BCF (roots)			1.6 b			1.8 a	1.7 ab
Tf			2.1 a			1.4 c	1.7 b
Ti			62.7 ab			60.2 b	63.0 a

BCF – bioconcentration factor, Tf – translocation factor, Ti – tolerance index

Summary of the chapter

- *CuO NPs decreased FRAP antioxidant power and TPC content in peas but increased SOD activity when peas were exposed to drought, Cu, and watered with CuO NPs.*
- *An increase in ABTS radical scavenging activity and TPC content in peas was determined, and a decrease in H₂O₂ and MDA content when peas were exposed to drought, Cu, and sprayed with CuO NPs.*
- *CuO NPs had a more significant effect on peas exposed only to drought than drought and Cu affected but NPs untreated plants. More statistically reliable effects on antioxidants were found after foliar CuO NPs application (spraying) (Table 3.3.2.2).*
- *During the combined stress of drought, Cu, and CuO NPs, CuO NPs did not positively affect peas.*

Table 3.3.2.2. The impact of drought stress, Cu, and CuO NPs (50 ppm) on *P. sativum* L. is presented. * Significant difference $p < 0,05$

Treatment with CuO NPs, 50 ppm	Watered				Sprayed	
	D	D x Cu	D x NPs	D x Cu x NPs	D x NPs	D x Cu x NPs
Plants height	*	*	*		*	
Leaf area	*	*	*	*		
Nodules	*	*	*		*	
RWC	*	*	*	*		*
Root/shoot	*			*		
SLA	*	*				
Yield	*	*			*	*
ABTS		*				*
DPPH	*	*	*		*	
TPC	*	*		*		*
FRAP	*	*	*	*	*	
Gr	*					
GPX	*	*	*		*	
APX	*	*	*		*	
SOD	*		*	*	*	
CAT	*				*	
HP	*	*				
MDA	*	*	*		*	*

3.3.3. Effects of molybdenum trioxide nanoparticles on peas exposed to the combined stress of drought and heavy metal copper

The results presented in Table 3.3.3.1 show the complex effects of drought, Cu, and MoO₃ NPs on peas. Pea height increased by 29%, leaf area by 28%, the number of nodules by 500%, RWC by 44%, yield by 88%, root to shoot ratio by 120%, and no effect was found on SLA when peas were exposed to drought, Cu and watered with 50 ppm MoO₃ NPs. A significant increase in the scavenging capacity of ABTS, DPPH, and FRAP radicals was determined. Besides, TPC content in peas increased by over 35%. Activation was found in GPX and CAT enzymes, but inhibition in GR and SOD activity was determined in peas exposed to drought, Cu, and watered with MoO₃ NPs. In addition, a 22% reduction in H₂O₂ and MDA content was found in pea leaves. MoO₃ NPs increased the Tf by 44% and the Ti by 76%.

Statistically reliable positive changes in the number of nodules, RWC, and yield were determined when peas were exposed to drought, Cu, and sprayed with MoO₃ NPs. The antioxidant capacity (DPPH, ABTS, and FRAP) increased to 13%, and TPC content in pea leaves increased by 32% after foliar application of MoO₃ NPs on plants exposed to drought and Cu. There, a 17% inhibitory effect was found only on GR activity, while others, GPX, APX, SOD, and CAT activity increased by 10, 35, 36, and 20%, respectively. A 23% reduction in

MDA content was found. Besides, the BCF in the roots increased by 25% and Ti - by 13%, while differences were not determined for Tf.

Table 3.3.3.1. The impact of drought stress, Cu, and MoO₃ NPs (50 ppm) on *P. sativum* L. is presented. H₂O – control plants watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean and different letters differed significantly by multiple comparisons of Tukey HSD Test ($P < 0.05$).

Treatment with MoO ₃ NPs, 50 ppm	80% SM		30 % SM				
	H ₂ O	H ₂ O	Cu	Watered NPs	Sprayed NPs	Cu+Watered NPs	Cu+Sprayed NPs
Morphological parameters							
Plants height	32.9 a	22.2 c	20.0 d	27.8 b	20.0 d	25.8 bc	23.7 bcd
Leaf area	76.9 a	49.9 c	43.9 cd	52.1 bc	43.9 cd	56.5 b	43.5 c
Nr. of nodules	8.3 b	3.7 d	2.0 e	7.0 bc	2.0 d	11.7 a	7.3 bc
RWC	80.6 a	41.7 c	29.3 e	49.4 b	43.0 bc	43.8 bc	33.6 d
Root/shoot	11.3 a	5.4 bc	4.1 c	6.1 bcd	9.3 ab	8.9 ab	6.1 bc
SLA	9.5 abc	11.0 ab	13.5 a	9.3 bc	14.0 a	9.4 ab	8.5 d
Yield	3.4 ab	2.4 c	1.8 d	4.2 a	3.8 a	3.4 ab	3.3 ab
Non-enzymatic antioxidant activity							
ABTS	113.2 a	103.1 bc	96.4 e	110.3 a	96.4 e	105.0 bc	101.8 d
DPPH	45.1 a	38.5 c	36.8 d	42.2 b	36.9 d	42.5 b	41.8 b
TPC	2.7 a	2.3 cd	1.9 d	2.5 b	1.8 e	2.5 b	2.4 bc
FRAP	1475.8 a	1158.6 c	939.5 de	1216.2 b	924.6 de	1184.2 b	1058.2 c
Enzymatic antioxidant activity							
Gr	0.053 c	0.074 b	0.084 a	0.089 a	0.070 b	0.072 b	0.070 b
GPX	11.1 c	15.5 bc	44.9 b	15.1 bc	19.6 bc	48.1 a	49.3 a
APX	25.1 d	51.3 c	54.3 c	58.2 b	74.5 a	71.8 ab	72.9 ab
SOD	46.8 e	67.2 c	74.5 b	73.3 b	74.2 b	64.4 cd	101.2 a
CAT	16.7 f	31.9 e	33.1 e	44.4 b	48.9 a	35.3 d	39.6 c
Oxidative stress biomarkers							
HP	12.3 e	25.5 bc	26.2 b	21.9 d	26.2 b	30.2 a	25.4 bc
MDA	54.7 e	62.3 c	97.3 a	58.6 d	63.3 c	75.7 b	75.3 b
Copper accumulation							
BCF _{Cu} (leaves)			0.04 b			0.06 a	0.05 ab
BCF _{Cu} (roots)			1.6 c			2.3 a	1.8 b
Tf _{Cu}			2.1 b			3.7 a	1.9 b
Ti _{Cu}			62.7 c			131.8 a	112.5 b

BCF – bioconcentration factor, Tf – translocation factor, Ti – tolerance index

Summary of the chapter

- *An increase in morphological parameters, antioxidant capacity, as well as GPX and CAT enzymes, an increase in the amount of TPC, and a decrease in the amount of H₂O₂ and MDA in peas irrigated with MoO₃ NPs suspension and exposed to drought with Cu were determined.*

- *Stimulation of the activity of ABTS, DPPH, and FRAP radicals scavenging, as well as GPX and APX, SOD, and CAT enzymes, increased TPC content, and decreased MDA content were found in peas sprayed with MoO₃ NPs suspension and exposed to drought with Cu. (Table 3.3.3.2).*

- *Irrigation with MoO₃ NPs had more statistically reliable effects on the results of peas exposed to drought as well as on the results of peas exposed to the combined stresses (drought, MH_{Cu}, and NPs).*

*Table 3.3.3.2. The impact of drought stress, Cu, and MoO₃ NPs (50 ppm) on P. sativum L. is presented. * Significant difference p<0,05*

Treatment with MoO ₃ NPs, 50 ppm	Watered				Sprayed	
	D	D x Cu	D x NPs	D x Cu x NPs	D x NPs	D x Cu x NPs
Plants height	*	*	*	*		
Leaf area	*	*		*		
Nodules	*	*		*		*
RWC	*	*	*	*		*
Root/shoot	*	*				
SLA		*		*		*
Yield	*	*	*	*	*	*
ABTS	*	*	*	*	*	*
DPPH	*	*	*	*	*	*
TPC	*	*	*	*	*	*
FRAP	*	*	*	*	*	*
Gr	*	*	*	*		*
GPX		*				*
APX	*	*	*		*	
SOD	*	*	*	*	*	*
CAT	*	*	*	*	*	*
HP	*	*	*		*	
MDA		*	*	*	*	*

3.3.4. Effects of boron nanoparticle on peas exposed to the combined stress of drought and heavy metal copper

Based on the results presented in Table 3.3.4.1, B₂O₃ NPs positively affect the morphological parameters of plants exposed to complex stress. After watering peas (affected by drought and Cu) with B₂O₃ NPs, it was found that the number of nodules increased by 1.5 times, RWC by 39%, root and shoot ratio by 75%, and yield by 56% compared to the data of peas grown in drought and Cu. Irrigation with a concentration of 12.5 ppm B₂O₃ NPs induced the activity of DPPH and FRAP radicals by 18 and 36% in complex-treated peas, but no statistically significant results were found on ABTS radical activity. The amount of TPC in pea leaves increased by 34%. SOD and CAT activities were positively influenced by 97 and 66% after watering complex-treated peas with 12.5 ppm B₂O₃ NPs while an inhibitory effect occurred on APX and GR enzymes. MDA concentration was suppressed by 22%, and no statistically significant difference was found for H₂O₂ content. No statistically reliable effect was found on the BCF. Still, an impact on the translocation factor was found after irrigation with B NPs it decreased by 14%, but the copper tolerance index increased by 42%.

Spraying 12.5 ppm B₂O₃ NPs suspension increased the resistance of peas to drought and heavy metal Cu by increasing pea height by 28%, RWC pre by 20%, root/shoot by 64%, and yield by 44%. The yield increase could be determined by a 20 and 10% increase in DPPH and FRAP radical activities and a 38% decrease in MDA content after spraying peas with B NPs compared to plants affected by drought and Cu. No statistically reliable results were found when analyzing ABTS radical activity, TPC content, Gr activity, and H₂O₂ content in pea leaves. Besides, a substantial increase of SOD over 57% and CAT 1.6 times activity was determined. Spraying peas with 12.5 ppm B₂O₃ NPs suspension affected the translocation factor and tolerance index by increasing them by 33 and 64%.

Table 3.3.4.1. The impact of drought stress, Cu, and B₂O₃ NPs (12.5 ppm) on *P. sativum* L. is presented. H₂O – control plants sprayed or watered with deionized water; substrate moisture (SM) 80%; drought stress – SM 30%. Values are mean and different letters differed significantly by Tukey HSD Test ($p < 0.05$)

Treatment with B ₂ O ₃ NPs, 12.5 ppm	80% SM		30 % SM				
	H ₂ O	H ₂ O	Cu	Watered NPs	Sprayed NPs	Cu+Watered NPs	Cu+Sprayed NPs
Morphological parameters							
Plants height	32.9 a	22.2 c	20.0 d	27.2 b	31.2 a	21.0 d	25.6 bc
Leaf area	76.9 a	49.9 c	43.9 cd	40.9 c	57.1 b	49.0 c	36.9 d
Nr. of nodules	8.3 a	3.7 b	2.0 c	5.0 b	7.0 a	5.0 a	3.7 bc
RWC	80.6 a	41.7 b	29.3 d	44.2 bc	38.1 bc	40.7 bc	35.3 bc
Root/shoot	11.3 a	5.4 bc	4.1 c	7.2 b	5.7 bc	7.2 b	6.7 b
SLA	9.5 bc	11.0 ab	13.5 a	7.4 d	11.1 ab	9.4 bc	8.5 cd
Yield	3.4 a	2.4 c	1.8 d	2.8 ab	2.7 ab	2.8 ab	2.6 ab
Non-enzymatic antioxidant activity							
ABTS	113.2 a	103.1 b	96.4 bc	112.3 a	112.1 a	91.9 c	94.9 c
DPPH	45.1 a	38.5 c	36.8 c	43.5 b	47.6 a	43.4 b	43.9 b
TPC	2.7 ab	2.3 cd	1.9 e	2.8 ab	2.9 a	2.6 bc	2.2 de
FRAP	1475.8 a	1158.6 b	939.5 de	979.6 d	1029.4 c	1280.3 a	1033.8 c
Enzymatic antioxidant activity							
Gr	0.053 b	0.074 a	0.084 a	0.031 c	0.078 a	0.050 b	0.073 a
GPX	11.1 d	15.5 c	44.9 b	65.2 a	49.4 b	15.1 c	13.6 cd
APX	25.1 f	51.3 d	74.3 a	70.0 ab	57.5 bc	66.5 b	34.2 e
SOD	46.8 e	67.2 d	74.5 cd	100.1 b	117.0 b	146.4 a	116.9 b
CAT	16.7 e	31.9 c	33.1 c	25.2 d	25.8 d	54.8 b	87.4 a
Oxidative stress biomarkers							
HP	12.3 c	25.5 ab	26.2 ab	26.4 ab	25.3 ab	23.8 b	26.7 ab
MDA	54.7 f	62.3 e	97.3 a	91.6 b	64.6 d	75.7 c	60.4 e
Copper accumulation							
BCF _{Cu} (leaves)			0.04 a			0.03 a	0.04 a
BCF _{Cu} (roots)			1.6 b			1.4 bc	1.9 ab
Tf _{Cu}			2.1 b			1.8 c	2.8a
Ti _{Cu}			62.7 c			89.2 b	103.0 a

BCF – bioconcentration factor, Tf – translocation factor, Ti – tolerance index

Summary of the chapter

- *Irrigation with B₂O₃ NPs positively affected morphological parameters, DPPH and FRAP values, TPC content, SOD, and CAT enzymes when peas were exposed to the stress complex.*

- *Foliar application with B₂O₃ NPs positively affected morphological parameters, DPPH and FRAP values, SOD and CAT enzymes, and reduced MDA content when peas were exposed to the combined stress.*

- *After watering with 12.5 ppm B₂O₃ NPs (D x Cu x NPs), the most statistically reliable changes in peas were determined compared to spraying (Table 3.3.4.2).*

Table 3.3.4.2. The impact of drought stress, Cu, and B₂O₃ NPs (12.5 ppm) on P. sativum L. is presented.

** Significant difference p<0,05*

Treatment with B ₂ O ₃ NPs, 12.5 ppm	Watered				Sprayed	
	D	D x Cu	D x NPs	D x Cu x NPs	D x NPs	D x Cu x NPs
Plants height	*	*	*	*	*	*
Leaf area	*	*	*	*	*	
Nodules	*	*	*	*	*	
RWC	*	*	*			
Root/shoot	*	*	*	*		
SLA		*	*	*		
Yield	*	*	*	*	*	*
ABTS	*	*		*	*	
DPPH	*	*		*	*	*
TPC	*	*		*	*	
FRAP	*	*	*	*	*	*
Gr	*	*	*	*		
GPX	*	*	*	*	*	
APX	*	*	*	*	*	*
SOD	*	*	*	*	*	*
CAT	*	*	*	*	*	*
HP	*	*		*		
MDA	*	*	*	*	*	*

4. DISCUSSION

This study elucidated the effects of SiO₂, MO₃, B₂O₃, and CuO NPs on pea plants grown under optimal agrometeorological drought and the combined stress of heavy metal Cu and agrometeorological drought conditions. The study considered NPs concentrations and various applications, such as foliar spray and root irrigation, to elucidate the underlying physiological, non-enzymatic, and enzymatic antioxidant mechanisms. The study showed that SiO₂, MO₃, B₂O₃, and CuO NPs positively affect plants grown in optimal conditions and reduce the harmful effects caused by drought and combined stresses. NPs physical and chemical properties, size, surface charge, concentration in suspensions, interaction with plants, and plant physiology determine the physical and chemical interaction of NPs with plants and can lead to the modification of specific membrane surface proteins, receptors, and transporters (Juárez-Maldonado et al., 2019). Biological membranes with hydrophobic and hydrophilic components and uneven fibers of lignin and cellulose create an uneven negative surface charge of -45 to -15 mV (Mittal et al., 2020). Negatively charged plant cell walls are observed to act as an ion exchange surface that potentially favors cationic permeation rather than anionic NPs. Considering the effect of irrigation when NPs reach plants through the soil (roots), it should be noted that soil particles are usually negatively charged, and NPs with a higher negative charge are more mobile in such soil. In contrast, positively charged NPs readily attract negatively charged soil surface particles. As the average soil grain size decreases, NP mobility decreases. The clay content in the soil can act as an anionic adjuvant, preventing the accumulation of NPs and increasing their mobility. Therefore, it is crucial to study the properties of NPs suspensions to understand the mechanisms of their entry and effect in the plant. NPs with a zeta potential of -10 to +10 mV are neutral and less stable than NPs with a zeta potential greater than +30 mV or less than -30 mV (strong cations and anions, respectively) (Clogston and Patri 2011).

NPs can cause various reactions in plants, increasing the activity of enzymes, promoting the conversion of nitrates into ammonia, intensifying the processes of respiration and photosynthesis, synthesizing enzymes and amino acids, enhancing carbon and nitrogen nutrition and/or directly affecting plant mineral nutrition. However, despite the potential advantages mentioned above, the use of nanotechnology in the agricultural sector is relatively limited and, compared to other industries, has not yet reached the market. Therefore, considering the possible practical application of NPs in the agricultural sector, a preliminary cost-benefit analysis should be considered, and risks to the environment and

human health should be adequately assessed before placing NPs-related products on the market.

Effects of different applications and concentrations of SiO₂ nanoparticles on pea plants

In our studies, the zeta potential of the aqueous suspension of SiO₂ NPs was -20.64 mV (Table 2.2.1.). Another scientific article reported that applying 10 g kg⁻¹ SiO₂ NPs resulted in a zeta potential of -40 mV. This indicates that the suspensions are stable and anionic.

Our study indicates that pea plant treatments with SiO₂ at higher concentrations (50 and 100 ppm) increased antioxidant activity and reduced H₂O₂ content and MDA in their leaves when grown under optimal conditions (Figure 3.1.2 A, B; Figure 3.2.1.1 A, B). In addition, the height of pea shoots, root length, and biomass accumulation was also increased when they were grown under normal conditions (Table 3.1.1; Table 3.2.1.1, 80% SM). A particular effect was observed on the accumulation of macroelements in pea plants; the content of P, K, Ca, and Mg in pea leaves and stems increased significantly (Table 3.2.1.2). SiO₂ NPs positively affected different plant species' growth, increased their biomass and physiological properties, modified tissue differentiation, activated antioxidant systems, and helped to adapt to stressful conditions (Luyckx et al. 2017). In general, the effects of SiO₂ NPs are widely studied in different plants (Lu et al., 2007; Janmohammadi et al., 2015). However, the SiO₂ NPs effects of the legume family have only been studied in lentils and soybeans. It has been reported that the nano-sized mixture of SiO₂ increased the nitrate reductase activity, stimulated the antioxidant system, enhanced the absorption and utilization of water and fertilizers, and accelerated the germination and growth of soybean (*Glycine max*) (Lu et al. 2007). Previous studies suggest that the higher SiO₂ NPs concentrations should be used with caution as an inhibitory effect of 120.16 ppm concentration of NPs on lentils germination was found (Janmohammadi et al. 2015).

Our established results indicate that deficiency of water causes a reduction in the pea plant's height, leaf area, SLA, RWC, and an increase in the root-to-shoot ratio (Table 3.2.1.1), and it is in agreement with other researchers (Arafa et al., 2021; Khatun et al., 2021; Bangar et al., 2019). It should be noted that after irrigating with NPs suspensions, SiO₂ NPs had a more significant effect on peas grown with optimal substrate moisture than sprayed peas. Peas have a group of aquaporins MIP - Si influx transporters (SiT1 and SiT2) and efflux transporters (SiT6) located in the central and lateral roots, facilitating the entry of Si into the xylem and enabling it to move freely in the plant (Maurel et al., 2015; Raoi and

Susmitha, 2017). However, in the case of drought, the movement of materials in the xylem is disrupted, and, therefore, Si irrigation becomes less effective. However, SiO₂ NPs foliar exposure may form an additional layer on the leaves, protecting plants from transpiration and turgor pressure changes, and allow SiO₂ NPs to penetrate the wax layer and diffuse directly into the plant and therefore have a more substantial positive effect against drought.

Based on the obtained data, we claim that SiO₂ NPs successfully activate antioxidant activity (Figure 3.2.1.2; Figure 3.2.1.3), thereby reducing the effects of oxidative stress on drought-affected peas and preserving their yield. SiO₂ NPs have a strong relationship with water content in plants, but there are just a few published scientific articles specifically on the effects of drought. For instance, spraying with Si NPs during drought stress increased the activity of CAT, APX, SOD, and GR enzymes and decreased the MDA and H₂O₂ in strawberries (Zahedi et al., 2020). In addition, an increase in leaf greenness, relative water content, and yield was found in wheat grown with SiO₂ NPs under drought conditions (Behboudi et al., 2018).

The relationship between Si NPs and abiotic factors such as salinity, exposure to heavy metals, and UV radiation in plants was also explored. Spraying salinity-treated potato seedlings with 50 ppm SiO₂ NPs suspension, the researchers found (Gowayed et al., 2017) that enzymatic antioxidants such as GPX and SOD were more active compared to potatoes untreated with SiO₂ NPs solution. SiO₂ (60.08 ppm) NPs significantly mitigated the adverse effects of salt stress on lentil seedling weight, germination, shoot and root length, and seedling vigor indices (Janmohammadi et al. 2015). Furthermore, SiO₂ NPs promoted the antioxidant system of soybeans, including the activity of SOD, POD, and CAT enzymes (Shen et al. 2010). A recent publication (Ismail et al., 2022) analyzed the effects of SiO₂ NPs on peas grown under salinity stress. Researchers found that SiO₂ NPs improved peas' morphological parameters and yielded under normal and salinity conditions. Additionally, they observed that SOD and POD genes were down-regulated at 200 mM salinity, but at higher salinity levels, these genes were induced by SiO₂ NPs. The CAT gene was strongly up-regulated by SiO₂ NPs regardless of salinity level. This was consistent with the activity of the enzymes themselves. The increase in H₂O₂ and MDA content was found in peas after exposure to heavy metal Cr (VI), but spraying them with Si NPs resulted in enzymatic antioxidants such as SOD, CAT, GR, APX stimulation and a decrease in oxidative stress biomarkers content (Tripathi et al., 2015). Moreover, when heavy metal Cd-treated common beans were sprayed with 30 ppm SiO₂ NPs, the activation of antioxidant enzymes and a decrease in the concentration of MDA and H₂O₂ were observed (Rizwan et al., 2019).

Results were similar when researchers studied (Hussain et al., 2020) the Cd-, Pb heavy metals effects and foliar application of SiO₂ NPs on rice. The sprayed plants with 20 ppm SiO₂ NPs accumulated smaller amounts of heavy metals and produced higher yields. Compared to the effects of heavy metals on peas studied by other researchers, our results also confirm that SiO₂ NPs increased the tolerance index to the heavy metal copper in peas. The SiO₂ NPs (Tripathi et al., 2017) protected wheat seedlings from UV-B stress by stimulating the antioxidant defense system, increasing fresh weight and chlorophyll content, and reducing tissue damage. The researchers noted that nitric oxide levels peaked after exposure to UV-B and Si NPs, suggesting that stress protection may be due to the modulation of NO levels.

Effects of different applications and concentrations of CuO nanoparticles on pea plants

In our studies, the zeta potential of the aqueous suspension of CuO NPs was -26.68 mV (Table 2.2.1.). Researchers reported that the zeta potential of CuO NPs is -34.4 ± 0.5 mV in suspension in deionized water at pH 7 when NPs were purchased at 10-100 nm in primary size (Keller et al. 2018; Adeleye et al. 2014; Hong et al. 2015). This indicates that the suspensions were stable and anionic.

Based on our research, significant changes were detected in the antioxidant activity of drought-affected plants. Exceptionally high activation was found in CAT, APX, and SOD (Figure 3.2.2.3). Also, FRAP antioxidant power was strongly influenced by CuO NPs. Using CuO NPs in small amounts can improve the quality of plants and protect them from environmental stresses. Our results established that the suspensions of CuO NPs significantly reduced the amount of H₂O₂ in pea plants but increased the concentration of MDA (Figure 3.2.2.1), indicating that CuO NPs are actively involved in lipid peroxidation processes. When mung bean seedlings were exposed to solutions of CuO NPs concentrations of 20, 50, 100, 200, and 500 ppm, researchers found that H₂O₂ and MDA content increased significantly in plant roots, but no effect was detected in plant leaves (Gopalakrishnan et al., 2014). They highlighted that the up-regulation of the SOD gene or Cu excess resulted in increased H₂O₂ formation in the roots of mung bean plants exposed to CuO NPs. CAT gene expression was also found to be up-regulated at lower concentrations of CuO NPs. In addition, the expression level of the APX gene was down compared to the CAT gene, which may have resulted in the attenuation of the activity of the H₂O₂ neutralization system. Therefore, the researchers hypothesized that plant antioxidant defense mechanisms were activated during CuO NPs stress; incomplete removal of H₂O₂ may cause

increased ROS levels in the roots of mung bean plants exposed to CuO NPs (Gopalakrishnan et al., 2014). The effect on Arabidopsis of 0, 0.5, 1, 2, 5, 10, 20, 50, and 100 mg L⁻¹ CuO NPs suspensions in the nutrient medium were investigated (Nair and Chung, 2014). This study demonstrated that up-regulation of genes encoding enzymatic and non-enzymatic antioxidant defense mechanisms, sulfur assimilation pathways, and proline biosynthesis showed increased ROS production and rapid activation of plant defense mechanisms to neutralize oxidative stress damage during exposure to CuO NPs. One year later, the same scientist investigated the effects of CuO NPs on pea plants. Our findings agree that the higher the concentration of CuO NPs, the higher the toxicity determined in peas, starting at 100 ppm (Nair and Chung 2015). Toxic effects were confirmed in soybean by elevated lipid peroxidation and H₂O₂ content in plants at concentrations of 100 ppm CuO NPs and higher (Yusefi-Tanha et al. 2020). A reduction in soybean yield was also observed using CuO NPs at concentrations of 50-100 ppm (Ochoa et al. 2017).

Our research showed that CuO NPs had a more positive effect on the antioxidant system and Cu accumulation when plants grown in water deficit were sprayed through the leaves. Root uptake and biotransformation of CuO NPs have been extensively investigated in rice plants (Peng et al., 2015). During the biotransformation of CuO NPs, dissolved Cu bound to cysteine, citrate, and phosphate ligands, and part of Cu (II) was transformed into Cu (I). It was determined that CuO NPs could move to the root epidermis, exoderm, and cortex and reach the endodermis, but CuO NPs have difficulty penetrating the Casparian strip (Peng et al., 2015). This hindered entry of CuO NPs into the plant through the roots could have resulted in a lower distribution through the plant and a localized effect on peas when they were exposed to the CuO NPs through the roots. This explains our results, in which peas watered with CuO NPs had a higher copper accumulation in the roots, and when the plants were sprayed, the copper content increased in all parts of the plant.

But there are also reports about the positive effects of CuO NPs on plants. For example, spraying lettuce with 0, 0.5, 1.0, 2.0, 4, and 6 ppm CuO NPs suspensions positively impacted total phenols, flavonoids, antioxidant capacity, and chlorophyll content (Gaucin-Delgado et al., 2022). In addition, CuO NPs initiated the activity of APX, SOD, and CAT and significantly increased Cu accumulation in lettuce leaves. In addition, when tomato seedlings were exposed to a concentration of 10 ppm CuO NPs suspension, significant increases in fresh biomass and total chlorophyll content, as well as an increase in sugar content, NR and CAT activity, and a decrease in MDA and electrolyte leakage were found (Singh et al., 2017). As we can see from the discussed publications, the effect of CuO

NPs on plants is usually determined by the concentration. In connection with the fact that NPs are many times smaller particles than bulk materials and have a much larger surface area, the concentrations used for plants should be reduced to achieve a positive response and avoid toxic effects.

Effects of different applications and concentrations of MoO₃ nanoparticles on pea plants

The zeta potential of the aqueous suspension of MoO₃ NPs was -24.92 mV (Table 2.2.1.). According to scientific publications, the zeta potential of MoO₃ NPs suspension can be -32 mV, confirming that the solution is stable and strongly anionic.

MoO₃ NPs significantly promoted the formation of nodules on pea roots (Table 3.2.3.1) and strongly affected non-enzymatic (Figure 3.2.3.2) and such enzymatic antioxidants as APX, and GPX (Figure 3.2.3.3), thereby increasing yield during drought. A comprehensive study was investigating MoO₃ NPs on hydroponically grown rice seedlings (Sharma et al., 2021). The researchers used high concentrations of 100, 500, and 1000 ppm. The most effective concentration for rice was found to be 100 ppm, the application of which induced hormesis in rice (Sharma et al., 2021). In rice, chlorophylls and carotenoids decreased with increasing concentration of MoO₃ NPs. Also, MoO₃ NPs in rice increased MDA concentration in shoots but dropped it in roots. Such adverse effects may be associated with the use of excessive concentration.

In comparison with our results, there is a remarkable agreement that SOD activity decreases in pea leaves treated with MoO₃ NPs, while APX, GPX, and CAT increase, which leads to faster neutralization of reactive oxygen species. Another study used suspensions of MoO₃ NPs at 200 and 1000 ppm concentrations (Huang et al., 2021). Researchers observed that corn leaves could accumulate higher amounts of Mo than wheat leaves. This highlights the differences in plant species' responses to MoO₃ NPs. They concluded that wheat is more sensitive than corn regarding the number of destabilized metabolites.

Mo is not biologically active, but a specific prosthetic group complexes it in small amounts. This Mo cofactor (Moco) is distributed after biosynthesis to both Mo-enzyme families of Moco-binding proteins (sulfite oxidase, nitrate reductase (NR), mitochondrial amidoxime reductant component) and Mo-enzymes (aldehyde oxidase (AO), xanthine dehydrogenase). However, the most crucial Mo-enzyme for plant survival is cytosolic NR, which catalyzes the first step in nitrate uptake (Kaiser et al. 2005; Mendel and Schwarz 2011). The conversion of nitrates to nitrites is essential for plant growth and development. A

scheme of the mechanism of Mo-induced oxidative tolerance under drought stress was proposed by Wu (2018). It is noted that when the moisture of the substrate decreases, the plant defense system is activated through ABA and NO signals. As a component of Mo-enzymes in the form of Moco, Mo stimulates ABA synthesis and NO production via AO and NR, respectively, and thus regulates oxidative tolerance. Furthermore, NO acts downstream of ABA in Mo-mediated oxidative tolerance in plants under drought stress (Wu et al., 2018). However, it is not clear whether the exact mechanism can be applied with nano-sized Mo. For instance, it is reported that MoO₃ NPs 0.1 and 1 ppm suspensions activate nitrate reductase activity in spinach and increase chlorophyll content (Abbasifar et al., 2020). Also, using a solution with a lower concentration of Mo NPs, such as 8 ppm, increased the antioxidant activity in chickpeas (Taran et al. 2014).

The research used MoO₃ NPs with a zeta potential of -6 mV, but they did not specify the solvent in which the tested suspensions were prepared (Yang et al., 2020). However, a lower zeta potential indicates an unstable suspension and its neutral charge. They found that adding 1, 10 mg kg⁻¹ MoO₃ NPs improved the morphological parameters of soybeans, but higher concentrations (100 and 1000 mg kg⁻¹ MoO₃ NPs) were toxic. Furthermore, the accumulation of Mo in soybean leaves increased with increasing concentration, and MoO₃ NPs increased the activity of enzymatic antioxidants such as POD, SOD, and CAT.

Spraying MoO₃ NPs concentrations of 10, 20, 30, 40, and 50 ppm on leaves of common bean plants significantly increased morphological parameters such as fresh and dry biomass, plant height, and root length compared to untreated plants (Osman et al., 2020). An essential parameter in that study was the percentage of genomic template stability (GTS%) directly related to the degree of DNA alteration and the competence of DNA repair and replication. Genomic instability includes changes in structure, such as an increased frequency of base pair mutations. The researchers found that GTS% values decreased with increasing MoO₃ NPs concentrations up to 10, 20, and 30 ppm, then raised up to 40 ppm, 84.61%, 80.77%, 73.08%, and 78.21%, respectively. In addition, the mutagenic effect of MoO₃ NPs on the common bean genomic DNA affected the expression of several genes encoding specific proteins, turning certain genes on or off during the seedling and flowering stages (Osman et al., 2020). This may have determined that all concentrations of MoO₃ NPs significantly affected the physiological characteristics of common bean plants, resulting in increased yield and yield quality.

Since tiny amounts of MoO₃ NPs are more effective than bulk molybdenum salts, they can be used in agriculture to increase plant productivity during optimal and stressful plant

growth conditions. As found in our results, pea yield can be increased by 6% with MoO₃ NPs under regular soil moisture, and the drought-stressed pea yield can be maintained up to 80% by exposure to 50 ppm suspension. Although producing MoO₃ NPs on an industrial scale is costly, the expanded plant yield per hectare covers the resulting difference and even increases profits.

Effects of different applications and concentrations of B₂O₃ nanoparticles on pea plants

In our studies, the zeta potential of the aqueous suspension of B₂O₃ NPs was -28.54 mV (Table 2.2.1.). However, other researchers found that the aqueous suspension of B₂O₃ NPs with 0.2% triton x-100 was -30.3 mV (Barreto et al., 2021). Such zeta potential values indicate that the solutions are stable and anionic. In addition, the PDI of this suspension was 0.23, while other scientists have found 0.4 (Barreto et al., 2021), indicating the suspensions are monodisperse.

The study shows that the number of nodules formed on pea roots (Table 3.2.4.1) increases strongly after exposure to B₂O₃ NPs in both regular and deficient substrate moisture. B is transported in the xylem, and about 90% is incorporated into plant cell walls (Goldbach and Wimmer 2007). B is the main element in forming esters with rhamnogalacturonan II (RG-II). This borate ester is required to maintain normal cell wall functions and structure (Ryden et al., 2003). Under normal conditions, when there is sufficient B in the soil, RGII-glycoproteins are also formed in the plasma membranes of pea root nodules and root cells. However, RGII-glycoproteins are not synthesized, and their absence destabilizes the plasma membrane and nodule formation in B deficiency (Bolaños et al. 2001). Moreover, B, as a component of glycoproteins, is essential for differentiating nodule bacteria into a nitrogen-fixing form (Bolaños et al., 2004).

B is also known to stimulate both enzymatic and non-enzymatic antioxidant activity. Many scientific publications are highlighting the benefits of bulk B on the plant antioxidant system during different stress conditions (Alpaslan and Gunes, 2001; Bonilla et al., 2004; Bastías et al., 2004), but only a few are investigating the effect of B₂O₃ NPs on plants (Dimkpa et al., 2019; Zewail et al., 2021; Mahmoud et al., 2020).

Recent research has shown that the application of 150 mg l⁻¹ arbuscular mycorrhiza (AM) with 100 mg l⁻¹ B₂O₃ NPs can significantly increase the height, the number of leaves, fresh and dry biomass, and herb g plant⁻¹ of stevia (Zewail et al., 2021). It also positively affected the content of chlorophyll, carotenoids, and the amount of nutrients N, P, K, Zn,

and B in stevia leaves. Another critical study highlighted that B₂O₃ NPs in soybean increased B content, grain yield, and nitrogen accumulation compared to untreated plants (Dimkpa et al., 2019). In addition, B₂O₃ NPs have a positive effect in protecting tomatoes against salinity stress. It increased shoot fresh and dry weight, chlorophyll content, photosynthesis rate, stomatal conductance, intercellular CO₂ concentration, water use efficiency, and decreased transpiration rate in saline soil-grown tomatoes (Mahmoud et al., 2020). It is worth mentioning that our study expands the knowledge about the effect of B NPs on the antioxidant system of plants. It should be emphasized that B₂O₃ NPs effectively protect plants from drought stress by stimulating ABTS free radical scavenging activity and FRAP antioxidant power non-enzymatic and APX, GPX, and CAT enzymatic antioxidants.

Effects of combined stress and applications of nanoparticles on pea plants

Scientists note that plants are more often affected by combined than single stresses. There are still many unexplored possible stress combinations between drought, salinity, heat waves, chilling, freezing, ozone, pathogens, UV, nutrient deficiency, excessive CO₂, excessive lighting, and heavy metal stress (Suzuki et al., 2014). One of the objectives of this scientific work was to investigate the potential interactions of the combined effects of drought, heavy metal stresses, and NPs on peas. In studies of combined strains in different plant species, transcriptomic and proteomic analyses have implicated antioxidant defense mechanisms as a significant pathway. The observed higher antioxidant capacity and/or lower accumulation of reactive oxygen species (ROS) appears to be a mechanism acting in plants to increase resistance to combined stress. When studying the combined effect of drought and heavy metal Cu, the results show reduced morphological parameters and the activity of non-enzymatic antioxidants but significantly increased activity of enzymatic antioxidants and the content of biomarkers of oxidative stress in plants compared to peas grown only under drought or normal conditions. A study was conducted on the combined effects of drought and heavy metal Cd on maize plants (Naz et al., 2021). In that study, the researchers found that such a combination did not affect plant morphological parameters but significantly reduced the total amount of chlorophylls, proteins, carotenoids, and relative water content in the plants. Combined stress significantly increased the levels of oxidative stress biomarkers and activated SOD, CAT, APX, and POD enzymatic antioxidants. Both drought and its combination with Cu heavy metal stress reduced pea growth and resulted in nearly identical plant changes. The effects were additive, with both strains acting simultaneously, resulting in smaller plants and more robust stress responses than each stress alone. The exception was only after using NPs.

Foliar treatment of SiO₂ and B₂O₃ NPs was more effective in peas grown under combined stress conditions, although different treatments did not differ under drought conditions. This can be linked to the negative effect of the heavy metal Cu on root growth. Because under drought conditions, plants focus most of their energy on root growth to find more water. The excessive presence of Cu in the soil inhibits this, hindering the uptake of both essential elements and inhibiting root growth (Kumar et al., 2021). The researchers showed that SiO₂ NPs improved the growth of wheat subjected to combined drought and heavy metal Cd stress and reduced the Cd concentration in different wheat tissues (Khan et al., 2020). The treatment reduced oxidative stress in plants and increased chlorophyll content, and the effect was significant at the highest ppm concentration of the treatment under both limited water and normal conditions. The researchers also note that the application of SiO₂ NPs can be an effective way to reduce Cd concentration in cereal grains. Significant changes in the accumulation of Cu in plants after the use of NPs were determined in our study. Both spraying and watering of SiO₂ NPs significantly increased Cu bioconcentration in pea leaves and roots, translocation factor, and tolerance index. This could be related to higher relative water content and increased antioxidant activity in plants exposed to combined stress and SiO₂ NPs.

Interesting results were found in peas exposed to combined stress and B₂O₃ NPs, as there was a substantial decrease in MDA concentration in pea leaves, while MDA was at its highest concentration in all cases with Cu excess. This is related to the fact that, as mentioned above, most of the B that enters the plant is bound to the plant cell walls and thus protects the cells from lipid peroxidation (Blevins and Lukaszewski, 1998). Also, B₂O₃ NPs increased the Cu tolerance index in peas and significantly increased the activity of the essential enzymatic antioxidants SOD and CAT. SOD breaks down the superoxide radical into H₂O₂; after that, CAT converts H₂O₂ into water. Significant activation of these enzymes helped peas to survive the combined stress.

Root treatment with MoO₃ NPs of peas under combined stress significantly affected nodulation, which was strongly reduced only by drought and Cu heavy metal exposure. A significant increase in the relative water content of the plants was also found. As a result, we see increased Cu bioaccumulation in roots, translocation factor, and Cu tolerance index. There was also a significant increase in the activity of enzymatic antioxidants GPX, APX, SOD, and CAT, which corresponded to the decreased levels of oxidative stress indicators. Another critical effect found is that in the case of the combined effects of drought and heavy

metal Cu, CuO NPs have strengthened their negative influence. This shows that CuO NPs in peas act synergistically with heavy metal Cu.

This work expands the knowledge about the potential effects of combined stresses on peas. Additionally, this study investigates and discusses in detail the effects of combined stresses and different NPs on pea morphological parameters, yield, antioxidant and oxidative systems, and elemental composition.

CONCLUSIONS

1. The most effective concentrations of SiO₂, CuO, and MoO₃ nanoparticle suspensions were 50 ppm and B₂O₃ – 12.5 ppm for pea seedlings and fully developed peas grown in a substrate of normal moisture or under drought conditions.

2. The used suspension of SiO₂ nanoparticles increased the drought resistance of peas. Under conditions of moisture deficiency, SiO₂ nanoparticles reduced the concentration of oxidative biomarkers by 27%. They raised the total amount of phenolic compounds and the activity of non-enzymatic antioxidants in peas by 30%. A significant activating effect was found on the catalase (159%), superoxide dismutase (37%), and glutathione reductase (128%) after exposure of plants to drought and a solution of 50 ppm SiO₂ nanoparticles. The yield of peas affected by drought was maintained by 45% after using a suspension of SiO₂ nanoparticles.

3. CuO nanoparticles reduced hydrogen peroxide by 37% but increased malondialdehyde by 66% in pea plants affected by drought. The highest concentration of 50 ppm CuO nanoparticles increased the activity of non-enzymatic antioxidants up to 2.5 times. It stimulated the activity of catalase (167%) and ascorbate peroxidase (72%) in peas grown under drought conditions. The highest accumulation of Cu in peas leaves, stems, and roots were determined after spraying the plants with a suspension of 50 ppm CuO nanoparticles.

4. After using the suspension of MoO₃ nanoparticles in drought conditions, a positive effect on the height of peas (40%) and leaf area (30%) was determined, and the indicators of oxidative stress decreased by up to 30%. A 37% increase in the total amount of phenolic compounds and a 145% increase in the activity of non-enzymatic antioxidants were determined in plants exposed to drought and MoO₃ nanoparticles. MoO₃ nanoparticles increased the activity of catalase and ascorbate peroxidase by 2 times and the activity of superoxide dismutase by 20%, and guaiacol peroxidase by 56%. The highest Mo accumulation was found in pea roots when the plants were watered with nanoparticle suspensions.

5. B₂O₃ nanoparticles positively affected pea height (26%) and leaf area (40%) and increased antioxidant activity by 2.5 times. Ascorbate peroxidase was stimulated 6 times, guaiacol peroxidase by 91%, and superoxide dismutase by 51%. The highest accumulation of B was found in the roots, regardless of the route of exposure. B accumulation occurred most intensively in stems and leaves of peas affected by drought and watered with B₂O₃ nanoparticles.

6. More substantial oxidative stress caused by the combined effects of Cu as heavy metal and drought on pea plants was inhibited by the used suspensions of SiO₂, MoO₃, and B₂O₃

nanoparticles compared to plants that were affected by drought and excess Cu. It effectively reduced the concentrations of hydrogen peroxide and malondialdehyde in the pea plant and increased the activity of the antioxidant system. Conversely, irrigation with CuO nanoparticles enhanced the adverse effects of drought and excess Cu on plants. In peas, the Cu tolerance index was statistically reliably increased by: irrigation and spraying of SiO₂, spraying B₂O₃, and irrigation with MoO₃ nanoparticle suspensions.

7. Watering with MoO₃ nanoparticle suspension more effectively reduced the adverse effects caused by stress in peas than spraying and saved 80% of the yield. On the other hand, the effectiveness of B₂O₃ and CuO nanoparticle suspensions was more substantial when sprayed on peas and preserved 92% and 47% of the yield, respectively. The effect of SiO₂ nanoparticles according to the application method did not differ substantially, and the yield was about 40% higher compared to plants affected by drought but not by nanoparticles.

LIST OF PUBLICATIONS

Articles in journals indexed in Clarivate Analytics Web of Science database:

1. **Sutulienė, R.**, Ragelienė, L., Samuolienė, G., Brazaitytė, A., Urbutis, M. and Miliauskienė, J., 2021. The Response of Antioxidant System of Drought-Stressed Green Pea (*Pisum sativum* L.) Affected by Watering and Foliar Spray with Silica Nanoparticles. *Horticulturae*, 8(1), 35.
2. **Sutulienė, R.**, Ragelienė, L., Duchovskis, P. and Miliauskienė, J., 2022. The Effects of Nano-copper, -molybdenum, -boron, and -silica on Pea (*Pisum sativum* L.) Growth, Antioxidant Properties, and Mineral Uptake. *Journal of Soil Science and Plant Nutrition*, 22(1), pp.801-814.

Articles in journals in Clarivate Analytics Web of Science database:

1. **(Sutulienė, R.) Paulauskaitė, R.**, Miliauskienė, J. and Ragelienė, L., 2019. Uptake and effect of nanoparticles in plants. *Sodininkyste ir Daržininkyste*, 38(1/2), pp.31-46.

Conferences:

1. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2022, poster: Influence of Silica Nanoparticles on the Enzymatic and Non-Enzymatic Antioxidant Activity of Green Pea (*Pisum sativum* L.) Affected by Drought. 31st International Horticultural Congress (IHC).
2. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2022, poster: The response of drought-stressed green pea (*Pisum sativum* L.) to boron nanoparticle application. 1st International Electronic Conference on Horticulturae (IECHo).
3. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2022, oral presentation: Impact of copper oxide nanoparticles on the antioxidant system of drought affected pea. 17th International Conference of young scientists on energy and natural sciences issues (CYSENI).
4. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2021, oral presentation: Growth properties of pea plants in response to the complex effects of different nanoparticles and drought stress. 17th International Conference of young scientists on energy and natural sciences issues (CYSENI).
5. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2020, poster: The effect of CuO, Mo, B and SiO₂ nanoparticles on pea (*Pisum sativum* L.) plants during the agrometeorological drought. Scandinavian Plant Physiology Society PhD Student (SPPS).

6. **Sutulienė, R.**, Ragelienė, L., Miliauskienė, J., 2020, oral presentation: The Effect of CuO, Mo, B, SiO₂ Nanoparticles and Agrometeorological Drought Stress on Growth Parameters of Pea (*Pisum sativum* L.). International Conference on Agriculture and Horticulture (ICAH).

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Rūta Sutulienė was born on March 1, 1992, in Marijampole, Lithuania. She successfully graduated from Marijampole "Sūduva" high school in 2011. She obtained a bachelor's degree in biology in 2015 at Vytautas the Great University, Faculty of Natural Sciences. She received a master's degree in applied biochemistry in 2018 at Vytautas Magnus University, Faculty of Natural Sciences. From 2019 until now, she has been working as a junior researcher at the Lithuanian Centre for Agricultural and Forestry Sciences, Institute of Horticulture and Horticulture. In addition, she participated in a short-term research internship at the University of Antwerp, Belgium, in 2021 and at the University of Agronomy in Krakow, Poland, in 2019.

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Article

The Response of Antioxidant System of Drought-Stressed Green Pea (*Pisum sativum* L.) Affected by Watering and Foliar Spray with Silica Nanoparticles

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Abstract: Abiotic stress caused by drought impairs plant growth and reduces yields. This study aimed to investigate the impact of silica nanoparticles (SiO₂ NPs) through the adverse effects of drought on the growth, oxidative stress, and antioxidative response of pea 'Respect'. Pea plants were grown in a greenhouse before being watered (100 ± 1 mL per pot) or foliar sprayed (ca. 14 ± 0.5 mL plant⁻¹) with suspensions containing SiO₂ NPs (0, 12.5 ppm, 25 ppm, and 50 ppm) and were exposed to drought stress for 10 days. Drought stress was created by maintaining 30% of the soil moisture while the control was 80%. The growth parameters of pea grown under drought stress conditions were improved by spraying or watering plants with SiO₂ NPs (12.5, 25, and 50 ppm). At drought stress, peas treated with SiO₂ NPs (50 ppm) increased their relative water content by 29%, specific leaf area by 17%, and decreased root/shoot ratio by 4% as compared to plant non-treated with SiO₂ NPs. In addition, spraying or watering of SiO₂ NPs increased peas tolerance to drought by increasing the activity of antioxidant enzymes at least three times including catalase, ascorbate peroxidase, glutathione reductase, and superoxide dismutase, as well as reducing hydrogen peroxide and lipid peroxidation in plant tissue. It was observed the increase in total phenolic compounds and non-enzymatic antioxidant activity (DPPH, ABTS, FRAP) in peas treated with SiO₂ NPs under drought stress. The physiological response of peas to drought and the effects of SiO₂ NPs studied in this experiment based on the use of the concentration of 50 ppm nanoparticles can protect peas from the damaging effects of drought and could help reduce global food shortages.

Keywords: antioxidant activity; drought; green pea; oxidative stress biomarkers; SiO₂ nanoparticles

1. Introduction

Field peas play an important role in crop rotation due to their ability to fix atmospheric nitrogen in a symbiotic association with *Rhizobium bacteria* and fulfill the nitrogen demand of the succeeding crops, as well as can be used for both human food and animal feed. However, climate change has led to increased heterogeneity of precipitation, ranging from heavy to drought [1], leading to unstable crop yields and seed protein content, thus contributing to the reduction in field pea cultivation in the world according to Food and Agriculture Organization Statistics (FAO/STAT). To restore pea cultivation in the world, new ways such as nanotechnology need to be discovered to maintain a constant pea harvest regardless of environmental factors, as well as to maintain soil quality [2–5].

Nanoparticle (NPs), according to their unique properties: size, surface charge, shape, and potential interaction with plants could help to reduce the impact of drought [6]. It has

been described that NPs ranging in size from 4 to 100 nm can cross the cuticle by disrupting the wax layer [7], and fluorescently tagged >50 nm NPs can accumulate in the epidermis under the cuticle where stomata are absent [7,8]. Negatively charged plant cell walls act as an ion exchange surface that potentially promotes the penetration of cationic NPs rather than anionic NPs [9], but show much higher adsorption on the root surface of positively charged NPs [10]. With increasing interest in silica NPs (SiO₂ NPs), it has been found that SiO₂ NPs through symplastic and apoplastic pathways (through cell wall microchannels) penetrate the roots and then reach other parts of the plant through conductive tissues [5,11].

Silicon (Si) is not essential for plants, but in drought stress, it can affect the water ratio in drought-affected plants, reduce stomatal conduction associated with protective cell turgor loss, improve the ability to extract water from the soil due to the promotion of root elongation and the regulation of aquaporin genes [12]. Si NPs can reduce oxidative stress by increasing antioxidant enzyme activities and decreasing reactive oxygen species (ROS) in plant leaf [7,13–15]. In a salinity experiment with sweet peppers and different forms of Si, it was found that using SiO₂ NPs resulted in 15% higher plants, 34% higher fresh, and 36% dry biomasses and 11% higher total chlorophyll content and 30% yield compared to bulk Si [16]. In wheat seedlings, Si NPs (size 20–95 nm) were found to be more effective than bulk SiO₂ in response to UV-B stress as well [17]; Si NPs successfully reduced lipid peroxidation by 82% and electrolyte leakage, as well as increased catalase (CAT) and superoxide dismutase (SOD) activity, resulting in increased antioxidant resistance of wheat to UV-B. More studies have demonstrated the effectiveness of SiO₂ NPs against abiotic stress in different plants. In strawberries, peroxide concentrations and total phenols, vitamin C, CAT, SOD, ascorbate peroxidase (APX), and glutathione peroxidase (GPX) were increased by spraying with 125 ppm SiO₂ NPs under severe to moderate drought stress [14]. SiO₂ NPs had a positive effect on photosynthesis and gas exchange in rice seedlings [18]. Besides 30 ppm of Si NPs foliar application reduced oxidative stress as evidenced by low levels of malondialdehyde (MDA) and electrolyte leakage and increased production of antioxidant enzymes such as SOD by 5%, APX by 41%, and CAT by 40% in rice tissues. In potatoes, increased GPX and SOD activity was observed after spraying with 50 ppm SiO₂ NPs under salinity stress [19]. Si NPs protected from oxidative stress in pea seedlings treated with Cr (VI) [20]; the application of SiO₂ NPs on Cr (VI) treated peas significantly increased the activity of enzymes such as SOD, APX, CAT, glutathione, and dehydroascorbate reductases. Another study [21] investigated the effects of SiO₂ NPs on marigold plants after spraying and watering them, observed that the CAT activity in plant leaves was higher when irrigated with the same concentration of SiO₂ NPs as compared to spray, but the opposite effect of the suspension was found for peroxidase—higher activity was found in sprayed plants.

In recent decades, the application of NPs to improve the resistance of various crops to abiotic stress was discussed, but there is still no consensus on the benefits of NPs for plant growth and yields. An array of plant species, different NPs, their particle size and concentrations, and application pathways have been used in the literature; however, knowing the beneficial role of Si to drought-affected plants [12], little data is available on the effects of SiO₂ NPs against drought stress in the Legumes family which are popular crops in the world. It was hypothesized that watering or foliar application of SiO₂ NPs alleviates oxidative stress in drought-affected peas by inducing non-enzymatic and enzymatic antioxidant activity. Therefore, the present study aimed to investigate the effect of watering and foliar application of different SiO₂ NPs concentrations on the growth traits of drought-affected green peas (*Pisum sativum* L.), to elucidate the influence of SiO₂ NPs on oxidative stress and antioxidant response in peas exposed to drought.

2. Materials and Methods

2.1. Study Site and Treatments

The research was carried out in a greenhouse (3 × 6 m, h = 2 m) at the Lithuanian Research Centre for Agriculture and Forestry, Institute of Horticulture, Bابتai, Lithuania

(55°05′08.4″ N 23°48′03.5″ E, at an altitude of 51 m; moderate climate zone of the northern hemisphere), from 6 May to 22 June. Before sowing, green pea (*Pisum sativum* L. cv. Respect, Maribo Seed International ApS, Holeby, Denmark) seeds were sterilized in 5% sodium hypochlorite solution for 15 min to assure surface sterility and rinsed gently with deionized water several times. Then, seeds were soaked in water for 24 h. Ten seeds were sown in 10-L volume plastic pots (7 pots per treatment, arranged randomized), filled with ~8 kg of soil mixture (7:1 soil to perlite ratio, respectively). The granulometric composition of the soil was heavy loam, pH 7.4 ± 0.1 ; concentration of humus— $3.6 \pm 0.1\%$; P_2O_5 — $243 \pm 8 \text{ mg kg}^{-1}$; K_2O — $348 \pm 37 \text{ mg kg}^{-1}$; NH_4 — $4 \pm 0.6 \text{ mg kg}^{-1}$; NO_3 — $22 \pm 0.9 \text{ mg kg}^{-1}$; SiO_2 — $39 \pm 0.8 \text{ mg kg}^{-1}$. Pea seedlings were thinned to 7 plants per pot 5 days after sowing. After 16 days of cultivation, the peas were fertilized with 7 g pot^{-1} ammonium nitrate. Pots were irrigated with water by graduated cylinder daily to 80% of substrate moisture (SM) using substrate moisture sensor (Delta-T devices, HH2 moisture meter, Cambridge, UK) for 23 days (6 May till 12 June). In the greenhouse with a natural day length photoperiod, the average day/night temperature was $24.2/14.4 \text{ }^\circ\text{C}$; relative air humidity— $54/75 \pm 5\%$ before exposure; during the drought treatment the average day/night temperature was $26.2/17.0 \text{ }^\circ\text{C}$ and the relative air humidity was $50/73 \pm 5\%$, data were measured throughout the experiment (Termio + data logger, Poland). When the peas reached the 40 BBCH growth stage [22], they were watered ($100 \pm 1 \text{ mL per pot}$) or foliar sprayed until full wetting (ca. $14 \pm 0.5 \text{ mL plant}^{-1}$) with solutions containing different concentrations of SiO_2 NPs: 0 (watered or sprayed with water, NPs-untreated), 12.5 ppm, 25 ppm, and 50 ppm. After the application of SiO_2 NPs, the watering of pea plants was stopped and drought stress was initiated (30% SM), while control plants were irrigated with water to maintain normal soil moisture (80% SM) throughout the experiment. These regimes were applied for 10 days (12 June till 22 June) until harvest. Plants were harvested after reaching the BBCH 50 growth stage from each treatment to assess their morphophysiological responses.

SiO_2 NPs solutions were prepared with silica (SiO_2) NPs (particle size: 20–30 nm; purity: 99%; US Research Nanomaterials, Inc, Houston, TX, USA). The NPs with concentrations of 12.5 ppm, 25 ppm, 50 ppm, were suspended in deionized (DI) water and ultrasonically dispersed for 60 min. The NPs size and suspension stability were measured using Delsa™ Nano Submicron Particle Size (Beckman Coulter Instruments Corporation, Fullerton, CA, USA) and Zeta Potential device (Dispersion Technology Inc., Bedford Hills, NY, USA). The data in Table 1 show the negative charge of NPs in the suspensions and the stability of the systems, furthermore the NPs suspensions tend to be more monodisperse according to the polydispersity index (PDI).

Table 1. Properties of SiO_2 NPs suspension in DI water (pH = 7.05): zeta potential, results represent the mean \pm standard error, polydispersity index, and percentage of nanoparticles between 1–100 nm.

Suspension of SiO_2 NPs	
Zeta potential (ζ , mV)	-20.64 ± 0.333
Polydispersity index (PI)	0.34
NPs size 1–100 nm in suspension (%)	70

2.2. Relative Water Content, Specific Leaf Area, and Root/Shoot Ratio

Ten pea plants were randomly selected from each treatment for biometric measurements. The shoots were separated from the roots and then fresh weight (FW) and dry weight (DW) were determined ($n = 10$). The FW and DW were measured with an electronic scale (Mettler Toledo AG64, Columbus, OH, USA) and DW was determined following forced-air convection drying at $105 \text{ }^\circ\text{C}$ to a constant dry weight (Venticell 222, MBT, Brno, Czech Republic). After shoot FW determination, 10 matured plants per each treatment

were floated on deionized water for 24 h and then turgid weights (TW) were measured. Relative water content (RWC) was calculated, using the following equation [23]:

$$\text{RWC, \%} = \frac{(\text{FW} - \text{DW})}{(\text{TW} - \text{DW})} \times 100 \quad (1)$$

The leaf area was measured using an automatic leaf area meter (AT Delta-T Devices, Wallingford, UK) expressed as $\text{cm}^2 \text{g}^{-1}$. For calculation of specific leaf area (SLA), the total plant leaf area ($n = 10$) was divided by the shoot DW. Root/shoot ratio was determined as the ratio of root DW to aboveground DW.

2.3. Antioxidant Properties and Total Phenolic Compounds

Antioxidant properties of pea leaves were evaluated as the DPPH (2-diphenyl-1-picrylhydrazyl), ABTS (2,2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid)) diammonium salt, radical scavenging activities, and Fe^{2+} reducing antioxidant power assay (FRAP); the total contents of phenolic compounds were also determined. Extracts were prepared by grinding 0.3 g of plant leaves with liquid nitrogen and diluting with 5 mL of 80% methanol. After 24 h, the samples were centrifuged for 10 min at 3000 rpm (Hermle Z300K, Baden-Württemberg, Germany), extracts were filtered through cellulose filters and the supernatant was used for further analyses. All biochemical analysis was performed in 3 biological replications. Each of three biological replicates consisted of at least three conjugated plants and was repeated in three analytical replicates.

The total content of phenolic compounds was determined as gallic acid equivalents. A 250 μL aliquot of the sample extract was mixed with 250 μL of 10% (*w/v*) Folin-Ciocalteu reagent, 500 μL of 1 M Na_2CO_3 solution, and 2 mL of distilled water [24]. After incubation for 20 min in the dark, the absorbance was measured at 765 nm (M501, Spectronic Camspec Ltd., Leeds, UK). The total phenolic compounds quantity mg g^{-1} was calculated from the calibration curve of the gallic acid (0.01–0.1 mg mL^{-1} , $R^2 = 0.99$).

The ABTS (2,2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid) radical cation was obtained by incubating the 7 mM ABTS stock solution (100 mL) with 2.45 mM potassium persulfate ($\text{K}_2\text{S}_2\text{O}_8$; final concentration) and allowing the mixture to stand in the dark at room temperature for 12–16 h before use [25]. Thereafter, 50 μL of the prepared sample was mixed with 2 mL of ABTS solution (ABTS stock solution was diluted 1:7) and the absorbance was measured after 11 min (plateau phase) at 734 nm (M501, Spectronic Camspec Ltd., Leeds, UK). The ABTS scavenging activity of pea leaves extracts was calculated as the difference between the initial absorbance and after reacting for 10 min. A calibration curve was determined using Trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid; 97% purity; Sigma-Aldrich, Burlington, MA, USA) as an external standard with a range of concentrations from 0.1 to 0.8 mM ($R^2 = 0.99$). It was expressed as ABTS μmol scavenged per 1 g of fresh weight ($\mu\text{mol g}^{-1}$ FW).

For DPPH (2-diphenyl-1-picrylhydrazyl) assay, a stable 126.8 μM DPPH (100% purity; Sigma-Aldrich, Burlington, MA, USA) solution was prepared in methanol [26]. Subsequently, 1 mL of the DPPH solution was transferred to a test tube and mixed with 100 μL of the diluted pea extract with 400 μL methanol. The absorbance was scanned at 515 nm (M501, Spectronic Camspec Ltd., Leeds, UK) while reacting for 16 min. The free radical scavenging capacity was expressed as μmol of DPPH radicals scavenged per 1 g of fresh weight ($\mu\text{mol g}^{-1}$ FW). A calibration curve was determined using Trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid; 97% purity; Sigma-Aldrich, Burlington, MA, USA) as an external standard with a range of concentrations from 0.1 to 0.6 mM ($R^2 = 0.99$).

The FRAP method is based on reducing ferric ion (Fe^{3+}) to ferrous ion (Fe^{2+}). The fresh working solution was prepared by mixing 300 mM, pH 3.6 acetate buffer, 10 mM TPTZ (2,4,6-tripyridyl-s-triazine) solution in 40 mM HCl, and 20 mM $\text{FeCl}_3 \times 6\text{H}_2\text{O}$ at 10:1:1 (*v/v/v*) [27]. 20 μL of the sample was mixed with 3 mL of working solution and incubated in the dark for 30 min. Readings of the colored product (ferrous tripyridyl-triazine complex) were then taken at 593 nm. A calibration curve was determined using $\text{Fe}_2(\text{SO}_4)_3$ (Iron

(III) sulfate; 97% purity; Sigma-Aldrich, Burlington, MA, USA) as an external standard with a range of concentrations from 0.005 to 0.5 mM ($R^2 = 0.99$). The antioxidant power is expressed as Fe^{2+} antioxidant capacity (Fe^{2+} $\mu\text{mol g}^{-1}$ FW).

2.4. Malondialdehyde and Hydrogen Peroxide

The extracts used to determine the concentration of lipid peroxidation and hydrogen peroxide (H_2O_2) in pea leaves were prepared by grinding 0.1 g of fresh sample with liquid nitrogen and diluting with 4 mL of 0.1% TCA (trichloroacetic acid). After centrifugation for 10 min at 3000 rpm (Hermle Z300K, Baden-Württemberg, Germany), the supernatant was used for further analyses.

For H_2O_2 measurements in plant leaves, 500 μL of the supernatant was added to 1 mL of 1 M potassium iodide (KI). The absorbance of the mixture was scanned at 390 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK). A calibration curve was determined using H_2O_2 (30% hydrogen peroxide) as an external standard with a range of concentrations from 0.6–24.3 mM ($R^2 = 0.99$). The content of H_2O_2 is expressed in fresh weight ($\mu\text{mol g}^{-1}$ FW) [28].

The TBARS test determines malondialdehyde (MDA) content in pea leaves samples as the end product of lipid peroxidation. 500 μL of the supernatant was added to 1 mL 0.5% (*w/v*) thiobarbituric acid (TBA) in 20% trichloroacetic acid (TCA). The mixture was incubated in boiling water for 30 min. The reaction stopped after the samples have cooled. The samples were centrifuged at $10,000\times g$ for 5 min, and the absorbance of the supernatant was measured at 532 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK). The value for non-specific absorbance at 600 nm was subtracted [29]. The amount of MDA–TBA complex (red pigment) in leaves was calculated and expressed as nmol g^{-1} FW:

$$C_{\text{MDA}} = (A_{532} - A_{600})/E_{\text{MDA}} \quad (2)$$

C_{MDA} —concentration of MDA, μM

A_{532} , A_{600} —Absorbance at wavelengths

E_{MDA} —MDA extinction coefficient $155 \text{ mM}^{-1} \text{ cm}^{-1}$

2.5. Antioxidant Enzymes Activities

The extracts used to determine the activity of superoxide dismutase (SOD), catalase (CAT), ascorbate peroxidase (APX), and glutathione reductase (GR) in pea leaves were prepared by grinding 0.5 g of fresh sample with liquid nitrogen and diluting within 5 mL extraction buffer (100 mM potassium-phosphate buffer, pH 7.8, containing 0.1 mM EDTA). After centrifugation for 10 min at 3000 rpm (Hermle Z300K, Baden-Württemberg, Germany), the supernatant was collected and used for the assays of enzymatic activities. All steps in the preparation of the enzyme extract were carried out at 4°C .

For soluble protein determination, the dye-binding method and bovine serum albumin as standard were used. A volume of 30 μL of enzyme extract was mixed with 1.5 mL of Bradford reagent diluted by 1:5 with DI water. Absorbance was read after 2 min. through a spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK) at 595 nm [30].

Total SOD activity was estimated by the inhibition of the photochemical reduction of nitroblue tetrazolium (NBT) by the enzyme [31]. Three mL of reaction mixture consisted of 13 mM methionine, 75 μM NBT, 100 mM potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 50 μL enzyme extract, and 13 μM riboflavin. The tubes were under $150 \mu\text{mol m}^{-2} \text{ s}^{-1}$ for 1 min to initiate the reaction and then covered. The absorbance was recorded after 30 min by spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK) at 560 nm, and one unit of enzyme activity was taken as that amount of enzyme, which reduced the absorbance reading to 50% in comparison with tubes lacking enzyme expressed as $\text{unit mg}^{-1} \text{ protein min}^{-1}$.

CAT activity was measured as the disappearance of H_2O_2 [32]. A volume of 100 μL enzyme extract was added in 1.275 mL of 0.1 M phosphate buffer (pH 7.8, containing 0.1 mM EDTA); the reaction started by adding 125 μL of 30 mM H_2O_2 . The decrease in

absorbance measured by spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK) at 240 nm was observed for 1 min and enzyme activity was computed by calculating the amount of H_2O_2 decomposed ($\mu\text{mol } H_2O_2 \text{ mg}^{-1} \text{ protein min}^{-1}$).

APX activity was assayed by recording the decrease in optical density due to ascorbic acid at 290 nm [33]. The 1 mL assay mixture contained 0.1 M potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 0.5 mM ascorbic acid, 0.1 mL enzyme extract, and 0.1 mL of 30 mM H_2O_2 was added to initiate the reaction. The decrease in absorbance was measured spectrophotometrically (M501, Spectronic Camspec Ltd., Leeds, UK) for 1 min and the extinction coefficient of $2.8 \text{ mM}^{-1} \text{ cm}^{-1}$ for reduced ascorbate was used in calculating the enzyme activity that was expressed as $\mu\text{mol AsA mg}^{-1} \text{ protein min}^{-1}$.

Measuring GR activity based on the rate of decrease in the absorbance of oxidized glutathione (GSSG), at 340 nm [34]. The reaction mixture contained 0.1 M potassium phosphate buffer (pH 7.8, containing 0.1 mM EDTA), 1 mM GSSG, 100 μL enzyme extract and 75 μL 0.1 mM NADPH added last to initiate the reaction. The decrease in absorbance measured by spectrophotometer (M501, Spectronic Camspec Ltd., Leeds, UK) was recorded every 5 min until 20 min. An absorption coefficient of $6.22 \text{ mM}^{-1} \text{ cm}^{-1}$ was used for calculations, and GR activity was defined as $\mu\text{mol NADPH mg}^{-1} \text{ protein min}^{-1}$.

2.6. Statistical Analysis

All the values were expressed as mean \pm standard error. Data were analyzed using the Analysis of Variance (ANOVA) test followed by Tukey HSD at $p < 0.001$ to identify significant differences. All statistical analyses were performed using XLSTAT (XLstat, Addinsoft, Paris, France, 2021).

3. Results

3.1. Effects of Drought Stress and SiO_2 NPs on Plants Specific Leaf Area, Relative Water Content, and Root/Shoot Ratio

The relative water content (RWC) and specific leaf area (SLA) decreased during drought stress (Figure 1a,b).

Drought stress significantly decreased SLA and RWC (38 and 27%, respectively) compared with control plants. Watering or spraying peas with SiO_2 NPs (50 ppm) increased the values of these two parameters compared to plants watered or sprayed with water. Moreover, watering SiO_2 NPs had a better effect on SLA than spraying at all concentration levels.

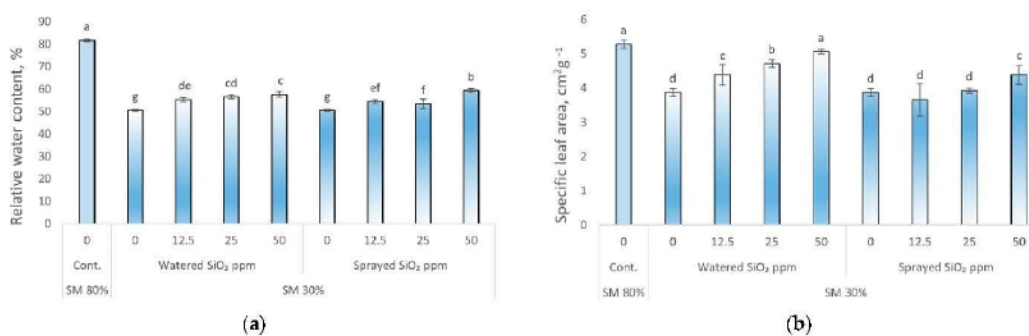


Figure 1. Cont.

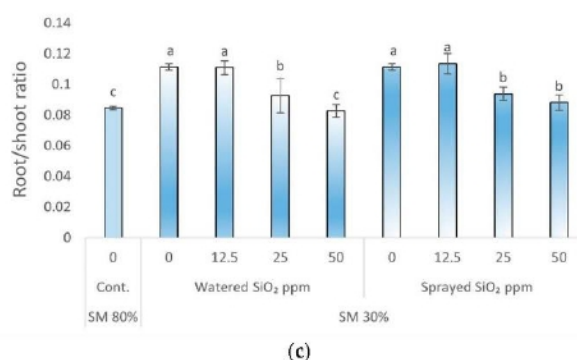


Figure 1. Effect of drought stress and SiO₂ NPs (0; 12.5; 25; and 50 ppm) on relative water content (%), specific leaf area (cm² g⁻¹), and root/shoot ratio in *P. sativum* L. Cont.—control plants, substrate moisture (SM) 80%; drought stress—SM 30%, RWC (a); specific leaf area (b); root/shoot ratio (c). Values are mean ± SE of 10 replicates and different letters are differed significantly by Tukey HSD Test ($p < 0.001$).

There was a statistically significant increase in the root/shoot ratio (Figure 1c) when comparing drought-affected (0 ppm SiO₂ NPs, SM 30%) and control plants (SM 80%). Comparing the effects of 0 ppm and 12.5 ppm SiO₂ on plant root/shoot ratio when watered or sprayed in drought treatment, no statistically significant difference was found. However, a statistically significant effect on root/shoot ratio was found by watering or spraying peas at a concentration of 25 ppm SiO₂ compared to SiO₂ NPs untreated plants (0 ppm) in drought-affected peas. Meanwhile, the root/shoot ratio of peas grown under drought did not differ significantly from control plants when were watered or sprayed with 50 ppm SiO₂ NPs.

3.2. Influence of Drought Stress and SiO₂ NPs on Lipid Peroxidation and Hydrogen Peroxide

The malondialdehyde (MDA) and hydrogen peroxide (H₂O₂) content in peas increased significantly under drought stress (0 ppm SiO₂ NPs, SM 30%) compared to control plants, i.e., 24 and 132%, respectively (Figure 2a,b).

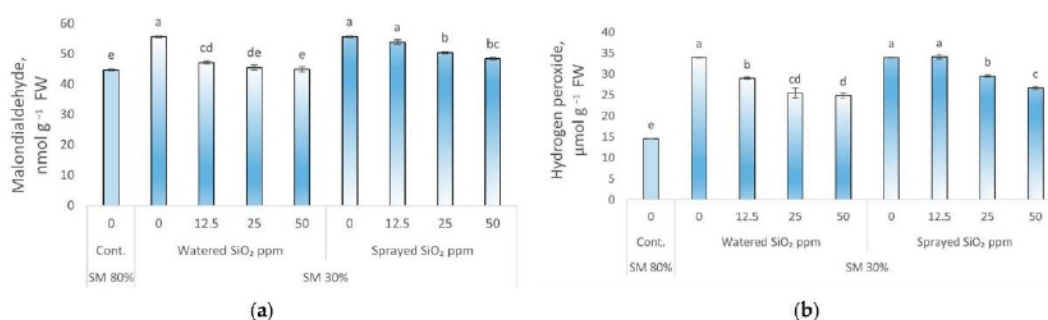


Figure 2. Influence of drought stress and SiO₂ NPs (0; 12.5; 25; and 50 ppm) on malondialdehyde content (MDA), (a); hydrogen peroxide (H₂O₂), (b) in *P. sativum* L. Cont.—control plants, substrate moisture (SM) 80%; drought stress—SM 30%. Values are mean ± SE of three replicates and different letters are differed significantly by Tukey HSD Test ($p < 0.001$).

All applied SiO₂ NPs concentrations significantly reduced the content of MDA and H₂O₂ in peas compared to NPs-untreated (0 ppm SiO₂ NP, SM 30%) plants as they were

watered with SiO₂ NPs. However, no statistical reliability was found when comparing MDA levels in plants watered with 25 and 50 ppm SiO₂ NPs concentrations with control plants. Spraying peas with SiO₂ NPs resulted in a statistically significant reduction in MDA and H₂O₂ content at 25 ppm (9 and 13%, respectively) and 50 ppm (13 and 21%, respectively) SiO₂ NPs concentrations compared to NP-untreated plants under drought stress (0 ppm SiO₂ NPs, SM 30%), but still exceeded the level of control plants.

3.3. Effect of Drought Stress and SiO₂ NPs on Antioxidant Activity

The total phenolic compounds (TPC), DPPH and ABTS free radical scavenging activity, and FRAP antioxidant power in pea leaves were significantly reduced by drought stress (0 ppm SiO₂ NPs, SM 30%) (24, 16, 39, and 17%, respectively) compared to control plants (Figure 3). There were no statistically significant changes in TPC and DPPH value between control plants and peas sprayed with 50 ppm SiO₂ NPs. Moreover, there was no significant effect on FRAP value when peas were watered with 25, 50 ppm and sprayed with 50 ppm SiO₂ NPs as compared to control plants. ABTS free radical scavenging activity in peas sprayed with 12.5 and 25 ppm SiO₂ NPs under drought stress was significantly reduced compared to control plants. However, a statistically significant increase in ABTS value was found in plants watered with 12.5, 25, and 50 ppm SiO₂ NPs under drought exposure.

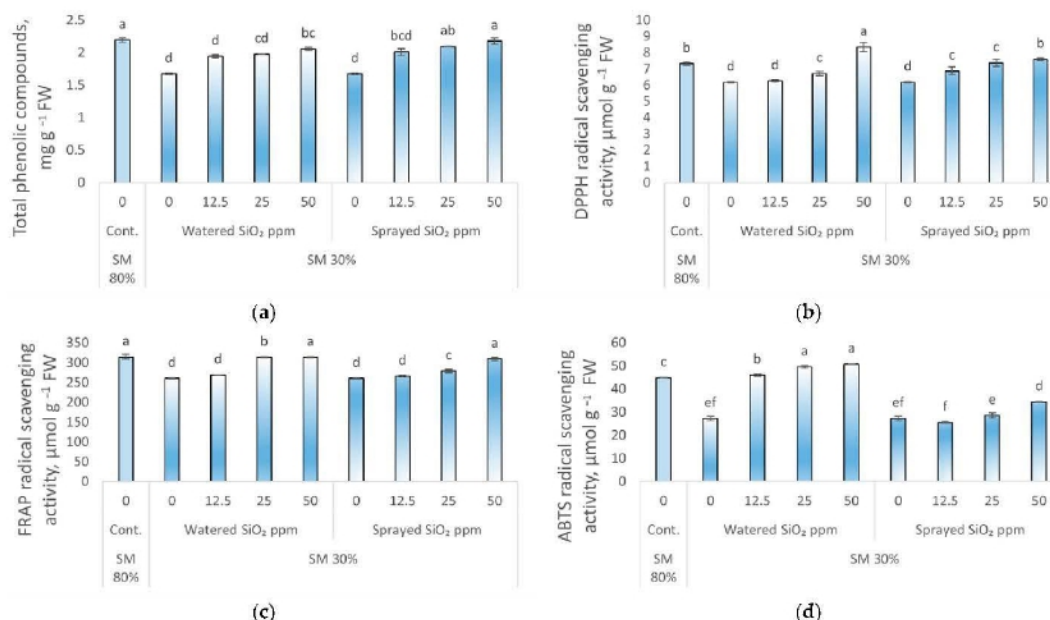


Figure 3. Influence of drought stress and SiO₂ NPs (0; 12.5; 25; and 50 ppm) on total phenolic compounds (a); DPPH radical scavenging activity (b); FRAP radical scavenging activity (c); ABTS radical scavenging activity (d) in *P. sativum* L. Cont.—control plants, substrate moisture (SM) 80%; drought stress—SM 30%. Values are mean ± SE of three replicates and different letters are differed significantly by Tukey HSD Test ($p < 0.001$).

3.4. Effect of Drought Stress and SiO₂ NPs on Antioxidant Enzymes Activities

Superoxide dismutase (SOD), catalase (CAT), ascorbate peroxidase (APX), and glutathione reductase (GR) activity increased in peas affected by drought compared with control plants (Figure 4). The most active antioxidants were found when plants were wa-

tered with SiO₂ NPs suspensions. SOD activity was most increased by watering plants with 25 ppm (95%) and 50 ppm (103%) or by spraying with 50 ppm (82%) SiO₂ NPs compared with control plants (Figure 4a). Increased CAT activity was observed in both watered and sprayed plants at all applied concentrations but was most pronounced when peas were exposed to 50 ppm SiO₂ NPs suspension (Figure 4b). The activity of the APX enzyme in peas was also increased in drought conditions (Figure 4c) when plants were watered at 0, 12.5, 25, and 50 ppm NPs concentrations, APX activity increased by 158, 244, 288, and 309% respectively, compared to control plants. As with all enzymes tested, GR activity in peas also increased under drought conditions (Figure 4d) when plants were watered at 0, 12.5, 25, and 50 ppm SiO₂ NPs suspension, GR activity increased by 39, 41, 69, and 178%, respectively, compared to control plants.

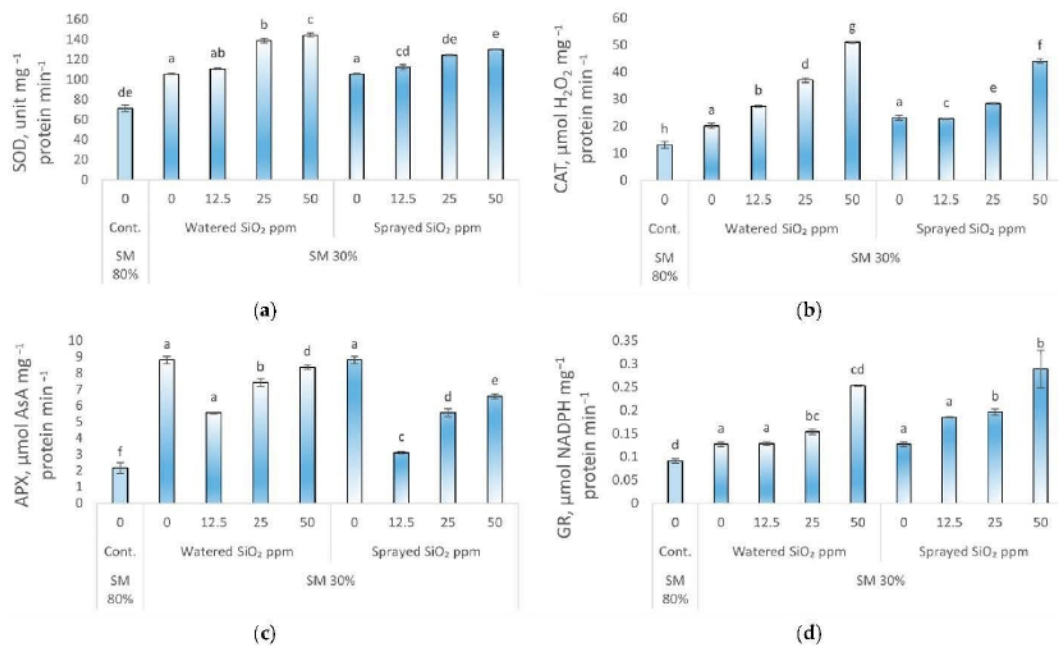


Figure 4. Response of (a), superoxide dismutase (SOD); (b), catalase (CAT); (c), ascorbate peroxidase (APX); (d), glutathione reductase (GR) activity to drought stress and SiO₂ NPs (0; 12.5; 25; and 50 ppm) in *P. sativum* L. Cont.—control plants, substrate moisture (SM) 80%; drought stress—SM 30%. Values are mean ± SE of three replicates and different letters are differed significantly by Tukey HSD Test ($p < 0.001$).

4. Discussion

This study elucidated the effects of SiO₂ NPs on drought stress in pea plants based on NP concentration and different applications (leaf application and root irrigation) to explain the main physiological, non-enzymatic, and enzymatic antioxidant defense mechanisms. The results of this study showed that SiO₂ NPs can alleviate drought-induced stress in plants. The above-presented research data indicate that the studied defense mechanisms significantly varied in SiO₂ NPs non-treated and treated pea plants under drought stress.

4.1. Effects of Different Applications and Concentration of NPs

Several factors are responsible for the transformation and uptake of NPs in plants: physicochemical properties of NP itself, size, surface charge, concentration in the suspen-

sion, potential interaction with plants, and plant physiology. Dissolved bulk silicon (Si) is known to be absorbed by plants in the form of mono-silicic acid ($\text{Si}(\text{OH})_4$) and ultimately polymerizes with the loss of water molecules to form hydrated silica ($\text{SiO}_2 \cdot n\text{H}_2\text{O}$) [35]. In plants with a high capacity for metalloid accumulation, it is assisted by one of aquaporin group MIP [36]—Si influx transporters (SiT1 and SiT2) and efflux transporters (SiT6) located in the main and lateral roots, responsible for the transport of Si from the cortical cells to the xylem [35–38]. The researchers found that pea plants tend to the silicification process using Si transporters [39]. However, NPs can have different properties than their bulk materials [40]. Plants can be affected by NPs in two ways, sprayed through shoots and watered through roots. Spraying—exposure through the leaves can form an additional layer on the leaves [17] to protect plants from increased transpiration and disease [41], but can also allow some NPs to penetrate through the wax layer and diffuse directly into the plant (limited to pore size 5–20 nm) [42]. Considering the composition of the cell wall, which has hydrophobic and hydrophilic components and unequal distribution of fixed negative charges (cellulose fibers and lignin surface potential are -15 and -45 mV, respectively) [9,43], it is noticeable that negatively charged plant cell walls act as an ion exchange surface that potentially promotes the penetration of cationic NPs rather than anionic ones. Considering the effects of irrigation when NPs are reached through the soil, it should be noted that the soil particles are usually negatively charged, and NPs with a higher negative charge, are more agile in such soil [44]. On the contrary, positively charged NPs easily attract negatively charged soil surface particles. In general, NPs mobility decreases as the average soil grain size decreases. The clay content of the soil can act as an anionic adjuvant to prevent NPs accumulation and increase their mobility [45]. Zeta potential measurement is a technique for determining the surface charge in a colloidal solution [46]. NPs with a zeta potential of -10 to $+10$ mV are approximately neutral, and NPs with a zeta potential greater than $+30$ mV or less than -30 mV are considered to be strongly cationic and anionic, respectively [47]. The zeta potential of the plasma membrane of various plants can vary from 20 to -39 mV, but the zeta potential of the plasma membrane of pea has not been studied, making it difficult to estimate [48]. In future studies, it would be valuable to evaluate the zeta potential of the pea leaf plasma membrane when grown under normal and drought conditions. However, in our results, the zeta potential of the SiO_2 suspension in distilled water was -20 meaning anion (Table 1). There is another scientific article describing that the application of 10 g kg^{-1} SiO_2 NPs with a zeta potential of -40 mV through the soil had a positive effect on photosynthesis, yield quality, and increased productivity of maize plants [49]. Moreover, the researchers [50] found that negatively charged Au NPs accumulate less on the root surface but move most efficiently through the root epidermis to plant shoots, especially in rice and ryegrass.

The effect of NPs on plants is highly dependent on their concentration. In this study, the most effective concentration against drought-induced effects in peas was spraying and watering with 50 ppm SiO_2 NPs. The researchers found that spraying strawberries with 125 ppm SiO_2 NPs increased their resistance to drought [14]. Using concentrations of 100, 200, 300, and 400 ppm SiO_2 NPs, a statistically significant effect was found for common bean germination [51]. Other researchers used Si NPs as leaf sprays at concentrations of 0, 5, 10, 20, 30 ppm and found that the Cd-induced negative growth rates of rice were significantly reduced [18]. Previous comparative studies showed that approximately 140 ppm Si NPs were effective in reducing Cr (VI) toxicity in pea (*Pisum sativum*) [20], arsenate toxicity in maize (*Zea mays*) [13], and UV-B stress in wheat (*Triticum aestivum*) [17]. In another experiment with cucumbers [52], SiO_2 NPs concentrations at 100, 200, 300, 400 ppm were used, of which the concentration at 200 ppm caused the greatest positive effect in plants. In addition, researchers [19] sprayed potatoes with SiO_2 NPs concluded that a concentration of 50 ppm had a positive, but a concentration of 100 ppm had a statistically significant negative effect on plants. In contrast, the opposite, negative effect was found for SiO_2 NPs at concentrations 10, 100, 500, and 2000 ppm on Bt-transgenic cotton height and weight [5].

These results confirm that the effects of NPs depend not only on the NPs and applied concentration, but also on their interaction with the plant species.

4.2. Influence of Drought Stress and SiO₂ NPs on Oxidative Stress Markers and Antioxidant Enzymes Activities in Peas

The effects of appropriately selected and studied doses of NPs on plants and the soil ecosystem could address the food insecurity caused by climate change, which is challenging agriculture. In our research, the deficiency of water in soil caused oxidative stress resulting in increased malondialdehyde (MDA) and hydrogen peroxide (H₂O₂) levels in pea plants (Figure 2a,b), but watering or spraying plants with SiO₂ NPs reduced the harmful effects of drought stress. Drought stress increased the activity of antioxidant enzymes, including CAT, APX, SOD, and GR (Figure 4), in the leaves of pea plants compared to control plants grown with insufficient moisture supply, while exposure to SiO₂ NPs further strengthened the activity of these enzymes. Previous studies [53,54] have shown that severe drought causes oxidative stress due to the accumulation of reactive oxidative species (ROS) in plant cells, including superoxide radicals (O₂^{•-}), alkoxy radicals (RO•) and hydroxyl radicals (OH•), hydrogen peroxide (H₂O₂), and singlet oxygen (¹O₂). However, plants have adapted to eliminate excessive oxidative stress agents using antioxidant defense systems consisting of enzymatic (SOD, CAT, GSH, etc.) and non-enzymatic (ascorbic acid, phenolic compounds, carotenoids, etc.) antioxidants [53,55,56]. As the amount of oxidative stress products in plants increases, the antioxidant system activates, for example, superoxide dismutase converts the superoxide radical to hydrogen peroxide (which is less harmful to the plant) then the peroxide decomposes into the water using antioxidants such as CAT, APX, GPX (using different reaction catalysts) [54]. As can be seen in Figure 3, the application of SiO₂ NPs tends to stimulate a non-enzymatic antioxidant response by increasing TPC, DPPH, FRAP, and most strongly ABTS radical scavenging activity in pea plants exposed to drought stress. The ABTS [57] and FRAP methods show a greater correlation with non-enzymatic hydrophilic antioxidants such as α-tocopherols, flavonoids, and ascorbic acid, and the DPPH method showed a higher correlation with lipophilic antioxidants such as carotenoids [25]. According to the results of ABTS, FRAP, and TPC, it can be stated that peas contain more hydrophilic antioxidants than lipophilic in terms of DPPH results.

In general, increased levels of MDA, H₂O₂, and other antioxidant enzymes have been reported in various plant species under drought stress [14,56,58,59]. In this study, a significant increase in MDA and H₂O₂ in pea leaves (Figure 2) in response to water deficiency caused strong oxidative stress resulting in retarded plant growth, decreased RWC, SLA, and increased root/shoot ratio (Figure 1), indicating that pea plants in the lack of soil moisture encourage root growth. MDA can generally be described as a process in which oxidants, such as free radicals, attack lipids having a carbon–carbon double bond(s), especially polyunsaturated fatty acids (PUFAs) [60]. According to the results obtained by other researchers [61], PUFAs in peas are less than in other plants of the Legumes family. Besides, the amount of PUFAs had a positive correlation with DPPH antioxidants, which in pea plants also had lower activity compared to ABTS (Figure 3b,d). In this study, a very strong increase was found in the activity of enzymatic antioxidants such as SOD, CAT, APX, and GR (Figure 4) when pea plants were watered or sprayed with SiO₂ NPs especially at 50 ppm (enzyme activity increased with increasing SiO₂ NPs concentration). In response to water deficit, due to stomatal closure, low CO₂ availability, and limited fixation, oxygen saturation of ribulose 1,5-bisphosphate (RuBP) is preferred that enhances photorespiration [59] leading to more than 70% of the H₂O₂ generation in plants. Besides, H₂O₂ is formed in the plant when the enzyme SOD performs the monovalent reduction and protonation of the superoxide radical [53]. In this study, significant activation of SOD along with other enzymatic antioxidants activity was observed in pea plants under water deficit. We suppose that, due to this, a significant decrease in H₂O₂ (Figure 2b) was observed in peas as plants were watered with SiO₂ NPs in the presence of a water deficit. As mentioned above, SiO₂ NPs may enter the pea plant through special channels in the roots, then move

in a symplastic pathway in the plant and enter the cell cytosol containing most of the enzymatic antioxidants and thus affect plant with abnormal particle surface area, size, and charge. This explains the more efficient watering of SiO₂ NPs than foliar application to plants.

SiO₂ NPs have a very strong relationship with water content in plants; there are just a few published scientific articles specifically on the effects of drought. For instance, in strawberries, spraying with Si NPs during drought stress increased the activity of CAT, APX, SOD, and GR enzymes and decreased the content of MDA and H₂O₂ [14]. The relationship of Si NPs and other abiotic factors such as salinity, exposure to heavy metals, and UV radiation to plants were also explored by researchers. For example, when spraying with 50 ppm SiO₂ NPs suspension in salinity-treated potato seedlings, the researchers found [19] that enzymatic antioxidants such as GPX and SOD were more active compared to potatoes not sprayed with SiO₂ suspension. The negative effects of Cr (VI) were found in peas with increased H₂O₂ and MDA concentrations but spraying them with Si NPs resulted in a clear decrease in oxidative stress biomarkers and activated enzymatic antioxidants such as SOD, CAT, GR, APX [20]. Moreover, when common beans were sprayed with 30 ppm SiO₂ NPs, a statistically positive increase in the activity of antioxidant enzymes and a decrease in the concentration of MDA and H₂O₂ were observed under Cd stress [18]. Similar results were obtained when researchers studied [62] the effects of Cd-, Pb-affected rice and foliar application of SiO₂ NPs, the sprayed plants with 20 ppm SiO₂ NPs accumulated less heavy metals and produced the highest yields. The SiO₂ NPs have been shown [17] to protect wheat seedlings from UV-B stress by stimulating the antioxidant defense system and reducing the negative effects of UV-B stress, such as low fresh weight, chlorophyll content, and tissue damage. Researchers noted that as nitric oxide levels reached a peak after exposure to UV-B and Si NPs, the protection arises from modulating NO levels.

4.3. Effects of Drought Stress and SiO₂ NPs on Peas Specific Leaf Area, Relative Water Content, and Root/Shoot Ratio

In plants, decreased growth parameters may be due to decreased relative water content (RWC) and corresponding cell contraction, decreased meristematic cell division, decreased leaf growth, accelerated aging, blocked leaf production, and leaf fall. Water stress can also directly affect the biochemical processes involved in photosynthesis and indirectly reduce the uptake of carbon dioxide into the stoma, which closes during drought. In this study, relative water content (RWC) (Figure 1a) and specific leaf area (SLA) (Figure 1b) of pea plants significantly decreased during drought stress, but watering or spraying with a suspension containing SiO₂ NPs increased their contents. Plant membranes are the first place in the cell which are influenced under stress conditions and the ability of plants to protect the integrity of membranes under drought stress determines the tolerance of the plants to drought stress [63]. Under drought stress, the water potential of soil decreases, and plants prevent transpiration phenomenon using different mechanisms such as closing stomata, increasing stomatal resistance, and decreasing stomatal conductivity. Scientists showed that the improving effect of Si on the hydration status of plants may help to reduce leaf and stem transpiration or the deposition of phytolith under epidermal cells, resulting in a decreased waste of water from cuticle layers [64]. According to that concentration of 50 ppm, SiO₂ NPs play an important role in increasing RWC in pea plants (Figure 1a) under stressful conditions. Similar RWC results were found in drought-affected strawberries [14], wheat [58], and cucumber [52]. The researchers also found a statistically positive increase in rice height, shoot, and root weight under Cd stress and spraying of SiO₂ NPs [18]. A study by other scientists has shown that drought stress has negatively affected wheat growth rates and yields [58], but the use of SiO₂ NPs, especially through soil application at 30 and 60 ppm, reduced the negative effects of drought stress by reducing transpiration, improving the rate of photosynthesis, increasing chlorophyll content and RWC. Moreover, similar results were found in peas [20], maize [13], and wheat [17] affected by various stressors,

with a statistically positive increase in fresh leaf and root biomass, also leaf area using Si NPs.

Previous studies [65] indicate that the root/shoot ratio increases with drought, as plants tend to grow roots and absorb more water over a larger root surface, resulting in reduced shoot growth. This trend was also observed in our study, with a significant increase in root/shoot ratio in drought-affected peas (Figure 1c), but the application of SiO₂ NPs at 50 ppm reduced the negative effects of drought stress on the root/shoot ratio. These results confirm that the application of SiO₂ NPs can strongly affect plant development under drought exposure.

5. Conclusions

In conclusion, watering or foliar spraying solutions containing 50 ppm SiO₂ NPs on pea plants is an efficient way to improve their resistance to drought stress. Favorable effects of SiO₂ NPs on pea growth efficiency under drought conditions have been attributed to the activation of the enzymatic and non-enzymatic antioxidative system to eliminate ROS and enhance relative water content, specific leaf area, and decreased root/shoot level. Based on the results obtained, the use of SiO₂ NPs is recommended for managing the drought stress in pea plants and even in other agricultural plants.

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The Effects of Nano-copper, -molybdenum, -boron, and -silica on Pea (*Pisum sativum* L.) Growth, Antioxidant Properties, and Mineral Uptake

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Abstract

Before applying nanoparticles (NPs) for outdoor use, their effects on plants must be verified in a controlled chamber. Molybdenum (Mo) is important for nitrate uptake and boron (B) is necessary for peas (*Pisum sativum* L.) growth, copper (Cu) is important for cellular redox state and silicon (Si) makes the plant tissue stronger. However, it is not clear what concentrations would be beneficial or toxic to pea if used Mo, B, Cu, and Si in the form of NPs. This experiment aimed to determine the concentration-dependent effects of silica (SiO₂), Cu, Mo, and B NPs on pea growth, antioxidant properties, and mineral uptake. The study was performed in a controlled environment chamber in 16-h daylight cycles to evaluate the significance of different concentrations of SiO₂, CuO, Mo, and B NPs for the response of pea plants. In this study were performed growth, non-destructive measurements, chemical antioxidant efficiency (total phenolic compounds, DPPH, ABTS), lipid peroxidation, and hydroxide amount analyzed, also macro- and microelement uptake was evaluated. The most effective and statistically reliable effects were obtained using the following NPs suspension concentrations: SiO₂ – 100, 50 ppm; CuO – 12.5 ppm; Mo – 100, 50 ppm; B – 12.5 ppm. The experiment showed that Mo and SiO₂ NPs had a positive effect at all concentrations, and B and CuO NPs at 100 ppm were toxic to peas. It has been found that SiO₂, Cu, Mo, B NPs can have both positive and negative effects on pea growth, antioxidant system, and mineral uptake depending on the concentration.

Keywords Nanoparticles · Field pea · Antioxidant activity · Oxidative stress

1 Introduction

The purpose of nanoparticles (NPs) in agriculture is to reduce the number of common chemicals, reduce losses in fertilizer use, and increase yields by regulating the use of pesticides and fertilizers (Prasad et al. 2017). However, the actual effects of nanomaterials on plants depend on their composition, concentration, size, surface load, and physico-chemical properties, in addition to the sensitivity of plant species (Ma et al. 2015). Silicon (Si), copper (Cu), molybdenum (Mo), and boron (B) are trace elements beneficial to plants, but effects as NPs have been poorly studied.

Silica (SiO₂) NPs exhibit great potential in agriculture on plant growth and development. It may work better in alleviating different abiotic stresses than bulk material (Rastogi et al. 2019). This was confirmed by the scientists (Tripathi et al. 2015) who studied the effects of Si NPs (75–125 nm, 10 μM concentration) on pea seedlings in response to chromium (Cr (VI)) stress and found that SiO₂ NPs regulated the antioxidant defense system in pea under Cr (VI) stress.

The effects of copper oxide (CuO) NPs on peas have been studied in several studies (Ochoa et al. 2017; Nair and Chung 2015). In research (Ochoa et al. 2017), pea plants were cultivated to full maturity in soil amended with CuO NPs (74.3% NPs of 10–100-nm size was in the suspension) at concentrations 0, 50, 100 mg kg⁻¹, and indoleacetic acid (IAA) at 10 and 100 μM and was found that interaction of IAA (10 μM) and CuO NPs (50 mg kg⁻¹) reduced stem and pod biomass of pea plants. Another study (Nair and Chung 2015) observed that the effects of CuO NPs on pea germination depended on applied CuO NPs concentration; toxicity to peas was observed by using 100, 200, 400, and 500 mg L⁻¹ CuO

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NPs, resulting in significantly reduced length of plant shoots and roots, and increased generation of reactive oxygen species (ROS) and lipid peroxidation.

Molybdenum (Mo) is one of the essential elements required for nitrogen metabolism in rhizobium bacteria in pea roots. There is only one scientific article on the effects of Mo NP on chickpeas [8]. Researchers stated that Mo NPs suspension (100–250nm; 8 mg L⁻¹) stimulated the formation of a nodule and increased their number by two times compared to control plants (Mo NPs untreated plants) (Taran et al. 2014).

Boron (B) is one of the key nutrients for optimal crop growth, development, yield, and quality (Brown et al. 2002). To our knowledge, the effect of B on pea seedlings in NP size has not yet been studied. The effects of B NPs (0.6 mg kg⁻¹) on soybean have been studied, indicating that B NPs affected soybean growth, yield, and grain micronutrient quality under drought conditions (Dimkpa et al. 2017). In addition, B NPs are more effective than bulk for soybean dry weight (Dimkpa et al. 2019).

Based on several studies, the beneficial role of SiO₂, CuO, Mo, and B NPs in plants is not well documented and the effects on green pea are not yet well known. Peas are a widely grown crop. The major pea producers are China, India, the USA, France, and Egypt. Besides, pea seeds are high in protein and a good source of vitamins (Khan et al. 2016). Legume crops are incorporated into agricultural systems to improve soil fertility, plant growth, and limit the use of chemical fertilizers. Pea plants have the beneficial property to form nodules on their roots, in which *Rhizobium* bacteria fix nitrogen from the air and thus enrich the soil. Thus, peas play a significant role in crop rotation, and the overall economic value in the production system and were therefore chosen as model plants. This study aimed to explore the effects of different concentrations of SiO₂, CuO, Mo, and B nanoparticles on green pea (*Pisum sativum* L.) concerning growth traits; chlorophyll; and nitrogen balance index, antioxidant defense system, and regulation of mineral elements. It was hypothesized that NPs of SiO₂, CuO, Mo, and B will have a significant effect on pea plants, causing a different response depending on the concentration of NPs applied.

2 Materials and Methods

2.1 Preparation of Nanoparticle Suspension

Silica (SiO₂) NPs (particle size: 20–30nm; purity: 99%), copper oxide (CuO) NPs (particle size: 25–55nm; purity: 99.95%), molybdenum (Mo) NPs (particle size: 35–45 nm; purity: 95%), and boron (B) NPs (particle size: 100nm; purity: 99.9%) were used for this experiment (US Research Nanomaterials, Inc, Houston, TX, USA). The NPs with concentrations of 12.5 ppm, 25 ppm, 50 ppm, and 100 ppm were suspended in deionized water and ultrasonically dispersed for 60 min. The NPs size and suspension stability were measured using Delsa™ Nano Submicron Particle Size (Beckman Coulter Instruments, Corporation, Fullerton, California) and Zeta Potential device (Dispersion Technology Inc., Bedford Hills, New York, USA).

The obtained data showed that all prepared NPs suspensions were strongly anionic and stable, also monodispersed according to the polydispersity index (PDI) (Table 1).

2.2 Plants and Growth Conditions

The research was carried out in a walk-in controlled environment growth chamber (4 × 6 m, h = 3.2 m) located at the Institute of Horticulture, Lithuanian Research Centre for Agriculture and Forestry. Before sowing, green pea (*Pisum sativum* L. cv. Respect, Maribo Seed International ApS, Denmark) seeds were sterilized in 5% sodium hypochlorite solution for 15 min to assure surface sterility (Lehotai et al. 2011) and rinsed gently with deionized water several times. Then, seeds were soaked in water for 24 h. Sixteen pots were used for each treatment. The 3 seeds were sown in each 500-mL plastic pot which was treated as a single biological replication. Plants were seeded and grown in peat substrate, pH 6 (Profi 1, JSC Durpeta, Lithuania) for 30 days. The average amounts of nutrients (mg L⁻¹) in the substrate were as follows: N, 110; P₂O₅, 50; K₂O, 160. The average amounts of microelements (mg L⁻¹), Fe (4.5), Mn (0.5), Cu (0.1), B (0.02), Mo (0.03), and Zn (0.04), were also present. Electrical conductivity (EC) varied between 2.0 and 2.5 mS cm⁻¹ (± 0.03 mS cm⁻¹). The growth conditions were as follows: 16-h photoperiod, 20 ± 2/16 ± 2 °C day/night ambient air temperature, 60 % relative air humidity, ~220 μmol m⁻²

Table 1 Properties of NPs suspensions in DI water (pH=7.05) zeta potential, results represent the mean ± standard deviation, polydispersity index, and presents of nanoparticles between 1 and 100 nm

	Suspension of NPs of 50 ppm concentration			
	CuO	Mo	SiO ₂	B
Zeta potential (ζ; mV)	-26.68 ± 0.631	-24.92 ± 0.314	-20.64 ± 0.333	-28.54 ± 0.359
Polydispersity index (PDI)	0.211	0.218	0.34	0.237
NPs size 1-100 nm in suspension, %	55	68	70	50

s^{-1} photon flux density of photosynthetically active radiation (PAR) provided by high-pressure sodium (HPS) lamp (SON-T Agro, 400 W; Philips, Somerset, NJ, USA). At the 14 BBCH growth stage, when plants unfolded their 3–4 true leaf or had 3–4 tendrils developed (14 days after sowing) (Meier 1997), peas were watered with 100 mL of SiO_2 , CuO, Mo, or B NPs suspension containing different concentrations. After that, all plants were watered daily with deionized water and no additional fertilizer or nutrient solution was added. To minimize the potential effect of the growth chamber on plant response, randomization and regular spatial rearrangement of 10 pots per treatment were applied. Others were left unrotated for avoiding border effects. At the end of the experiment when plants reached the 31 BBCH growth stage (beginning of stem elongation), the peas were harvested for biometric, non-destructive measurements of leaf chlorophyll and nitrogen balance (NBI) indexes, and biochemical analyses.

2.3 Biometric and Non-destructive Measurements

For biometric measurements, the shoots were separated from the roots, and then shoot height and root length and fresh and dry weight were determined for randomly selected 10 plants per treatment ($n = 10$). The fresh biomass was measured with an electronic scale (Mettler Toledo AG64, Columbus, OH, USA), and dry biomass was determined following forced-air convection drying at 105°C to a constant dry weight (VENTICELL 222, MBT, Czech Republic). Non-destructive measurements of leaf chlorophyll and nitrogen balance (NBI) indexes in the fully developed leaves (10 plants per treatment, $n = 10$) were performed using a chlorophyll and flavonoid meter (Force-A Dualex® 4 Scientific, Ocala, FL, USA).

2.4 Antioxidant Properties and Total Phenolic Compounds

The extracts used to determine the total phenolic content and antioxidant activity were prepared by grinding 0.5 g of fresh sample with liquid nitrogen and diluting with 5 mL of 80% methanol. After 24 h, the samples were centrifuged for 10 min at 3000 rpm (Herrmle Z300K, Germany), extracts were filtered through cellulose filters, and the supernatant was used for further analyses. All biochemical analysis was performed in 3 biological replications.

The total content of phenolic compounds was determined as gallic acid equivalents. A 250 μ L aliquot of the sample extract was mixed with 250 μ L of 10% (w/v) Folin-Ciocalteu reagent, 500 μ L of 1 M Na_2CO_3 solution, and 2 mL of distilled water. After incubation for 20 min in the dark, the absorbance was measured at 765 nm (Ainsworth and Gillespie 2007) (M501, Spectronic Camspec Ltd., UK). The

total phenolic compounds quantity $mg\ g^{-1}$ was calculated from the calibration curve of the gallic acid.

Antioxidant properties of pea leaves were evaluated as the ABTS (2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) and DPPH (2-diphenyl-1-picrylhydrazyl), radical-scavenging activities.

The ABTS radical cation was obtained by incubating the 2 mM ABTS stock solution (0.112 g in 100 mL deionized water) with 400 μ L $K_2S_2O_8$ (0.1982 g in 10 mL deionized water) in the dark for 24 h. Thereafter, 50 μ L of the prepared sample was mixed with 2 mL of ABTS solution (ABTS stock solution was diluted 1:7), and the absorbance was measured after 11 min (plateau phase) at 734 nm (M501, Spectronic Camspec Ltd., UK). The ABTS scavenging activity of green pea extracts was calculated as the difference between the initial absorbance and after reacting for 10 min. It was expressed as ABTS mmol scavenged per 1 g of fresh weight (μ mol g^{-1} FW). Methanol was used as the blank solution (Re et al. 1999).

For the DPPH assay, a stable 0.1268 μ M DPPH (2,2-diphenyl-1-picrylhydrazyl radical) solution was prepared in methanol. Subsequently, 1 mL of the DPPH solution was transferred to a test tube and mixed with 100 μ L of the diluted green pea extract with 400 μ L methanol. The absorbance was scanned at 515 nm (M501, Spectronic Camspec Ltd., UK) while reacting for 16 min. The free radical scavenging capacity was expressed as μ mol of DPPH radicals scavenged per 1 g of fresh weight (μ mol g^{-1} FW) (Sharma and Bhat 2009).

2.5 Oxidative Stress Markers

The extracts used to determine the concentration of H_2O_2 and lipid peroxidation in pea tissue were prepared by grinding 0.1 g of fresh sample with liquid nitrogen and diluting with 4 mL of 0.1% TCA (trichloroacetic acid). After centrifugation for 10 min at 3000 rpm, the supernatant was used for further analyses.

Five hundred microliters of the supernatant were added to 1000 μ L of 1 M potassium iodide (KI). The absorbance of the mixture was scanned at 390 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., UK). The content of H_2O_2 was determined from a standard curve prepared with known concentrations of H_2O_2 and expressed in fresh weight (μ mol g^{-1} FW) (Velikova et al. 2000).

For the measurement of lipid peroxidation in pea leaves, the thiobarbituric acid (TBARS) test was used. The TBARS test determines malondialdehyde (MDA) content in pea leaves samples as the end product of lipid peroxidation. Five hundred microliters of the supernatant was added to 1000 μ L 0.5% (w/v) thiobarbituric acid (TBA) in 20% trichloroacetic acid (TCA). The mixture was incubated in boiling water for 30 min. The reaction stopped after the samples have cooled.

Then, the samples were centrifuged at 10 000×g for 5 min, and the absorbance of the supernatant was measured at 532 nm using a spectrophotometer (M501, Spectronic Camspec Ltd., UK). The value for non-specific absorbance at 600 nm was subtracted (Heath and Packer 1968). The amount of MDA–TBA complex (red pigment) in leaves was calculated:

$$C_{\text{MDA}} = (A_{532} - A_{600})/E_{\text{MDA}} \quad (1)$$

C_{MDA} concentration of MDA, μM
 A_{532} , A_{600} absorbance at wavelengths
 E_{MDA} MDA extinction coefficient $155 \text{ mM}^{-1} \text{ cm}^{-1}$

2.6 Determination of Macro- and Microelements

The macro- and microelements quantity in pea leaves, stem, and roots were determined using the microwave digestion technique combined with inductively coupled plasma optical emission spectrometry. Complete digestion of dry plant material (0.3 g) was achieved with 8 mL 65% HNO_3 using a microwave digestion system Multiwave GO (Anton Paar GmbH, Graz, Austria). The digestion program was as follows: (1) 170 °C reached within 3 min, digested for 10 min; (2) 180 °C reached within 10 min, digested for 10 min. After, full digestion samples were diluted to 50 mL with deionized water. The elemental profile was analyzed by an ICP–OES spectrometer (Spectro Genesis, SPECTRO Analytical Instruments, Kleve, Germany). The operating conditions employed for ICP–OES determination were 1300 W RF power, 12 L min^{-1} plasma flow, 1 L min^{-1} auxiliary flow, 0.8 L min^{-1} nebulizer flow, and 1 mL min^{-1} sample uptake rate. The analytical wavelengths chosen were P I 213.618 nm, K I 766.491 nm, S I 182.034 nm, Ca II 445.478 nm, Mg II 279.079 nm, Fe II 259.941 nm, Zn I 213.856 nm, Mn II 259.373 nm, and Cu I 324.754 nm. The calibration standards were prepared by diluting a stock multi-elemental standard solution (1000 mg L^{-1}) in 6.5% (v/v) nitric acid and by diluting stock phosphorus and standard sulfur solutions (1000 mg L^{-1}) in deionized water. The calibration curves for all the studied elements were in the range of 0.01–400 mg L^{-1} . The contents of macro- and microelements in the dry weight of pea are presented (Viršilė et al. 2020).

2.7 Statistical Analysis

All the values were expressed as mean \pm standard deviation. Data were analyzed using the analysis of variance (ANOVA) test followed by Tukey HSD at $p \leq 0.05$ to identify significant differences between control peas that have not been affected by NPs and plants that were exposed to different NPs concentrations. Pearson's correlation coefficient was used to determine relationships between antioxidant

activity assays. All statistical analyses were performed using XLSTAT (XLstat, Addinsoft, Paris, France, 2020).

3 Results

3.1 The Effects of Nanoparticles on the Pea Growth Parameters

Plants subjected to CuO, SiO_2 , B, Mo NPs at different concentrations showed distinct morphological responses (Table 2). Peas irrigated with 12.5 ppm CuO NPs suspension had the most positive effect on plant shoot height (40%), root length (104%), aboveground fresh (87%) and dry biomass (47%), root fresh (109%), and dry biomass (56%) compared to control plants (watered with water). Using the 100 ppm of Mo NPs, it had the greatest positive effect on pea shoot height (48%), root length (85%), aboveground fresh (83%), dry biomass (61%), root fresh (122%), and dry biomass (72%) compared to control plants. The suspension of B NPs had a similar effect to CuO NPs, as the 12.5 ppm concentration had the greatest positive effect on pea shoot height (57%), root length (92%), aboveground fresh (95%), dry biomass (116%), root fresh (104%), and dry biomass (41%) compared to control plants. Suspension of SiO_2 NPs had a comparable effect to Mo NPs, as concentrations of 100 and 50 ppm had the greatest positive effect on pea shoot height (44%), root length (75%), aboveground fresh (109%), dry biomass (135%), root fresh (119%), and dry biomass (136%) compared to control.

Chlorophyll and flavanol index showed statistically significant differences between plants irrigated with different NPs suspensions (Table 3). Both indices was most dependent on applied NPs concentration on pea plants; CuO at 12.5 ppm, Mo at 50ppm, B at 12.5ppm, and SiO_2 at 50ppm increased chlorophyll index by 85%, 76%, 90%, and 100% and flavanol index by 37%, 28%, 44%, and 45%, accordingly. CuO (12.5), Mo (50ppm), B (12.5ppm), and SiO_2 (50ppm) NPs had a statistically significant effect on the NBI, increasing it by 38%.

3.2 The Effects of NPs on the Antioxidative System in Pea Leaves

In pea plants, a significant effect on phenolic compounds was found under all applied NPs within the different concentrations, except for SiO_2 under 12.5 ppm (Fig. 1). The strongest statistically significant effects on DPPH radical scavenging activity (Fig. 2) were found using SiO_2 (100 ppm), CuO (12.5 ppm), Mo (50 ppm) NPs; it was increased by 35%, 12%, 34% respectively. However, B NPs did not have a statistically significant difference in DPPH radical scavenging activity in pea leaves. The greatest statistically significant impact on ABTS radical scavenging activity (Fig. 3) in pea

Table 2 Effects of NPs 100, 50, 25, 12.5 ppm suspension on pea shoot height, root length, fresh and dry shoot, and root biomass compared to control 0 ppm. Mean values within columns followed by different letters differ significantly at $p < 0.05$ ($n=10$) according to Tukey (HSD) test

NPs	Concentration, ppm	Shoot height, cm	Root length, cm	Aboveground fresh biomass, g	Aboveground dry biomass, g	Root fresh biomass, g	Root dry biomass, g
CuO	100	20.34 d	8.26 b	1.240 c	0.217 d	0.469 bc	0.037 c
	50	23.28 c	8.44 b	1.868 b	0.323 b	0.501 b	0.055 b
	25	26.34 b	9.32 a	2.353 a	0.361 a	0.668 a	0.058 b
	12.5	28.14 a	9.10 a	2.472 a	0.353 a	0.763 a	0.069 a
	0	20.02 d	4.46 c	1.322 c	0.240 c	0.365 c	0.044 c
Mo	100	29.64 a	8.26 a	2.429 a	0.388 a	0.809 a	0.076 a
	50	27.66 b	7.26 b	2.220 b	0.380 a	0.667 b	0.065 b
	25	24.70 c	7.42 b	1.525 d	0.304 b	0.577 b	0.058 b
	12.5	24.64 c	6.94 c	1.672 c	0.259 c	0.627 b	0.063 b
	0	20.02 d	4.46 d	1.322 e	0.240 c	0.365 c	0.044 c
B	100	19.34 e	5.46 c	1.364 c	0.270 d	0.341 d	0.043 b
	50	20.44 c	4.98 d	1.502 c	0.366 b	0.444 c	0.044 b
	25	24.30 b	7.26 b	2.112 b	0.323 c	0.598 b	0.053 ab
	12.5	30.34 a	8.58 a	2.578 a	0.518 a	0.696 a	0.061 a
	0	20.02 d	4.46 e	1.322 c	0.240 d	0.365 d	0.044 b
SiO ₂	100	28.66 a	9.20 a	2.857 a	0.501 b	1.049 b	0.089 b
	50	28.84 a	9.20 a	2.769 a	0.564 a	1.166 a	0.105 a
	25	28.20 a	9.10 a	2.698 a	0.465 c	0.782 c	0.082 c
	12.5	25.70 b	7.84 b	2.396 b	0.388 d	0.624 d	0.073 d
	0	20.02 c	4.46 c	1.322 c	0.240 e	0.365 e	0.044 e

plants was found using SiO₂ (100 ppm), CuO (12.5 ppm), Mo (50 ppm), and B (50 ppm) NPs, which increased by 22%, 17%, 25%, and 19% respectively compared to control plants.

The highest and positive correlation between the content of phenolic compounds, DPPH and ABTS scavenging activity, was found in pea plants exposed to SiO₂ NPs suspension ($r_{(DPPH)} = 0.954$, $p < 0.001$; $r_{(ABTS)} = 0.964$, $p < 0.001$) (Table 4). The lowest but still positive correlation between the content of phenolic compounds, DPPH and ABTS scavenging activity, was found in pea plants exposed to B NPs suspension ($r_{(DPPH)} = 0.590$, $p < 0.05$; $r_{(ABTS)} = 0.435$ not significant).

3.3 The Effects of NPs on the Oxidation System in Pea Leaves

The results showed that the accumulation of H₂O₂ in the pea leaves was mostly significantly lower or comparable to the control plants, and these effects depended mainly on the applied NPs and concentration (Fig. 4). It is worth noting that the accumulation of H₂O₂ in pea leaves was significantly induced when plants were exposed to B NPs and CuO NPs at 100 ppm and SiO₂ NPs at 12.5 ppm.

The level of damage related to oxidative stress was determined by monitoring the differences in lipid peroxidation in terms of MDA formation. Figure 5 shows a significant

difference in MDA concentration in the pea leaves of all treatments. The results indicate that the plant treated with 100 ppm B NPs had significantly higher MDA content compared to control plants, while the lowest MDA content was the 12.5 ppm and 25 ppm B NPs-treated plants. For Mo NPs results, 50 ppm and 100 ppm treatments to the plants had reduced MDA content compared to untreated plants. Comparing the effects of the four concentrations of CuO NPs suspension, it was obvious that the lowest applied concentration (12.5 ppm) showed the lowest MDA content in pea leaves. All the treated plants with SiO₂ NPs show a lower MDA content when compared to the untreated plants.

3.4 NPs Effect on Macroelements (Phosphorus (P), Potassium (K), Calcium (Ca), magnesium (Mg)) and Microelements (Sulfur (S), Iron (Fe), Zinc (Zn), Manganese (Mn), Copper (Cu), Molybdenum (Mo), Boron (B)) of Peas Leave, Stem, and Root

SiO₂, CuO, Mo, and B NPs had the greatest statistically significant effect on pea leaf trace elements. They increased the amount of Ca and Mg in pea leaves compared to control plants. Moreover, the amount of Ca, K, Fe, Zn, and Mn was increased in pea stems. Furthermore, it was observed that the use of SiO₂ and CuO NPs

Table 3 Non-destructive measurement parameters in green peas leave, chlorophyll and flavanol index, nitrogen balance index (NBI). Different letters indicate statistically significant differences between means according to the Tukey (HSD) test at the confidence level $p = 0.05$ ($n=10$).

NPs	Concentration, ppm	Chlorophyll index	Flavanol index	NBI
CuO	100	22.842 c	1.030 a	23.116 b
	50	27.884 b	1.026 a	27.248 ab
	25	31.488 ab	1.089 a	29.098 ab
	12.5	34.224 a	1.053 a	32.500 a
	0	18.503 c	0.768 b	24.154 b
Mo	100	31.889 a	1.126 a	28.369 b
	50	32.584 a	0.983 b	33.217 a
	25	25.129 b	0.877 bc	28.715 b
	125	26.039 b	0.888 bc	29.274 b
	0	18.503 c	0.768 c	24.154 c
B	100	26.285 b	1.036 a	25.531 b
	50	24.726 b	0.820 b	30.177 a
	25	25.129 b	0.877 bc	28.715 b
	12.5	35.238 a	1.106 a	32.118 a
	0	18.503 c	0.768 b	24.154 b
SiO ₂	100	34.369 ab	1.152 a	29.806 ab
	50	37.084 a	1.117 a	33.762 a
	25	34.258 ab	1.071 a	32.467 a
	12.5	33.369 bc	1.078 a	32.223 a
	0	18.503 c	0.768 b	24.154 b

increased the S content in the leaves and roots by 3 times. Mo NPs increased P, Ca, Mg, and Fe two times; besides, a strong accumulation of Mo content in pea roots was determined (Table 5). B NPs affected the amount of Ca, Fe, and Mn in pea roots by 36%, 38%, and 476%, respectively (Table 6). Exposure of peas to 100 ppm CuO NPs (Table 7) increased the copper content 14 times in the leaves, 4 times in the stem, and decreased 10 times in the

roots. SiO₂ NPs affected the accumulation of macronutrients in pea leaves and stems more than in roots (Table 8).

4 Discussion

The physicochemical interaction of NPs with plants in terms of their energy and surface charge results in the modification of specific membrane surface proteins, transporters, and receptors (Juárez-Maldonado et al. 2019). Inside the leaves, there is a biological membrane with hydrophobic and hydrophilic components and unequal distribution of negative charges (lignin surface potential and cellulose fibers are -45 and -15 mV, respectively) (Mittal et al. 2020). NPs with a zeta potential between -10 and $+10$ mV are approximately neutral, and NPs with a zeta potential greater than $+30$ mV or less than -30 mV are considered strongly cationic and anionic, respectively, and stable due to particles and particle repulsion (Clogston and Patri 2011). According to International Standards Organizations (ISOs), polydispersity index values <0.05 are more common for monodisperse samples and values >0.7 are common for a wide (e.g., polydisperse) particle distribution (ISO standards ISO 22,412:2017 and ISO 22,412:2017). Researchers reported that the zeta potential of CuO NPs is -34.4 ± 0.5 mV in suspension in deionized water at pH 7 when NPs were purchased at 10–100nm in primary size (Keller et al. 2018; Adeleye et al. 2014; Hong et al. 2014). In this research, the zeta potential of CuO NPs was found to be -26.68 ± 0.6 mV in suspension in deionized water when NPs were purchased at 25–55nm in primary size. The zeta potential of SiO₂ NPs was obtained to be -40 mV by other scientists (El-Naggar et al. 2020); meanwhile, the zeta potential of SiO₂ NPs was -20.64 ± 0.3 in our study. Unfortunately, there are no scientific articles with already performed measurements of zeta potential using Mo and B nanoparticles. The negative value showed to be an anionic type of suspensions in all studied systems.

Fig. 1 Effects of SiO₂, CuO, Mo, B NPs on the content of phenolic compounds in pea leaves. Data are means \pm SE of 3 replicates, $n = 3$, Tukey (HSD): * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$ compared with control

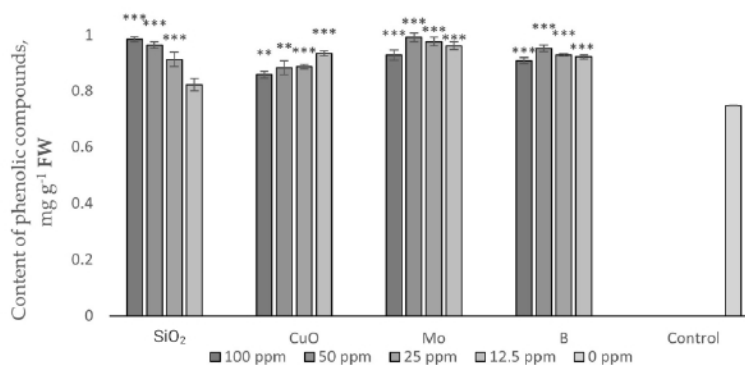


Fig. 2 Effects of SiO₂, CuO, Mo, B NPs on the DPPH radical scavenging activity. Data are means \pm SE of 3 replicates, $n = 3$, Tukey (HSD): * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$ compared with control

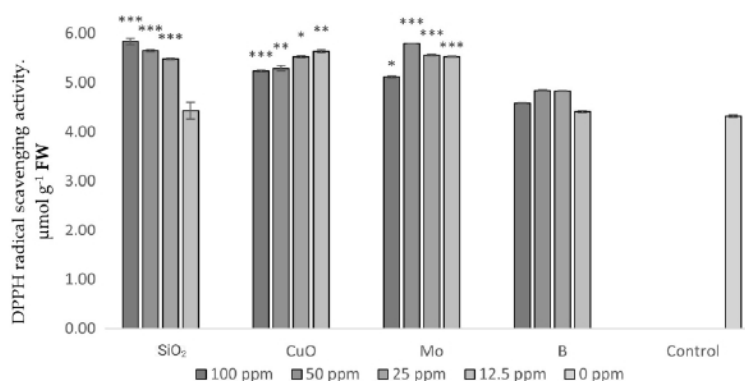
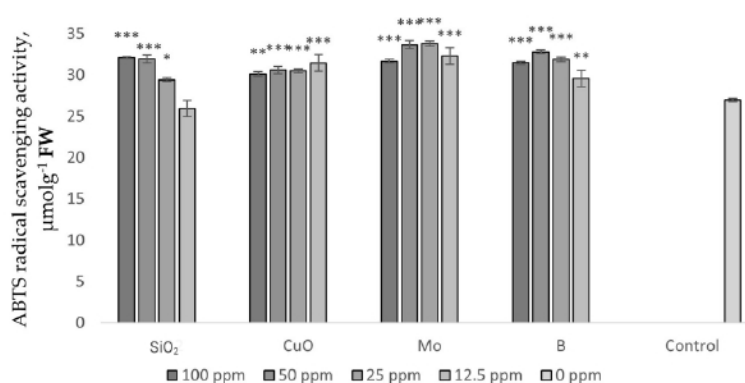


Fig. 3 Effects of SiO₂, CuO, Mo, B NPs on the ABTS radical scavenging activity. Data are means \pm SE of 3 replicates, $n = 3$, Tukey (HSD): * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$ compared with control



By participating in electron transfer, NPs induce various responses in plants, increase the activity of plant enzymes, promote the conversion of nitrates to ammonia, intensify plant respiration and photosynthesis processes, synthesize enzymes and amino acids, enhance carbon, and nitrogen nutrition, and thus directly affect plant mineral nutrition.

The effects of CuO NPs on plants are extensively studied compared to other NPs researched in this study. Comparing the results of other authors (Nair and Chung 2015), it is observed that the effect of CuO NPs on plants is similar; the higher the CuO concentration, the higher the toxicity determined in peas, starting at 100 ppm. Toxic effects were confirmed in soybean by elevated levels of lipid peroxidase and peroxide in plants at concentrations of 100 ppm CuO NPs and higher (Yusefi-Tanha et al. 2020). A reduction in soybean yield was also observed with the use of CuO NPs at concentrations of 50–100 ppm (Ochoa et al. 2017). Does the question, therefore, arise as to what effect could a lower concentration of CuO NPs have on plants? The experiment shows that using a concentration of 12.5 ppm can produce a positive response both in the antioxidant system and in

reducing oxidative stress such as lipid peroxidation and peroxide content in peas. Researchers have found (Ogunkunle et al. 2018; Yusefi-Tanha et al. 2020) that using CuO NPs up to 25 nm has a stronger effect on lipid peroxidation and peroxide concentration than using 60–80 nm CuO NPs, so using smaller particles would be beneficial to reduce the peroxide concentration in pea leaves. By the way in agreement with the data of other authors (Ogunkunle et al. 2017, 2018), it can be seen from the macro-microelements that the higher the concentration of CuO NPs the more it accumulates in pea leaves and roots. Cu belongs to the micronutrients that are faced with the problem of poor bioavailability; though this element is essential for the vital function of plants, the use of Cu as an NPs size could have an appropriate solution. However, it should be mentioned that excess Cu can affect plants as heavy metal and can therefore be harmful to plants and increase the level of oxidative stress.

Mo is imported into plants by special molybdenum transporters. Mo is biologically inactive unless it is complexed by a specific prosthetic group (except bacterial nitrogenase). This Mo cofactor (Moco) after biosynthesis is distributed to

Table 4 Pearson's correlation coefficients (*r*) for the relationships between antioxidant assays and phenolic contents ($^*p < 0.05$; $^{**}p < 0.01$; $^{***}p < 0.001$)

Control	Phenols	ABTS	DPPH
Phenols	1		
ABTS	0.890**	1	
DPPH	0.970**	0.753***	1
CuO NPs	Phenols	ABTS	DPPH
Phenols	1		
ABTS	0.694*	1	
DPPH	0.793*	0.683*	1
B NPs	Phenols	ABTS	DPPH
Phenols	1		
ABTS	0.435	1	
DPPH	0.590*	0.754*	1
SiO ₂ NPs	Phenols	ABTS	DPPH
Phenols	1		
ABTS	0.964***	1	
DPPH	0.954***	0.927***	1
Mo NPs	Phenols	ABTS	DPPH
Phenols	1		
ABTS	0.727*	1	
DPPH	0.837**	0.635*	1

both Mo-enzyme families of Moco-binding proteins (sulfite oxidase (SO), nitrate reductase (NR), mitochondrial amidoxime reducing component (mARC)) and Mo-enzymes (aldehyde oxidase (AO), xanthine dehydrogenase (XDH)). The most important Mo-enzyme for plant survival is cytosolic NR, which catalyzes the first step in nitrate uptake (Kaufholdt et al. 2017; Mendel and Schwarz 2011). The conversion of nitrates to nitrites is essential for pea plant growth and development as shown by studies with Mo NPs (Taran et al. 2014). The results of our research showed a statistically significant increase in pea aboveground biomass, shoot height, and root height as compared to control plants. It is

also important to note that higher antioxidant activity and lower levels of H₂O₂ and MDA in peas were found as plants were treated with 50 and 100 ppm Mo NPs. Similar changes were observed in another study [8] showing increased activity of the chickpea antioxidant system, particularly increased catalase activity at 8 ppm Mo NPs.

In our study, all mineral uptakes in pea plants were significantly affected by the Mo NPs treatments. In leaves, Mo NPs induced the accumulation of P, K, Ca, Mg, and in stems; Fe also contributes to these elements. In the roots, the largest accumulation was observed of all identified elements, as well as an exceptional effect was observed in the increase of Cu and Fe content compared to the effect of other used NPs. Also, the application of Mo NPs to pea plants induced the accumulation of Mo in pea leaves, stems, and roots.

One of the main functions of B in plants has been described as its ability to produce esters with rhamnolacturonan II (RGII). The formation of this borate ester is necessary to perform cell wall functions and maintain structure (Ryden et al. 2003). Ninety percent of all B entering the plant is carried to the plant cell wall (Goldbach and Wimmer 2007). Plasma membranes of pea root nodules and uninfected pea root cells also contain RGII glycoproteins, but these proteins do not occur in B-deficient cells, suggesting that borate stabilizes them in peri bacteroid and plasma membranes (Bolaños et al. 2001). B is also required for the regulation of glycoproteins, which are necessary as indicators for the differentiation of bacteria into a nitrogen-fixing form (Bolaños et al. 2004). The results of this study showed that low concentrations of B NPs have a positive effect on pea plant growth parameters and increased efficiency of the antioxidant activity was observed. When higher concentrations of B NPs were used, an increase in MDA and H₂O₂ levels in pea leaves was observed. H₂O₂ plays an important role in plants, especially because photosynthesis provides an additional source compared to non-photosynthetic organisms. H₂O₂ is one of the signals of the state

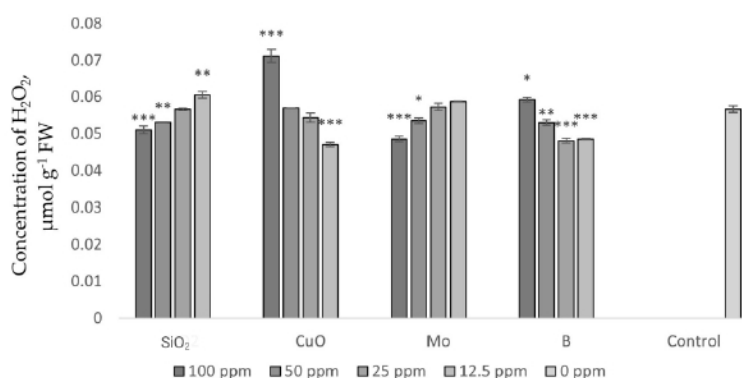
Fig. 4 Effects of SiO₂, CuO, Mo, B NPs on the peroxide amount. Data are means \pm SE of 3 replicates, $n = 3$, Tukey (HSD): * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$ compared with control

Fig. 5 Effects of SiO₂, CuO, Mo, B NPs on the MDA amount. Data are means ± SE of 3 replicates, *n* = 3, Tukey (HSD): **p* < 0.05; ***p* < 0.01; ****p* < 0.001 compared with control

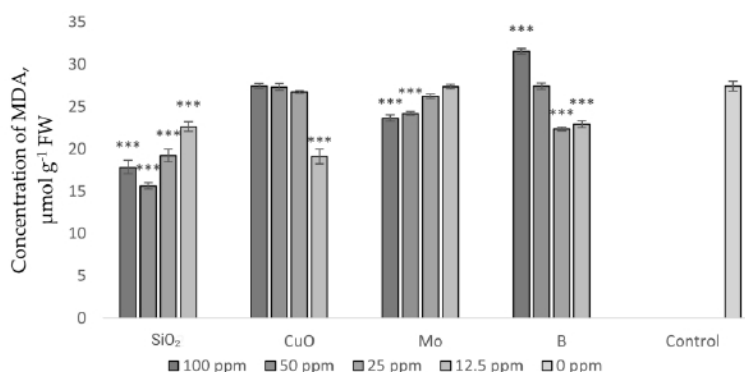


Table 5 Effects of Mo NPs suspension on macroelements (phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg)) and microelements (sulfur (S), iron (Fe), zinc (Zn), manganese (Mn), copper (Cu), molybdenum (Mo), boron (B)) of peas leave, stem, and root compared to control 0 ppm. Mean values within columns followed by different letters differ significantly at *p* < 0.05 according to Tukey (HSD) test (*n*=3)

Treatment Mo NPs	Macroelements, mg g ⁻¹ DW				Microelements, μg g ⁻¹ DW						
	P	K	Ca	Mg	S	Fe	Zn	Mn	Cu	Mo	B
Leaves											
100 ppm	3.897 a	15.953 a	40.871 a	10.761 a	0.736 b	104.827 a	19.366 a	48.636 b	8.803 a	26.408 a	6.382 a
50 ppm	2.952 b	11.001 b	23.268 b	6.629 b	0.443 c	89.566 c	10.983 b	37.068 c	5.107 b	8.512 b	1.098 c
25 ppm	1.545 c	6.166 c	17.426 c	4.751 d	0.298 d	41.099 d	3.214 e	20.999 d	2.786 bc	5.033 c	0.429 d
12.5 ppm	1.308 d	4.054 d	12.636 d	3.113 e	0.253 e	31.394 e	3.552 d	8.353 e	3.936 b	4.512 c	1.440 c
0 ppm	1.607 e	11.333 b	10.138 e	5.545 e	4.515 a	94.868 b	7.905 c	97.233 a	1.186 e	1.858 d	4.534 b
Stem											
100 ppm	2.523 a	9.108 b	20.743 a	5.808 a	0.326 b	80.251 a	9.049 c	29.635 a	6.391 ab	28.077 a	4.298 a
50 ppm	1.629 b	9.472 a	18.125 b	5.621 a	0.314 b	64.176 b	11.375 a	16.583 c	6.286 ab	11.840 b	3.772 a
25 ppm	1.620 b	7.451 d	12.770 c	4.319 b	0.260 d	58.838 b	9.545 b	14.800 d	7.611 a	10.573 c	0.967 c
12.5 ppm	1.359 c	6.198 e	20.869 a	5.787 a	0.282 c	51.886 bc	1.526 d	24.907 b	4.524 b	8.596 d	0.491 c
0 ppm	1.069 d	7.725 c	12.787 c	4.220 b	0.970 a	39.531 c	1.019 e	11.931 e	2.333 c	2.694 e	1.547 b
Roots											
100 ppm	2.020 a	19.477 a	20.643 b	7.331 a	2.378 b	7217.571 a	31.399 a	180.463 a	34.253 a	22.766 c	2.253 c
50 ppm	1.911 b	16.868 b	23.189 a	7.233 a	2.372 b	4894.413 b	26.005 b	129.406 b	26.623 b	17.263 c	5.250 c
25 ppm	1.641 d	8.873 d	16.873 c	5.117 c	2.321 c	7061.15 a	6.371 d	20.979 c	8.081 c	158.103 a	24.263 a
12.5 ppm	1.941 b	11.773 c	18.975 b	5.876 b	2.336 c	7093.522 a	15.216 c	26.428 c	10.322 c	110.892 b	17.849 b
0 ppm	1.859 c	15.999 b	9.293 d	5.788 b	2.607 a	2975.111 b	7.037 d	121.188 b	21.241 b	19.807 c	7.688 c

of photosynthesis and stomatal movements (Smirnov and Arnaud 2019). Therefore, chlorophyll and flavanol indices were not adversely affected even at 100 ppm B NPs concentration, but shoot height and root length of pea were lower than control plants. The suspension of B NPs was observed to have a positive effect on Ca and Mg in pea leaves, K, Ca, and B in stems, Ca, Mg, S, Fe, Mn, and B in roots.

SiO₂ NPs have a positive effect on plant growth; increase their biomass, anatomy, and physiology; modify tissue differentiation; activate defense systems, and help to adapt to stressful conditions (Luyckx et al. 2017). The effects of SiO₂ NPs are widely studied in different plants; but to our

knowledge, there is only one research on the effects of these NPs on lentils and soybeans from the legume plant family. It has been reported that the nano-sized mixture of SiO₂ and TiO₂ increased the nitrate reductase activity of soybean (*Glycine max*), enhanced the absorption and utilization of water and fertilizers, stimulated the antioxidant system, and accelerated germination and growth (Lu et al. 2002). The concentration of 60.08 ppm SiO₂ NPs significantly mitigated the negative effects of salt stress on germination, shoot and root length, seedling weight, mean germination time in lentil seedlings, and seedling vigor indices (Janmohammadi et al. 2015), promoting the antioxidant system of soybeans,

Table 6 Effects of B NPs suspension on macroelements (phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg)) and microelements (sulfur (S), iron (Fe), zinc (Zn), manganese (Mn), copper (Cu), molybdenum (Mo), boron (B)) of peas leave, stem, and root compared to control 0 ppm. Mean values within columns followed by different letters differ significantly at $p < 0.05$ according to Tukey (HSD) test ($n=3$)

Treatment B NPs	Macroelements, mg g ⁻¹ DW						Microelements, µg g ⁻¹ DW					
	P	K	Ca	Mg	S	Fe	Zn	Mn	Cu	Mo	B	
Leaves												
100 ppm	3.073 a	10.098 d	23.255 d	5.158 e	0.783 a	70.943 d	5.331 c	29.026 d	1.256 c	2.001 b	1.929 d	
50 ppm	2.732 b	10.918 c	25.218 a	6.535 c	0.545 d	79.723 c	2.147 d	41.959 b	4.148 a	2.537 a	11.421 a	
25 ppm	2.553 c	10.901 c	25.190 a	6.764 b	0.699 c	87.903 b	9.525 a	30.220 c	1.305 c	2.563 a	4.297 c	
12.5 ppm	3.035 a	11.756 a	24.768 b	7.088 a	0.742 b	93.364 b	5.191 c	28.339 d	3.157 b	2.525 a	1.797 d	
0 ppm	1.607 d	11.333 b	10.138 c	5.545 d	0.515 d	94.868 a	7.905 b	97.233 a	1.186 c	1.858 b	4.534 c	
Stem												
100 ppm	1.003 e	7.731 b	12.651 e	4.195 d	0.305 d	39.956 c	1.006 d	10.431 d	1.208 d	2.385 b	19.759 a	
50 ppm	1.260 c	7.719 b	14.279 c	4.462 c	0.457 c	50.103 b	6.808 a	12.685 b	1.175 d	2.498 ab	1.574 d	
25 ppm	1.408 b	8.166 a	14.744 b	4.984 b	0.311 d	67.360 a	1.949 c	17.869 a	6.606 a	2.166 b	6.943 c	
12.5 ppm	1.441 a	7.396 c	17.380 a	5.768 a	0.502 b	67.335 a	2.149 b	12.655 b	4.298 b	2.229 b	9.405 d	
0 ppm	1.069 d	7.725 b	12.787 d	4.220 d	0.970 a	39.531 c	1.019 d	11.931 c	2.333 c	2.694 a	1.547 c	
Roots												
100 ppm	1.943 b	15.587 c	5.255 e	6.170 c	1.478 b	4221.563 b	6.874 a	117.043 ab	8.098 b	9.416 b	18.411 b	
50 ppm	2.533 a	16.794 a	8.542 c	6.442 b	1.376 c	3357.684 d	2.165 b	114.313 b	14.475 ab	9.064 b	13.741 c	
25 ppm	1.892 b	16.174 b	5.902 d	6.782 a	1.471 b	3780.957 c	7.363 a	103.916 c	11.044 b	9.538 b	10.404 d	
12.5 ppm	2.503 a	14.690 d	11.109 a	5.133 e	1.124 d	4816.125 a	2.549 b	121.256 a	8.764 b	9.560 b	36.078 a	
0 ppm	1.859 b	15.999 b	9.293 b	5.788 d	2.607 a	2975.111 e	7.037 a	121.188 a	21.241 a	19.807 a	7.688 e	

Table 7 Effects of CuO NPs suspension on macroelements (phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg)) and microelements (sulfur (S), iron (Fe), zinc (Zn), manganese (Mn), copper (Cu), molybdenum (Mo), boron (B)) of peas leave, stem, and root compared to control 0 ppm. Mean values within columns followed by different letters differ significantly at $p < 0.05$ according to Tukey (HSD) test ($n=3$)

Treatment CuO NPs	Macroelements, mg g^{-1} DW					Microelements, $\mu\text{g g}^{-1}$ DW					
	P	K	Ca	Mg	S	Fe	Zn	Mn	Cu	Mo	B
Leaves											
100 ppm	1.774 d	10.138 c	22.741 a	7.462 a	6.021 a	55.703 d	0.996 c	25.445 e	13.993 a	1.051 b	0.608 e
50 ppm	1.994 b	10.441 b	15.697 c	5.452 e	4.546 c	45.781 e	0.352 d	30.731 c	7.243 b	0.914 bc	1.899 b
25 ppm	1.879 c	11.467 a	21.135 b	6.691 c	5.007 b	70.072 c	4.700 b	28.393 d	3.933 c	0.624 c	1.007 d
12.5 ppm	2.125 a	11.220 a	22.608 a	6.858 b	4.358 c	90.982 b	4.249 b	32.558 b	4.355 c	0.637 c	1.381 c
0 ppm	1.607 e	11.333 a	10.138 d	5.545 d	4.515 c	94.868 a	7.905 a	97.233 a	1.186 d	1.858 a	4.534 a
Stem											
100 ppm	1.262 a	9.863 a	15.337 a	5.449 b	2.503 b	42.700 bc	2.842 a	11.693 c	5.233 d	0.517 b	1.421 a
50 ppm	1.215 b	6.613 d	10.970 d	4.058 e	3.007 a	52.464 b	1.064 c	10.493 d	6.995 c	0.836 b	0.228 d
25 ppm	0.965 e	7.812 c	13.298 b	4.732 c	2.527 b	65.627 a	2.555 b	10.307 e	8.721 a	0.617 b	1.145 b
12.5 ppm	1.088 c	8.998 b	15.580 a	6.165 a	2.984 a	53.511 b	3.024 a	13.105 a	8.317 b	0.588 b	0.924 c
0 ppm	1.069 d	7.725 c	12.787 c	4.220 d	0.970 c	39.531 c	1.019 c	11.931 b	2.333 e	2.694 a	1.547 a
Roots											
100 ppm	1.940 b	20.067 b	9.035 b	8.120 a	9.116 a	1948.500 e	7.557 a	51.500 e	2.405 b	1.141 c	4.939 d
50 ppm	1.976 a	15.083 d	5.424 e	5.960 d	6.210 d	3405.684 a	1.145 c	82.136 c	1.312 c	1.249 b	7.846 b
25 ppm	1.805 d	21.459 a	6.475 d	7.412 b	7.860 b	3172.238 b	0.969 c	78.852 d	1.242 c	1.332 b	8.851 a
12.5 ppm	1.810 d	19.829 b	7.057 c	6.608 c	6.721 c	2792.113 d	3.765 b	87.297 b	1.305 c	1.685 b	6.191 c
0 ppm	1.859 c	15.999 c	9.293 a	5.788 e	2.607 e	2975.111 c	7.037 a	121.188 a	21.241 a	19.807 a	7.688 b

Table 8 Effects of SiO₂ NPs suspension on macroelements (phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg)) and microelements (sulfur (S), iron (Fe), zinc (Zn), manganese (Mn), copper (Cu), molybdenum (Mo), boron (B)) of pea leave, stem, and root compared to control 0 ppm. Mean values within columns followed by different letters differ significantly at *p* < 0.05 according to Tukey (HSD) test (*n*=3)

Treatment SiO ₂ NPs	Macroelements, mg g ⁻¹ DW										Microelements, µg g ⁻¹ DW												
	P	K	Ca	Mg	S	Fe	Zn	Mn	Cu	Mo	B	P	K	Ca	Mg	S	Fe	Zn	Mn	Cu	Mo	B	
Leaves																							
100 ppm	3.239 b	12.226 a	22.537 b	6.286 b	1.373 d	77.914 c	5.290 c	34.783 b	3.954 b	1.684 b	9.373 a												
50 ppm	3.335 a	9.651 e	23.122 a	6.494 a	1.662 b	73.659 d	3.870 e	32.583 c	4.549 a	1.561 b	9.492 a												
25 ppm	2.496 c	10.648 c	20.695 c	6.369 ab	1.563 c	91.719 b	9.402 a	29.202 d	1.537 c	1.673 b	7.685 b												
12.5 ppm	2.296 d	10.185 d	19.653 d	6.081 c	1.429 d	66.054 e	4.874 d	32.058 c	4.062 b	1.312 c	9.624 a												
0 ppm	1.607 e	11.333 b	10.138 e	5.545 d	4.515 a	94.868 a	7.905 b	97.233 a	1.186 a	1.858 a	4.534 c												
Stem																							
100 ppm	1.567 a	9.026 b	15.896 a	4.847 a	0.854 d	50.378 b	10.746 a	13.778 c	2.234 b	1.543 b	8.033 b												
50 ppm	1.354 d	7.470 e	14.747 b	4.841 a	0.896 c	55.198 b	8.629 b	15.063 a	2.451	1.328 bc	4.545 c												
25 ppm	1.434 c	9.397 a	12.391 e	4.504 b	0.885 c	95.673 a	5.617 d	11.892 d	3.406b a	1.195 c	10.577 a												
12.5 ppm	1.491 b	7.973 c	14.496 c	4.486 b	1.042 a	50.335 b	8.466 c	14.356 b	1.235 c	1.481 bc	1.532 d												
0 ppm	1.069 e	7.725 d	12.787 d	4.220 c	0.970 b	39.531 c	1.019 e	11.931 d	2.333 b	2.694 a	1.547 d												
Roots																							
100 ppm	1.717 d	15.169 c	6.919 b	6.154 b	2.696 a	3581.036 c	6.637 c	91.028 c	1.532 b	3.355 b	1.167 c												
50 ppm	1.871 b	14.771 d	6.100 c	6.653 a	2.316 d	4147.903 a	8.246 a	87.196 d	1.818 b	3.636 b	7.467 a												
25 ppm	1.904 a	16.522 a	5.469 d	5.967 c	2.331 d	3037.381 d	3.556 e	115.883 b	1.185 b	2.462 c	2.188 b												
12.5 ppm	1.810 c	16.583 a	6.165 c	6.068 bc	2.415 c	4060.693 b	4.762 d	84.242 e	0.519 c	3.550 b	0.087 d												
0 ppm	1.859 b	15.999 b	9.293 a	5.788 d	2.607 b	2975.111 d	7.037 b	121.188 a	21.241 a	19.807 a	7.688 a												

including the activity of SOD, POD, and CAT enzymes (Shen et al. 2010). However, previous studies suggest that the higher SiO₂ NPs concentrations should be used with caution as the inhibitory effect of these NPs (120.16 ppm) was found for lentils germination (Janmohammadi et al. 2015). Our study indicates that treatments with SiO₂ at higher concentrations (50 and 100 ppm) increased antioxidant activity and reduced H₂O₂ content and MDA. In addition, the height of pea shoots, root length, and biomass accumulation was also increased as well as chlorophyll and flavanol indices. There was observed a particular effect on the accumulation of macroelements in pea plants. The content of P, K, Ca, and Mg in pea leaves and stems increased.

Despite the potential benefits mentioned above, the use of nanotechnologies in the agricultural sector is relatively limited and has not yet entered the market compared to other industries. Given the possible practical application of NPs in the agricultural sector, a preliminary cost-benefit analysis should be considered before placing NP-related products on the market, with a proper risk assessment. More extensive research is planned to conduct with greenhouse-grown peas to harvest maturity using CuO, Mo, B, and SiO₂ NPs, and it would be valuable to delve into abiotic stress reduction using various NPs.

5 Conclusions

Nanotechnology is a promising area in agronomy to reduce fertilizer levels. The use of nano-sized particles as fertilizer needs to be thoroughly analyzed, as their effects may depend not only on concentration, surface area, and size but also on the species of plant. In this experiment, studies of pea plants exposed to nanoparticles of silica, copper, molybdenum, and boron revealed that both positive and negative effects could occur. According to biometric indices, non-destructive measurements of leaf chlorophyll and nitrogen balance, chemical antioxidant efficiency analyses, and oxidative stress biomarkers, the most effective and statistically reliable effects were obtained using 100, 50 ppm concentrations of silica and molybdenum nanoparticles as well as using 12.5 ppm concentration of copper oxide and boron nanoparticles. Besides this study, it showed that higher concentrations of copper oxide and boron nanoparticles can induce oxidative stress and be toxic to pea plants.

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CRedit authorship contribution statement Rūta Sutulienė: conceptualization, methodology, investigation, formal analysis, writing—original draft, visualization. Lina Ragelienė: methodology, resources, writing—review and editing. Pavelas Duchovskis: conceptualization, methodology, resources. Jurga Miliauskienė: conceptualization, methodology, resources, writing—review and editing, supervision.

Declarations

Competing Interest The authors declare no competing interests.

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SANTRAUKA

ĮVADAS

Žemės ūkio pasėliams vis didesnę neigiamą poveikį daro smarkios liūtys, sausros, ekstremalios temperatūros, stiprūs vėjai ir naujų tipų patogenų plitimas, atsirandantis dėl besikeičiančio klimato (Shahzad ir kt., 2021; Rivero ir kt., 2022). Sausros stresas yra vienas labiausiai paplitusių aplinkos veiksnių, ribojančių pasėlių produktyvumą (Basu ir kt., 2016; Liliane ir kt., 2020). Agrometeorologinės sausros atsiranda sumažėjus kritulių kiekiui, o tai nulemia dirvožemio išdžiūvimą. Tokios sausros poveikis augalams priklauso nuo jos trukmės, dirvožemio tipo ir savybių (Seleiman ir kt., 2021). Sausros stresas neigiamai paveikia fotosintezę, nes augalai mažindami vandens praradimą per transpiraciją, uždaro žioteles. Tai sulėtina fotosintezės intensyvumą, sumažėja tarpląstelinio CO₂ kiekis ir maistinių medžiagų pasisavinimas, dėl ko augaluose vyksta morfologiniai ir biocheminiai pakitimai (Khan ir kt., 2018; Gambetta ir kt., 2020; Kapoor ir kt., 2020; Ozturk ir kt., 2021). Vienas iš pagrindinių augalų atsakų ir redokso balanso sutrikimo ląstelėje sukėlėjų yra suintensyvėjusi aktyvių deguonies junginių (ADJ) gamyba. ADJ sukelia oksidacinį stresą augalų ląstelėse (Yang ir kt., 2021). Augaluose išsivysčiusi antioksidacinė sistema padeda jiems prisitaikyti ir išgyventi nepalankiomis aplinkos sąlygomis (Basu ir kt., 2016; Kapoor ir kt., 2020), o oksidacinių ir redukcinių reakcijų balansas padeda išlaikyti redokso homeostazę juose. Dėl to augaluose susiformuoja fermentinė ir nefermentinė antioksidantų sistema. Siekiant sumažinti žalingą neigiamų aplinkos veiksnių poveikį augalams, gali būti naudojamos išorinės priemonės, tokios kaip įvairių nanodalelių (ND) panaudojimas, tikslingai sustiprinant antioksidacinės sistemos veiklą (Kandhol at el., 2022; Ghani et al., 2022; Ahmad ir kt., 2022). Pastaraisiais dešimtmečiais buvo diskutuojama apie ND taikymą, siekiant pagerinti įvairių pasėlių atsparumą nepalankiems aplinkos veiksniams ir augalų mineralinės mitybos optimizavimą, tačiau iki šiol vis dar trūksta pagrįstų mokslinių žinių apie ND naudą augalų augimui ir derliui. Be to, vis dar nėra pilnai ištirtas įvairių ND poveikis aplinkai, augalų oksidacinio streso biožymenims ir antioksidacinės sistemos aktyvumui.

Nanodalelės – medžiagos, kurių dydis iki 100 nm, pasižyminčios unikaliomis formomis, paviršiaus krūviu ir dideliu paviršiaus plotu (Modena ir kt., 2019; Raval ir kt., 2019), o šios savybės nulemia jų poveikį aplinkai. Tiriant ND poveikį augalams, būtina ištirti ND suspensijų savybes, įvertinti skirtingus pritaikymo būdus, tokius kaip augalų laistymą ar purškimą (Tarafdar ir kt., 2012; Mittal ir kt., 2020). Ypač svarbu parinkti optimalią naudojamų ND koncentraciją, kuri kiekvienai augalų rūšiai gali skirtis ir sukelti priešingą efektą, nei tikimasi.

Sėjamas žirnis (*Pisum sativum* L.) – vienas populiariausių ankštinių šeimos augalų, itin jautrus drėgmės trūkumui (Nadeem ir kt., 2019). Nano- dydžio mikroelementų molibdeno (MoO_3), boro (B_2O_3), vario (CuO) ir silicio (SiO_2) poveikio sėjamųjų žirnių tyrimams buvo parinkti atsižvelgiant į jų svarbą augalų fiziologijai. MoO_3 ir B_2O_3 yra ypač svarbūs žirniams būdingoms gumbelių bakterijoms ir jų diferenciacijai į azotą fiksuojančią formą. SiO_2 yra svarbus reguliuojant oksidacinį stresą ir aktyvuojant antioksidacinę sistemą, o CuO pasižymi kaip elementas naudojamas prieš patogenus. Atsižvelgiant į jų naudą žirniams, yra būtina ieškoti naujų, efektyvesnių šių mikroelementų pritaikymo būdų, todėl šiame moksliniame darbe pirmą kartą ištirtas nano- dydžio molibdeno, boro, silicio ir vario poveikis žirnių augalams. Šis darbas pasižymi tiek moksline, tiek praktine reikšme, nes tokių inovatyvių technologijų pritaikymas agronomijoje gali padėti išsaugoti derlių esant nepalankioms aplinkos sąlygoms, apsaugoti aplinką nuo sunkiųjų metalų susidarymo dirvoje, sumažinant naudojamų trąšų normas.

Hipotezė. Tikėtina, kad nanodalelėmis stiprinama fermentinė ir nefermentinė antioksidacinė sistema neutralizuoja žalingo oksidacinio streso poveikį sausros veikiamuose žirnių augaluose.

Tyrimų tikslas. Parinkti tinkamiausią SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelių koncentraciją ir taikymo būdą sėjamesiems žirniams (*Pisum sativum* L.) sustiprinant jų atsparumą sausrui ir kompleksiniam sunkiojo metalo vario ir sausros poveikiui.

Tyrimų uždaviniai:

1. Nustatyti efektyviausias SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelių koncentracijas žirnių morfologiniams parametrams, poveikį oksidaciniam stresui ir antioksidacinei sistemai.
2. Ištirti agrometeorologinės sausros ir SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelių poveikį žirnių oksidaciniam stresui ir antioksidacinei sistemai.
3. Įvertinti kompleksinį aplinkos veiksnių (agrometeorologinės sausros, sunkiojo metalo Cu ir SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelių) poveikį žirnių oksidaciniam stresui ir antioksidacinei sistemai.
4. Įvertinti ND skirtingų panaudojimo būdų efektyvumą, stiprinant žirnių antioksidacinės sistemos veiklą ir atsparumą nepalankiems aplinkos veiksniams.

Disertacijos ginamieji teiginiai:

1. Pritaikius optimalias SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelių suspensijų koncentracijas sėjamesiems žirniams, augantiems normalaus drėgno substrate, galima suaktyvinti

antioksidacinės ir oksidacinės sistemos aktyvumą, kuri teigiamai veikia morfologinius parametrus ir padidina augalų produktyvumą.

2. Sėjamųjų žirnių atsparumą agrometeorologinei sausrai padidina SiO_2 , MoO_3 , B_2O_3 , CuO nanodalelės. Nanodalelių suspensijos praturtina augalų mineralinę mitybą bei suaktyvina fermentinius ir nefermentinius antioksidantus, kurie sumažina oksidacinių biožymenų kiekį bei turi teigiamą poveikį sėjamųjų žirnių morfologiniams parametrams ir derliui.
3. SiO_2 , MoO_3 , B_2O_3 nanodalelių suspensijos padidina sėjamųjų žirnių atsparumą kompleksiniam agrometeorologinės sausras ir sunkiojo metalo vario poveikiui stiprinant antioksidacinę sistemą silpninant oksidacinio streso poveikį bei didinant augalų tolerancijos indeksą variui. CuO nanodalelių poveikis veikia sinergiškai su pertekliniu vario kiekiu padidindamas oksidacinių biožymenų kiekį ir sumažindamas produktyvumą sėjamuosiuose žirniuose.
4. Nanodalelių SiO_2 , MoO_3 , B_2O_3 , CuO suspensijų poveikis antioksidacinei sistemai, morfologiniams parametrams ir mikro- ir makro- elementų akumuliacijai ir derliui sėjamuosiuose žirniuose priklauso nuo jų stabilumo, dydžio, paviršiaus krūvio ir poveikio būdo.

Mokslinio darbo naujumas. Tyrimas suteikia naujų žinių apie SiO_2 , CuO , MoO_3 , B_2O_3 nanodalelių poveikį žirnių augalų oksidacinio streso biožymenims, antioksidacinei sistemai, makro- ir mikroelementų pokyčiams, kai augalai augo normalaus drėgno substrate, drėgno trūkumo sąlygomis bei buvo veikiami kompleksinio sausras ir sunkiojo metalo vario streso. Nustatyta, kad nanodalelių suspensijos gali turėti tiek teigiamą, tiek neigiamą poveikį, kuris priklauso nuo taikomų dalelių savybių, koncentracijos ir pritaikymo metodo. Ištirtas SiO_2 , CuO , MoO_3 , B_2O_3 nanodalelių dzeta potencialas parodė, kad visos suspensijos buvo stabilios ir anijoninės, o nano-dydžio dalelių buvo nustatyta visose vandeninėse suspensijose. Atrinktos optimaliausios teigiamą poveikį žirnių oksidacinio streso biožymenims, antioksidacinei sistemai, makro- ir mikroelementų pokyčiams ir produktyvumui turėjusios nanodalelių koncentracijos: 50 ppm – SiO_2 , CuO , MoO_3 , 12,5 ppm – B_2O_3 . Taip pat nustatyta, kad purškimas su CuO ir B_2O_3 nanodalelėmis žirnių produktyvumą veikia efektyviau nei laistymas, o augalų laistymas su MoO_3 nanodalelių suspensija buvo efektyvesnis nei purškimas.

Praktinė darbo reikšmė. Metalų pagrindu pagamintų ND poveikio žirniams tyrimai praturtina žinias apie ND naudą ar riziką augalams, padeda įtraukti platesnę ND įvairovę ir plėsti nanotechnologijų naudojimą augalininkystėje, tuo pačiu prisidedant prie bendros žemės ūkio tobulinimo praktikos. Visa tai turi didelę vertę agronomijos mokslo raidai. Be to, disertacinis

darbas turi ir socialinę naudą, nes tokių tyrimų plėtra padėtų patenkinti augalinio maisto poreikius visuomenėje ir prisidėtų prie augalinės kilmės produktų kokybės gerinimo.

Disertacijos apimtis ir struktūra. Disertaciją sudaro įvadas, literatūros analizė, tyrimo metodų aprašymas, rezultatų analizė, aptarimas, išvados, panaudotos literatūros sąrašas ir publikacijų, išleistų kartu su bendraautoriais, sąrašas ir priedai. Literatūros analizė, tyrimo metodikos aprašymas ir rezultatai iliustruojami 15 paveikslų ir 21 lentele. Bibliografijoje yra 295 šaltiniai, disertacijos apimtis 155 puslapiai.

2. TYRIMO OBJEKTAS, SĄLYGOS IR METODAI

2.1 Tyrimo objektas

Sėjamasis žirnis (*Pisum sativum* L.)

2.2 Nanodalelių suspensijos paruošimas. Eksperimentams buvo naudojamos silicio dioksido ((SiO₂) dalelių dydis: 20–30 nm; grynumas: 99%), vario oksido ((CuO) dalelių dydis: 25–55 nm; grynumas: 99,95%), molibdeno trioksido ((MoO₃) dalelių dydis: 35–45 nm; grynumas: 95%) ir boro ((B₂O₃) dalelių dydis: 100 nm; grynumas: 99,9%) nanodalelės (ND) įsigytos US Research Nanomaterials (Inc, Hiustonas, TX USA). ND suspenduotos dejonizuotame vandenyje (12,5; 25; 50; 100 ppm) ir ultragarsu (37 kHz) disperguotos 60 min. ND dydis ir suspensijos stabilumas išmatuoti Delsa™ nano submikrono dalelių dydžio (Beckman Coulter Instruments, Corporation, Fullerton, Kalifornija) ir dzeta potencialo (Dispersion Technology Inc., Bedford Hills, Niujorkas) matuokliais. Gauti duomenys parodė, kad visos paruoštos ND suspensijos buvo stipriai anijoninės ir stabilios, taip pat monodispersinės pagal polidispersiškumo indeksą (PDI) (1 lentelė).

1 lentelė. SiO₂, B₂O₃, CuO, MoO₃ ND suspensijų dejonizuotame vandenyje savybės: dzeta potencialas (vidurkis ± standartinę paklaidą), polidispersiškumo indeksą, nanodalelių dydžio nuo 1 iki 100 nm procentą suspensijoje.

	SiO ₂ ND	B ₂ O ₃ ND	CuO ND	MoO ₃ ND
Dzeta potencialas (ζ; mV)	-20,64 ± 0,333	-28,54±0,223	-26,68±0,631	-24,92±0,314
Polidispersinis indeksas (PI)	0,34	0,237	0,245	0,218
ND dydis 1–100 nm suspensijoje, %	70%	43%	54%	68%

2.3 Eksperimentų schema

Vizualiai apibendrinta metodika pateikta 1 paveiksle. Prieš sėją žirnių sėklos buvo sterilizuojamos 5% natrio hipochlorito tirpale 15 min., kad būtų užtikrintas paviršiaus sterilumas (Lehotai ir kt., 2011) ir kelis kartus skalaujamos dejonizuotu vandeniu. Tada sėklos 24 valandas mirkytos vandenyje.

Atlikti šie eksperimentai:

Pradinis (pirminis) eksperimentas atliktas siekiant nustatyti tinkamas SiO_2 , B_2O_3 , CuO , MoO_3 ND koncentracijas ir jų poveikį žirnių augalams. Tyrimas atliktas Lietuvos agrarinių ir miškų mokslo centro Sodininkystės ir daržininkystės institute esančioje kontroliuojamos aplinkos augimo kameroje (4×6 m, $h = 3,2$ m). Po 3 sėklas buvo sėjama į 500 ml plastikinį vazoną, kuris buvo traktuojamas kaip vienas biologinis pakartojimas. Vienam variantui buvo naudojama 16 vazonų. Augalai pasėti ir augo durpių substrate Profi 1 (pH 6, EL 2,0 - 2,5 mS cm^{-1} ($\pm 0,03$ mS cm^{-1}), vidutiniai maistinių medžiagų kiekiai (mg L^{-1}) substrate: N, 110; P_2O_5 , 50; K_2O , 160, Fe (4,5), Mn (0,5), Cu (0,1), B (0,02), Mo (0,03), ir Zn (0,04), UAB Durpeta, Lietuva) 30 dienų. Žirniai augo esant 16 h fotoperiodui, $20 \pm 2/16 \pm 2$ °C dienos/nakties aplinkos oro temperatūrai, 60 % santykinei oro drėgmei, ~ 220 $\mu\text{mol m}^{-2} \text{s}^{-1}$ fotosintetiškai aktyvios spinduliuotės fotonų srauto tankiui (PAR) naudojant aukšto slėgio natrio (HPS) lempas (SON-T Agro, 400 W; Philips, Somerset, NJ, JAV). Žirniai laistyti 100 ml SiO_2 , B_2O_3 , CuO , MoO_3 ND suspensijomis, su skirtingomis koncentracijomis (12,5; 25; 50; 100 ppm), kai augalai turėjo 3–4 tikruosius lapus arba suformavo 3–4 ūselius (14 dienų po sėjos, 14 BBCH augimo tarpsniu) (Meier, 2018). Po to, visi augalai kasdien buvo laistomi dejonizuotu vandeniu jų papildomai netrešiant. Siekiant sumažinti galimą augimo kameros poveikį augalų atsakui, buvo taikomas reguliarus 10 vazonų erdvinis pertvarkymas kiekviename variante. Siekiant išvengti krašto efekto, likę 6 vazonai buvo palikti kraštuose ir nejudinami. Eksperimento pabaigoje, augalams pasiekus 31 BBCH augimo stadiją – stiebo ilgėjimo pradžią (Meier, 2018), žirniai nuskinti biometriniais matavimams ir biocheminėms analizėms. Išsamiai ištyrus ir išanalizavus paveiktus augalus, buvo parinktos 12,5; 25; 50 ppm kiekvienos ND koncentracijos, skirtos naudoti tolesniuose eksperimentuose. Šiais eksperimentais buvo siekiama parinkti efektyviausias skirtingų ND koncentracijas drėgmės trūkumo veikiamiems žirnių augalams.

Dvejuose šiltnamiuose (3×6 m, $h = 2$ m; $55^\circ 05' 08.4''\text{N}$ $23^\circ 48' 03.5''\text{E}$, 51 m aukštyje; vidutinio klimato zona šiaurės pusrutulio) dvejų metų pavasario-vasaros laikotarpiu (2019–2020 m.) buvo atlikti žemiau aprašomi eksperimentai.

Eksperimentams žirnių sėklos buvo paruoštos sėjai, kaip aprašyta aukščiau. Dešimt sėklų buvo pasėta į 10 l tūrio plastikinius vazonus (7 vazonai vienam variantui, išdėstyti atsitiktine

tvarka), užpildyti ~ 8 kg dirvožemio mišinio (atitinkamai 7:1 dirvožemio ir perlito tūrio santykis). Dirvožemis buvo sunkus priemolis, pH $7,4 \pm 0,1$; humuso koncentracija – $3,6 \pm 0,1$ %; P_2O_5 – 243 ± 8 mg kg^{-1} ; K_2O – 348 ± 37 mg kg^{-1} ; NH_4^+ – $4 \pm 0,6$ mg kg^{-1} ; NO_3^- – $22 \pm 0,9$ mg kg^{-1} ; SiO_2 – $39 \pm 0,8$ mg kg^{-1} . Praėjus 5 dienoms po sėjos žirnių daigai išretinti iki 7 augalų vazone. Po 16 dienų auginimo kiekvienas žirnių vazonas patręštas 7 g amonio salietra. Žirniai nupurkšti fungicidais, nes veislė ‘Respect’ yra jautresnė miltligei. Vazonai buvo laistomi vandeniu kasdien iki 80% substrato drėgmės (SD) su graduotu cilindru bei matuojama SD su substrato drėgmės jutikliu (Delta-T prietaisai, HH2 drėgmės matuoklis, Kembridžas, Jungtinė Karalystė) 35 dienas. Augalai augo natūralaus paros ilgio fotoperiodu, vidutinė dienos/nakties temperatūra buvo $22,2/14,4$ °C; santykinė oro drėgmė – $58/77 \pm 5\%$ prieš sausrą; 10 dienų agrometeorologinės sausras metu vidutinė dienos/nakties temperatūra buvo $25,4/16,6$ °C, o santykinė oro drėgmė – $53/75 \pm 5\%$, duomenys buvo matuojami viso eksperimento metu (Termio+ duomenų kaupiklis, Lenkija).

Žirniams pasiekus 40 BBCH augimo tarpsnį (Meier, 2018), jie buvo palaistyti su 100 ± 1 ml vienam vegetaciniam indui arba purškiami augalai iki visiško sušlapimo (apie $14 \pm 0,5$ ml augalui) tirpalais, kuriuose buvo skirtingos SiO_2 , B_2O_3 , CuO , MoO_3 ND koncentracijos: 0 (laistyti arba apipurkšti vandeniu, neapdoroti su ND), 12,5 ppm, 25 ppm ir 50 ppm. Po ND panaudojimo, vienos dalies žirnių augalų laistymas buvo sustabdytas ir inicijuotas sausras stresas (30% SD), o kita dalis – laistomi vandeniu, palaikant normalią substrato drėgmę (80% SD). Šie poveikio režimai buvo taikomi 10 dienų. Po to augalai atsitiktinai buvo atrenkami įvertinti jų morfologinius parametrus ir biocheminį atsaką.

Vėlesniuose eksperimentuose siekėme nustatyti, ar ND suspensijos gali apsaugoti augalus nuo kompleksinio agrometeorologinės sausras ir vario, kaip sunkiojo metalo, poveikio.

Šio eksperimento sąlygos: vidutinė dienos/nakties temperatūra $24,2/14,4$ °C; santykinė oro drėgmė – $54/75 \pm 5\%$ prieš ekspoziciją; 10 dienų sausras metu vidutinė dienos/nakties temperatūra buvo $26,2/17,0$ °C, o santykinė oro drėgmė – $50/73 \pm 5\%$. Dirvožemio paruošimas, žirnių sėjimas ir apdorojimas ND buvo toks pats, kaip aprašyta aukščiau. Norint sukelti Cu kaip sunkiojo metalo įtampą žirniuose, į pusės vegetacinių indų substratą buvo pridėta 160 mg kg^{-1} $CuSO_4$ vandeninio tirpalo. Kai žirniai pasiekė 40 BBCH augimo stadiją (Meier, 2018), jie buvo laistomi (100 ± 1 ml viename vazone) arba purškiami iki visiško sudrėkimo (apie $14 \pm 0,5$ ml augalo⁻¹) ND tirpalais, naudojant ankstesnių eksperimentų nustatytas veiksmingiausias koncentracijas: SiO_2 – 50 ppm, CuO – 50 ppm, MoO_3 – 50 ppm, B_2O_3 – 12,5 ppm. Žirnius paveikus ND, vienos dalies augalų laistymas buvo nutrauktas ir pradėtas sausras stresas (30 % SM), o kiti – buvo laistomi vandeniu, kad būtų palaikoma normali dirvožemio drėgmė (80 % SM). Šie režimai buvo taikomi 10 dienų iki derliaus nuėmimo. Augalai buvo atsitiktinai atrenkami pasiekus BBCH 50 augimo stadiją (Meier, 2018) siekiant įvertinti jų morfofiziologinį atsaką.

2.4 Morfologinių parametru matavimai

Biometriniams matavimams atsitiktinai atrinkta 10 augalų iš varianto ($n = 10$). Antžeminė žirnio dalis atskirta nuo šaknų, išmatuotas stiebo aukštis ir šaknies ilgis, pasverta augalų žalia biomasė (ŽM) ir sausas svoris (SM) po džiovavimo. ŽM ir SM pasverti elektroninėmis svarstyklėmis (Mettler Toledo AG64, Columbus, OH, JAV). SM nustatytas augalus džiovinant 105°C temperatūroje konvekciniame džiovavimo spintoje (VENTICELL 222, MBT, Čekija). Pasvėrus antžeminės žirnio dalies ŽM, po 10 augalų iš kiekvieno varianto buvo paliekami 24 valandoms plūduriuoti dejonizuotame vandenyje, po to augalai pasverti nustatyti turgorinei masei (TM), kad būtų galima išskaičiuoti santykinį vandens kiekį (SVK) (Baris ir Weatherley, 1962):

$$\text{SVK, \%} = \frac{(\text{ŽM} - \text{SM})}{(\text{TM} - \text{SM})} \times 100. \quad (1)$$

Lapų plotas išmatuotas automatiniu lapų ploto matuokliu (AT Delta-T Devices, Wallingford, JK). Apskaičiuotas specifinis lapų plotas (SLP), kur bendras augalo lapų plotas ($n=10$) padalintas iš antžeminės dalies SM ir išreikštas $\text{cm}^2 \text{g}^{-1}$. Šaknų ir antžeminės augalų dalies santykis nustatytas kaip šaknies SM ir antžeminės SM santykis. Žirnių derlius apskaičiuotas pagal Koiter ir Bill Ashton (2021) aprašytą metodiką. Bio-koncentracijos faktorius, translokacijos faktorius ir tolerancijos indeksas apskaičiuoti pagal Zacchini su bendraautoriais pateiktą metodiką (2009).

2.5 Antioksidacinis aktyvumas

Nefermentiniai antioksidantai

Žirnių lapų antioksidacinės savybės įvertintos spektrofotometriškai (spektrofotometru M501, Spectronic Camspec Ltd., UK) kaip DPPH (2-difenil-1-pikrilhidrazilas), ABTS (2,20-azino-bis (3-etilbenzotiazolin-6-sulfonrūgštis)) diamonio druskos) radikalų surišimo geba (Sharma ir Bhat, 2009; Re ir kt., 1999) ir FRAP redukcijos antioksidacinė geba (Benzie ir Strain, 1996). Bendras fenolinių junginių kiekis, išreikštas galo rūgšties ekvivalentu, nustatytas Folin–Ciocalteu metodu (Ainsworth ir Gillespie, 2007) absorbciją matuojant ties 765 nm.

Fermentiniai antioksidantai

Ekstraktai buvo ruošiami iš 0,5 g šviežio mėginio, jį homogenizuojant su skystu azotu ir praskiedžiant 5 mL ekstrahavimo buferiu (100 mM kalio fosfato buferis, pH 7,8, kuriame yra 0,1 mM EDTA). Po 10 minučių centrifugavimo 3000 aps./min. (Hermle Z300K, Vokietija), supernatantas buvo surinktas ir naudojamas fermentinio aktyvumo tyrimams. Visi ekstrakto paruošimo etapai buvo atlikti 4 °C temperatūroje. Šie ekstraktai buvo naudojami

spektrofotometriniais (M501, Spectronic Camspec Ltd., UK) superoksido dismutazės (SOD), katalazės (CAT), askorbato peroksidazės (APX), glutationo reduktazės (GR) ir gvajakolio peroksidazės (GPX) aktyvumui žirnių lapuose nustatyti.

SOD aktyvumas nustatytas pagal formazano mėlio susidarymo slopinimą esant 560 nm bangos ilgiui (Dhindsa et al., 1981) ir išreiškiamas UI mg^{-1} baltymų min^{-1} . APX aktyvumas išmatuotas prie 290 nm bangos ilgio (Nakano and Asada, 1981), naudotas ekstinkcijos koeficientas 2,8 mM/cm, išreikštas $\mu\text{mol AsA mg}^{-1}$ baltymų min^{-1} ; GR aktyvumas – pagal oksiduoto glutationo (GSSG) kiekį, kurio koncentracija nustatoma prie 340 nm bangos ilgio (Sofa et al., 2005); naudotas absorbcijos koeficientas $6.22 \text{ mM}^{-1} \text{ cm}^{-1}$, išreikštas $\mu\text{mol NADPH mg}^{-1}$ baltymų min^{-1} . CAT aktyvumas nustatytas naudojant H_2O_2 kaip substratą prie 240 nm ilgio bangos (Aebi, 1984); naudotas absorbcijos koeficientas $6.22 \text{ mM}^{-1} \text{ cm}^{-1}$, išreikštas $\text{mmol H}_2\text{O}_2 \text{ mg}^{-1}$ baltymų min^{-1} ; GPX aktyvumas nustatytas pagal gvajakolio oksidacijos sukeliama absorbcijos esant 470 nm bangos ilgiui padidėjimą (Kvaratskhelia et al., 1997).

Suminis baltymų kiekis nustatytas Bradfordo metodu prie 595 nm bangos ilgio (Bradford, 1976). Pagal kalibracinę kreivę, kurioje jaučio serumo albumino tirpalas naudotas kaip standartas, išskaičiuotas baltymų kiekis mg mL^{-1} .

2.6 Oksidacinio streso biožymenys

Lipidų peroksidacijai ir vandenilio peroksido (H_2O_2) koncentracijai žirnių lapuose nustatyti naudojami ekstraktai paruošti iš 0,1 g šviežio mėginio sumalant skystu azotu ir atskiedus 4 ml 0,1 % trichloracto rūgšties (TCA). Po 10 minučių centrifugavimo 3000 aps./min. (Hermle Z300K, Vokietija), supernatantas naudotas tolimesnėms analizėms. H_2O_2 matavimams naudotas 1 M kalio jodidas (KI), absorbcija matuota su spektrofotometru (M501, Spectronic Camspec Ltd., JK) esant 390 nm bangai. H_2O_2 kiekis buvo išskaičiuotas iš kalibracinės kreivės ir išreiškiamas $\mu\text{mol g}^{-1}$ ŽM (Velikova ir kt., 2000). Tiobarbitūrinės rūgšties (TBARS) testo pagalba (Heath ir Packer, 1968) nustatytas malondialdehido (MDA) kiekis žirnių lapų mėginiuose, kaip galutinis lipidų peroksidacijos produktas. Absorbcija buvo matuojama esant 532 nm ir 600 nm bangoms, naudojant spektrofotometrą (M501, Spectronic Camspec Ltd., JK). MDA-TBA komplekso (raudonojo pigmento) kiekis lapuose buvo apskaičiuotas ir išreikštas nmol g^{-1} ŽM:

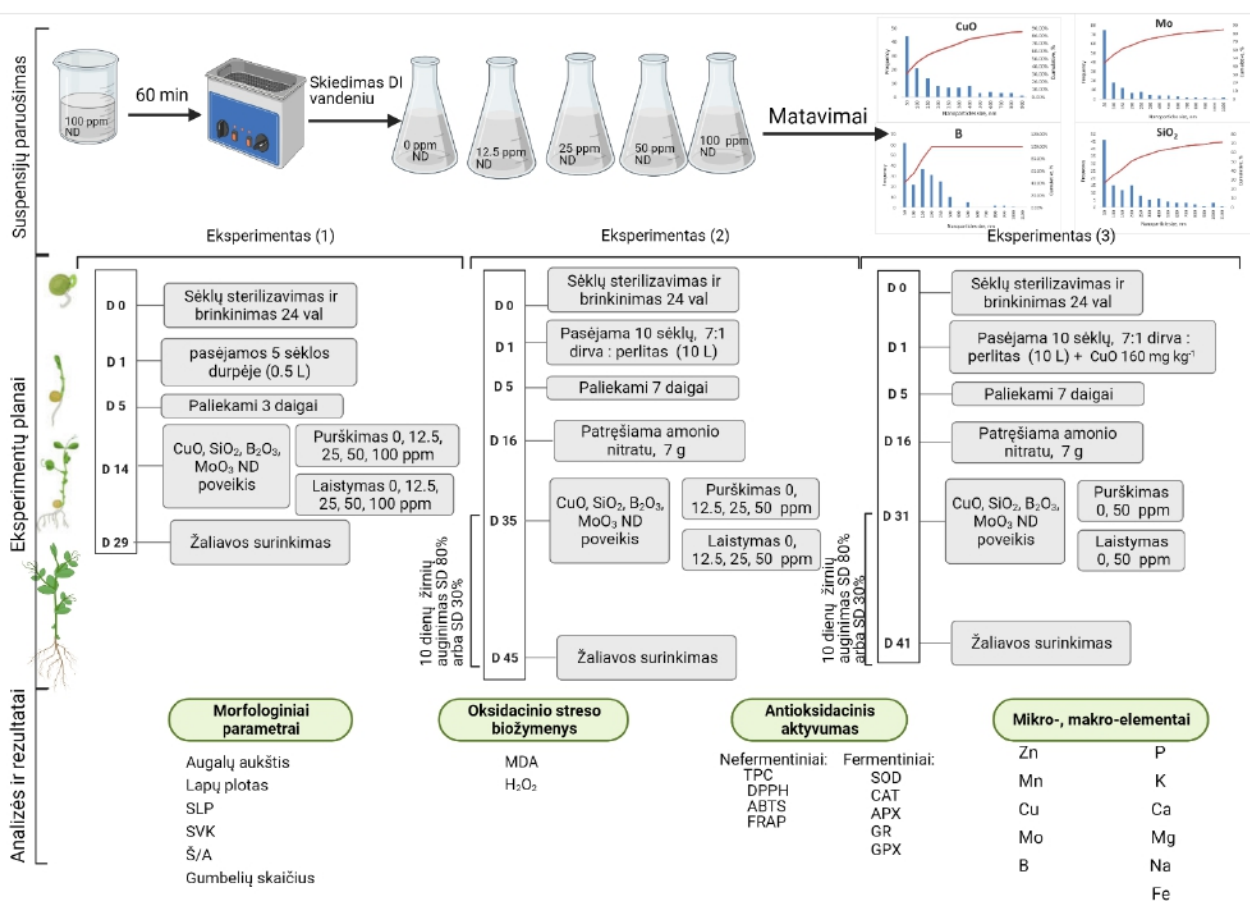
2.7 Makro- ir mikroelementų nustatymas

Makro- ir mikroelementų kiekis žirnių lapuose, stiebe ir šaknyse nustatytas naudojant mikrobangų skaidymo techniką kartu su indukciniu būdu susietos plazmos optinės emisijos spektrometrija. Visiškas sausos augalinės medžiagos (0,3 g) skaidymas gautas naudojant 8 ml

65 % HNO₃ su mikrobangų skaidymo sistema Multiwave GO (Anton Paar GmbH, Gracas, Austrija). Po to mėginiai nufiltruoti praskiesti iki 50 ml ir analizuoti su ICP–OES spektrometru (Spectro Genesis, SPECTRO Analytical Instruments, Kleve, Germany).

2.8 Statistinė analizė

Visos vertės buvo išreikštos kaip vidurkis ± standartinis nuokrypis. Siekiant nustatyti reikšmingus skirtumus duomenys buvo analizuojami naudojant dispersinės analizės (ANOVA) testą Tukey HSD esant $p \leq 0,05$. Analizuojami skirtumai tarp variantų, kai žirniai buvo auginami normaliomis sąlygomis ir atskirai tarp žirnių, kurie buvo auginami sausros sąlygomis. Mikro- ir makroelementų analizės rezultatai taip pat buvo lyginami atskirai tarp kontrolinių – sausros metu augintų žirnių, neapdorotų su ND (SD 30%) ir apdorotų su ND, ir tarp kontrolinių žirnių, augintų normaliomis sąlygomis (SD 80%) neapdorotų su ND ir paveiktų su ND.



1 pav. Tyrimų schema: nanodalelių suspensijų ruošimas, eksperimentų planai, matavimai ir analizės

3. REZULTATAI

Šiame skyriuje analizuojama SiO₂, MoO₃, B₂O₃ ir CuO nanodalelių (ND) poveikis žirnių daigams augusiems optimaliomis sąlygomis (3.1), pilnai išsivysčiusiems žirnių augalams augusiems agrometeorologinės sausras (3.2) bei kompleksinio vario ir agrometeorologinės sausras (3.3) sąlygomis.

3.1 Silicio dioksido, vario oksido, molibdeno trioksido ir boro trioksido nanodalelių poveikis žirnių morfologiniams parametrams ir antioksidacinei sistemai

Siekiant išsiaiškinti, kaip žirnių daigai reaguotų į skirtingos koncentracijos nanodalelių (ND) poveikį juos auginant optimaliomis sąlygomis, buvo atliktas pirmasis eksperimentas. Šiame skyriuje analizuojamas silicio dioksido (SiO₂), vario oksido (CuO), molibdeno trioksido (MoO₃) ir boro trioksido (B₂O₃) ND poveikis žirnių daigų morfologiniams parametrams ir antioksidaciniam aktyvumui. Pateikti rezultatai yra lyginami su augalais, kurie nebuvo paveikti ND.

Nanodalelių poveikis žirnių morfologiniams parametrams

Rezultatai parodė, kad naudojant SiO₂ ND, žirnių aukštis padidėjo 28–45 %, priklausomai nuo didėjančios koncentracijos (2 lentelė). Žirnius palaisčius su SiO₂ ND 12,5, 25, 50 ir 100 ppm suspensijomis, žirnių vidutinė ŽM padidėjo atitinkamai 81, 104, 116 ir 109 %. Didėjanti SiO₂ ND koncentracija tirpale, padidino žirnių SM iki 135%. Augalų palaistymas su 100 ppm koncentracijos suspensija, šaknis pailgino apie 76 %, o su kitomis koncentracijomis - iki 106%. Didėjanti SiO₂ ND koncentracija teigiamai paveikė šaknų ŽM ir SM. Rezultatai patvirtino teigiamą SiO₂ ND poveikį žirnių aukščiui, ŽM ir SM bei šaknų ilgiui esant visoms tirpalo koncentracijoms (1 lentelė).

MoO₃ ND turėjo teigiamą poveikį žirniams, didėjant jų koncentracijai tirpale (2 lentelė). Didesnė MoO₃ ND koncentracija tirpale padidino žirnių aukštį iki 48%, ŽM – iki 61%, o SM – iki 84 %. Šaknų ilgis padidėjo 56–85 %, o jų biomasė padidėjo tiek šviežia 58–121 %, tiek sausa 30–71 % esant 12,5–100 ppm koncentracijoms tirpale.

Nustatyta, kad mažiausia CuO ND koncentracija tirpale turėjo didžiausią teigiamą poveikį žirnio morfologiniams parametrams (2 lentelė). Naudojant 12,5, 25 ir 50 ppm koncentracijas, žirnių aukštis padidėjo 41, 32 ir 16 %, ŽM - 87, 78 ir 41 %, o SM – 47, 50 ir 35 % lyginant su ND nepaveiktais augalais. Šaknų ilgiui reikšmingą poveikį turėjo visos naudotos CuO ND koncentracijos. Didėjančios CuO ND koncentracijos tirpale turėjo teigiamą poveikį šaknų ŽM tačiau didėjant koncentracijai biomasės didėjimas sulėtėjo.

Mažesnės B₂O₃ ND koncentracijos turėjo teigiamą poveikį žirnių morfologiniams parametrams (2 lentelė). Žirnių aukštis padidėjo 52, 21 ir 2 %, ŽM – 95, 60 ir 14 %, o SM –

116, 35 ir 52 %, kai jie buvo palaistyti su 12,5, 25 ir 50 ppm B₂O₃ ND suspensijomis. Šaknų ilgis padidėjo 92, 62 ir 11 %, šaknų ŽM – 91, 64 ir 21 % panaudojus aukščiau nurodytas B₂O₃ ND koncentracijas.

2 lentelė. SiO₂, CuO, MoO₃, B₂O₃ ND (12,5; 25; 50; 100 ppm) poveikis *P. sativum* L. morfologiniams parametrams: augalų antžeminės dalies aukštis, šviežia ir sausa biomase (ŽM, SM), šaknų ilgis, ŽM ir SM. 0 – kontroliniai augalai, laistomi dejonizuotu vandeniu. Vidutinės reikšmės stulpeliuose su skirtingomis raidėmis reikšmingai skiriasi nuo kontrolinės vertės, kai $p < 0,05$ ($n=10$) pagal Tukey (HSD) testą

Nanodalelių koncentracija, ppm	Antžeminė dalis				Šaknys		
	Aukštis, cm	ŽM, g	SM, g	Ilgis, cm	ŽM, g	SM, g	
H ₂ O	0	20,0 b	1,3 b	0,240 b	4,5 b	0,365 b	0,044 b
SiO ₂	12,5	25,7 a	2,4 a	0,388 a	9,1 a	0,624 a	0,073 a
	25	28,2 a	2,7 a	0,465 a	9,2 a	0,782 a	0,082 a
	50	28,7 a	2,9 a	0,501 a	9,2 a	1,049 a	0,089 a
	100	28,8 a	2,8 a	0,564 a	7,8 a	1,166 a	0,105 a
CuO	12,5	28,1 a	2,5 a	0,353 a	9,1 a	0,763 a	0,069 a
	25	26,3 a	2,4 a	0,361 a	8,3 a	0,668 a	0,058 b
	50	23,3 a	1,9 a	0,323 a	8,4 a	0,501 a	0,055 b
	100	20,3 b	1,2 b	0,217 c	9,3 a	0,469 a	0,037 c
MoO ₃	12,5	24,6 a	1,7 b	0,259 b	6,9 a	0,627 a	0,063 b
	25	24,7 a	1,5 b	0,304 a	7,4 a	0,577 a	0,058 b
	50	27,7 a	2,2 a	0,380 a	7,3 a	0,667 a	0,065 a
	100	29,6 a	2,5 a	0,388 a	8,3 a	0,809 a	0,076 a
B ₂ O ₃	12,5	30,3 a	2,6 a	0,518 a	8,6 a	0,696 a	0,061 a
	25	24,3 a	2,1 a	0,323 a	7,3 a	0,598 a	0,053 b
	50	20,4 b	1,5 b	0,366 a	4,9 b	0,444 a	0,044 b
	100	19,3 c	1,4 b	0,270 b	5,5 a	0,341 b	0,043 b

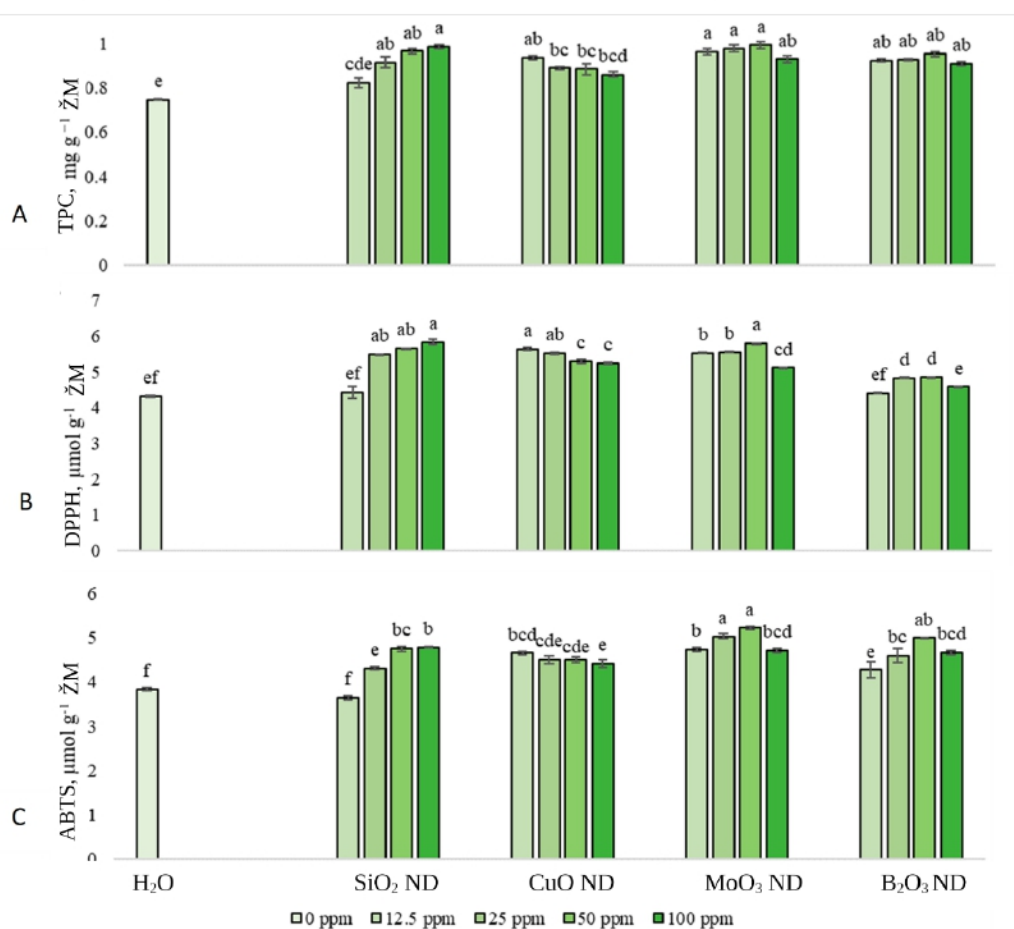
ŽM – šviežia biomase; SM – sausa biomase

Nanodalelių poveikis žirnių antioksidaciniam aktyvumui

Visos naudotos ND suspensijos paveikė antioksidacinį atsaką žirnių lapuose (2 pav.). Didėjant SiO₂ ND koncentracijai, žirnių lapuose bendras fenolinių junginių (TPC) kiekis padidėjo 10 (12,5 ppm), 22 (25 ppm), 30 (50 ppm) ir 32 % (100 ppm) (2A pav.). Panašus poveikis buvo pastebėtas 2-difenil-1-pikrilhidrazilo (DPPH) ir 2,20-azino-bis (3-etilbenzotiazolin-6-sulfonrūgšties) (ABTS) laisvųjų radikalų surišimo aktyvume (2B, C pav.).

DPPH laisvųjų radikalų surišimo aktyvumas padidėjo iki 35 % naudojant SiO₂ ND suspensiją. ABTS laisvųjų radikalų surišimo aktyvumas padidėjo apie 12 %, 24 % ir 25 %, kai žirniai buvo paveikti 25 ppm 50 ppm ir 100 ppm SiO₂ ND suspensijomis.

Nustatyta, kad MoO₃ ND didino TPC kiekį apie 29, 31, 33 ir 24 % po poveikio 12,5, 25, 50 ir 100 ppm MoO₃ ND suspensijomis. (2A pav.). DPPH laisvųjų radikalų pašalinimo aktyvumas intensyviai didėjo esant 12,5 ppm, tačiau aktyvumas padidėjo ne taip stipriai, kai augalai buvo palaistyti su 100 ppm suspensija (2B pav.). ABTS laisvųjų radikalų surišimo aktyvumui taip pat įtakos turėjo MoO₃ ND koncentracijos didėjimas tirpale (2C pav.). Apskritai buvo pastebėta, kad 100 ppm MoO₃ ND suspensija silpniau paveikė TPC kaupimąsi ir antioksidacinį pajėgumą žirnių lapuose nei mažesnės koncentracijos.

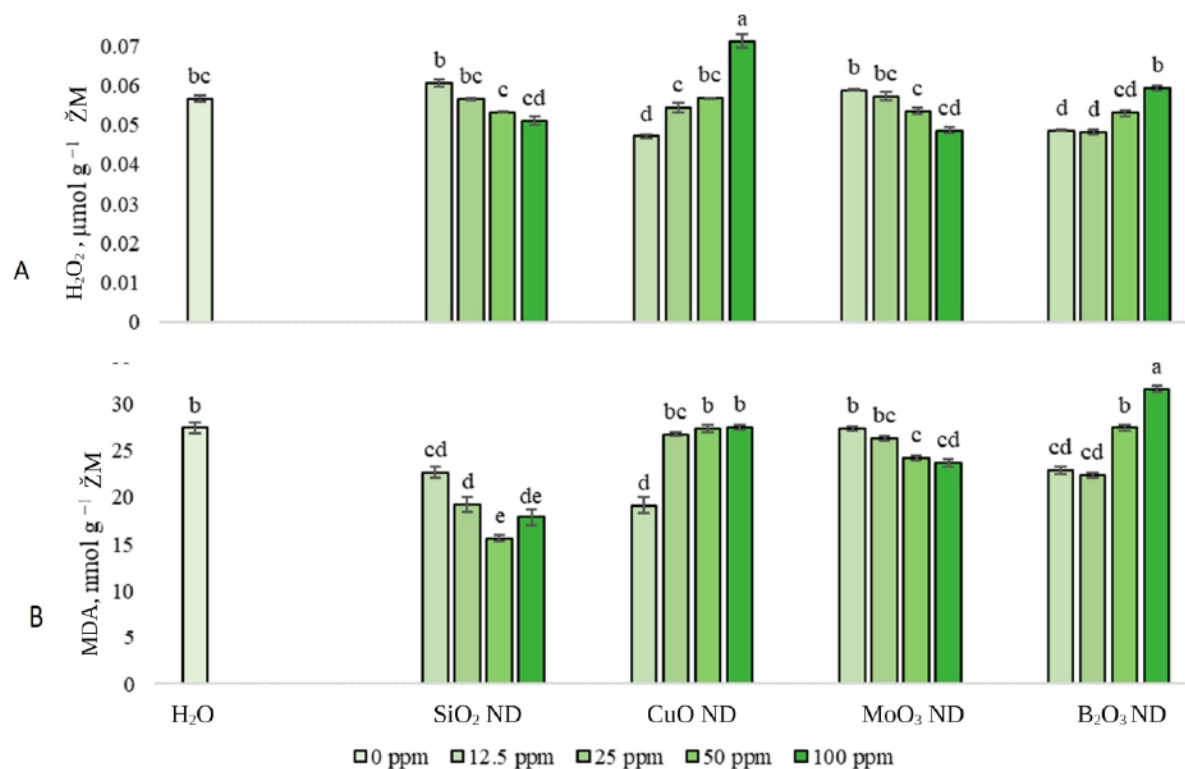


2 pav. SiO₂, CuO, MoO₃ ir B₂O₃ ND (0; 12,5; 25; 50, 100 ppm) poveikis bendram fenolinių junginių kiekiui (A, TPC), DPPH (B) laisvųjų radikalų surišimo aktyvumui, ir ABTS (C) laisvųjų radikalų surišimo aktyvumui sėjamaajame žirnyje (*P. sativum L.*) H₂O – kontroliniai augalai, laistyti dejonizuotu vandeniu. Vertės yra vidurkiai ± SE apskaičiuoti iš devynių pakartojimų. Skirtingos raidės rodo reikšmingus skirtumus pagal Tukey HSD testą ($p < 0,05$)

TPC kiekis žirnių lapuose padidėjo 25, 19, 18 ir 15 % po poveikio su 12,5, 25, 50 ir 100 ppm CuO ND suspensijomis (2A pav.). Taip pat naudotos skirtingų koncentracijų CuO ND suspensijos teigiamai paveikė DPPH ir ABTS laisvųjų radikalų surišimo gebą. (2B, C pav.). Rezultatai parodė, kad didėjant CuO ND koncentracijai suspensijoje, susilpnėjo antioksidantų aktyvumas žirniuose.

12,5, 25, 50 ir 100 ppm B₂O₃ ND koncentracijos stipriai stimuliuo TPC kaupimąsi žirniuose, padidinant jį atitinkamai apie 23, 24, 28 ir 22 % (2A pav.). DPPH laisvųjų radikalų surišimo aktyvumas buvo silpnai paveiktas (2B pav.), o ABTS - B₂O₃ ND suspensijos padidino apie 12, 20, 30 ir 22 % (2C pav.) augalus paveikus 12,5, 25, 50 ir 100 ppm tirpalais.

SiO₂ ND efektyviai sumažino vandenilio peroksido (H₂O₂) ir malondialdehido (MDA) koncentracijas žirnių lapuose (3A, B pav.). H₂O₂ koncentracija žirnių lapuose sumažėjo iki 10 % po poveikio 25, 50 ir 100 ppm SiO₂ tirpalais. Nustatytas nežymus H₂O₂ kiekio padidėjimas po augalų palaistymo su 12,5 ppm SiO₂ ND suspensija. MDA kiekis žirnių lapuose žymiai sumažėjo, palaisčius augalus skirtingų koncentracijų SiO₂ ND tirpalais.



3 pav. SiO₂, CuO, MoO₃ ir B₂O₃ ND (0; 12,5; 25; 50, 100 ppm) poveikis vandenilio peroksido (A, H₂O₂) ir malondialdehido (B, MDA) kiekiui sėjamaajame žirnyje (*P. sativum* L.). H₂O – kontroliniai augalai, laistomi dejonizuotu vandeniu. Vertės yra vidurkiai ± SE apskaičiuoti iš devynių pakartojimų. Skirtingos raidės rodo reikšmingus skirtumus pagal Tukey HSD testą ($p < 0,05$)

MoO₃ ND mažino H₂O₂ kiekį žirnių lapuose, augalus palaisčius 50 ir 100 ppm koncentracijos tirpalais. Taip pat nustatyta, kad lipidų peroksidacija sumažėjo didėjant MoO₃ ND koncentracijai suspensijoje.

12,5 ppm CuO ND suspensija turėjo stipriausią poveikį H₂O₂ (17 %) ir MDA (30 %) kiekiui žirnių lapuose. Šiek tiek silpnesnis poveikis nustatytas augalus paveikus 25 ppm koncentracija. O 100 ppm CuO ND tirpalas reikšmingai didino (26 %) H₂O₂ kiekį žirnių lapuose.

12,5 ir 25 ppm B₂O₃ ND suspensijos efektyviai mažino H₂O₂ kiekį žirnių lapuose. MDA kiekis žirnių lapuose reikšmingai sumažėjo, kai augalai buvo palaistyti 12,5 ir 25 ppm B₂O₃ ND tirpalais. O 100 ppm B₂O₃ ND suspensija, reikšmingai didino MDA kiekį žirniuose.

3.2 Agrometeorologinės sausros ir nanodalelių poveikis žirniams

Šiame skyriuje aptariamas ND poveikis agrometeorologinės sausros paveiktiems žirniams, augintiems esant 30 % substrato drėgniui (SD 30 %) ir optimaliomis sąlygomis esant normaliam 80 % substrato drėgniui (SD 80 %). Išsamiai aptariama ND poveikis morfologiniams parametrų, antioksidacinės sistemos aktyvumui, makro- ir mikroelementų kaupimuisi žirniuose. ND poveikis žirniams buvo ištirtas augalus veikiant skirtingomis ND koncentracijomis bei taikant du poveikio būdus – purškiant augalus ir laistant per šaknis.

3.2.1 Silicio dioksido nanodalelių poveikis žirniams augusiems skirtingomis substrato drėgnio sąlygomis

Nanodalelių poveikis morfologiniams parametrų

Žirnių laistymas 12,5, 25 ir 50 ppm silicio dioksido (SiO₂) ND suspensijomis turėjo reikšmingą poveikį žirnių aukščiui (3 lentelė, SD 80 %) esant pakankamai substrato drėgniui. Be to, laistymas 50 ppm SiO₂ ND suspensija skatino lapų vystymąsi. Augalus purškiant 12,5, 25 ir 50 ppm SiO₂ ND suspensija, jų lapų plotas padidėjo iki 15 %. Santykinis vandens kiekis (SVK) didėjo, kai augalai buvo palaistyti 50 ppm SiO₂ ND suspensija. Po palaistymo 12,5, 25 ir 50 ppm suspensijomis atitinkamai padidėjo šaknų ir antžeminės dalies santykis apie 7, 34 ir 68 %. Specifinio lapų ploto (SLP) skirtumų žirniuose po jų laistymo su skirtingų koncentracijų SiO₂ ND tirpalais nenustatyta, tačiau po nupurškimo su 12,5 ppm – SLA sumažėjo apie 13 %. Nustatytas teigiamas poveikis gumbelių formavimuisi, gumbelių skaičius padidėjo iki 5 kartų, kai žirniai buvo palaistyti SiO₂ ND tirpalais ir augo normaliam substrato drėgnyje. Tuo tarpu žirnius nupurškus SiO₂ ND tirpalais, šaknyse gumbelių skaičius padidėjo nežymiai iki 1,4 karto. Nustatyta, kad augalus paveikus 50 ppm SiO₂ ND suspensija, žirnių derlius gali padidėti iki 40 %.

SiO₂ ND parodė reikšmingą teigiamą poveikį žirnių augalams, kai jie augo substrate, kuriame nėra pakankamai drėgmės (3 lentelė, SD 30 %). Po palaistymo 12,5, 25 ir 50 ppm SiO₂ ND suspensijomis, žirnių aukštis padidėjo apie 23, 20 ir 18 % atitinkamai, o apipurškus – 8, 11 ir 24 %. Lapų plotas padidėjo apie 11 ir 13 %, kai žirniai buvo palaistyti 25 ir 50 ppm tirpalais, bei 12, 18 ir 10 % augalus nupurškus 12,5, 25 ir 50 ppm SiO₂ ND tirpalais. Nustatytas SVK padidėjimas po žirnių palaistymo 12,5 ppm ir 50 ppm SiO₂ ND suspensijomis, kai augalai augo esant 30 % substrato drėgniui. Žirnius paveikus 50 ppm SiO₂ ND tirpalu laistant, reikšmingai padidėjo šaknų ir antžeminės dalies santykis (iki 30 %). Tačiau šaknų ir antžeminės dalies santykis sumažėjo žirnius nupurškus 12,5 ir 25 ppm SiO₂ ND tirpalais. SiO₂ ND nepaveikė žirnių augusių sausros sąlygomis SLP, išskyrus augalus, nupurškčius 25 ppm tirpalu. Sausros paveiktų žirnių šaknyse susiformavo daugiau gumbelių (30 %) juos palaisčius 50 ppm SiO₂ ND suspensija, o nupurškus 12,5, 25 ir 50 ppm tirpalais – 33 %, 50 % ir 180 %. Žirnių augusių drėgmės trūkumo sąlygomis derlius reikšmingai padidėjo (iki 45 %) juos paveikus SiO₂ ND suspensijas.

Nanodalelių poveikis oksidacinio streso biožymenims

SiO₂ ND padidino H₂O₂ koncentraciją žirnių lapuose, palaisčius 12,5 ppm 25 ppm ir 50 ppm, bei nupurškus 12,5 ppm tirpalais, kai augalai augo esant 80 % substrato drėgniui (3 lentelė, SD 80 %). Žinant, kad H₂O₂ augalams reikia nedideliais kiekiais, galima teigti, kad toks padidėjimas yra optimalus augalams ir nesukėlė neigiamo atsako. MDA kiekis žirnių lapuose žymiai sumažėjo, kai žirniai buvo paveikti 50 ppm SiO₂ ND suspensija (3 lentelė, SD 80 %).

Žirniuose paveiktuose sausros streso, H₂O₂ kiekis padidėjo iki pavojingos augalui koncentracijos, sukeliančios ląstelėse disbalansą (3 lentelė, SD 30 %). Juose H₂O₂ koncentracija padidėjo 2,5 karto, lyginant su augalais, augusiais normalaus drėgnio substrate. Tačiau augalų laistymas SiO₂ ND tirpalais sumažino H₂O₂ koncentraciją iki 27 %, o purškimas – 21 %. Žirnių lapuose MDA koncentracija padidėjo daugiau nei 26 %, kai augalai buvo veikiami sausros, palyginti su augusiais 80 % SD. Pažymėtina, kad žirniuose paveiktuose sausros ir SiO₂ ND tirpalų, MDA kiekis žymiai sumažėjo, lyginant su ND nepaveiktais augalais.

Nanodalelių poveikis nefermentiniams antioksidantams

Nustatytas SiO₂ ND slopinamas poveikis žirnių antioksidacinei sistemai, kurie augo 80 % drėgnio substrate (3 lentelė, SD 80 %). TPC kiekis sumažėjo 17 %, augalus palaisčius 12,5 ppm koncentracijos suspensija, o nupurškus SiO₂ ND tirpalais sumažėjo iki 23 %. Statistiškai patikimą slopinantį poveikį FRAP antioksidacinei galiai sukėlė augalų apdorojimas SiO₂ ND suspensijomis. Poveikis DPPH laisvųjų radikalų surišimo aktyvumui buvo daugialypis, po poveikio 25 ppm tirpalu aktyvumas sumažėjo apie 13 %, tačiau po augalų nupurškimo 12,5 ppm

suspensija padidėjo 25 %. ABTS laisvųjų radikalų šalinimo aktyvumas žymiai padidėjo žirniuose, augusiuose normalaus drėgnio substrate, nupurškus 50 ppm SiO₂ ND suspensija.

Priešingi rezultatai gauti tiriant SiO₂ ND poveikį žirnių augalams, augusiems sumažinto drėgnio substrate (3 lentelė, 30 % SD). Žirnių lapuose nustatytas reikšmingas 16, 18 ir 23 % TPC padidėjimas palaisčius, o 20, 25 ir 30 % TPC padidėjimas nupurškus augalus 12,5, 25 ir 50 ppm SiO₂ ND suspensijomis. Rezultatai parodė, kad laistymas 50 ppm SiO₂ ND tirpalu, DPPH laisvųjų radikalų surišimo gebą padidino apie 35 %, o purškimas iki 23 %. FRAP antioksidacinė geba padidėjo iki 20 % augalus palaisčius 25 ir 50 ppm SiO₂ ND ir iki 19 % nupurškus, lyginant su sausra paveiktais žirniais. SiO₂ ND ir sausra ABTS laisvųjų radikalų surišimo aktyvumą padidino 69, 82 ir 86 %, augalus palaisčius 12,5, 25 ir 50 ppm SiO₂ ND tirpalais. Tačiau po nupurškimo tik 50 ppm suspensija padidino ABTS aktyvumą iki 26 %.

Nanodalelių poveikis fermentiniams antioksidantams

12,5 ir 25 ppm SiO₂ ND suspensijos statistiškai patikimai stimuliuo CAT aktyvumą nupurškus žirnius, augusius esant pakankamam substrato drėgniui (3 lentelė, SD 80 %). APX ir GR fermentams nustatytas slopinantis poveikis, APX aktyvumas sumažėjo iki 38 % po poveikio 50 ppm suspensija, o GR aktyvumas – iki 13 %, kai žirniai buvo palaistyti 50 ppm SiO₂ ND suspensija. Ypač stiprus poveikis nustatytas GPX ir SOD fermentų veiklai. GPX aktyvumas padidėjo 9, 8 ir 5 kartus palaisčius žirnius 12,5, 25 ir 50 ppm SiO₂ ND suspensijomis, o po purškimo visos naudotos koncentracijos didino GPX aktyvumą 9 kartus. SOD aktyvumas taip pat padidėjo augalus paveikus SiO₂ ND tirpalais: laistant iki 47 %, purškiant iki 48 %.

Žirniuose augusiuose drėgmės trūkumo sąlygomis, CAT aktyvumas padidėjo 36, 84 ir 153 % palaisčius ir 13, 41 ir 119 % nupurškus augalus 12,5, 25 ir 50 ppm SiO₂ ND suspensijomis (3 lentelė, 30 % SD). Didėjant SiO₂ ND koncentracijai, SOD aktyvumas po laistymo padidėjo iki 37 %, o po purškimo – iki 23 % sausros paveiktuose žirnių augaluose. GR aktyvumui teigiamos įtakos turėjo laistymas 25 ir 50 ppm SiO₂ ND suspensijomis, o nupurškus SiO₂ ND suspensija, sausros paveiktus žirnius, GR aktyvumas padidėjo per 128 %. Pažymėtina, kad buvo nustatytas APX ir GPX aktyvumo sumažėjimas. APX aktyvumas sumažėjo apie 37 % ir 16 % po laistymo 12,5 ir 25 ppm SiO₂ ND suspensijomis. GPX aktyvumo slopinimas (19 %) nustatytas sausros paveiktus žirnius laistant 12,5 ppm SiO₂ ND suspensija, o purškimas slopino aktyvumą iki 30 %.

3 lentelė. SiO₂ ND (12,5; 25; 50 ppm) poveikis sėjamajam žirniui (P. sativum L.) augusiam skirtingo drėgnio substratuose (SD 80 %; SD 30 %), išreiškiamas procentiniu pokyčiu (%), lyginant su kontrole (80 % SD kontrolė reiškia augalus, auginamus SD 80 % ir neapdorotus su ND; SD 30 % kontrolė reiškia sausros paveiktus, bet neapdorotus su ND augalus) poveikio stiprumo žemėlapyje. Statistiškai reikšmingi skirtumai pažymėti pajuodintu šriftu Tukey HSD testą (p<0,05)

SiO ₂ ND poveikis, ppm	SD 80 %						SD 30 %					
	Laistymas			Purškimas			Laistymas			Purškimas		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Augalų aukštis	6	15	21	4	3	3	23	20	18	8	11	24
Lapų plotas	1	4	31	8	9	15	0	11	13	12	18	10
Gumbeliai	200	300	460	100	140	100	0	0	33	33	50	183
SVK	1	4	6	3	2	3	7	7	2	3	4	10
Š/A	7	34	68	70	23	4	27	14	30	-3	-8	16
SLP	-11	-3	-9	-13	6	-3	-24	-20	-4	2	23	-25
Derlius	-2	-5	40	-14	4	38	37	37	44	40	35	45
ABTS	-1	2	6	3	1	8	69	82	86	-7	5	26
DPPH	2	-13	-3	25	-15	5	1	9	35	11	19	23
TPC	-17	-4	-4	-21	-23	-12	16	18	23	20	25	30
FRAP	-16	-12	-19	-18	-11	-28	3	20	20	2	7	19
HP	26	20	12	78	0	-9	-15	-25	-27	1	-13	-21
MDA	15	5	-9	4	1	-5	-15	-18	-19	-3	-9	-13
GR	2	2	-13	7	-11	-19	2	21	100	46	55	128
GPX	914	841	525	919	965	912	-19	0	6	-29	-30	-4
APX	16	-12	-32	13	-24	-38	-37	-16	-5	-65	-37	-25
SOD	49	44	47	42	47	48	5	32	37	7	18	23
CAT	15	8	-16	93	89	19	36	84	153	13	41	119

SVK – santykinis vandens kiekis, Š/A - šaknų ir antžeminės dalies santykis, SLP – specifinis lapų plotas, TPC – suminis fenolinių junginių kiekis, HP – vandenilio peroksidas, MDA – malondialdehidai, GR – glutatono reduktazė, GPX – gvajakolio peroksidazė, APX – askorbato peroksidazė, SOD – superoksido dismutazė, CAT – katalazė, Cu kiekis lapuose, stiebe ir šaknyje. 0 – kontroliniai augalai laistomi dejonizuotu vandeniu, sausros stresas – 30 % substrato drėgmės.

3.2.2 Vario oksido nanodalelių poveikis žirniams augusiems skirtingomis substrato drėgnio sąlygomis

Nanodalelių poveikis žirnių morfologiniams parametrams

4 lentelėje pateikti rezultatai rodo, kad vario oksido (CuO) ND neturėjo įtakos augalų, augusių 80 % SD, aukščiui, SLP ir šaknų bei antžeminės dalies santykiui. Lapų plotas sumažėjo 13 % palaisčius 25 ar 50 ppm CuO ND suspensijomis. Žirnius nupurškus 12,5 ir 25 ppm CuO ND suspensijomis, lapų plotas šiek tiek sumažėjo. SVK sumažėjo, augalus paveikus 50 ppm CuO ND tirpalu. Žirnių, augusių su 80 % substrato drėgme, derliui nebuvo nustatytas reikšmingas pokytis.

Didžiausia CuO ND koncentracija (50 ppm) ženkliai sumažino žirnių aukštį laistant ar purškiant CuO ND jiems augant nepakankamo drėgnio substrate (4 lentelė, 30 % SD). Augalus laistant 25 ir 50 ppm suspensijomis, lapų plotas sumažėjo 14 ir 18 %, o purškiant 25 ppm suspensija – 20 %. SLP sumažėjo 12 % augalus palaisčius 12,5 ir 25 ppm CuO ND tirpalais. CuO ND turėjo teigiamą poveikį gumbelių susidarymui ant žirnių šaknų, kai augalai buvo

paveikti 50 ppm CuO ND koncentracija. Nustatytas derliaus padidėjimas apie 27 % augalus laistant ir 47 % juos purškiant CuO ND lyginant su sausros streso paveiktais augalais.

Nanodalelių poveikis oksidacinio streso biožymenims

CuO ND turėjo stiprų poveikį tiek H₂O₂, tiek MDA kaupimuisi žirnių augaluose, kurie augo pakankamos drėgmės substrate (4 lentelė, SD 80 %). H₂O₂ kiekis padidėjo 70, 41 ir 16 % žirnius laistant 12,5, 25 ir 50 ppm CuO ND suspensijomis, o purškiant – 45, 74 ir 72 %, atitinkamai. MDA kiekis taip pat reikšmingai padidėjo (54 %) žirnius paveikus CuO ND tirpalais. Iš šių rezultatų galime daryti išvadą, kad CuO ND stipriai aktyvuoja lipidų peroksidaciją ir gali sukelti oksidacinį stresą augaluose.

Tačiau CuO ND poveikis H₂O₂ kiekiui augale pasikeitė kartu su sausros poveikiu (4 lentelė. 30 % SD). H₂O₂ kiekis žymiai sumažėjo, žirnius paveikus skirtingų CuO ND koncentracijų tirpalais. Tiek po laistymo, tiek po purškimo CuO ND tirpalais, MDA kiekis žirniuose didėjo iki 66 %.

Nanodalelių poveikis nefermentiniams antioksidantams

Žirnių laistymas 12,5 ppm CuO ND tirpalu žymiai sumažino TPC kiekį, bet nustatytas padidėjimas paveikus 25 ir 50 ppm koncentracijų tirpalais (4 lentelė. 80 % SD). Žirnių purškimas 12,5 ir 25 ppm CuO ND tirpalais, sumažino TPC kiekį 17 ir 10 %, atitinkamai. ABTS laisvųjų radikalų surišimo aktyvumo slopinimas buvo nustatytas palaisčius žirnius 25 ir 50 ppm arba nupurškus 12,5 ir 25 ppm CuO ND suspensijomis, lyginant su kontroliniais augalais. DPPH laisvųjų radikalų šalinimo aktyvumas sumažėjo 36 ir 13 %, kai žirniai buvo palaistyti 12,5 ir 50 ppm CuO ND suspensijomis (4 lentelė. 80 % SD). Tačiau reikšmingas DPPH laisvųjų radikalų šalinimo aktyvumo stimuliavimas nustatytas žirnius nupurškus 25 ir 50 ppm CuO ND suspensijomis. FRAP antioksidacinė geba žirnių lapuose padidėjo iki 3 kartų juos palaisčius ir iki 2 kartų nupurškus CuO ND suspensijomis.

Žirniuose, kurie buvo auginti 30 % SD ir CuO ND buvo nustatyta daugiau teigiamų rodiklių lyginant su augalais augintais. Pavyzdžiui, TPC kiekis žirnių lapuose žymiai padidėjo, kai sausros paveikti žirniai buvo laistomi arba purškiami bet kokia CuO ND koncentracija. Be to, DPPH laisvųjų radikalų surišimo aktyvumas didėjo didėjant CuO ND koncentracijai. ABTS laisvųjų radikalų surišimo aktyvumas buvo stimuliuojamas iki 70 % priklausomai nuo naudotos koncentracijos laistant ir iki 50 % purškiant CuO ND tirpalais. Nustatytas reikšmingas ABTS laisvųjų radikalų surišimo aktyvumo sumažėjimas nupurškus augalus 12,5 ppm CuO ND suspensija. Žirnių lapuose teigiamas poveikis FRAP antioksidacinei galiai pastebėtas augalus paveikus visomis CuO ND koncentracijoms.

Nanodalelių poveikis fermentiniams antioksidantams

Stiprus slopinamasis poveikis CAT ir GR aktyvumui nustatytas žirnių lapuose, kai augalai augo pakankamos drėgmės substrate (4 lentelė. 80 % SD). CAT aktyvumas sumažėjo iki 65 % žirnius paveikus CuO ND suspensijomis. Be to, GR aktyvumas sumažėjo 37, 27 ir 17 %, žirnius palaisčius 12,5, 25 ir 50 ppm CuO ND suspensijomis, atitinkamai. Žymus aktyvumo skatinimas nustatytas APX fermentui, nes palaisčius CuO ND suspensijomis ji padidėjo iki 5,6 karto, o nupurškus – 4,6 karto. GPX aktyvumas reikšmingai padidėjo iki 94 % palaisčius augalus bet kokios koncentracijos CuO ND tirpalu, o purškiant reikšmingą poveikį turėjo tik 25 ppm CuO ND tirpalas. SOD aktyvumas padidėjo, žirnius laistant 25 ir 50 ppm, ir purškiant 12,5 ppm CuO ND tirpalais.

Esant sausrai, žirnių lapuose nustatytas CuO ND neigiamas poveikis GR aktyvumui (4 lentelė. 30 % SD), sumažinant jį iki 93 %. Taip pat slopinantis CuO ND suspensijų poveikis nustatytas GPX aktyvumui. GPX aktyvumo padidėjimas nustatytas tik augalus palaisčius 25 ppm CuO ND suspensija. CAT aktyvumas padidėjo iki 1,8 karto naudojant bet kurią CuO ND koncentraciją. Be to, APX aktyvumas taip pat buvo žymiai paskatintas žirnius paveikus CuO ND suspensijomis. Žirnių laistymas CuO ND reikšmingai padidino SOD aktyvumą (iki 33 %), kuris priklausė nuo naudotos koncentracijos, be to, purškiant 50 ppm tirpalu SOD aktyvumas padidėjo 53 %.

Makro- ir mikroelementų sudėties pokyčiai

Vario (Cu) kiekis žirnių lapuose, augusiuose substrate su pakankamu drėgmės kiekiu, padidėjo 24 % juos palaisčius 50 ppm CuO ND suspensija (4 lentelė. 80 % SD). Cu kiekis žirnių lapuose padidėjo juos paveikus CuO ND tirpalais: 12,5 ppm – 49 %, 25 ppm – 112 %, o 50 ppm – 233 %, ta pati tendencija buvo pastebėta Cu kaupimuisi žirnio stiebe. Nustatytas Cu kiekio padidėjimas žirnių šaknyse 59, 101 ir 159 % , augalus palaisčius, o dar daugiau jo susikaupė – 77, 113 ir 230 % juos nupurškus 12,5, 25 ir 50 ppm CuO ND suspensijomis, atitinkamai.

4 lentelė. CuO ND (12,5; 25; 50 ppm) poveikis sėjamajam žirniui (*P. sativum L.*) augusiam skirtingo drėgnio substratuose (SD 80 %; SD 30 %), išreiškiamas procentiniu pokyčiu (%), lyginant su kontrole (80 % SD kontrolė reiškia augalus, auginamus SD 80 % ir neapdorotus su ND; SD 30 % kontrolė reiškia sausros paveiktus, bet neapdorotus su ND augalus) poveikio stiprumo žemėlapyje. Statistiškai reikšmingi skirtumai pažymėti pajuodintu šriftu Tukey HSD testą ($p < 0,05$)

CuO ND poveikis, ppm	SD 80 %						SD 30 %					
	Laistymas			Purškimas			Laistymas			Purškimas		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Augalų aukštis	-2	0	-4	-6	-4	-8	5	0	-9	2	5	-17
Lapų plotas	-7	-12	-13	-21	-19	8	1	-14	-18	12	-20	-1
Gumbeliai	20	-40	220	-60	120	180	-83	-17	83	17	-100	283
SVK	2	2	-5	-3	-3	-4	-1	-7	-15	-3	-4	-9
Š/A	25	-29	-1	62	-20	-36	22	-24	-24	-18	-9	-34
SLP	-26	-24	-17	-24	12	10	-12	-13	4	13	-7	-3
Derlius	-4	-2	-1	-9	-11	-7	2	1	27	7	14	47
ABTS	7	-15	-45	-13	-13	26	55	70	60	-54	50	48
DPPH	-36	-5	-13	-7	6	18	17	21	56	6	13	70
TPC	-4	4	4	-17	-10	1	6	10	27	7	4	15
FRAP	177	304	-13	136	100	190	103	182	224	224	253	220
HP	70	41	16	45	74	72	-20	-32	-23	-28	-35	-37
MDA	52	54	50	52	77	81	66	51	27	61	62	38
GR	-37	-27	-17	2	-27	-21	-70	-93	-92	-68	-81	-78
GPX	90	83	94	15	59	-4	-9	16	2	-23	-37	-22
APX	364	556	173	463	320	308	72	25	58	71	19	33
SOD	1	18	21	25	-5	-3	33	29	20	-3	-31	53
CAT	-57	-37	-63	-48	-68	-65	176	166	128	167	153	153
Cu (lapai)	1	-11	24	49	112	233	-8	-4	-17	50	153	419
Cu (stiebai)	13	0	19	50	88	199	14	35	55	69	89	359
Cu (šaknyse)	59	101	159	77	113	230	29	86	94	64	101	166

SVK – santykinis vandens kiekis, Š/A - šaknų ir antžeminės dalies santykis, SLP – specifinis lapų plotas, TPC – suminis fenolinių junginių kiekis, HP – vandenilio peroksidas, MDA – malondialdehidai, GR – glutatono reduktazė, GPX – gvajakolio peroksidazė, APX – askorbato peroksidazė, SOD – superoksido dismutazė, CAT – katalazė. , Cu kiekis lapuose, stiebe ir šaknyje. 0 – kontroliniai augalai laistomi dejonizuotu vandeniu, sausros stresas – 30 % substrato drėgmės.

Laistymas CuO ND nesukėlė reikšmingų Cu kiekio skirtumų sausros paveiktuose žirnių augalų lapuose (4 lentelė. 30 % SD). Tačiau Cu kiekis lapuose padidėjo iki 55 %, kai sausros veikiami žirniai buvo nupurkšti CuO ND suspensijomis. Sausros paveiktų žirnių šaknyse Cu susikaupė daugiau juos laistant CuO ND tirpalais, lyginant su ND nepaveiktais augalais. Purškimas CuO ND turėjo teigiamą poveikį Cu kaupimuisi visose žirnio dalyse, o Cu kiekis didėjo didėjant CuO ND koncentracijai. Lapuose Cu kiekis padidėjo iki 4 kartų, stiebuose – iki 3,5, šaknyse – iki 1,7 karto.

3.2.3 Molibdeno trioksido nanodalelių poveikis žirniams augusiems skirtingomis substrato drėgnio sąlygomis

Nanodalelių poveikis morfologiniams parametrams

Molibdeno trioksido (MoO_3) ND suspensijos turėjo reikšmingą poveikį žirnių, augusių 80 % drėgnio substrate, morfologiniams parametrams (5 lentelė). Nustatytas padidėjęs žirnių aukštis palaisčius bet kokios koncentracijos MoO_3 ND suspensijomis. 50 ppm MoO_3 ND suspensija paskatino lapų ploto augimą tiek augalus palaisčius (25 %), tiek nupurškus (15 %). 50 ppm MoO_3 ND suspensija turėjo ypatingai stiprų teigiamą poveikį SVK, taip pat šaknų ir antžeminės dalies santykiui. SLP sumažėjo apie 28 % žirniams augant pakankamai drėgname substrate ir juos palaisčius 12,5 ppm MoO_3 ND tirpalu, o nupurškus 12,5 ir 50 ppm suspensijomis, SLP sumažėjo iki 30 %. MoO_3 ND tirpalai turėjo stiprų poveikį gumbelių susidarymui ant šaknų, nes jų skaičius padidėjo 3 ir 5,6 karto palaisčius 25 ir 50 ppm tirpalais ir 1,4 karto didesnis nupurškus 50 ppm MoO_3 ND suspensija, lyginant su kontroliniais augalais. Žirnius nupurškus 50 ppm MoO_3 ND tirpalu pastebėtas reikšmingas derliaus padidėjimas (5 lentelė. SD 80 %).

MoO_3 ND efektyviai sumažino sausros poveikį žirnių morfologiniams parametrams (5 lentelė, 30 % SD). Žirnius palaisčius ar nupurškus MoO_3 ND tirpalais jų aukštis padidėjo iki 40 % ir iki 24 % atitinkamai. MoO_3 ND suspensijos didino augalų lapų plotą, sausros paveiktuose žirniuose. SVK padidėjo 10 %, kai augalai buvo palaistyti 50 ppm MoO_3 ND. Visos MoO_3 ND koncentracijos sumažino SLP. Pastebėta, kad ir sausros sąlygomis buvo nustatytas teigiamas MoO_3 ND poveikis gumbelių formavimuisi. Nustatytas 1,2–5 kartų padidėjimas, kai žirniai buvo laistomi 25 ir 50 ppm, ir 1,2–2 kartų – nupurškus MoO_3 ND. Žirnių, paveiktų sausros, derlius stipriai priklausė nuo naudotos MoO_3 ND koncentracijos, nes padidėjo iki 80 % ir 64 % po palaistymo ir purškimo 50 ppm MoO_3 ND suspensija.

Nanodalelių poveikis oksidacinio streso biožymenims

H_2O_2 kiekis žirnių lapuose padidėjo iki 2 kartų augalus paveikus skirtingų koncentracijų MoO_3 ND suspensijomis (SD 80 %). MoO_3 ND skatino lipidų peroksidaciją žirnių augaluose, augusiuose 80 % SM (5 lentelė. 80 % SD). Tai rodo MDA koncentracijos padidėjimas žirniuose, kai jie buvo paveikti 12,5 ir 25 ppm MoO_3 ND suspensijomis.

Sausros paveiktuose žirniuose H_2O_2 kiekis sumažėjo 8, 12 ir 20 %, augalus palaisčius 12,5, 25 ir 50 ppm MoO_3 ND suspensijomis, bet MDA koncentracija padidėjo 9, 11 ir 25 %, 196

atitinkamai (5 lentelė. 30 % SD). Purškimas MoO₃ ND suspensijomis turėjo panašų poveikį mažinant H₂O₂ kiekį žirnių lapuose. Sumažėjęs MDA kiekis buvo nustatytas žirniuose, purkštuose 50 ppm MoO₃ ND suspensija, lyginant su augalais, augusiais sausros sąlygomis ir neapdorotais ND.

Nanodalelių poveikis nefermentiniams antioksidantams

Pastebėta, kad MoO₃ ND turi dvejopą poveikį nefermentiniams antioksidantams žirniuose, augusiems 80 % SD (5 lentelė). TPC kiekio sumažėjimas iki 20 % nustatytas augalus palaisčius 12,5 ppm MoO₃ ND tirpalu, o nupurškus 12,5 ir 25 ppm tirpalu TPC kiekis sumažėjo 23 ir 15 %, atitinkamai. Žirnių lapuose FRAP antioksidacinė geba žymiai padidėjo augalus palaisčius ar nupurškus bet kokios koncentracijos MoO₃ ND tirpalu. Žirnių palaistymas 25 ir 50 ppm MoO₃ ND tirpalais paskatino DPPH laisvųjų radikalų surišimo aktyvumą 27 ir 36 %, atitinkamai. Purškimas 25 ppm MoO₃ ND suspensija sumažino DPPH laisvųjų radikalų surišimo aktyvumą. ABTS laisvųjų radikalų surišimo aktyvumas buvo slopinamas žirnius palaisčius 12,5 ppm ir nupurškus 25 ppm MoO₃ ND suspensijomis.

Sausros paveiktuose žirniuose, MoO₃ ND poveikis nefermentiniams antioksidantams buvo reikšmingas (5 lentelė. 30 % SD). TPC kiekis sausros paveiktuose žirniuose juos palaisčius MoO₃ ND tirpalais padidėjo iki 37 %, o nupurškus – iki 15 %. ABTS laisvųjų radikalų surišimo aktyvumas žirniuose ypač padidėjo – 60, 67 ir 105 % augalus palaisčius 12,5, 25 ir 50 ppm MoO₃ ND tirpalais, atitinkamai. Be to, MoO₃ ND išpurškimas ant augalų padidino ABTS laisvųjų radikalų surišimo aktyvumą iki 70 %. Rezultatai rodo (5 lentelė. 30 % SD), kad laistant MoO₃ ND DPPH laisvųjų radikalų surišimo aktyvumas padidėjo iki 145 %, o purškiant – iki 81 %. Bet kokios koncentracijos MoO₃ ND poveikis buvo reikšmingas FRAP antioksidacinei galiai žirnių lapuose.

5 lentelė. MoO₃ ND (12,5; 25; 50 ppm) poveikis sėjamajam žirniui (P. sativum L.) augusiam skirtingo drėgumo substratuose (SD 80 %; SD 30 %), išreiškiamas procentiniu pokyčiu (%), lyginant su kontrole (80 % SD kontrolė reiškia augalus, auginamus SD 80 % ir neapdorotus su ND; SD 30 % kontrolė reiškia sausros paveiktus, bet neapdorotus su ND augalus) poveikio stiprumo žemėlapyje. Statistiškai reikšmingi skirtumai pažymėti pajuodintu šriftu pagal Tukey HSD testą (p<0,05)

MoO ₃ ND poveikis, ppm	SD 80 %						SD 30 %					
	Laistymas			Purškimas			Laistymas			Purškimas		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50

Augalų aukštis	6	21	25	7	5	17	10	18	40	4	10	24
Lapų plotas	-3	15	25	-9	2	15	-7	-8	30	12	10	10
Gumbeliai	40	320	560	40	20	140	-50	117	533	-17	117	183
SVK	5	6	8	3	5	7	13	9	21	7	7	10
Š/A	-1	2	22	33	-15	11	-45	1	23	36	-8	16
SLP	-28	-22	-24	-34	-6	-30	-12	-30	-29	-12	-18	-25
Derlius	6	5	6	1	-12	7	11	26	80	3	15	64
ABTS	-11	-5	2	-3	-8	4	60	67	105	43	65	70
DPPH	5	27	36	6	-7	13	59	53	145	36	76	81
TPC	-20	4	4	-23	-15	1	20	28	37	-1	22	15
FRAP	117	141	202	111	146	190	231	241	242	55	100	142
HP	109	101	122	68	120	65	-8	-12	-20	-9	-12	-30
MDA	18	-2	13	19	24	5	9	11	25	13	-1	-17
GR	-70	-59	-56	-33	-52	-62	-56	-57	-31	-88	-79	-71
GPX	161	76	28	110	142	168	-11	43	55	17	-2	56
APX	692	899	607	423	481	748	198	490	622	216	276	422
SOD	25	65	19	5	24	35	-2	-15	17	-10	0	20
CAT	-16	-19	-4	41	-10	-2	234	215	224	161	191	183
Mo (lapai)	14	27	43	14	26	47	160	234	310	27	421	454
Mo (stiebai)	10	34	34	29	44	46	201	205	215	114	122	136
Mo (šaknys)	200	213	360	130	177	184	21	28	69	38	56	73

SVK – santykinis vandens kiekis, Š/A - šaknų ir antžeminės dalies santykis, SLP – specifinis lapų plotas, TPC – suminis fenolinių junginių kiekis, HP – vandenilio peroksidas, MDA – malondialdehidai, GR – glutatono reduktazė, GPX – gvajakolio peroksidazė, APX – askorbato peroksidazė, SOD – superoksido dismutazė, CAT – katalazė. , Cu kiekis lapuose, stiebe ir šaknyje. 0 – kontroliniai augalai laistomi dejonizuotu vandeniu, sausros stresas – 30 % substrato drėgmės.

Nanodalelių poveikis fermentiniams antioksidantams

Stiprus MoO₃ ND poveikis fermentinių antioksidantų aktyvumui nustatytas augaluose, augusiuose pakankamo drėgnio substrate (80 %) (5 lentelė). CAT aktyvumas padidėjo 41 % žirnius nupurškus 12,5 ppm MoO₃ ND suspensija. 12,5, 25 ir 50 ppm MoO₃ ND koncentracijos sukėlė žymiai didesnę APX aktyvumą lyginant su ND nepaveiktas augalais. Nustatyta, kad GR aktyvumas labai sumažėjo po poveikio MoO₃ ND suspensijomis. MoO₃ ND turėjo stiprų poveikį GPX aktyvumui žirnių lapuose, kuris priklausė nuo koncentracijos ir naudojimo būdo.

Sausros poveikis kartu su MoO₃ ND skatino APX aktyvumą žirnių lapuose (5 lentelė. 30 % SD). Žirnius palaisčius 12,5, 25 ir 50 ppm MoO₃ ND suspensijomis, nustatytas APX aktyvumo padidėjimas iki 6 kartų, o nupurškus iki 4 kartų. Panašūs rezultatai nustatyti ir CAT aktyvumui žirnių lapuose. Žirnius palaisčius 12,5, 25 ir 50 ppm MoO₃ ND tirpalais, CAT aktyvumas padidėjo 2 kartus, o nupurškus CAT aktyvumas padidėjo nuo 1,6 iki 1,9 karto. Žirnių laistymas 12,5, 25 ir 50 ppm MoO₃ ND suspensijomis GR aktyvumą sumažino apie 56,

57 ir 31 %, o purškimas - 88, 79 ir 71 %, atitinkamai. Sausros paveiktuose žirniuose GPX aktyvumas padidėjo iki 43 ir 55 % augalus palaisčius 25 ir 50 ppm MoO₃ ND suspensijomis. GPX aktyvumas padidėjo 17 ir 56 % kai augalai buvo nupurkšti 12,5 ir 50 ppm MoO₃ ND suspensijomis. SOD aktyvumas buvo slopinamas, žirnius palaisčius 25 ppm, ir skatinamas palaisčius ar nupurškus 50 ppm MoO₃ ND suspensija.

Makro- ir mikroelementų sudėties pokyčiai

Mo kiekis žirnių lapuose didėjo didėjant MoO₃ ND koncentracijai suspensijoje (5 lentelė). Žirnių lapuose, augusiuose normaliomis sąlygomis (SD 80 %), ir juos laistant MoO₃ ND nustatytas Mo kiekio padidėjimas iki 43 %, o purškiant - iki 47 %. Panaši tendencija pastebėta ir Mo kiekio kaupimuisi žirnio stiebe. Šaknyse Mo susikaupė iki 3,6 karto daugiau, kai žirniai buvo laistomi ir iki 1,8 karto daugiau juos purškiant MoO₃ ND tirpalais.

Sausros ir MoO₃ ND poveikis ypatingai padidino Mo kiekį žirnių lapuose ir stiebuose. Laistymas MoO₃ ND suspensijomis nulėmė iki 3 kartų didesnę Mo kiekį lapuose, o purškimas padidino Mo kiekį iki 4,5 karto.

3.2.4 Boro nanodalelių poveikis žirniams augusiems skirtingomis substrato drėgnio sąlygomis

Poveikis morfologiniams parametrams

Žirnių augusių normalaus drėgnio substrate (80 % SD) aukštis padidėjo 14 ir 27 %, kai jie buvo palaistyti 12,5 ir 50 ppm B₂O₃ ND suspensijomis, o nupurškus – 28 ir 19 %, atitinkamai (6 lentelė, 80 % SD). Lapų plotas taip pat reikšmingai padidėjo, augalus palaisčius 12,5 ir 50 ppm B₂O₃ ND suspensijomis, o nupurškus lapų ploto padidėjimas nustatytas naudojant visas koncentracijas. Taip pat, buvo nustatytas teigiamas B₂O₃ ND poveikis SVK. Augalus nupurškus 12,5 ppm B₂O₃ ND tirpalu, SLP sumažėjo 15 %. Šaknų ir antžeminės dalies santykis statistiškai patikimai padidėjo augalus palaisčius šiomis suspensijomis 12,5 ppm (21 %), 25 ppm (36 %), ir 50 ppm (68 %). Tai rodo, kad augaluose, augusiuose normalaus drėgnio substrate (80 % SD) B₂O₃ ND koncentracijos didėjimas skatino šaknų augimą. Be to, teigiamas poveikis buvo pastebėtas ir augalus nupurškus 12,5 ppm, 25 ppm ir 50 ppm B₂O₃ ND suspensijomis, šaknų ir antžeminės dalies santykis padidėjo 68 %,– 18 %, ir– 34 %, atitinkamai. Rezultatai taip pat rodo, kad B₂O₃ ND teigiamai veikė gumbelių formavimąsi ant augalų šaknų, padidindami jų kiekį iki 5,6 karto žirnius palaisčius ir iki 3,4 juos nupurškus. Žirnių laistymas 50 ppm B₂O₃ ND tirpalu turėjo reikšmingą teigiamą poveikį derliui, o purškiant žirnių derlius padidėjo, kai buvo naudotos, 12,5 ir 25 ppm B₂O₃ ND suspensijos.

B₂O₃ ND stipriai paveikė žirnių augalus, augusius sausros streso sąlygomis (6 lentelė, 30 % SD). Skirtingų koncentracijų B₂O₃ ND suspensija padidino augalų aukštį, kai jie buvo laistomi ar purškiami. Be to, laistant augalus 25 ir 50 ppm B₂O₃ ND suspensijomis, lapų plotas padidėjo atitinkamai 30 ir 40 %. Esant didesnei B₂O₃ ND koncentracijai, statistiškai reikšmingai padidėjo SVK. Sausros paveiktuose žirniuose, šaknų ir antžeminės dalies santykis padidėjo iki 30 % augalus palaisčius ir iki 14 % juos nupurškus B₂O₃ ND tirpalais. 50 ppm B₂O₃ ND koncentracija padidino šaknų gumbelių skaičių 3 kartus augalus palaisčius ir iki 6 kartų - nupurškus. Laistymas 12,5 ir 25 ppm B₂O₃ ND suspensija turėjo teigiamą poveikį žirnių derliui, o purškiant B₂O₃ ND sausros paveiktus žirnius derlius padidėjo iki 16 %.

Nanodalelių poveikis oksidacinio streso biožymenims

Rezultatai rodo, kad B₂O₃ ND poveikis padidino H₂O₂ kiekį augaluose, juos laistant skirtingų koncentracijų tirpalais, kai žirniai augo esant pakankamam substrato drėgnumui (6 lentelė, 80 % SD). Nupurškus augalus, nustatytas statistiškai patikimas 65 % H₂O₂ kiekio padidėjimas, kai žirniai buvo paveikti 12,5 ppm B₂O₃ ND koncentracija. Žirnių lapuose nustatytas reikšmingas MDA koncentracijos sumažėjimas, augalus paveikus bet kokia B₂O₃ ND koncentracija.

Sausros ir B₂O₃ ND paveiktuose augaluose nustatytas reikšmingas H₂O₂ ir MDA slopinimas (6 lentelė, 30 % SD). Apipurškus augalus 12,5, 25 ir 50 ppm B₂O₃ ND tirpalais, H₂O₂ kiekis sumažėjo 18, 24 ir 45 %, o palaisčius – 22, 37 ir 9 %, atitinkamai. MDA koncentracija sumažėjo 22, 13 ir 17 %, žirnius palaisčius, o nupurškus – 20, 25 ir 22 % .kai B₂O₃ ND suspensijos buvo šių koncentracijų 12,5, 25 ir 50 ppm.

Nanodalelių poveikis nefermentiniams antioksidantams

Nustatyta, kad tiek laistymas, tiek purškimas B₂O₃ ND sumažino TPC kiekį žirnių lapuose iki 30 % (6 lentelė, 80 % SD). B₂O₃ ND nepaveikė ABTS laisvųjų radikalų surišimo aktyvumo žirniuose. Tačiau nustatyta, kad žirnius nupurškus 25 ir 50 ppm B₂O₃ ND suspensijomis, DPPH laisvųjų radikalų surišimo aktyvumas padidėjo 25 ir 24 %, atitinkamai. Be to, laistymas ar purškimas 12,5, 25 ir 50 ppm B₂O₃ ND suspensijomis didino FRAP antioksidacinę galią, kai augalai augo 80 % SD.

Rezultatai parodė(6 lentelė, 30 % SD), kad purškiant sausros paveiktus žirnius 12,5, 25 ir 50 ppm B₂O₃ ND suspensijomis turi teigiamą poveikį nefermentiniams antioksidantams, pavyzdžiui TPC kiekis padidėjo iki 18 %, tačiau laistymas 12,5 ppm suspensija žymiai sumažino TPC kiekį. ABTS laisvųjų radikalų surišimo aktyvumas padidėjo iki 73 % augalus laistant ir iki 96 % juos purškiant B₂O₃ ND tirpalais. Panaši tendencija buvo gauta FRAP antioksidacinės galios rezultatuose. Sausros ir 12,5 bei 25 ppm B₂O₃ ND suspensijų laistymas

padidino (20 %) DPPH laisvųjų radikalų surišimo aktyvumą. O nupurškšti 12,5, 25 ir 50 ppm B_2O_3 ND suspensijomis augalai su pasižymėjo DPPH laisvųjų radikalų surišimo aktyvumo padidėjimu apie 35, 24 ir 25 %, atitinkamai.

Nanodalelių poveikis fermentiniams antioksidantams

B_2O_3 ND sukėlė CAT, APX, SOD ir GPX aktyvumą žirnių lapuose, kai jie augo normalaus drėgčio substrate (80 % SD) (6 lentelė). APX aktyvumas ypač padidėjo augalus palaisčius B_2O_3 ND suspensijomis, kiek silpnesnį poveikį sukėlė purškimas. CAT aktyvumas padidėjo iki 2 kartų, augalus palaisčius B_2O_3 ND suspensijomis. Purškiant žirnius 12,5, 25 ir 50 ppm B_2O_3 ND tirpalais, CAT aktyvumas padidėjo 1,3, 1,8 ir 2 kartus. SOD aktyvumas padidėjo 41 % augalus laistant ir 46 % juos purškiant B_2O_3 ND tirpalais. GPX aktyvumas išsiskyrė tuo, kad mažesnės koncentracijos laistant turėjo stipresnį teigiamą poveikį, o purškiant – didesnės. 12,5 ir 50 ppm B_2O_3 ND suspensijos, sumažino GR aktyvumą žirniuose.

Stiprų GR aktyvumo sumažėjimą lėmė sausros ir B_2O_3 ND poveikis – laistant jis sumažėjo apie 55 %, o purškiant – 45 % (6 lentelė, 30 % SD). Taip pat buvo nustatytas slopinantis poveikis SOD aktyvumui žirnius palaisčius 25 ir 50 ppm B_2O_3 ND tirpalais, tačiau SOD aktyvumo padidėjimas (36 %) nustatytas paveikus 12,5 ppm B_2O_3 ND suspensija. Be to, SOD aktyvumo padidėjimas 51 %, gautas sausros streso sąlygomis augusius žirnius nupurškus B_2O_3 ND tirpalu. Palaisčius žirnius B_2O_3 ND suspensijomis, APX suaktyvėjo iki 1,4 karto, o nupurškus – iki 8 kartų. GPX aktyvumas sausros paveiktuose žirniuose padidėjo iki 91 % augalus palaisčius B_2O_3 ND tirpalais, CAT aktyvumas ypač padidėjo laistant arba purškiant augalus bet kokios koncentracijos B_2O_3 ND tirpalais.

Makro- ir mikroelementų sudėties pokyčiai

B kiekis žirnių lapuose, augusių pakankamos drėgmės substrate (6 lentelė, 80 % SD), padidėjo 3, 5 ir 12 kartų laistant ir 2, 3 ir 11 kartų juos nupurškus 12,5, 25 ir 50 ppm B_2O_3 ND suspensijomis, atitinkamai. B kiekis stiebe padidėjo 10 kartų augalus palaisčius ir iki 17 kartų juos nupurškus B_2O_3 ND tirpalais. Be to, B kiekis padidėjo 3, 7 ir 20 kartų augalų šaknyse, kurie augo normaliomis substrato drėgmės sąlygomis ir palaistyti 12,5, 25 ir 50 ppm B_2O_3 ND suspensijomis, atitinkamai. Žirnius nupurškus B_2O_3 ND tirpalais, B kiekis šaknyse padidėjo iki 3 kartų.

Sausros paveiktus (30 % SD) žirnius nupurškus 25 ir 50 ppm B_2O_3 ND tirpalais, B kiekis lapuose padidėjo 3 ir 5 kartus, atitinkamai. Be to, B kiekis lapuose padidėjo 3 kartus, kai žirniai buvo palaistyti B_2O_3 ND suspensijomis. Po žirnių palaistymo B_2O_3 ND suspensijomis, stiebuose nustatytas žymus (iki 18 kartų) B kiekio padidėjimas.

6 lentelė. B₂O₃ ND (12,5; 25; 50 ppm) poveikis sėjamajam žirniui (*P. sativum* L.) augusiam skirtingo drėgnio substratuose (SD 80 %; SD 30 %), išreiškiamas procentiniu pokyčiu (%), lyginant su kontrole (80 % SD kontrolė reiškia augalus, auginamus SD 80 % ir neapdorotus su ND; SD 30 % kontrolė reiškia sausros paveiktus, bet neapdorotus su ND augalus) poveikio stiprumo žemėlapyje. Statistiškai reikšmingi skirtumai pažymėti pajuodintu šriftu pagal Tukey testą ($p < 0,05$)

B ₂ O ₃ ND poveikis, ppm	SD 80 %						SD 30 %					
	Laistymas			Purškimas			Laistymas			Purškimas		
	12.5	25	50	12.5	25	50	12.5	25	50	12.5	25	50
Augalų aukštis	14	7	27	28	9	19	9	26	17	13	20	14
Lapų plotas	28	9	30	38	25	24	-17	30	40	-4	1	-6
Gumbeliai	560	200	480	340	240	200	17	200	317	0	67	583
SVK	2	4	5	6	4	4	-1	10	11	-2	4	9
Š/A	21	36	68	69	18	34	4	14	30	11	14	5
SLP	14	-5	-11	-15	-6	-3	-17	-4	-16	-13	-23	-29
Derlius	-21	2	20	12	11	-1	16	10	6	19	16	14
ABTS	-7	-11	-6	7	5	0	73	61	56	92	96	74
DPPH	-5	4	7	0	25	24	20	8	3	35	24	25
TPC	-29	-22	-22	-19	-21	-25	-6	-1	1	18	5	15
FRAP	106	81	83	132	109	166	194	174	141	217	151	148
HP	119	89	109	65	10	6	-22	-37	-9	-18	-24	-45
MDA	-31	-25	-14	-7	-15	-23	-22	-13	-17	-20	-25	-22
GR	-19	0	-19	-30	-8	-32	-53	-57	-54	-45	-42	-47
GPX	147	136	14	11	71	153	-3	-2	13	81	91	12
APX	1657	1100	1817	363	498	200	136	33	46	750	634	522
SOD	29	41	41	18	46	15	36	-20	-29	9	9	51
CAT	173	107	84	131	175	214	161	159	132	91	69	48
B (lapai)	18	16	7	8	4	20	-3	36	52	-18	-20	-24
B (stiebai)	210	794	1017	221	1508	1670	17	63	83	-9	-2	5
B (šaknys)	344	735	2062	87	202	315	47	167	253	61	158	217

SVK – santykinis vandens kiekis, Š/A – šaknų ir antžeminės dalies santykis, SLP – specifinis lapų plotas, TPC – suminis fenolinių junginių kiekis, HP – vandenilio peroksidas, MDA – malondialdehidai, GR – glutatono reduktazė, GPX – gvajakolio peroksidazė, APX – askorbato peroksidazė, SOD – superoksido dismutazė, CAT – katalazė, Cu kiekis lapuose, stiebe ir šaknyje. 0 – kontroliniai augalai laistomi dejonizuotu vandeniu, sausros stresas – 30 % substrato drėgmės.

3.3 Kompleksinio agrometeorologinės sausros ir sunkiojo metalo vario bei nanodalelių poveikis žirniams

Lauko sąlygomis augalai vienu metu gali patirti keletą skirtingų abiotinių įtampų, pavyzdžiui: sumažėjusį vandens prieinamumą sausros metu ir ekstremalią temperatūrą, pernelyg padidėjusį vandens kiekį potvynių metu, sunkiųjų metalų poveikį, padidėjusį dirvožemio kietumą, ribojantį šaknų augimą ir maistinių medžiagų trūkumą. Įvairūs stresai sukelia skirtingą augalų atsaką, todėl prisitaikydami prie nepalankių aplinkos veiksnių augalai naudoja skirtingus aklimatizacijos mechanizmus. Kai augalą vienu metu veikia keletas nepalankių aplinkos

veiksnių, jų poveikis gali sustiprėti ir augalai patiria stipresnį stresą. Šiame skyriuje analizuojams kompleksinis ND, agrometeorologinės sausras ir sunkiojo metalo vario (Cu) poveikis žirnių augalams.

Silicio dioksido nanodalelių poveikis žirniams, esant kompleksiniam sausras ir vario stresams

SiO₂ ND turėjo stiprų teigiamą poveikį žirniams paveiktiems sausras ir Cu (7 lentelė). Žirnių aukštis, šaknų ir antžeminės dalies santykis, SVK ir derlius padidėjo atitinkamai 24, 86, 24 ir 50 %, augalus palaisčius SiO₂ ND tirpalais esant sausras ir Cu (LxSxCu). Kai augalus paveikė LxSxCu, DPPH laisvųjų radikalų surišimo aktyvumas ir FRAP antioksidacinė geba žymiai padidėjo, atitinkamai 21 ir 10 %, o TPC kiekis padidėjo apie 21 %. Be to, SiO₂ ND stimuliuo GR ir CAT aktyvumą, atitinkamai 50 ir 67 %, kai augalai augo LxSxCu sąlygomis. Laistymas SiO₂ ND suspensijomis sumažino H₂O₂ ir MDA kiekį lapuose, kai augalai buvo paveikti sausras ir Cu. Taip pat buvo nustatytas bio-koncentracijos faktorius (BCF) padidėjimas tiek lapuose (75 %), tiek šaknyse (94 %), taip pat translokacijos faktorius (Tf) – padidėjo apie 150 %, o tolerancijos indeksas (Ti) – 85 %.

Sausros ir Cu paveiktiems žirniams nustatytas teigiamas SiO₂ ND poveikis, kai jie buvo nupurkšti SiO₂ ND tirpalais (PxSxCu). Augalų aukštis, SVK, šaknų ir antžeminės dalies santykis bei derlius padidėjo atitinkamai 34, 23, 99 ir 44 %. Taip pat SiO₂ ND turėjo teigiamą poveikį antioksidaciniam aktyvumui (ABTS, DPPH, FRAP) ir padidino TPC kiekį žirnių lapuose esant PXSxCu sąlygomis. Tačiau PXSxCu stipriai sumažino fermentų, tokių kaip GPX, APX ir SOD aktyvumą žirniuose. Purškimas SiO₂ ND tirpalais žymiai sumažino H₂O₂ (57) ir MDA (55 %) kiekį žirnių lapuose. Nustatytas BCF padidėjimas tiek lapuose (99 %), tiek šaknyse (81 %), taip pat padidėjo Tf iki 200 %, Ti iki 90 % po poveikio PXSxCu.

7 lentelė. Agrometeorologinės sausras, sunkiojo metalo vario (Cu) ir SiO₂ ND (50 ppm) poveikis sėjamajam žirniui (*P. sativum L.*). H₂O – kontroliniai augalai nupurkšti arba palaistyti dejonizuotu vandeniu, substrato drėgnis SD) 80 %; sausras stresas – SD 30 %. Rezultatuose pateikiami vidurkiai, o skirtingos raidės pažymi statistiškai patikimus skirtumus tarp variantų pagal Tukey HSD testą ($p < 0,05$)

50 ppm SiO ₂ ND poveikis	80 % SD		30 % SD				
	H ₂ O	H ₂ O	Cu	Laistymas ND	Purškimas ND	Cu + Laistymas ND	Cu + Purškimas ND
Morfologiniai parametrai							
Augalo aukštis	32,9 a	22,2 cd	20,0 d	28,3 ab	30,4 a	24,9 c	26,8 bc

Lapų plotas	76,9 a	49,9 c	43,9 cd	55,9 b	62,7 ab	44,8 d	50,4 c
Gumbelių sk.	8,3 a	3,7 d	2,0 e	6,0 bc	6,3 b	3,6 d	2,6 de
SVK	80,6 a	41,7 c	29,3 e	47,3 b	45,8 bc	36,3 cd	35,9 cd
Š/A	11,3 a	5,4 bc	4,1 c	6,0 bc	8,9 a	6,8 bc	8,3 ab
SLP	9,5 bc	11,0 ab	13,5 a	8,5 b	9,4 bc	8,4 b	9,6 bc
Derlius	3,4 a	2,4 b	1,8 c	3,3 a	2,6 b	2,7 b	2,6 b
Nefermentinių antioksidantų aktyvumas							
ABTS	113,2 a	103,1 b	96,4 bc	111,4 a	109,1 ab	93,9 c	106,6 a
DPPH	45,1 ab	38,5 c	36,8 cd	45,9 ab	49,2 a	44,5 ab	44,8 ab
TPC	2,7 ab	2,3 bc	1,9 d	2,7 ab	2,9 a	2,3 bc	2,4 b
FRAP	1475,8 a	1158,6 b	939,5 d	1380,3 a	1405,9 a	1027,6 bc	1117,3 bc
Fermentinių antioksidantų aktyvumas							
Gr	0,053 c	0,074 bc	0,084 b	0,041 c	0,027 c	0,126 a	0,094 b
GPX	11,1 c	15,5 b	44,9 a	8,4 d	8,1d	19,3 b	10,6 cd
APX	25,1 e	51,3 b	74,3 a	61,5a	55,6 ab	48,2 cd	34,5 d
SOD	46,8 c	67,2 b	74,5 a	73,9 a	79,7 a	14,7 de	11,1 e
CAT	16,7 d	31,9 c	33,1 c	37,7 ab	15,2 d	55,8 a	43,2 ab
Oksidacinio streso rodikliai							
HP	12,3 e	25,5 ab	26,2 a	21,6 bc	19,6 d	23,1 b	20,3 bc
MDA	54,7 f	62,3 de	97,3 a	89,9 b	76,8 d	83,4 bc	74,7 d
Vario akumuliacija							
BCF (lapai)			0,04 b			0,07 a	0,08 a
BCF (šaknys)			1,6 b			3,1 a	2,9 a
Tf			2,1 b			5,2 a	6,1 a
Ti			62,7 c			116,2 b	119,3 a

BCF – bio-koncentracijos faktorius, Tf – translokacijos faktorius, Ti – tolerancijos indeksas

Vario oksido nanodalelių poveikis žirniams, esant kompleksiniam sausros ir vario stresams

Žirnius veikiant sausrai, Cu ir palaisčius CuO ND tirpalais, nustatytas reikšmingas šaknų ir antžeminės dalies santykio padidėjimas (8 lentelė), be to, reikšmingai sumažėjo lapų plotas ir derlius. Taip pat, nustatytas FRAP antioksidacinės galios ir TPC kiekio sumažėjimas bei statistiškai reikšmingas DPPH ir ABTS laisvųjų radikalų surišimo aktyvumo sumažėjimas žirnių lapuose. CuO ND padidino SOD aktyvumą augaluose paveiktuose sausra ir Cu, o GR, GPX, APX ir CAT fermentų aktyvumo pokyčiams statistinis patikimumas nenustatytas lyginant su augalais paveiktais sausra ir Cu. CuO ND reikšmingai sumažino H₂O₂ kiekį, tačiau MDA

kiekiui žirniuose jokio poveikio nenustatyta. Augalų laistymas CuO ND padidino BCF (13 %) šaknyse, sumažino Tf (33 %) ir neturėjo poveikio Ti.

Kai žirnių augalai buvo nupurkšti CuO ND, gauta daugiau teigiamų rezultatų lyginant su laistymu. Nustatytas SVK padidėjimas iki 44 % augaluose paveiktuose sausra ir Cu bei nupurkštuose CuO ND tirpalais. Šiuose augaluose taip pat nustatytas ABTS laisvųjų radikalų surišimo aktyvumo ir TPC kiekio padidėjimas, o fermentiniams antioksidantams poveikio nenustatyta, be to lapuose žymiai sumažėjo H₂O₂ ir MDA. CuO ND išpurškimas ant augalų neturėjo įtakos lapų ir šaknų BCF bei Ti, tačiau žymiai sumažino Tf.

8 lentelė. Agrometeorologinės sausros, sunkiojo metalo vario (Cu) ir CuO ND (50 ppm) poveikis sėjamajam žirniui (*P. sativum* L.). H₂O – kontroliniai augalai nupurkšti arba palaistyti dejonizuotu vandeniui, substrato drėgnis SD) 80 %; sausros stresas – SD 30 %. Rezultatuose pateikiami vidurkiai, o skirtingos raidės pažymi statistiškai patikimus skirtumus tarp variantų pagal Tukey HSD testą ($p < 0,05$)

50 ppm CuO ND poveikis	80 %	30 % SM					
	SM	H ₂ O		Laistymas	Purškimas	Cu +	Cu + Purškimas
	H ₂ O	H ₂ O	Cu	ND	ND	Laistymas ND	ND
Morfologiniai parametrai							
Augalo aukštis	32,9 a	22,2 cd	20,0 d	25,0 b	24,3 b	19,9 d	21,8 cd
Lapų plotas	76,9 a	49,9 b	43,9 c	40,6 cd	46,8 bc	24,3 e	40,2 cd
Gumbelių sk.	8,3 a	3,7 b	2,0 cd	2,1 cd	3,1 c	1,7 d	1,9 cd
SVK	80,6 a	41,7 bc	29,3 d	48,1 b	47,2 b	38,2 c	42,2 bc
Š/A	11,3 a	5,4 bc	4,1 c	10,2 a	6,8 b	8,1 b	6,1 bc
SLP	9,5 cd	11,0 c	13,5 ab	10,0 c	9,2 cd	13,2 b	14,7 a
Derlius	3,4 a	2,4 b	1,8 c	2,4 b	2,7 ab	1,4 de	1,6 cd
Nefermentinių antioksidantų aktyvumas							
ABTS	113,2 a	103,1 ab	96,4 bc	108,7 b	111,5 a	90,7 bc	110,2 a
DPPH	45,1 a	38,5 b	36,8 c	36,6 c	35,9 c	35,2 c	35,3 c
TPC	2,7 a	2,3 bc	1,9 c	2,6 ab	2,6 ab	1,2 d	2,4 b
FRAP	1475,8 a	1158,6 b	939,5 de	1082,9 d	1201,8 c	683,4 f	890,0 e
Fermentinių antioksidantų aktyvumas							
Gr	0,053 c	0,074 b	0,084 ab	0,073 b	0,066 bc	0,094 a	0,083 ab
GPX	11,1 f	15,5 e	44,9 ab	21,4 cd	19,6 cd	47,0 a	37,5 b
APX	25,1 d	51,3 c	74,3 a	63,0 b	63,7 b	70,3 ab	72,3 ab
SOD	46,8 e	67,2 d	74,5 cd	133,3 a	82,6 c	105,8 b	73,8 cd
CAT	16,7 d	31,9 bc	33,1 b	28,0 c	38,5 a	34,0 ab	32,2 bc
Oksidacinio streso rodikliai							

HP	12,3 d	25,5 ab	26,2 ab	24,8 ab	22,4 bc	19,7 c	20,1 c
MDA	54,7 e	62,3 d	97,3 a	84,5 b	81,9 bc	97,1 a	73,3 c
Vario akumuliacija							
BCF (lapai)			0,04 ab			0,05 ab	0,06 a
BCF (šaknys)			1,6 b			1,8 a	1,7 ab
Tf			2,1 a			1,4 c	1,7 b
Ti			62,7 ab			60,2 b	63,0 a

BCF – bio-koncentracijos faktorius, Tf – translokacijos faktorius, Ti – tolerancijos indeksas

Molibdeno trioksido nanodalelių poveikis žirniams, esant kompleksiniam sausros ir vario stresams

9 lentelėje pateikti rezultatai rodo kompleksinio sausros, Cu ir MoO₃ ND poveikį žirniams. Žirnių aukštis padidėjo 29 %, lapų plotas 28 %, gumbelių skaičius 500 %, SVK 44 %, derlius 88 %, šaknų ir ūglių santykis 120 %, o poveikio SLP nenustatyta, kai žirniai buvo veikiami sausros, Cu ir palaistyti 50 ppm MoO₃ ND suspensija (LxSxCu). Taip pat nustatytas reikšmingas ABTS, DPPH ir FRAP verčių padidėjimas. Be to, TPC kiekis žirniuose padidėjo daugiau nei 35 %. Nustatytas GPX ir CAT fermentų aktyvumo padidėjimas žirniuose, kurie buvo veikiami LxSxCu, tačiau nustatytas GR ir SOD aktyvumo sumažėjimas. Be to, žirnių lapuose H₂O₂ ir MDA kiekis sumažėjo apie 22 %. MoO₃ ND padidino Tf iki 44 %, o Ti – 76 %.

Statistiškai patikimi teigiami rezultatai nustatyti gumbelių skaičiui, SVK ir derliui, žirnius veikiant sausra, Cu ir purškiant MoO₃ ND suspensija (PxSxCu). Žirnių lapuose antioksidacinis pajėgumas (DPPH, ABTS ir FRAP) padidėjo iki 13 %, o TPC kiekis – 32 % po poveikio su PxsxCu. Fermentiniams antioksidantams slopinantis poveikis nustatytas tik GR aktyvumui (17 %), o kitų, GPX, APX, SOD ir CAT aktyvumas padidėjo atitinkamai 10, 35, 36 ir 20 %. Nustatyta, kad MDA kiekis žirnių lapuose sumažėjo 23 %. Be to, BCF šaknyse padidėjo apie 25 %, Ti – 13 %, o poveikio Tf nenustatyta.

9 lentelė. Agrometeorologinės sausros, sunkiojo metalo vario (Cu) ir MoO₃ ND (50 ppm) poveikis sėjamajam žirniui (P. sativum L.). H₂O – kontroliniai augalai nupurkšti arba palaistyti dejonizuotu vandeniu, substrato drėgnis SD) 80 %; sausros stresas – SD 30 %. Rezultatuose pateikiami vidurkiai, o skirtingos raidės pažymi statistiškai patikimus skirtumus tarp variantų pagal Tukey HSD testą (p < 0,05)

50 ppm MoO ₃ ND poveikis	80 % SD		30 % SD				
	H ₂ O		H ₂ O	Cu	Laistymas ND	Purškimas ND	Cu + Laistymas ND
Morfologiniai parametrai							
Augalo aukštis	32,9 a	22,2 c	20,0 d	27,8 b	20,0 d	25,8 bc	23,7 bcd

Lapų plotas	76,9 a	49,9 c	43,9 cd	52,1 bc	43,9 cd	56,5 b	43,5 c
Gumbelių sk.	8,3 b	3,7 d	2,0 e	7,0 bc	2,0 d	11,7 a	7,3 bc
SVK	80,6 a	41,7 c	29,3 e	49,4 b	43,0 bc	43,8 bc	33,6 d
Š/A	11,3 a	5,4 bc	4,1 c	6,1 bcd	9,3 ab	8,9 ab	6,1 bc
SLP	9,5 abc	11,0 ab	13,5 a	9,3 bc	14,0 a	9,4 ab	8,5 d
Derlius	3,4 ab	2,4 c	1,8 d	4,2 a	3,8 a	3,4 ab	3,3 ab
Nefermentinių antioksidantų aktyvumas							
ABTS	113,2 a	103,1 bc	96,4 e	110,3 a	96,4 e	105,0 bc	101,8 d
DPPH	45,1 a	38,5 c	36,8 d	42,2 b	36,9 d	42,5 b	41,8 b
TPC	2,7 a	2,3 cd	1,9 d	2,5 b	1,8 e	2,5 b	2,4 bc
FRAP	1475,8 a	1158,6 c	939,5 de	1216,2 b	924,6 de	1184,2 b	1058,2 c
Fermentinių antioksidantų aktyvumas							
Gr	0,053 c	0,074 b	0,084 a	0,089 a	0,070 b	0,072 b	0,070 b
GPX	11,1 c	15,5 bc	44,9 b	15,1 bc	19,6 bc	48,1 a	49,3 a
APX	25,1 d	51,3 c	54,3 c	58,2 b	74,5 a	71,8 ab	72,9 ab
SOD	46,8 e	67,2 c	74,5 b	73,3 b	74,2 b	64,4 cd	101,2 a
CAT	16,7 f	31,9 e	33,1 e	44,4 b	48,9 a	35,3 d	39,6 c
Oksidacinio streso rodikliai							
HP	12,3 e	25,5 bc	26,2 b	21,9 d	26,2 b	30,2 a	25,4 bc
MDA	54,7 e	62,3 c	97,3 a	58,6 d	63,3 c	75,7 b	75,3 b
Vario akumuliacija							
BCF _{Cu} (leaves)			0,04 b		0,06 a		0,05 ab
BCF _{Cu} (roots)			1,6 c		2,3 a		1,8 b
Tf _{Cu}			2,1 b		3,7 a		1,9 b
Ti _{Cu}			62,7 c		131,8 a		112,5 b

BCF – bio-koncentracijos faktorius, Tf – translokacijos faktorius, Ti – tolerancijos indeksas

Boro nanodalelių poveikis žirniams, esant kompleksiniam sausros ir sunkiojo metalo vario stresams

Remiantis 10 lentelėje pateiktais rezultatais, B₂O₃ ND teigiamai veikė kompleksinio streso paveiktų augalų morfologinius parametrus. Palaisčius žirnius, paveiktus sausros ir Cu, B₂O₃ ND suspensijomis (LxSxCu), nustatyta, kad gumbelių skaičius padidėjo 1,5 karto, SVK – 39 %, šaknų ir antžeminės dalies santykis – 75 %, derlius – 56 %. Laistymas 12,5 ppm B₂O₃ ND suspensija, sausros ir Cu paveiktuose žirniuose padidino DPPH laisvųjų radikalų surišimo aktyvumą 18 % ir FRAP antioksidacinę galią 36 %, tačiau ABTS laisvųjų radikalų surišimo aktyvumui statistiškai reikšmingų rezultatų nenustatyta. TPC kiekis žirnių lapuose padidėjo 34 %. SOD ir CAT aktyvumą teigiamai veikė LxSxCu, o slopinamasis poveikis pasireiškė APX

ir GR fermentams. MDA koncentracija sumažėjo 22 %, o H₂O₂ kiekiui statistiškai reikšmingo poveikio nenustatyta. Nenustatyta statistiškai patikimo poveikio BCF, tačiau Tf po palaistymo B₂O₃ ND tirpalais sumažėjo 14 %, o vario Ti padidėjo 42 %.

Nupurškus žirnius 12,5 ppm B₂O₃ ND suspensija (PxSxCu), žirnių atsparumas sausras ir Cu padidėjo, nes nustatytas teigiamas poveikis žirnių aukščiui (28 %), SVK (20 %), šaknų ir antžeminės dalies santykiui (64 %) bei derliui (44 %). Derliaus padidėjimą galima sieti su DPPH laisvųjų radikalo surišimo gebos padidėjimu (20 %) ir FRAP antioksidacinės galios padidėjimu (10 %) bei sumažėjusiu MDA kiekiu (38 %) žirnius nupurškus B₂O₃ ND suspensijomis, lyginant su augalais, nepaveiktais ND, bet paveiktais sausras ir Cu. Analizuojant ABTS laisvųjų radikalų surišimo aktyvumą, TPC kiekį, GR aktyvumą ir H₂O₂ kiekį žirnių lapuose, statistiškai patikimų rezultatų nerasta tarp žirnių paveiktų sausra ir Cu ir žirnių paveiktų PXSxCu. Be to, nustatytas stiprus 57 % SOD aktyvumo padidėjimas ir 160 % padidėjęs CAT aktyvumas. Žirnių purškimas 12,5 ppm B₂O₃ ND suspensija paveikė Tf ir Ti, jie padidėjo 33 ir 64 %, atitinkamai.

10 lentelė. Agrometeorologinės sausras, sunkiojo metalo vario (Cu) ir B₂O₃ ND (12,5 ppm) poveikis sėjamajam žirniui (*P. sativum* L.). H₂O – kontroliniai augalai nupurškšti arba palaistyti dejonizuotu vandeniu, substrato drėgnis SD) 80 %; sausras stresas – SD 30 %. Rezultatuose pateikiami vidurkiai, o skirtingos raidės pažymi statistiškai patikimus skirtumus tarp variantų pagal Tukey HSD testą ($p < 0,05$)

12.5 ppm B ₂ O ₃ ND poveikis	80 % SD		30 % SD				
	H ₂ O	H ₂ O	Cu	Laistymas ND	Purškimas ND	Cu + Laistymas ND	Cu + Purškimas ND
	Morfologiniai parametrai						
Augalo aukštis	32,9 a	22,2 c	20,0 d	27,2 b	31,2 a	21,0 d	25,6 bc
Lapų plotas	76,9 a	49,9 c	43,9 cd	40,9 c	57,1 b	49,0 c	36,9 d

Gumbelių sk.	8,3 a	3,7 b	2,0 c	5,0 b	7,0 a	5,0 a	3,7 bc
SVK	80,6 a	41,7 b	29,3 d	44,2 bc	38,1 bc	40,7 bc	35,3 bc
Š/A	11,3 a	5,4 bc	4,1 c	7,2 b	5,7 bc	7,2 b	6,7 b
SLP	9,5 bc	11,0 ab	13,5 a	7,4 d	11,1 ab	9,4 bc	8,5 cd
Derlius	3,4 a	2,4 c	1,8 d	2,8 ab	2,7 ab	2,8 ab	2,6 ab
Nefermentinių antioksidantų aktyvumas							
ABTS	113,2 a	103,1 b	96,4 bc	112,3 a	112,1 a	91,9 c	94,9 c
DPPH	45,1 a	38,5 c	36,8 c	43,5 b	47,6 a	43,4 b	43,9 b
TPC	2,7 ab	2,3 cd	1,9 e	2,8 ab	2,9 a	2,6 bc	2,2 de
FRAP	1475,8 a	1158,6 b	939,5 de	979,6 d	1029,4 c	1280,3 a	1033,8 c
Fermentinių antioksidantų aktyvumas							
Gr	0,053 b	0,074 a	0,084 a	0,031 c	0,078 a	0,050 b	0,073 a
GPX	11,1 d	15,5 c	44,9 b	65,2 a	49,4 b	15,1 c	13,6 cd
APX	25,1 f	51,3 d	74,3 a	70,0 ab	57,5 bc	66,5 b	34,2 e
SOD	46,8 e	67,2 d	74,5 cd	100,1 b	117,0 b	146,4 a	116,9 b
CAT	16,7 e	31,9 c	33,1 c	25,2 d	25,8 d	54,8 b	87,4 a
Oksidacinio streso rodikliai							
HP	12,3 c	25,5 ab	26,2 ab	26,4 ab	25,3 ab	23,8 b	26,7 ab
MDA	54,7 f	62,3 e	97,3 a	91,6 b	64,6 d	75,7 c	60,4 e
Vario akumuliacija							
BCF _{Cu} (lapai)			0,04 a			0,03 a	0,04 a
BCF _{Cu} (šaknys)			1,6 b			1,4 bc	1,9 ab
Tf _{Cu}			2,1 b			1,8 c	2,8a
Ti _{Cu}			62,7 c			89,2 b	103,0 a

BCF – bio-koncentracijos faktorius, Tf – translokacijos faktorius, Ti – tolerancijos indeksas

APTARIMAS

Šiame tyrime buvo išaiškintas SiO₂, MoO₃, B₂O₃ ir CuO nanodalelių (ND) poveikis žirnių daigams augusiems optimaliomis sąlygomis, pilnai išsivysčiusiems žirnių augalams augusiems agrometeorologinės sausros bei kompleksinio vario ir agrometeorologinės sausros sąlygomis. Siekiant išsamiai įvertinti ND poveikį žirniams, buvo tiriamas ne tik nefermentinių ir fermentinių antioksidantų atsakas, bet ir kiti rodikliai, bei buvo įvertintas skirtingų ND koncentracijų poveikis augalus purškiant ar laistant. Tyrimai parodė, kad SiO₂, MoO₃, B₂O₃ ir CuO ND teigiamai veikia žirnius ir mažina žalingą agrometeorologinės sausros bei kompleksinių įtempčių poveikį. ND sąveika su augalais lemia fizikinės ir cheminės ND savybės (dydis, paviršiaus krūvis, koncentracija suspensijoje) bei pati augalų fiziologija. ND sąveika su

augalais gali lemti specifinių membranos paviršiaus baltymų, receptorių ir transporterių modifikaciją (Juárez-Maldonado ir kt., 2019). Biologinės membranos su hidrofobiniais ir hidrofiliniais komponentais bei netolygiai pasiskirsčiusiais lignino ir celiuliozės pluoštais sukuria netolygų neigiamą paviršiaus krūvį nuo -45 iki -15 mV (Mittal ir kt., 2020). Pastebėta, kad neigiamai įkrautos (neigiamą krūvį turinčios) augalų ląstelių sienelės veikia kaip jonų mainų paviršius, kuris potencialiai skatina katijonų prasiskverbimą. Atsižvelgiant į skirtingus ND pateikimo augalams būdus, pastebėtina, kad kai ND pasiekia augalus per dirvą (šaknis), o dirvožemio dalelės dažniausiai būna neigiamo krūvio, didesnę neigiamą krūvį turinčios ND tokioje dirvoje tampa dar judresnės. Priešingai, teigiamą krūvį turinčios ND lengvai pritraukia neigiamo krūvio dirvožemio paviršiaus daleles. Mažėjant vidutiniam dirvožemio dalelių dydžiui, mažėja ND judrumas. Molio kiekis dirvožemyje gali veikti kaip anijoninis adjuvantas, neleidžiantis kauptis ND ir padidinti jų mobilumą. Todėl labai svarbu iširti ND suspensijų savybes, kad būtų galima suprasti jų patekimo į augalą ir poveikio mechanizmus. Nanodalelių, kurių zeta potencialas yra nuo -10 iki +10 mV, yra neutralios ir mažiau stabilios nei ND, kurių zeta potencialas didesnis nei +30 mV arba mažesnis nei -30 mV (atitinkamai stiprūs katijonai ir anijonai) (Clogston ir Patri, 2011).

ND gali sukelti įvairias reakcijas augaluose, padidinti fermentų aktyvumą, paskatinti nitratų virtimą amoniaku, suintensyvinti kvėpavimo ir fotosintezės procesus, sintetinti fermentus ir aminorūgštis, sustiprinti anglies ir azoto mitybą ir/arba tiesiogiai paveikti augalų mineralinę mitybą. Tačiau, nepaisant jau minėtų galimų pranašumų, nanotechnologijų pagalba sukurtų produktų naudojimas žemės ūkio sektoriuje yra gana ribotas, lyginant su kitomis pramonės šakomis. Todėl, atsižvelgiant į galimą praktinį ND pritaikymą žemės ūkio sektoriuje, prieš pateikiant su ND susijusius gaminius į rinką, reikėtų atlikti preliminarią kaštų ir naudos analizę, taip pat tinkamai įvertinti riziką aplinkai ir žmonių sveikatai.

SiO₂ ND poveikis žirnių morfologiniams parametrams, oksidacinio streso biožymenims ir antioksidaciniam aktyvumui

Mūsų tyrimuose SiO₂ ND vandeninės suspensijos zeta potencialas buvo -20,64 mV. Kitame moksliniame straipsnyje nurodoma, kad naudojant 10 g kg⁻¹ SiO₂ ND, zeta potencialas buvo -40 mV. Tai rodo, kad suspensijos yra stabilios ir anijoninės.

Nustatyta, kad žirnių augalų apdorojimas didesnėmis SiO₂ (50 ir 100 ppm) ND koncentracijomis padidino antioksidacinį aktyvumą ir sumažino H₂O₂ kiekį ir MDA jų lapuose, kai jie augo optimaliomis sąlygomis. Be to, augant normalaus substrato drėgnio sąlygomis, padidėjo žirnių ūglių aukštis, šaknų ilgis bei biomasė. Ypatingas poveikis pastebėtas

makroelementų kaupimuisi žirnių augaluose; žymiai padidėjo P, K, Ca ir Mg kiekis žirnių lapuose ir stiebuose. SiO₂ ND teigiamai paveikė skirtingų augalų rūšių augimą, padidino jų biomasę ir fiziologines savybes, suaktyvino antioksidacinę sistemą, padėjo prisitaikyti prie stresinių sąlygų (Luyckx ir kt., 2017). Apskritai SiO₂ ND poveikis yra plačiai tiriamas skirtinguose augaluose (Lu ir kt., 2007; Janmohammadi ir kt., 2015). Tačiau iš ankštinių augalų šeimos SiO₂ ND poveikis iki šiol buvo ištirtas tik lęšiams ir sojos pupelėms. Nustatyta, kad nano dydžio SiO₂ suspensija padidino nitratų reduktazės aktyvumą, stimuliuo antioksidacinę sistemą, pagerino vandens ir trąšų įsisavinimą bei sojų pupelių (*Glycine max*) daigumą ir augimą (Lu et al. 2007). Ankstesni tyrimai rodo, kad didesnes SiO₂ ND koncentracijas reikia vartoti atsargiai, nes nustatytas neigiamas 120,16 ppm ND koncentracijos poveikis lęšių dygimui (Janmohammadi ir kt., 2015).

Rezultatai rodo, kad vandens trūkumas dirvožemyje sumažino žirnių aukštį, lapų plotą, SLP, SVK ir padidino šaknų ir antžeminės dalies santykį. Tokius gautus rezultatus patvirtina ir kitų tyrėjų atlikti tyrimai (Arafa ir kt., 2021). Khatun ir kt., 2021; Bangar ir kt., 2019). Pažymėtina, kad laistymas SiO₂ ND suspensijomis turėjo didesnę teigiamą poveikį žirniams, augusiems normalaus drėgčio substrate, nei purškimas. Žirniai turi grupę akvaporinų MIP – Si įtraukimo (eng.: influx) transporterių (SiT1 ir SiT2) ir ištekėjimo (eng.: efflux) transporterių (SiT6), esančių centrinėje ir šoninėse šaknyse, palengvinančių Si patekimą į ksilemą ir leidžiančių laisvai judėti augale (Maurel ir kt. al., 2015; Raoi ir Susmitha, 2017). Tačiau sausros atveju medžiagų judėjimas ksilemoje sutrinka, todėl laistymas Si ND tampa mažiau efektyvus. Purškiant ant augalų, SiO₂ ND padengia lapų paviršių dėl ko gali susidaryti papildomas Si sluoksnis, apsaugantis augalus nuo transpiracijos ir turgorinio slėgio pokyčių. Be to, augalus apipurškus mažo dydžio SiO₂ ND, jos gali prasiskverbti pro vaško sluoksnį ir taip laisvai judėti augale. Dėl tokio poveikio augalų purškimas SiO₂ ND tirpalais gali turėti didesnę teigiamą poveikį esant sausros stresui.

Remdamiesi gautais duomenimis, galima teigti, kad SiO₂ ND sėkmingai suaktyvina antioksidacinę sistemą, taip sumažėja oksidacinio streso poveikis sausros paveiktiems žirniams ir išsaugomas jų derlius. SiO₂ ND yra stipriai susijusios su vandens kiekiu augaluose palaikymu, tačiau yra tik keli moksliniai straipsniai nagrinėjantys SiO₂ ND ir sausros poveikį augaluose. Pavyzdžiui, braškių nupurškimas SiO₂ ND sausros streso metu padidino CAT, APX, SOD ir GR fermentų aktyvumą, sumažino MDA ir H₂O₂ kiekį (Zahedi ir kt., 2020). Be to, sausros sąlygomis su SiO₂ ND augintuose kviečiuose nustatytas lapų žalumo, santykinio vandens kiekio ir derliaus padidėjimas (Behboudi ir kt., 2018).

CuO ND poveikis žirnių morfologiniams parametrams, oksidacinio streso biožymenims ir antioksidaciniam aktyvumui

Mūsų tyrimuose CuO ND vandeninės suspensijos zeta potencialas buvo $-26,68$ mV. Kiti tyrėjai nurodė, kad CuO ND zeta potencialas gali būti $-34,4 \pm 0,5$ mV dejonizuoto vandens suspensijoje, kai pH 7, o pirminis ND dydis 10–100 nm (Keller ir kt., 2018; Adeleye ir kt., 2014; Hong ir kt. al. 2015). Tai rodo, kad suspensijos buvo stabilios ir anijoninės.

Nustatyti reikšmingi sausros paveiktų augalų antioksidacinio aktyvumo pokyčiai naudojant CuO ND. Ypatingai stiprus teigiamas poveikis nustatytas CAT, APX ir SOD fermentams. Taip pat CuO ND stipriai paveikė FRAP antioksidacinę galią. Atsižvelgiant į rezultatus galime teigti, kad CuO ND naudojimas nedideliais kiekiais gali padidinti augalų atsparumą ir apsaugoti juos nuo aplinkos įtempčių. Mūsų rezultatai parodė, kad CuO ND suspensijos žymiai sumažino H₂O₂ kiekį žirnių augaluose, bet padidino MDA koncentraciją, tai rodo, kad CuO ND aktyviai dalyvauja lipidų peroksidacijos procesuose. Mokslininkai nustatė, kad H₂O₂ ir MDA kiekis pupelių šaknyse žymiai padidėjo, o poveikis lapuose nenustatytas augalus paveikus 20, 50, 100, 200 ir 500 ppm CuO ND tirpalais (Gopalakrishnan ir kt., 2014). Jie pabrėžė, kad CuO ND paveikė SOD koduojančių genų raišką bei perteklinis Cu kiekis galėjo padidinti H₂O₂ susidarymą pupelių šaknyse. Taip pat nustatyta, kad padidėja ir CAT genų ekspresija naudojant mažesnes CuO ND koncentracijas. APX geno ekspresijos lygis buvo mažesnis, lyginant su CAT genu, todėl galėjo susilpnėti H₂O₂ neutralizavimas. Todėl mokslininkai iškėlė hipotezę, kad CuO ND streso metu suaktyvina augalų antioksidacinius gynybos mechanizmus; nepilnas H₂O₂ pašalinimas galėjo padidinti ROS kiekį pupelių šaknyse, veikiamose CuO ND (Gopalakrishnan ir kt., 2014). Nair ir Chung (2014) ištyrė 0, 0,5, 1, 2, 5, 10, 20, 50 ir 100 mg L⁻¹ CuO ND suspensijų poveikį *Arabidopsis* augalams. Jų tyrimas parodė, kad po poveikio padidėjo genų, koduojančių fermentinius ir nefermentinius antioksidacinius gynybos mechanizmus aktyvumas bei padidėjo ROS gamyba. Po metų tas pats mokslininkas ištyrė CuO ND poveikį žirnių augalams. Mūsų išvados sutinka, kad kuo didesnė CuO ND koncentracija, tuo pasireiškia didesnis toksiškumas žirniams, pradedant nuo 100 ppm (Nair ir Chung, 2015). Toksinis poveikis sojų pupelėms patvirtintas nustatčius padidėjusią lipidų peroksidaciją ir H₂O₂ kiekį, kai CuO ND koncentracija buvo 100 ppm ir didesnė (Yusefi-Tanha et al. 2020). Taip pat buvo pastebėtas sojų pupelių derlingumo sumažėjimas naudojant 50–100 ppm CuO ND koncentraciją (Ochoa ir kt., 2017).

Mūsų tyrimai parodė, kad CuO ND turėjo teigiamą poveikį antioksidacinei sistemai ir Cu kaupimuisi, kai augalai augo vandens trūkumo sąlygomis ir buvo purškiami per lapus. CuO ND įsisavinimas ir translokacija buvo plačiai ištirti ryžių augaluose (Peng ir kt., 2015). CuO ND

translokacijos metu ištirpęs Cu prijungiamas prie cisteino, citrato ir fosfato ligandų, o dalis Cu (II) redukuojama į Cu (I). Nustatyta, kad CuO ND gali judėti į šaknies epidermį, egzodermą ir žievę ir pasiekti endodermą, tačiau CuO ND sunkiai prasiskverbia pro Kasparijos juostelę (Peng ir kt., 2015). Ji trukdo CuO ND patekti iš šaknų į augalo antžeminę dalį. Tai galėjo lemti mažesnę pasiskirstymą augale ir lokalizuotą poveikį žirniams, kai jie buvo laistomi CuO ND tirpalais. Tai paaiškina mūsų gautus rezultatus, kur CuO ND laistytų žirnių šaknyse Cu kaupimasis buvo didesnis, tačiau lapuose jo kiekis nepadidėjo, o nupurškus augalus Cu kiekis padidėjo visose augalo dalyse.

Moksliniuose straipsniuose, taip pat, yra rašoma apie teigiamą CuO ND poveikį augalams. Pavyzdžiui, salotų purškimas 0, 0,5, 1,0, 2,0, 4 ir 6 ppm CuO ND suspensijomis teigiamai paveikė bendrą fenolinių junginių ir flavonoidų kiekį, antioksidacinį aktyvumą ir chlorofilų kiekį (Gaucin-Delgado ir kt., 2022). Be to, CuO ND inicijavo APX, SOD ir CAT aktyvumą ir žymiai padidino Cu kaupimąsi salotų lapuose. Pomidorų daigus paveikus 10 ppm CuO ND suspensija, žymiai padidėjo šviežia biomasė ir chlorofilų kiekis, taip pat cukraus kiekis, NR ir CAT aktyvumas bei sumažėjo MDA (Singh ir kt., 2017). Kaip matome iš aptartų publikacijų, CuO ND poveikį augalams dažniausiai lemia koncentracija. Atsižvelgiant į tai, kad ND yra daug kartų mažesnės dalelės nei įprastos medžiagos ir turi daug didesnę paviršiaus plotą, todėl augalams naudojama koncentracija turėtų būti sumažinta, kad būtų pasiektas teigiamas atsakas ir išvengta toksiško poveikio.

MoO₃ ND poveikis žirnių morfologiniams parametrams, oksidacinio streso biožymenims ir antioksidaciniam aktyvumui

MoO₃ ND vandeninės suspensijos zeta potencialas buvo -24,92 mV. Remiantis mokslinėmis publikacijomis, MoO₃ ND suspensijos zeta potencialas gali būti -32 mV, patvirtinantis, kad tirpalas yra stabilus ir stipriai anijoninis.

MoO₃ ND reikšmingai skatino gumbelių susidarymą ant žirnių šaknų ir stipriai paveikė nefermentinius ir fermentinius antioksidantus. Išsamiam tyrimui apie MoO₃ ND poveikį hidroponikoje augintiems ryžių daigams nustatyta, kad efektyviausia koncentracija buvo 100 ppm, kuri sukėlė hormezės būseną (Sharma ir kt., 2021). Taip pat buvo nustatyta, kad ryžiuose sumažėjo chlorofilų ir karotenoidų kiekis esant didesnei nei 100 ppm MoO₃ ND koncentracijai. Be to, MoO₃ ND ryžiuose padidino MDA koncentraciją ūgliuose, bet sumažino šaknyse. Toks neigiamas poveikis gali būti susijęs su per didelės koncentracijos naudojimu.

Gauti rezultatai rodo, kad SOD aktyvumas sumažėjo žirnių lapuose, paveiktuose MoO₃ ND, o APX, GPX ir CAT aktyvumas padidėjo, tai nulemia greitesnę (aktyvių deguonies junginių neutralizavimą). Yra paskelbta publikacija (Yang ir kt., 2020), kurioje mokslininkai nurodė, kad

pridėjus 1, 10 mg kg⁻¹ MoO₃ ND padidino fermentinių antioksidantų, tokių kaip POD, SOD ir CAT, aktyvumą sojų pupelėse, tačiau didesnės koncentracijos (100 ir 1000 mg kg⁻¹) buvo toksiškos. Kitame tyrime buvo naudotos 200 ir 1000 ppm koncentracijų MoO₃ ND suspensijos (Huang ir kt., 2021). Tyrėjai pastebėjo, kad kukurūzų lapai gali sukaupti didesnę Mo kiekį nei kviečių lapai, o tai rodo, kad skirtingos augalų rūšys gali skirtingai reaguoti į skirtingas MoO₃ ND koncentracijas.

Mo nėra biologiškai aktyvus elementas, todėl nedideliais kiekiais jis yra prijungiamas į specifines proteazių grupes. Šis Mo kofaktorius (Moco) yra priskiriamas abiem Moco surišančioms šeimoms: sulfito oksidazei (sulfito oksidazė, nitratų reduktazė (NR), mitochondrijų amidoksimo reduktorius) ir ksantino oksidoreduktazei (aldehido oksidazė (AO), ksantino dehidrogenazė). Svarbiausias Mo surišantis fermentas augalų išlikimui yra citozolinis NR, kuris katalizuoja pirminį nitratų įsisavinimo žingsnį (Kaiser ir kt., 2005; Mendel ir Schwarz 2011). Nitratų pavertimas nitritais yra būtinas augalų augimui ir vystymuisi. Wu (2018) pasiūlė Mo sukeltos oksidacinės tolerancijos sausros streso metu mechanizmo schemą. Pažymima, kad sumažėjus substrato drėgmeniui, augalų apsaugos sistema aktyvuojama per ABA ir NO signalus. Kaip Mo-fermentų komponentas, Mo stimuliuoja ABA sintezę ir NO gamybą atitinkamai per AO ir NR ir taip reguliuoja oksidacinę toleranciją sausros paveiktuose augaluose (Wu ir kt., 2018). Tačiau neaišku, ar toks pats mechanizmas gali būti pritaikytas naudojant nano dydžio Mo. Pavyzdžiui, pranešama, kad 0,1 ir 1 ppm MoO₃ ND suspensijos aktyvina nitratų reduktazės aktyvumą špinatuose ir padidina chlorofilų kiekį (Abbasifar et al., 2020). Be to, naudojant 8 ppm Mo ND koncentracijos tirpalą, padidėjo sėjamųjų avinžirnių antioksidacinis aktyvumas (Taran ir kt., 2014). Taip pat, 10, 20, 30, 40 ir 50 ppm koncentracijų MoO₃ ND purškimas ant paprastųjų pupelių augalų lapų žymiai pagerino morfologinius parametrus, tokius kaip šviežia ir sausa biomasa, augalo aukštis ir šaknų ilgis, lyginant su neapdorotais augalais (Osman ir kt., 2020). Esminis šio tyrimo parametras buvo genomo šablono stabilumo procentas (GTS%), tiesiogiai susijęs su DNR pakitimo laipsniu ir DNR atkūrimo bei replikacijos kompetencija. Genominis nestabilumas apima struktūros pokyčius, tokius kaip padidėjęs bazinių porų mutacijų dažnis. Tyrėjai nustatė, kad GTS% vertės sumažėjo, didėjant MoO₃ ND koncentracijai nuo 10 iki 40 ppm, atitinkamai 84,61 – 78,21 %. Be to, mutageninis MoO₃ ND poveikis paprastųjų pupelių DNR sukėlė kelių genų, koduojančių specifinius baltymus, ekspresiją, įjungdamas arba išjungdamas tam tikrus genus daigų ir žydėjimo stadijose (Osman ir kt., 2020). Tai galėjo lemti, kad visos MoO₃ ND koncentracijos reikšmingai paveikė paprastųjų pupelių augalų fiziologines savybes, padidino derlių ir jo kokybę.

Kadangi nedideli MoO₃ ND kiekiai yra veiksmingesni už įprastą molibdeno druskas, jų pritaikymas žemės ūkyje, galėtų padidinti augalų produktyvumą tiek optimaliomis, tiek

nepalankiomis augimo sąlygomis. Mūsų rezultatuose nustatyta, kad naudojant 50 ppm MoO₃ ND suspensiją, esant normaliai dirvožemio drėgnei, žirnių derlius padidėja 6 %, o sausros paveiktų žirnių derlius galimai padidėja iki 80 %. Nors gaminti MoO₃ ND pramoniniu būdu yra brangus procesas, padidintas augalų derlius iš hektaro padengia susidariusį skirtumą ir netgi padidina pelną.

Boro ND poveikis žirnių morfologiniams parametrams, oksidacinio streso biožymenims ir antioksidaciniam aktyvumui

Mūsų tyrimų metu B₂O₃ ND vandeninės suspensijos dzeta potencialas buvo -28,54 mV. Tačiau kiti mokslininkai nustatė, kad B₂O₃ ND vandeninė suspensija su 0,2 % tritono x-100 buvo -30,3 mV (Barreto ir kt., 2021). Tokios dzeta potencialo vertės rodo, kad tirpalai yra stabilūs ir anijoniniai. Be to, šios suspensijos PDI buvo 0,23, o kiti mokslininkai nustatė 0,4 (Barreto ir kt., 2021), tai rodo, kad suspensijos yra monodispersinės.

Mūsų atlikti tyrimai parodė, kad ant žirnių šaknų susidarantių gumbelių skaičius reikšmingai padidėjo po poveikio B₂O₃ ND tiek augalams augant normaliame, tiek esant drėgmės trūkumui substrate. B augale yra transportuojamas ksilemoje, o apie 90 % yra sujungiamas į augalų ląstelių sienelės (Goldbach ir Wimmer 2007). B yra pagrindinis elementas formuojant esterius su ramnogalakturonanu II (RG-II, eng.: rhamnogalacturonan II). Šis borato esteris reikalingas išlaikyti normalioms ląstelių sienelių funkcijoms ir struktūrai (Ryden ir kt., 2003). Optimaliomis sąlygomis, kai dirvoje pakanka B, RGII-glikoproteinai susidaro ir žirnių šaknų gumbelių bei šaknų ląstelių plazminėse membranose. Tačiau esant B trūkumui RGII-glikoproteinai nėra sintetinami, o jų trūkumas destabilizuoja plazminę membraną ir gumbelių susidarymą (Bolaños ir kt., 2001). Taip pat, B, kaip glikoproteinų komponentas, yra būtinas siekiant gumbelinių bakterijų diferenciacijos į azotą fiksuojančią formą (Bolaños ir kt., 2004). Taip pat tiek mūsų rezultatuose, tiek kitų mokslininkų publikacijose nurodoma, kad B gali stimuliuoti tiek fermentinį, tiek nefermentinį antioksidacinį aktyvumą. Daugelyje mokslinių publikacijų pabrėžiama B nauda augalų antioksidacinei sistemai esant įvairioms streso sąlygoms (Alpaslan ir Gunes, 2001; Bonilla ir kt., 2004; Bastías ir kt., 2004), tačiau tik keliuose tyrimuose buvo tirtas B₂O₃ ND poveikis (Dimkpa ir kt., 2019; Zewail ir kt., 2021; Mahmoud ir kt., 2020).

Kompleksinio streso ir ND poveikis žirniams

Mokslininkai pastebi, kad augalus dažniau veikia kompleksiniai neigiami aplinkos veiksniai nei pavieniai. Vis dar yra daug neištirtų galimų streso derinių tarp sausros, druskingumo, karščio bangų, atšalimo, užšalimo, ozono, patogenų, UV, maistinių medžiagų trūkumo, per didelio CO₂ kiekio, per didelio apšvietimo ir sunkiųjų metalų (Suzuki ir kt., 2014).

Vienas iš šio mokslinio darbo tikslų buvo ištirti galimą kompleksinį agrometeorologinės sausras ir sunkiųjų metalų įtempčių poveikį žirniams kartu su skirtingų ND sąveiką. Atliekant skirtingų augalų rūšių kompleksinių įtempčių tyrimus, transkriptomines ir proteomines analizės parodė, kad svarbiausias yra antioksidacinės sistemos atsakas. Suaktyvėjęs antioksidacinis pajėgumas ir/arba sumažėjęs aktyvių deguonies junginių (ROS) kaupimasis yra mechanizmas, veikiantis augaluose, kuris padidina atsparumą kompleksiniam stresui. Tiriant bendrą sausras ir sunkiojo metalo Cu poveikį, rezultatai rodo, kad labiau sumažėja morfologiniai parametrai ir nefermentinių antioksidantų aktyvumas, tačiau stipriau padidėja fermentinių antioksidantų aktyvumas ir oksidacinio streso biožymenų kiekis žirniuose, lyginant augalus paveiktus tik sausra ir žirnius paveiktus sausra bei sunkiuoju metalu Cu. Tokį persidengiantį kompleksinio streso poveikį patvirtina ir kiti moksliniai tyrimai. Pavyzdžiui, buvo ištirtas kompleksinis sausras ir sunkiojo metalo Cd poveikis kukurūzų augalams (Naz ir kt., 2021). Šiame tyrime mokslininkai nustatė, kad toks derinys nepaveikė augalų morfologinių parametru, tačiau žymiai sumažino bendrą chlorofilų, baltymų, karotinoidų kiekį ir santykinį vandens kiekį augaluose. Kombinuotas stresas žymiai padidino oksidacinio streso biožymenų ir suaktyvavo SOD, CAT, APX ir POD fermentinius antioksidantus. Tiek sausra, tiek jos derinys su sunkiojo metalo Cu stresu sumažino žirnių augimą ir lėmė beveik identiškų augalų pokyčius. Poveikis buvo adityvus, abiem stresiniams veiksniams veikiant vienu metu, todėl augalai buvo mažesni ir atsakas į stresą buvo stipresnis nei kiekvienam stresui veikiant atskirai. Tačiau buvo pastebėta išimtis panaudojus ND.

Žirnių augalų purškimas SiO_2 ir B_2O_3 ND buvo veiksmingesnis jiems augant kompleksinio streso sąlygomis, nors prieš tai atliktame tyrime tik su sausras stresu skirtumų tarp ND panaudojimo būdų nebuvo nustatyta. Tai gali būti siejama su neigiamu sunkiojo metalo Cu poveikiu šaknų augimui. Kadangi sausras sąlygomis augalai didžiąją dalį energijos skiria šaknų augimui, kad surastų daugiau vandens, o per didelis Cu kiekis dirvožemyje tai slopino ir trukdė pasisavinti tiek esminius elementus, tiek stabdė šaknų augimą (Kumar ir kt., 2021). Tyrėjai ištyrė, kad SiO_2 ND pagerino kviečių augimo rodiklius, kurie buvo paveikti kompleksinio sausras ir sunkiojo metalo Cd stresų, taip pat SiO_2 ND sumažino Cd koncentraciją skirtinguose kviečių audiniuose, slopino oksidacinį stresą ir padidino chlorofilo kiekį. (Khan ir kt., 2020) Tyrėjai taip pat pažymi, kad SiO_2 ND taikymas gali būti veiksmingas būdas siekiant sumažinti Cd koncentraciją javų grūduose. Mūsų tyrime buvo nustatyti reikšmingi Cu kaupimosi augaluose pokyčiai, žirnius paveikus SiO_2 ND suspensijomis Tiek purškimas, tiek laistymas SiO_2 ND reikšmingai padidino Cu biokoncentraciją žirnių lapuose ir šaknyse, translokacijos faktorių ir tolerancijos indeksą. Tai gali būti susiję su didesniu santykiniu vandens kiekiu ir

padidėjusiu antioksidaciniu aktyvumu augaluose, kurie buvo veikiami kompleksinio streso ir SiO_2 ND.

Įdomūs rezultatai gauti žirniuose, kurie buvo paveikti kompleksinio streso ir B_2O_3 ND, nes šių augalų lapuose MDA koncentracija labai sumažėjo, tačiau MDA koncentracija buvo didžiausia tais atvejais, kai nustatytas Cu perteklius. Kaip minėta aukščiau, tai gali būti susiję su tuo, kad, didžioji dalis B, patenkančio į augalą, yra surišta su augalo ląstelių sienelėmis apsaugant ląsteles nuo lipidų peroksidacijos, kurią gali sukelti per didelis Cu kiekis (Blevins ir Lukaszewski, 1998). Be to, B_2O_3 ND padidino Cu tolerancijos indeksą bei pagrindinių fermentinių antioksidantų SOD ir CAT aktyvumą žirniuose. Siekiant ląstelėse išlaikyti homeostazę SOD katalizuoja superoksido radikalo virtimą į H_2O_2 , o CAT H_2O_2 paverčia vandeniu. Toks reikšmingas šių fermentų suaktyvėjimas galėjo padėti žirniams lengviau išverti kompleksinio streso poveikį.

Žirnių laistymas su MoO_3 ND, esant kompleksiniam stresui, reikšmingai paveikė gumbelių susidarymą, kurį stipriai sumažino sausros ir sunkiojo metalo Cu poveikis. Taip pat nustatytas reikšmingas augalų santykinio vandens kiekio padidėjimas. Dėl to matome padidėjusią Cu bioakumuliaciją šaknyse, bei padidėjusius translokacijos faktorių ir Cu tolerancijos indeksą. Taip pat reikšmingai padidėjo fermentinių antioksidantų GPX, APX, SOD ir CAT aktyvumas, kuris nulėmė oksidacinio streso rodiklių sumažėjimą. Reikia paminėti, kad CuO ND sustiprino stresorių kompleksinį neigiamą poveikį žirniams. Tai rodo, kad CuO ND žirniuose veikia sinergiškai su sunkiuoju metalu Cu.

Šis darbas praplečia žinias apie galimą kompleksinių įtempčių poveikį žirniams. Be to, šis tyrimas išsiskiria savo svarba dėl išsamiai aptariamo kompleksinių įtempčių ir skirtingų ND poveikio žirnių morfologiniams parametrams, derliui, antioksidacinės sistemos aktyvumui bei elementų akumuliacijai.

IŠVADOS

1. Žirnių daigams ir pilnai išsivysčiusiems žirniams, augusiems normalaus drėgčio substrate ar sausros sąlygomis, efektyviausios SiO_2 , CuO ir MoO_3 nanodalelių suspensijų koncentracijos buvo 50 ppm, o B_2O_3 – 12,5 ppm.

2. Naudojama SiO_2 nanodalelių suspensija padidino žirnių atsparumą sausrai. Drėgmės trūkumo sąlygomis SiO_2 nanodalelės 27 % sumažino oksidacinių biožymenų koncentraciją ir 30 % padidino bendrą fenolinių junginių kiekį bei nefermentinių antioksidantų aktyvumą žirniuose. Reikšmingas aktyvinantis poveikis nustatytas fermentiniams antioksidantams katalazei (159 %), superoksido dismutazei (37 %) ir glutationo reduktazei (128 %), augalus paveikus 50 ppm SiO_2 nanodalelių tirpalu. Sausros paveiktų žirnių derlius išlaikytas 45 % panaudojus SiO_2 nanodalelių suspensiją.

3. CuO nanodalelės 37 % sumažino vandenilio peroksido, bet 66 % padidino malondialdehido koncentraciją substrato drėgmės trūkumo paveiktuose žirnių augaluose. Didžiausia 50 ppm CuO nanodalelių koncentracija iki 2,5 karto padidino nefermentinių antioksidantų aktyvumą bei paskatino katalazės (167 %) ir askorbato peroksidazės (72 %) aktyvumą žirniuose, augusiuose sausros sąlygomis. Didžiausia Cu akumuliacija žirnių lapuose, stiebe bei šaknyse nustatyta nupurškus augalus 50 ppm CuO nanodalelių suspensija.

4. Panaudojus MoO_3 nanodalelių suspensiją sausros sąlygomis, nustatytas teigiamas poveikis žirnių aukščiui (40 %) ir lapų plotui (30 %) bei iki 30 % sumažėję oksidacinio streso rodikliai. Nustatytas 37 % bendro fenolinių junginių kiekio ir 145 % nefermentinių antioksidantų aktyvumo padidėjimas augaluose paveiktuose sausros ir MoO_3 nanodalelių. MoO_3 nanodalelės 2 kartus padidino katalazės ir askorbato peroksidazės aktyvumą bei 20 % superoksido dismutazės ir 56 % gvajakolio peroksidazės aktyvumą. Didžiausia Mo akumuliacija nustatyta žirnių šaknyse, kai augalai buvo laistomi nanodalelių suspensijomis.

5. B_2O_3 nanodalelės turėjo teigiamą poveikį žirnių aukščiui (26 %), lapų plotui (40 %) ir 2,5 karto padidino antioksidantų aktyvumą. Augaluose paveiktuose sausros ir B_2O_3 nanodalelių 6 kartus suaktyvėjo askorbato ir 91 % gvajakolio peroksidazės bei 51 % padidėjo superoksido dismutazės aktyvumas. B didžiausia akumuliacija nustatyta šaknyse nepriklausomai nuo poveikio būdo. Žirnių, paveiktų sausros ir palaistytų su B_2O_3 nanodalelėmis, stiebuose ir lapuose B akumuliacija vyko intensyviausiai.

6. Tiriant kompleksinį Cu , kaip sunkiojo metalo, ir sausros poveikį žirnių augalams nustatyta, kad stipresnį kelių stresorių sukeltą oksidacinį stresą slopino naudojamos SiO_2 , MoO_3 ir B_2O_3 nanodalelių suspensijos. Jos efektyviai sumažino vandenilio peroksido ir malondialdehido koncentracijas augale bei padidino antioksidacinės sistemos aktyvumą lyginant

su augalais kurie buvo paveikti sausros ir Cu pertekliaus. Laistymas CuO nanodalelėmis sustiprino sausros ir perteklinio Cu neigiamą poveikį augalams. Žirniuose tolerancijos indeksą Cu statistiškai patikimai didino tiek laistymas, tiek purškimas SiO₂, purškimas B₂O₃ ir laistymas MoO₃ nanodalelių suspensijomis.

7. Atsižvelgiant į nanodalelių savybes bei patekimo į augalą būdus, nustatyta, kad laistymas MoO₃ nanodalelių suspensija efektyviau sumažino stresų sukeltą neigiamą poveikį žirniuose nei purškimas ir išsaugojo 80 % derliaus. B₂O₃ ir CuO nanodalelių suspensijų efektyvumas buvo stipresnis jomis purškiant žirnius, šie poveikiai, atitinkamai, išsaugojo 92 % ir 47 % derliaus. O SiO₂ nanodalelių poveikis pagal taikymo būdą iš esmės nesiskyrė ir derlius buvo apie 40 % didesnis lyginant su augalais paveiktais sausra, bet nepaveiktais nanodalelėmis.

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